

SUPPORTING INFORMATION

Testing the Conjugative Properties of Benzodithiophene and Benzo-trithiophene in Charge Transfer Multi(Ferrocenyl) Systems

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Table of Contents

Figure S1. UV-vis spectra of $\mathbf{1}^{2+}$ in CH_2Cl_2 / 0.1 M $n\text{Bu}_4\text{NPF}_6$	page S2
Figure S2. UV-vis spectra of $\mathbf{2}^{3+}$ in CH_2Cl_2 / 0.1 M $n\text{Bu}_4\text{NPF}_6$	page S3
Figure S3. NIR spectra of $\mathbf{1}^{2+}$ in CH_2Cl_2 / 0.1 M $n\text{Bu}_4\text{NPF}_6$	page S4
Figure S4. NIR spectra of $\mathbf{2}^{3+}$ in CH_2Cl_2 / 0.1 M $n\text{Bu}_4\text{NPF}_6$	page S5
Figura S5. NIR spectra of $\mathbf{1}^+$ and $\mathbf{1}^{2+}$ in acetone / 0.1 M $n\text{Bu}_4\text{NB}(\text{C}_6\text{F}_5)$	page S6
Figura S6. NIR spectra of $\mathbf{2}^+$ and $\mathbf{2}^{3+}$ in acetone / 0.1 M $n\text{Bu}_4\text{NB}(\text{C}_6\text{F}_5)$	page S7

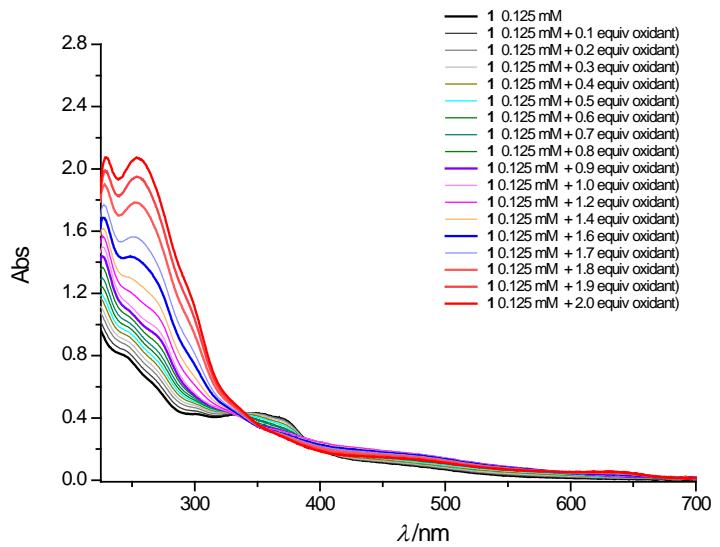


Figure S1. UV-vis spectra of **1**²⁺ in CH₂Cl₂ / 0.1 M *n*Bu₄NPF₆ obtained by incremental additions of acetylferrocenium(BF₄) from 0.1 to 2 equivalents to 0.125 mM **1**.

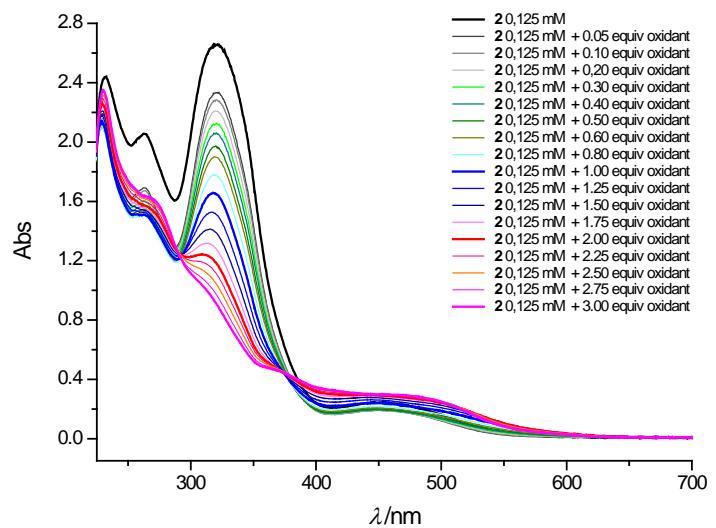


Figure S2. UV-vis spectra of **2²⁺** in CH₂Cl₂ / 0.1 M *n*Bu₄NPF₆ obtained by incremental additions of acetylferrocenium(BF₄) from 0.05 to 3 equivalents to 0.125 mM **2**.

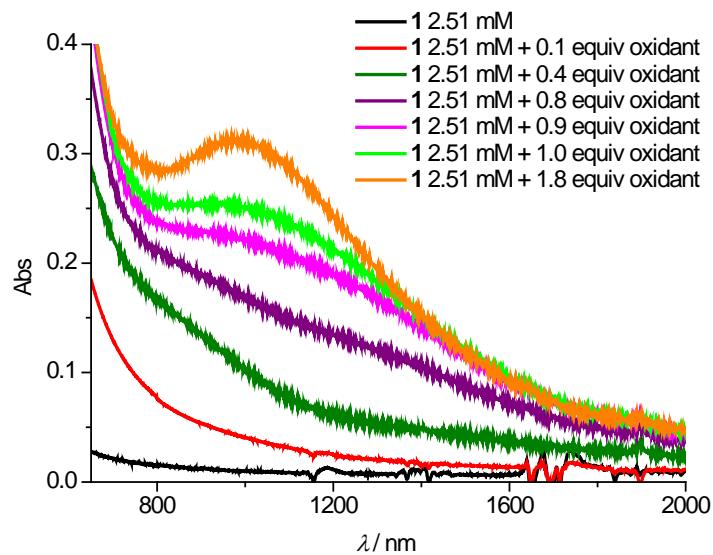


Figure S3. NIR spectra of **1**²⁺ in CH₂Cl₂ / 0.1 M *n*Bu₄NPF₆ obtained by incremental additions of acetylferrocenium(BF₄) from 0.05 to 1.8 equivalents to 2.51 mM **1**. Further addition of the oxidant caused precipitation of **1**²⁺.

