# Pyrtriazoles, a Novel Class of Store-Operated Calcium Entry <br> Modulators: Discovery, Biological Profiling and In Vivo Proof-ofConcept Efficacy in Acute Pancreatitis 

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## Molecular Formula Strings

| Compound | SMILE | Residual SOCE activity (\% of Ctrl) | IC50 ( $10 \mu \mathrm{M}$ ) |
| :---: | :---: | :---: | :---: |
| 1H | $\mathrm{O}=\mathrm{C}(\mathrm{Cl}=\mathrm{C}(\mathrm{C}(\mathrm{F})(\mathrm{F}) \mathrm{F}) \mathrm{N}(\mathrm{C} 2=\mathrm{CC}=\mathrm{C}(\mathrm{N} 3 \mathrm{~N}=\mathrm{NC}(\mathrm{C} 4=\mathrm{CC}=\mathrm{CC}(\mathrm{N})=\mathrm{C} 4)=\mathrm{C} 3) \mathrm{C}=\mathrm{C} 2) \mathrm{N}=\mathrm{C} 1) \mathrm{OCC}$ | $81.7 \pm 12.1$ | ND |
| 1N | $\mathrm{O}=\mathrm{C}(\mathrm{C} 1=\mathrm{C}(\mathrm{C}(\mathrm{F})(\mathrm{F}) \mathrm{F}) \mathrm{N}(\mathrm{C} 2=\mathrm{CC}=\mathrm{C}(\mathrm{N} 3 \mathrm{~N}=\mathrm{NC}(\mathrm{C} 4=\mathrm{CC}=\mathrm{C}(\mathrm{O}) \mathrm{C}(\mathrm{OC})=\mathrm{C} 4)=\mathrm{C} 3) \mathrm{C}=\mathrm{C} 2) \mathrm{N}=\mathrm{C} 1) \mathrm{OCC}$ | $43.2 \pm 3.7$ | ND |
| 1 S | $\mathrm{O}=\mathrm{C}(\mathrm{O}) \mathrm{C} 1=\mathrm{CC}=\mathrm{CC}(\mathrm{C} 2=\mathrm{CN}(\mathrm{C} 3=\mathrm{CC}=\mathrm{C}(\mathrm{N} 4 \mathrm{~N}=\mathrm{CC}(\mathrm{C}(\mathrm{OCC})=\mathrm{O})=\mathrm{C} 4 \mathrm{C}(\mathrm{F})(\mathrm{F}) \mathrm{F}) \mathrm{C}=\mathrm{C} 3) \mathrm{N}=\mathrm{N} 2)=\mathrm{C} 1$ | $9.8 \pm 0.5$ | $0.6 \pm 0.1$ |
| $1 T$ | $\mathrm{O}=\mathrm{C}(\mathrm{O}) \mathrm{C} 1=\mathrm{CC}=\mathrm{C}(\mathrm{C} 2=\mathrm{CN}(\mathrm{C} 3=\mathrm{CC}=\mathrm{C}(\mathrm{N} 4 \mathrm{~N}=\mathrm{CC}(\mathrm{C}(\mathrm{OCC})=\mathrm{O})=\mathrm{C} 4 \mathrm{C}(\mathrm{F})(\mathrm{F}) \mathrm{F}) \mathrm{C}=\mathrm{C} 3) \mathrm{N}=\mathrm{N} 2) \mathrm{C}=\mathrm{C} 1$ | $52.1 \pm 4.2$ | ND |
| 2 S | $\mathrm{O}=\mathrm{C}(\mathrm{O}) \mathrm{C} 1=\mathrm{CC}=\mathrm{CC}(\mathrm{C} 2=\mathrm{CN}(\mathrm{C} 3=\mathrm{CC}=\mathrm{C}(\mathrm{N} 4 \mathrm{~N}=\mathrm{C}(\mathrm{C}(\mathrm{F})(\mathrm{F}) \mathrm{F}) \mathrm{C}=\mathrm{C} 4 \mathrm{C}(\mathrm{F})(\mathrm{F}) \mathrm{F}) \mathrm{C}=\mathrm{C} 3) \mathrm{N}=\mathrm{N} 2)=\mathrm{C} 1$ | $53.0 \pm 8.7$ | ND |
| 2 T | $\mathrm{O}=\mathrm{C}(\mathrm{O}) \mathrm{C} 1=\mathrm{CC}=\mathrm{C}(\mathrm{C} 2=\mathrm{CN}(\mathrm{C} 3=\mathrm{CC}=\mathrm{C}(\mathrm{N} 4 \mathrm{~N}=\mathrm{C}(\mathrm{C}(\mathrm{F})(\mathrm{F}) \mathrm{F}) \mathrm{C}=\mathrm{C} 4 \mathrm{C}(\mathrm{F})(\mathrm{F}) \mathrm{F}) \mathrm{C}=\mathrm{C} 3) \mathrm{N}=\mathrm{N} 2) \mathrm{C}=\mathrm{C} 1$ | $213.0 \pm 0.8$ | ND |
| 3D | $\mathrm{O}=\mathrm{C}(\mathrm{O}) \mathrm{CCCN} 1 \mathrm{~N}=\mathrm{NC}(\mathrm{C} 2=\mathrm{CC}=\mathrm{C}(\mathrm{N} 3 \mathrm{~N}=\mathrm{CC}(\mathrm{C}(\mathrm{OCC})=\mathrm{O})=\mathrm{C} 3 \mathrm{C}(\mathrm{F})(\mathrm{F}) \mathrm{F}) \mathrm{C}=\mathrm{C} 2)=\mathrm{C} 1$ | $95.5 \pm 10.1$ | ND |
| 3G | $\mathrm{O}=\mathrm{C}(\mathrm{Cl}=\mathrm{C}(\mathrm{C}(\mathrm{F})(\mathrm{F}) \mathrm{F}) \mathrm{N}(\mathrm{C} 2=\mathrm{CC}=\mathrm{C}(\mathrm{C} 3=\mathrm{CN}(\mathrm{C} 4=\mathrm{CC}=\mathrm{C}(\mathrm{Cl}) \mathrm{C}=\mathrm{C} 4) \mathrm{N}=\mathrm{N} 3) \mathrm{C}=\mathrm{C} 2) \mathrm{N}=\mathrm{C} 1) \mathrm{OCC}$ | $141.6 \pm 1.5$ | ND |