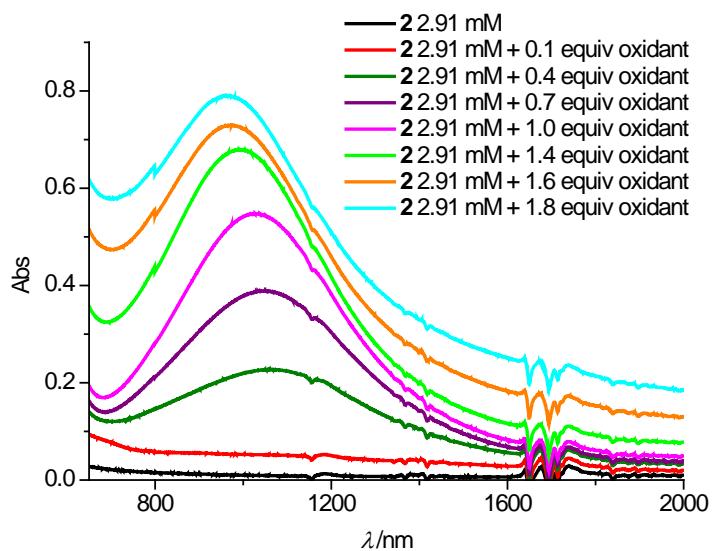


Figure S4. NIR spectra of **2**³⁺ in CH_2Cl_2 / 0.1 M $n\text{Bu}_4\text{NPF}_6$ obtained by incremental additions to 2.91 mM **2** of acetylferrocenium(BF_4^-) from 0.05 to 1.8 equivalents to 2.91 mM **2**. Further oxidant addition caused precipitation of **2**³⁺.

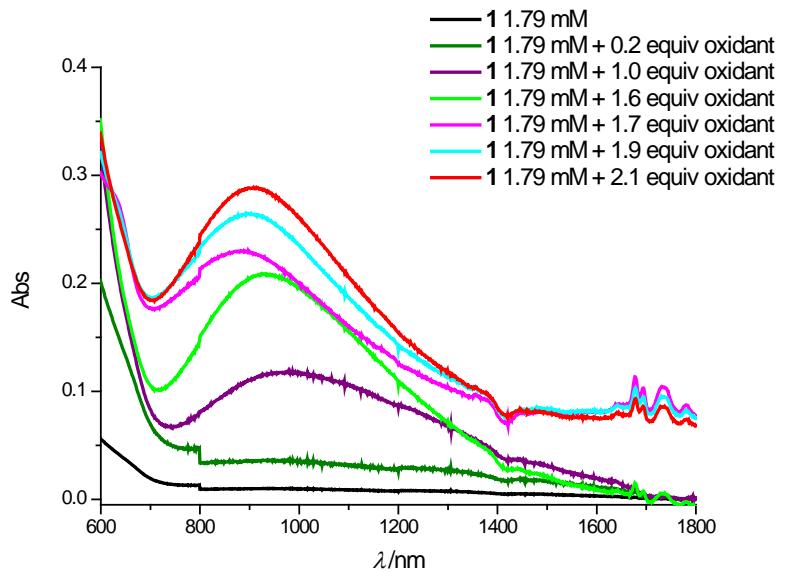


Figure S5. Oxidation of 1.79 mM **1** solutions in acetone / 0.1 M *n*Bu₄NB(C₆F₅)₄ by incremental addition of acetylferrocenium(BF₄).

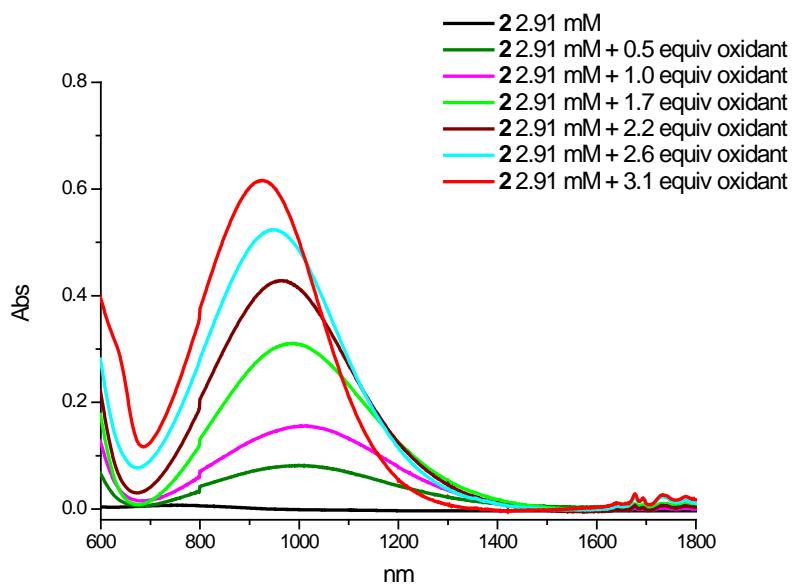


Figure S6. Oxidation of 2.91 mM **2** solutions in acetone / 0.1 M $n\text{Bu}_4\text{N}^+\text{B}(\text{C}_6\text{F}_5)_4^-$ by incremental addition of acetylferrocenium(BF_4^-).