| 3H | $\mathrm{O}=\mathrm{C}(\mathrm{C} 1=\mathrm{C}(\mathrm{C}(\mathrm{F})(\mathrm{F}) \mathrm{F}) \mathrm{N}(\mathrm{C} 2=\mathrm{CC}=\mathrm{C}(\mathrm{C} 3=\mathrm{CN}(\mathrm{C} 4=\mathrm{CC}=\mathrm{CC}(\mathrm{N})=\mathrm{C} 4) \mathrm{N}=\mathrm{N} 3) \mathrm{C}=\mathrm{C} 2) \mathrm{N}=\mathrm{C} 1) \mathrm{OCC}$ | $73.2 \pm 3.3$ | ND |
| 3J | $\mathrm{O}=\mathrm{C}(\mathrm{C} 1=\mathrm{C}(\mathrm{C}(\mathrm{F})(\mathrm{F}) \mathrm{F}) \mathrm{N}(\mathrm{C} 2=\mathrm{CC}=\mathrm{C}(\mathrm{C} 3=\mathrm{CN}(\mathrm{C} 4=\mathrm{C}(\mathrm{F}) \mathrm{C}=\mathrm{NC}=\mathrm{C} 4) \mathrm{N}=\mathrm{N} 3) \mathrm{C}=\mathrm{C} 2) \mathrm{N}=\mathrm{C} 1) \mathrm{OCC}$ | $81.4 \pm 9.2$ | ND |
| 3 S | $\mathrm{O}=\mathrm{C}(\mathrm{O}) \mathrm{C} 1=\mathrm{CC}=\mathrm{CC}(\mathrm{N} 2 \mathrm{~N}=\mathrm{NC}(\mathrm{C} 3=\mathrm{CC}=\mathrm{C}(\mathrm{N} 4 \mathrm{~N}=\mathrm{CC}(\mathrm{C}(\mathrm{OCC})=\mathrm{O})=\mathrm{C} 4 \mathrm{C}(\mathrm{F})(\mathrm{F}) \mathrm{F}) \mathrm{C}=\mathrm{C} 3)=\mathrm{C} 2)=\mathrm{C} 1$ | $30.0 \pm 6.0$ | ND |
| 3T | $\mathrm{O}=\mathrm{C}(\mathrm{O}) \mathrm{C} 1=\mathrm{CC}=\mathrm{C}(\mathrm{N} 2 \mathrm{~N}=\mathrm{NC}(\mathrm{C} 3=\mathrm{CC}=\mathrm{C}(\mathrm{N} 4 \mathrm{~N}=\mathrm{CC}(\mathrm{C}(\mathrm{OCC})=\mathrm{O})=\mathrm{C} 4 \mathrm{C}(\mathrm{F})(\mathrm{F}) \mathrm{F}) \mathrm{C}=\mathrm{C} 3)=\mathrm{C} 2) \mathrm{C}=\mathrm{C} 1$ | $69.3 \pm 10.0$ | ND |
| 19 | $\mathrm{O}=\mathrm{C}(\mathrm{O}) \mathrm{Cl}=\mathrm{CC}=\mathrm{CC}(\mathrm{NC}(\mathrm{C} 2=\mathrm{CC}=\mathrm{C}(\mathrm{N} 3 \mathrm{~N}=\mathrm{CC}(\mathrm{C}(\mathrm{OCC})=\mathrm{O})=\mathrm{C} 3 \mathrm{C}(\mathrm{F})(\mathrm{F}) \mathrm{F}) \mathrm{C}=\mathrm{C} 2)=\mathrm{O})=\mathrm{C} 1$ | > 90\% | ND |
| 22 | $\mathrm{O}=\mathrm{C}(\mathrm{O}) \mathrm{C} 1=\mathrm{CC}=\mathrm{CC}(\mathrm{C}(\mathrm{NC} 2=\mathrm{CC}=\mathrm{C}(\mathrm{N} 3 \mathrm{~N}=\mathrm{CC}(\mathrm{C}(\mathrm{OCC})=\mathrm{O})=\mathrm{C} 3 \mathrm{C}(\mathrm{F})(\mathrm{F}) \mathrm{F}) \mathrm{C}=\mathrm{C} 2)=\mathrm{O})=\mathrm{C} 1$ | $63.5 \pm 3.2$ | ND |
| 23 | $\mathrm{O}=\mathrm{C}(\mathrm{O}) \mathrm{Cl}=\mathrm{CC}=\mathrm{CC}(\mathrm{C} 2=\mathrm{CN}=\mathrm{NN} 2 \mathrm{C} 3=\mathrm{CC}=\mathrm{C}(\mathrm{N} 4 \mathrm{~N}=\mathrm{CC}(\mathrm{C}(\mathrm{OCC})=\mathrm{O})=\mathrm{C} 4 \mathrm{C}(\mathrm{F})(\mathrm{F}) \mathrm{F}) \mathrm{C}=\mathrm{C} 3)=\mathrm{C} 1$ | > 90\% | ND |
| 30 | $\mathrm{O}=\mathrm{C}(\mathrm{O}) \mathrm{C} 1=\mathrm{CC}=\mathrm{CC}(\mathrm{C} 2=\mathrm{CN}(\mathrm{C} 3=\mathrm{CC}=\mathrm{C}(\mathrm{N} 4 \mathrm{~N}=\mathrm{CC}(\mathrm{C}(\mathrm{OC})=\mathrm{O})=\mathrm{C} 4 \mathrm{C}(\mathrm{F})(\mathrm{F}) \mathrm{F}) \mathrm{C}=\mathrm{C} 3) \mathrm{N}=\mathrm{N} 2)=\mathrm{C} 1$ | $7.9 \pm 1.8$ | ND |
| 31 | $\mathrm{O}=\mathrm{C}(\mathrm{O}) \mathrm{C} 1=\mathrm{CC}=\mathrm{CC}(\mathrm{C} 2=\mathrm{CN}(\mathrm{C} 3=\mathrm{CC}=\mathrm{C}(\mathrm{N} 4 \mathrm{~N}=\mathrm{CC}(\mathrm{C}(\mathrm{OC}(\mathrm{C}) \mathrm{C})=\mathrm{O})=\mathrm{C} 4 \mathrm{C}(\mathrm{F})(\mathrm{F}) \mathrm{F}) \mathrm{C}=\mathrm{C} 3) \mathrm{N}=\mathrm{N} 2)=\mathrm{C} 1$ | $3.5 \pm 0.3$ | $3.1 \pm 1.3$ |
| 39 | $\mathrm{O}=\mathrm{C}(\mathrm{O}) \mathrm{C} 1=\mathrm{CC}=\mathrm{CC}(\mathrm{C} 2=\mathrm{CN}(\mathrm{C} 3=\mathrm{CC}=\mathrm{C}(\mathrm{N} 4 \mathrm{~N}=\mathrm{CC}(\mathrm{C}(\mathrm{OC}(\mathrm{CCC}) \mathrm{C})=\mathrm{O})=\mathrm{C} 4 \mathrm{C}(\mathrm{F})(\mathrm{F}) \mathrm{F}) \mathrm{C}=\mathrm{C} 3) \mathrm{N}=\mathrm{N} 2)=\mathrm{C} 1$ | $17.5 \pm 1.6$ | $4.4 \pm 1.2$ |
| 40 | $\mathrm{O}=\mathrm{C}(\mathrm{O}) \mathrm{C} 1=\mathrm{CC}=\mathrm{CC}(\mathrm{C} 2=\mathrm{CN}(\mathrm{C} 3=\mathrm{CC}=\mathrm{C}(\mathrm{N} 4 \mathrm{~N}=\mathrm{CC}(\mathrm{C}(\mathrm{OCCC}(\mathrm{C}) \mathrm{C})=\mathrm{O})=\mathrm{C} 4 \mathrm{C}(\mathrm{F})(\mathrm{F}) \mathrm{F}) \mathrm{C}=\mathrm{C} 3) \mathrm{N}=\mathrm{N} 2)=\mathrm{Cl}$ | > 90\% | ND |
| 41 | $\mathrm{O}=\mathrm{C}(\mathrm{O}) \mathrm{C} 1=\mathrm{CC}=\mathrm{CC}(\mathrm{C} 2=\mathrm{CN}(\mathrm{C} 3=\mathrm{CC}=\mathrm{C}(\mathrm{N} 4 \mathrm{~N}=\mathrm{CC}(\mathrm{C}(\mathrm{OC} / \mathrm{C}=\mathrm{C}(\mathrm{C}) / \mathrm{CC} / \mathrm{C}=\mathrm{C}(\mathrm{C}) / \mathrm{C})=\mathrm{O})=\mathrm{C} 4 \mathrm{C}(\mathrm{F})(\mathrm{F}) \mathrm{F}) \mathrm{C}=\mathrm{C} 3) \mathrm{N}=\mathrm{N} 2)=\mathrm{Cl}$ | > $90 \%$ | ND |
| 42 | $\mathrm{O}=\mathrm{C}(\mathrm{O}) \mathrm{C} 1=\mathrm{CC}=\mathrm{CC}(\mathrm{C} 2=\mathrm{CN}(\mathrm{C} 3=\mathrm{CC}=\mathrm{C}(\mathrm{N} 4 \mathrm{~N}=\mathrm{CC}(\mathrm{C}(\mathrm{OCC} 5=\mathrm{CC}=\mathrm{NC}=\mathrm{C} 5)=\mathrm{O})=\mathrm{C} 4 \mathrm{C}(\mathrm{F})(\mathrm{F}) \mathrm{F}) \mathrm{C}=\mathrm{C} 3) \mathrm{N}=\mathrm{N} 2)=\mathrm{C} 1$ | > 90\% | ND |
| 43 | $\mathrm{O}=\mathrm{C}(\mathrm{O}) \mathrm{C} 1=\mathrm{CC}=\mathrm{CC}(\mathrm{C} 2=\mathrm{CN}(\mathrm{C} 3=\mathrm{CC}=\mathrm{C}(\mathrm{N} 4 \mathrm{~N}=\mathrm{CC}(\mathrm{C}(\mathrm{OCC} 5=\mathrm{CC}=\mathrm{CC}=\mathrm{C} 5)=\mathrm{O})=\mathrm{C} 4 \mathrm{C}(\mathrm{F})(\mathrm{F}) \mathrm{F}) \mathrm{C}=\mathrm{C} 3) \mathrm{N}=\mathrm{N} 2)=\mathrm{C} 1$ | $10.5 \pm 1.4$ | ND |
| 44 | $\mathrm{O}=\mathrm{C}(\mathrm{O}) \mathrm{C} 1=\mathrm{CC}=\mathrm{CC}(\mathrm{C} 2=\mathrm{CN}(\mathrm{C} 3=\mathrm{CC}=\mathrm{C}(\mathrm{N} 4 \mathrm{~N}=\mathrm{CC}(\mathrm{C}(\mathrm{OCCN} 5 \mathrm{CCOCC} 5)=\mathrm{O})=\mathrm{C} 4 \mathrm{C}(\mathrm{F})(\mathrm{F}) \mathrm{F}) \mathrm{C}=\mathrm{C} 3) \mathrm{N}=\mathrm{N} 2)=\mathrm{C} 1$ | >90\% | ND |
| 46 | $\mathrm{O}=\mathrm{C}(\mathrm{O}) \mathrm{C} 1=\mathrm{CC}=\mathrm{CC}(\mathrm{C} 2=\mathrm{CN}(\mathrm{C} 3=\mathrm{CC}=\mathrm{C}(\mathrm{N} 4 \mathrm{~N}=\mathrm{CC}(\mathrm{CO})=\mathrm{C} 4 \mathrm{C}(\mathrm{F})(\mathrm{F}) \mathrm{F}) \mathrm{C}=\mathrm{C} 3) \mathrm{N}=\mathrm{N} 2)=\mathrm{C} 1$ | > 90\% | ND |
| 48 | $\mathrm{O}=\mathrm{C}(\mathrm{O}) \mathrm{C} 1=\mathrm{CC}=\mathrm{CC}(\mathrm{C} 2=\mathrm{CN}(\mathrm{C} 3=\mathrm{CC}=\mathrm{C}(\mathrm{N} 4 \mathrm{~N}=\mathrm{CC}(\mathrm{C}(\mathrm{NC}(\mathrm{C}) \mathrm{C})=\mathrm{O})=\mathrm{C} 4 \mathrm{C}(\mathrm{F})(\mathrm{F}) \mathrm{F}) \mathrm{C}=\mathrm{C} 3) \mathrm{N}=\mathrm{N} 2)=\mathrm{C} 1$ | > 90\% | ND |
| 49 | $\mathrm{O}=\mathrm{C}(\mathrm{O}) \mathrm{C} 1=\mathrm{CC}=\mathrm{CC}(\mathrm{C} 2=\mathrm{CN}(\mathrm{C} 3=\mathrm{CC}=\mathrm{C}(\mathrm{N} 4 \mathrm{~N}=\mathrm{CC}(\mathrm{C}(\mathrm{N})=\mathrm{O})=\mathrm{C} 4 \mathrm{C}(\mathrm{F})(\mathrm{F}) \mathrm{F}) \mathrm{C}=\mathrm{C} 3) \mathrm{N}=\mathrm{N} 2)=\mathrm{C} 1$ | > $90 \%$ | ND |
| 50 | $\mathrm{O}=\mathrm{C}(\mathrm{Cl}=\mathrm{C}(\mathrm{C}(\mathrm{F})(\mathrm{F}) \mathrm{F}) \mathrm{N}(\mathrm{C} 2=\mathrm{CC}=\mathrm{C}(\mathrm{N} 3 \mathrm{~N}=\mathrm{NC}(\mathrm{C} 4=\mathrm{CC}=\mathrm{CC}(\mathrm{C}(\mathrm{O})=\mathrm{O})=\mathrm{C} 4)=\mathrm{C} 3) \mathrm{C}=\mathrm{C} 2) \mathrm{N}=\mathrm{C} 1) \mathrm{O}$ | > $90 \%$ | ND |
| 51 | $\mathrm{O}=\mathrm{C}(\mathrm{Cl}=\mathrm{C}(\mathrm{C}(\mathrm{F})(\mathrm{F}) \mathrm{F}) \mathrm{N}(\mathrm{C} 2=\mathrm{CC}=\mathrm{C}(\mathrm{N} 3 \mathrm{~N}=\mathrm{NC}(\mathrm{C} 4=\mathrm{CC}=\mathrm{CC}(\mathrm{C}(\mathrm{NS}(=\mathrm{O})(\mathrm{C})=\mathrm{O})=\mathrm{O})=\mathrm{C} 4)=\mathrm{C} 3) \mathrm{C}=\mathrm{C} 2) \mathrm{N}=\mathrm{C} 1) \mathrm{OCC}$ | > $90 \%$ | ND |
| 58 | $\mathrm{O}=\mathrm{C}(\mathrm{C} 1=\mathrm{C}(\mathrm{C}(\mathrm{F})(\mathrm{F}) \mathrm{F}) \mathrm{N}(\mathrm{C} 2=\mathrm{CC}=\mathrm{C}(\mathrm{N} 3 \mathrm{~N}=\mathrm{NC}(\mathrm{C} 4=\mathrm{CC}=\mathrm{CC}(\mathrm{C}(\mathrm{OC})=\mathrm{O})=\mathrm{C} 4)=\mathrm{C} 3) \mathrm{C}=\mathrm{C} 2) \mathrm{N}=\mathrm{C} 1) \mathrm{OCC}$ | > 90\% | ND |
| 59 | $\mathrm{O}=\mathrm{C}(\mathrm{C} 1=\mathrm{C}(\mathrm{C}(\mathrm{F})(\mathrm{F}) \mathrm{F}) \mathrm{N}(\mathrm{C} 2=\mathrm{CC}=\mathrm{C}(\mathrm{C} 3=\mathrm{CN}(\mathrm{C} 4=\mathrm{CC}=\mathrm{CC}(\mathrm{C}(\mathrm{C})=\mathrm{O})=\mathrm{C} 4) \mathrm{N}=\mathrm{N} 3) \mathrm{C}=\mathrm{C} 2) \mathrm{N}=\mathrm{C} 1) \mathrm{OCC}$ | > 90\% | ND |
| 60 | $\mathrm{O}=\mathrm{C}(\mathrm{C} 1=\mathrm{C}(\mathrm{C}(\mathrm{F})(\mathrm{F}) \mathrm{F}) \mathrm{N}(\mathrm{C} 2=\mathrm{CC}=\mathrm{C}(\mathrm{N} 3 \mathrm{~N}=\mathrm{NC}(\mathrm{C} 4=\mathrm{CC}=\mathrm{CC}(\mathrm{C}(\mathrm{N})=\mathrm{O})=\mathrm{C} 4)=\mathrm{C} 3) \mathrm{C}=\mathrm{C} 2) \mathrm{N}=\mathrm{C} 1) \mathrm{OCC}$ | $72.1 \pm 2.5$ | ND |
| 61 | $\mathrm{O}=\mathrm{C}(\mathrm{C} 1=\mathrm{C}(\mathrm{C}(\mathrm{F})(\mathrm{F}) \mathrm{F}) \mathrm{N}(\mathrm{C} 2=\mathrm{CC}=\mathrm{C}(\mathrm{C} 3=\mathrm{CN}(\mathrm{C} 4=\mathrm{CC}=\mathrm{CC}(\mathrm{CO})=\mathrm{C} 4) \mathrm{N}=\mathrm{N} 3) \mathrm{C}=\mathrm{C} 2) \mathrm{N}=\mathrm{C} 1) \mathrm{OCC}$ | > 90\% | ND |
| 62 | $\mathrm{O}=\mathrm{C}(\mathrm{C} 1=\mathrm{C}(\mathrm{C}(\mathrm{F})(\mathrm{F}) \mathrm{F}) \mathrm{N}(\mathrm{C} 2=\mathrm{CC}=\mathrm{C}(\mathrm{N} 3 \mathrm{~N}=\mathrm{NC}(\mathrm{C} 4=\mathrm{CC}=\mathrm{CC}(\mathrm{S}(=\mathrm{O})(\mathrm{N})=\mathrm{O})=\mathrm{C} 4)=\mathrm{C} 3) \mathrm{C}=\mathrm{C} 2) \mathrm{N}=\mathrm{C} 1) \mathrm{OCC}$ | $83.6 \pm 4.1$ | ND |
| 63 | $\mathrm{O}=\mathrm{C}(\mathrm{C} 1=\mathrm{C}(\mathrm{C}(\mathrm{F})(\mathrm{F}) \mathrm{F}) \mathrm{N}(\mathrm{C} 2=\mathrm{CC}=\mathrm{C}(\mathrm{N} 3 \mathrm{~N}=\mathrm{NC}(\mathrm{C} 4=\mathrm{CC}=\mathrm{CC}(\mathrm{C}(\mathrm{NO})=\mathrm{O})=\mathrm{C} 4)=\mathrm{C} 3) \mathrm{C}=\mathrm{C} 2) \mathrm{N}=\mathrm{C} 1) \mathrm{OCC}$ | $6.4 \pm 5.7$ | ND |

## Chemistry

## Synthesis and characterization of alkyne $\mathbf{J}$



Reagents and conditions: (a) Ethynyltrimethylsilane, DIPEA, $\mathrm{CuI}, \mathrm{Pd}\left(\mathrm{PPh}_{3}\right)_{2} \mathrm{Cl}_{2}$, DMF, rt, 2 h . (b) TBAF, THF, $0^{\circ} \mathrm{C}, 30 \mathrm{~min}, 45 \%$.

4-Ethynyl-3-fluoropyridine (alkyne J). Step 1: 3-Fluoro-4-iodopyridine (118 mg; 0.53 mmol ), DMF ( 1.2 $\mathrm{mL}), \operatorname{Pd}\left(\mathrm{PPh}_{3}\right)_{2} \mathrm{Cl}_{2}(35 \mathrm{mg} ; 0.05 \mathrm{mmol}), \mathrm{CuI}(9.52 \mathrm{mg} ; 0.05 \mathrm{mmol})$, DIPEA ( $\left.0.36 \mathrm{~mL} ; 2.11 \mathrm{mmol}\right)$ and ethynyltrimethylsilane ( 0.22 mL ; 1.59 mmol ) were added in a Schlenk apparatus under nitrogen atmosphere. After 2 h the mixture was filtered over a pad of celite, the volatile was removed under reduced pressure and the reaction was worked up by diluition with ethyl acetate and washed with water (x1). The organic layer was washed with brine, dried over sodium sulfate and evaporated. The crude product was used in the next step without further purification. Step 2: 3-fluoro-4((trimethylsilyl)ethynyl) pyridine was dissolved in THF ( 3.3 mL ) at $0^{\circ} \mathrm{C}$. After 5 min TBAF $(0.62 \mathrm{~mL}$; 0.62 mmol ) was added. After 30 min the volatile was removed under reduced pressure and the reaction was worked up by dilution with ethyl acetate and washing with water (x1). The crude material was purified by column chromatography using petroleum ether/ethyl acetate $98: 2$ and then petroleum ether/ethyl acetate $95: 5$ as eluents yielding 4-ethynyl-3-fluoropyridine ( $28.9 \mathrm{mg}, 0.23 \mathrm{mmol}, 45 \%$ ) as a brown oil. ${ }^{1} \mathrm{H}$ NMR ( $300 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta=8.50(\mathrm{~s}, 1 \mathrm{H}), 8.39(\mathrm{~d}, J=4.7 \mathrm{~Hz}, 1 \mathrm{H}), 7.38(\mathrm{~d}, J=5.5 \mathrm{~Hz}$, 1H), $3.51(\mathrm{~s}, 1 \mathrm{H})$. MS (ESI): $m / z: 122[\mathrm{M}+\mathrm{H}]^{+}$.

## Synthesis and characterization of azide $\mathbf{K}$



Reagents and conditions: (a) $\mathrm{NaN}_{3}, \mathrm{NH}_{4} \mathrm{Cl}, \mathrm{DMF}, 110^{\circ} \mathrm{C}, 10 \mathrm{~h}, 77 \%$.
2-azidopyridine, (azide K). To a solution of 2-chloropyridine ( $2.00 \mathrm{~g}, 17.6 \mathrm{mmol}$ ) in DMF ( 20 mL ) $\mathrm{NaN}_{3}$ $(2.29 \mathrm{~g}, 35.2 \mathrm{mmol})$ and $\mathrm{NH}_{4} \mathrm{Cl}(1.88 \mathrm{~g}, 35.2 \mathrm{mmol})$ were added under nitrogen atmosphere. The mixture was heated at $110^{\circ} \mathrm{C}$ overnight. The reaction was worked up by dilution with ethyl acetate and washing with water ( x 4 ), with brine, dried over sodium sulfate and evaporated to give 2 -azidopyridine $(1.63 \mathrm{~g}$, $13.55 \mathrm{mmol}, 77 \%)$ as a pale brown oil. ${ }^{1} \mathrm{H}$ NMR ( $300 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta=8.83(\mathrm{~d}, J=6.9 \mathrm{~Hz}, 1 \mathrm{H}), 8.34$ $(\mathrm{d}, J=6.9 \mathrm{~Hz}, 1 \mathrm{H}), 7.65(\mathrm{t}, J=6.9 \mathrm{~Hz}, 1 \mathrm{H}), 7.24(\mathrm{t}, J=6.9 \mathrm{~Hz}, 1 \mathrm{H})$.

## Synthesis and characterization of compound 18



Reagents and conditions: (a) EtOH, THF, rt, 3 h, $93 \%$.
4-(4-(Ethoxycarbonyl)-5-(trifluoromethyl)-1H-pyrazol-1-yl)benzoic acid, (18).
To a solution of 4-hydrazinylbenzoic acid ( $1 \mathrm{~g}, 6.58 \mathrm{mmol}$ ) in EtOH ( 10 mL ), THF ( 2 mL ) was added and the reaction mixture was cooled at $-8{ }^{\circ} \mathrm{C}$ under nitrogen atmosphere. Then ( $E$ )-ethyl 2-(ethoxymethylene)-4,4,4-trifluoro-3-oxobutanoate ( $1.28 \mathrm{~mL}, 6.58 \mathrm{mmol}$ ) was added dropwise. After stirring at room temperature for 3 h the mixture was filtered under vacuum and rinsed with heptane to give compound $18(2 \mathrm{~g}, 6.10 \mathrm{mmol}, 93 \%)$ as a dark white solid. ${ }^{1} \mathrm{H}$ NMR ( $300 \mathrm{MHz}, \mathrm{DMSO}-d_{6}$ ): $\delta=$ $8.36(\mathrm{~s}, 1 \mathrm{H}), 8.14(\mathrm{~d}, J=7.5 \mathrm{~Hz}, 2 \mathrm{H}), 7.71(\mathrm{~d}, J=7.5 \mathrm{~Hz}, 2 \mathrm{H}), 4.36(\mathrm{q}, J=7.1 \mathrm{~Hz}, 2 \mathrm{H}), 1.33(\mathrm{t}, J=$ 7.1 Hz, 3H). MS (ESI): $m / z: 329[\mathrm{M}+\mathrm{H}]^{+}$.

## Synthesis and characterization of alkyne 52



Reagents and conditions: (a) Ethynyltrimethylsilane, DIPEA, $\mathrm{CuI}, \mathrm{Pd}\left(\mathrm{PPh}_{3}\right)_{2} \mathrm{Cl}_{2}$, toluene, $100{ }^{\circ} \mathrm{C}, 6 \mathrm{~h}$, $89 \%$. (b) TBAF, $\mathrm{CH}_{3} \mathrm{COOH}, \mathrm{THF}, 0^{\circ} \mathrm{C}, 1 \mathrm{~h}, 98 \%$.

Methyl 3-ethynylbenzoate, (52).
Step 1: To a solution of methyl 3-bromobenzoate ( $2 \mathrm{~g}, 9.30 \mathrm{mmol}$ ) in toluene ( 50 mL ) DIPEA ( 6.5 mL , $37.2 \mathrm{mmol}), \mathrm{CuI}(0.32 \mathrm{~g}, 1.67 \mathrm{mmol}), \mathrm{Pd}\left(\mathrm{PPh}_{3}\right)_{2} \mathrm{Cl}_{2}(0.39 \mathrm{~g}, 0.56 \mathrm{mmol})$ and ethynyltrimethylsilane ( 5.25 $\mathrm{mL}, 37.2 \mathrm{mmol}$ ) were added in a Schlenk apparatus. The reaction was stirred at reflux for 6 h under nitrogen atmosphere. Then, the mixture was filtered over a pad of celite and rinsed with ethyl acetate. The organic phase was washed with water (x1), dried over sodium sulfate and evaporated. The crude material was purified by column chromatography using petroleum ether and petroleum ether/ethyl acetate $95: 5$ as eluents to give methyl 3-((trimethylsilyl)ethynyl)benzoate as a brown oil $(1.95 \mathrm{~g}, 8.30$ mmol, $89 \%$ ). ${ }^{1} \mathrm{H}$ NMR ( $300 \mathrm{MHz} ; \mathrm{CDCl}_{3}$ ): $\delta=8.10(\mathrm{~s}, 1 \mathrm{H}), 7.93(\mathrm{~d}, J=7.7 \mathrm{~Hz}, 1 \mathrm{H}), 7.59(\mathrm{~d}, J=7.7$ $\mathrm{Hz}, 1 \mathrm{H}), 7.32(\mathrm{t}, J=7.7 \mathrm{~Hz}, 1 \mathrm{H}), 3.88(\mathrm{~s}, 3 \mathrm{H}), 0.22(\mathrm{~s}, 9 \mathrm{H})$.

Step 2: Methyl 3-((trimethylsilyl)ethynyl)benzoate ( $0.60 \mathrm{~g}, 2.55 \mathrm{mmol}$ ) was dissolved in THF ( 6 mL ). The mixture was cooled at $0^{\circ} \mathrm{C}$ and $\mathrm{CH}_{3} \mathrm{COOH}(175 \mu \mathrm{~L}, 3.06 \mathrm{mmol})$ and TBAF ( $3.06 \mathrm{~mL}, 3.06 \mathrm{mmol}$ ) were added. The reaction was stirred at $0^{\circ} \mathrm{C}$ for 1 h . The volatile was removed under vacuum, ethyl acetate was added and the organic layer was washed with water (x1). After drying over sodium sulfate and evaporation of the solvent, the crude material was purified by column chromatography using petroleum ether and petroleum ether/ethyl acetate $95: 5$ as eluents to give compound $\mathbf{5 2}(0.40 \mathrm{~g}, 2.50$ mmol, $98 \%$ ) as a yellow oil. ${ }^{1} \mathrm{H}$ NMR ( $300 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta=8.17(\mathrm{~s}, 1 \mathrm{H}), 7.96(\mathrm{~d}, J=6.5 \mathrm{~Hz}, 1 \mathrm{H})$,
$7.67(\mathrm{~d}, J=6.5 \mathrm{~Hz}, 1 \mathrm{H}), 7.40(\mathrm{t}, J=6.5 \mathrm{~Hz}, 1 \mathrm{H}), 3.92(\mathrm{~s}, 3 \mathrm{H}), 3.12(\mathrm{~s}, 1 \mathrm{H}) . \mathrm{MS}(\mathrm{ESI}): m / z: 161[\mathrm{M}+$ $\mathrm{H}]^{+}$.

## Synthesis and characterization of alkyne 53



Reagents and conditions: (a) Ethynyltrimethylsilane, TEA, $\mathrm{CuI}, \mathrm{Pd}_{\left(\mathrm{PPh}_{3}\right)_{2} \mathrm{Cl}_{2}, \mathrm{THF}, 50{ }^{\circ} \mathrm{C}, 6 \mathrm{~h}, 98 \% \text {. (b) }}$ $\mathrm{K}_{2} \mathrm{CO}_{3}, \mathrm{MeOH}, \mathrm{rt}, 1 \mathrm{~h}, 98 \%$.

1-(3-Ethynylphenyl)ethanone, (53).
Step 1: To a solution of 1-(3-bromophenyl)ethanone ( $2 \mathrm{~g}, 10.05 \mathrm{mmol}$ ) in THF ( 20 mL ) TEA ( 2.1 mL , $15.07 \mathrm{mmol}), \mathrm{CuI}(7.62 \mathrm{mg}, 0.04 \mathrm{mmol}), \operatorname{Pd}\left(\mathrm{PPh}_{3}\right)_{2} \mathrm{Cl}_{2}(141.08 \mathrm{mg}, 0.20 \mathrm{mmol})$ and ethynyltrimethylsilane $(2.78 \mathrm{~mL}, 20.09 \mathrm{mmol})$ were added in a Schlenk apparatus. The reaction was stirred at $50{ }^{\circ} \mathrm{C}$ for 6 h under nitrogen atmosphere. Then, the mixture was filtered under vacuum and rinsed with ethyl acetate. The volatile was then removed under vacuum and the crude material was purified by column chromatography using petroleum ether/ethyl acetate $98: 2$ as eluent to give 1-(3((trimethylsilyl)ethynyl)phenyl)ethanone as a brown oil ( $2.13 \mathrm{~g}, 9.85 \mathrm{mmol}, 98 \%) .{ }^{1} \mathrm{H}$ NMR ( 300 MHz ; $\left.\mathrm{CDCl}_{3}\right): \delta=7.96(\mathrm{~s}, 1 \mathrm{H}), 7.78(\mathrm{~d}, J=7.7 \mathrm{~Hz}, 1 \mathrm{H}), 7.56(\mathrm{~d}, J=7.7 \mathrm{~Hz}, 1 \mathrm{H}), 7.28(\mathrm{t}, J=7.7 \mathrm{~Hz}, 1 \mathrm{H}), 3.02$ $(\mathrm{s}, 3 \mathrm{H}), 0.24(\mathrm{~s}, 9 \mathrm{H})$.

Step 2: 1-(3-((Trimethylsilyl)ethynyl)phenyl)ethanone (1.84 g, 8.52 mmol ) was dissolved in MeOH (18 $\mathrm{mL})$ and $\mathrm{K}_{2} \mathrm{CO}_{3}(2.96 \mathrm{~g}, 21.4 \mathrm{mmol})$ was added. The reaction was stirred for 1 h . The volatile was
removed under vacuum, ethyl acetate was added and the organic layer was washed with water (x1). After drying over sodium sulfate and evaporation of the solvent, the crude material was purified by column chromatography using petroleum ether/ethyl acetate $95: 5$ as eluent to give compound $\mathbf{5 3}(1.20 \mathrm{~g}, 8.33$ mmol, $98 \%$ ) as a yellow solid. ${ }^{1} \mathrm{H}$ NMR ( $300 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta=8.01(\mathrm{~s}, 1 \mathrm{H}), 7.88(\mathrm{~d}, J=7.6 \mathrm{~Hz}, 1 \mathrm{H})$, $7.61(\mathrm{~d}, J=7.6 \mathrm{~Hz}, 1 \mathrm{H}), 7.37(\mathrm{t}, J=7.6 \mathrm{~Hz}, 1 \mathrm{H}), 3.13(\mathrm{~s}, 1 \mathrm{H}), 2.56(\mathrm{~s}, 3 \mathrm{H}) . \mathrm{MS}(\mathrm{ESI}): m / z: 145[\mathrm{M}+$ $\mathrm{H}]^{+}$.

## Synthesis and characterization of alkyne 54



Reagents and conditions: (a) Oxalyl chloride, conc. $\mathrm{NH}_{4} \mathrm{OH}, \mathrm{THF}, \mathrm{DMF}$, rt, 30 min , $98 \%$. (b) ethynyltrimethylsilane, DIPEA, $\mathrm{CuI}, \mathrm{Pd}\left(\mathrm{PPh}_{3}\right)_{2} \mathrm{Cl}_{2}$, toluene, $100{ }^{\circ} \mathrm{C}, 6 \mathrm{~h}, 48 \%$. (c) TBAF, THF, $0^{\circ} \mathrm{C}, 30$ $\min , 79 \%$.

3-Bromobenzamide, (70).
3-Bromobenzoic acid ( $2 \mathrm{~g}, 9.95 \mathrm{mmol}$ ) was dissolved in THF ( 20 mL ) and oxalyl chloride ( 1.74 mL , $19.90 \mathrm{mmol})$ and DMF $(62 \mu \mathrm{~L})$ were added. After stirring at room temperature for 2 h , conc. $\mathrm{NH}_{4} \mathrm{OH}$ $(12 \mathrm{~mL})$ was added. The reaction was stirred for additional 30 min and then was evaporated, affording compound $70(2.11 \mathrm{~g}, 9.75 \mathrm{mmol}, 98 \%)$ as a white solid. ${ }^{1} \mathrm{H}$ NMR $\left(300 \mathrm{MHz} ; \mathrm{CD}_{3} \mathrm{OD}\right): \delta=8.04(\mathrm{~s}, 1 \mathrm{H})$ $7.84(\mathrm{~d}, J=7.6 \mathrm{~Hz}, 1 \mathrm{H}), 7.70(\mathrm{~d}, J=7.6 \mathrm{~Hz}, 1 \mathrm{H}), 7.39(\mathrm{t}, J=7.6 \mathrm{~Hz}, 1 \mathrm{H})$.

3-Ethynylbenzamide, (54).

Step 1: 3-((Trimethylsilyl)ethynyl)benzamide was synthesized as previously reported for the preparation of compound 52 in step 1, starting from compound $69(1.30 \mathrm{~g}, 6.02 \mathrm{mmol})$. The crude material was subjected to chromatography column using petroleum ether/ethyl acetate 7:3 and petroleum ether/ethyl acetate 5:5 as eluents, yielding 3-((trimethylsilyl)ethynyl)benzamide ( $0.63 \mathrm{~g}, 2.91 \mathrm{mmol}, 48 \%)$ as a brown oil. ${ }^{1} \mathrm{H}$ NMR ( $300 \mathrm{MHz} ; \mathrm{CDCl}_{3}$ ): $\delta=7.88(\mathrm{~s}, 1 \mathrm{H}), 7.76(\mathrm{~d}, J=7.9 \mathrm{~Hz}, 1 \mathrm{H}), 7.60(\mathrm{~d}, J=7.9 \mathrm{~Hz}$, $1 \mathrm{H}), 7.42(\mathrm{t}, J=7.9 \mathrm{~Hz}, 1 \mathrm{H}), 0.26(\mathrm{~s}, 9 \mathrm{H})$.

Step 2: 3-((Trimethylsilyl)ethynyl)benzamide ( $0.63 \mathrm{~g}, 2.89 \mathrm{mmol}$ ) was dissolved in THF ( 13 mL ). The mixture was cooled at $0^{\circ} \mathrm{C}$ and TBAF ( $3.47 \mathrm{~mL}, 3.47 \mathrm{mmol}$ ) was added. The reaction was stirred at 0 ${ }^{\circ} \mathrm{C}$ for 30 min . The volatile was removed under vacuum, water was added and the aqueous layer was extracted with ethyl acetate (x3). The organic phase was dried over sodium sulfate and evaporated. The crude material was purified by column chromatography using petroleum ether/ethyl acetate $5: 5$ as eluent to give compound $54(0.33 \mathrm{~g}, 2.25 \mathrm{mmol}, 79 \%)$ as a yellow solid. ${ }^{1} \mathrm{H}$ NMR ( $300 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta=7.91$ (s, 1 H), $7.80(\mathrm{~d}, J=7.7 \mathrm{~Hz}, 1 \mathrm{H}), 7.64(\mathrm{~d}, J=7.7 \mathrm{~Hz}, 1 \mathrm{H}), 7.43(\mathrm{t}, J=7.7 \mathrm{~Hz}, 1 \mathrm{H}), 3.13(\mathrm{~s}, 1 \mathrm{H}) . \mathrm{MS}$ (ESI): $m / z: 146[\mathrm{M}+\mathrm{H}]^{+}$.

Synthesis and characterization of alkyne 55


Reagents and conditions: (a) Ethynyltrimethylsilane, $\mathrm{PPh}_{3}, \mathrm{Pd}(\mathrm{OAc})_{2}$, TEA, $94{ }^{\circ} \mathrm{C}, 1 \mathrm{~h}, 65 \%$. (b) $\mathrm{K}_{2} \mathrm{CO}_{3}$, $\mathrm{MeOH}, \mathrm{rt}, 30 \mathrm{~min}, 49 \%$.
(3-Ethynylphenyl)methanol, (55).
Step 1: In a Schlenk apparatus (3-bromophenyl)methanol (1.5 g, 8.02 mmol ), TEA ( 7.5 mL ), $\mathrm{PPh}_{3}(42.02$ $\mathrm{mg}, 0.16 \mathrm{mmol}), \mathrm{Pd}(\mathrm{OAc})_{2}(18.01 \mathrm{mg}, 0.08 \mathrm{mmol})$ and ethynyltrimethylsilane $(2.23 \mathrm{~mL}, 16.04 \mathrm{mmol})$ were added. The reaction was stirred at $94^{\circ} \mathrm{C}$ for 1 h under nitrogen atmosphere. Then, the mixture was
filtered under vacuum and rinsed with ethyl acetate. The organic phase was washed with water (3x) and dried over sodium sulfate. The volatile was then removed under vacuum and the crude material was purified by column chromatography using petroleum ether/ethyl acetate $95: 5$ and petroleum ether/ethyl acetate $9: 1$ as eluents to give (3-((trimethylsilyl)ethynyl)phenyl)methanol as a brown oil (1.07 g, 5.25 mmol, 65\%). ${ }^{1} \mathrm{H}$ NMR ( $300 \mathrm{MHz} ; \mathrm{CDCl}_{3}$ ): $\delta=7.96(\mathrm{~s}, 1 \mathrm{H}), 7.26(\mathrm{~d}, J=7.2 \mathrm{~Hz}, 1 \mathrm{H}), 7.18(\mathrm{~d}, J=7.2$ $\mathrm{Hz}, 1 \mathrm{H}), 6.98(\mathrm{t}, J=7.2 \mathrm{~Hz}, 1 \mathrm{H}), 4.50(\mathrm{~s}, 2 \mathrm{H}), 0.21(\mathrm{~s}, 9 \mathrm{H})$.

Step 2: The title compound was synthesized as previously reported for the preparation of compound $\mathbf{5 3}$ in step 2, starting from (3-((trimethylsilyl)ethynyl)phenyl)methanol (1.07 g, 5.25 mmol$)$. The crude material was subjected to chromatography column using petroleum ether/ethyl acetate 95:5 and petroleum ether/ethyl acetate $9: 1$ as eluents, yielding compound $55(0.34 \mathrm{~g}, 2.52 \mathrm{mmol}, 49 \%)$ as a yellow oil. ${ }^{1} \mathrm{H}$ NMR ( $300 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta=7.42(\mathrm{~s}, 1 \mathrm{H}), 7.32(\mathrm{~d}, J=6.2 \mathrm{~Hz}, 1 \mathrm{H}), 7.22-7.11(\mathrm{~m}, 2 \mathrm{H}), 4.56(\mathrm{~s}$, 2H), 3.01 (s, 1H). MS (ESI): $m / z: 133[\mathrm{M}+\mathrm{H}]^{+}$.

## Synthesis and characterization of alkyne 56



Reagents and conditions: (a) Ethynyltrimethylsilane, DIPEA, $\mathrm{CuI}, \mathrm{Pd}\left(\mathrm{PPh}_{3}\right)_{2} \mathrm{Cl}_{2}$, toluene, $100{ }^{\circ} \mathrm{C}, 6 \mathrm{~h}$, $99 \%$. (b) $\mathrm{K}_{2} \mathrm{CO}_{3}, \mathrm{MeOH}, \mathrm{rt}, 30 \mathrm{~min}, 89 \%$.

3-Ethynylbenzenesulfonamide, (56).
Step 1: 3-((Trimethylsilyl)ethynyl)benzenesulfonamide was synthesized as previously reported for the preparation of compound 52 in step 1, starting from compound $72(0.50 \mathrm{~g}, 2.12 \mathrm{mmol})$. The crude material was subjected to chromatography column using petroleum ether/ethyl acetate 95:5 and petroleum ether/ethyl acetate 8:2 as eluents, yielding 3-((trimethylsilyl)ethynyl)benzenesulfonamide
$(0.53 \mathrm{~g}, 2.09 \mathrm{mmol}, 99 \%)$ as a brown oil. ${ }^{1} \mathrm{H}$ NMR ( $300 \mathrm{MHz} ; \mathrm{CDCl}_{3}$ ): $\delta=8.02(\mathrm{~s}, 1 \mathrm{H}), 7.85(\mathrm{~d}, J=7.6$ $\mathrm{Hz}, 1 \mathrm{H}), 7.64(\mathrm{~d}, J=7.6 \mathrm{~Hz}, 1 \mathrm{H}), 7.46(\mathrm{t}, J=7.6 \mathrm{~Hz}, 1 \mathrm{H}), 0.25(\mathrm{~s}, 9 \mathrm{H})$.

Step 2: The title compound was synthesized as previously reported for the preparation of compound $\mathbf{5 3}$ in step 2, starting from 3-((trimethylsilyl)ethynyl)benzenesulfonamide ( $0.53 \mathrm{~g}, 2.11 \mathrm{mmol}$ ). After evaporation, ethyl acetate wad added and the organic phase was washed with 3 N HCl , dried over sodium sulfate and evaporated. The crude material was subjected to chromatography column using petroleum ether/ethyl acetate 7:3 as eluent, yielding compound $56(0.34 \mathrm{~g}, 1.88 \mathrm{mmol}, 89 \%)$ as a yellow oil. ${ }^{1} \mathrm{H}$ NMR ( $300 \mathrm{MHz} ; \mathrm{CD}_{3} \mathrm{OD}$ ): $\delta=8.03(\mathrm{~s}, 1 \mathrm{H}), 7.87(\mathrm{~d}, J=7.7 \mathrm{~Hz}, 1 \mathrm{H}), 7.67(\mathrm{~d}, J=7.7 \mathrm{~Hz}, 1 \mathrm{H}), 7.53(\mathrm{t}$, $J=7.7 \mathrm{~Hz}, 1 \mathrm{H}), 3.67(\mathrm{~s}, 1 \mathrm{H}) . \mathrm{MS}(\mathrm{ESI}): m / z: 182[\mathrm{M}+\mathrm{H}]^{+}$.

## Synthesis and characterization of alkyne 57



Reagents and conditions: (a) O -(Trimethylsilyl)hydroxylamine, TEA, EDCI, dry $\mathrm{CH}_{2} \mathrm{Cl}_{2}$, $\mathrm{rt}, 4 \mathrm{~h}$. (b) TBAF, THF, $0^{\circ} \mathrm{C}, 1 \mathrm{~h}, 70 \%$.

3-Ethynyl-N-hydroxybenzamide, (57).
To a solution of alkyne $\mathbf{S}(99 \mathrm{mg}, 0.68 \mathrm{mmol})$ in dry $\mathrm{CH}_{2} \mathrm{Cl}_{2}(1 \mathrm{~mL})$, TEA $(94 \mu \mathrm{~L}, 0.68 \mathrm{mmol})$, EDCI (143 mg, 0.68 mmol ) and $O$-(trimethylsilyl)hydroxylamine $(99.6 \mathrm{mg}, 0.68 \mathrm{mmol})$ were added in order under nitrogen atmosphere. After 4 h , the reaction was finished. The volatile was removed under vacuum and the crude material was dissolved in THF ( 3.2 mL ). The mixture was cooled at $0^{\circ} \mathrm{C}$ and TBAF $(0.68$
$\mathrm{mL}, 0.68 \mathrm{mmol}$ ) was added. The reaction was stirred for 1 h at $0^{\circ} \mathrm{C}$ and then ethyl acetate was added. The organic phase was washed with $3 \mathrm{~N} \mathrm{HCl}(1 \mathrm{x})$, dried over sodium sulfate and evaporated. The crude material was purified by column chromatography using ethyl acetate and ethyl acetate/methanol 8:2 as eluents, yielding compound $57(76.6 \mathrm{mg}, 0.48 \mathrm{mmol}, 70 \%)$ as a white solid. ${ }^{1} \mathrm{H} \mathrm{NMR}(300 \mathrm{MHz}$; $\left.\mathrm{CD}_{3} \mathrm{OD}\right): \delta=7.84(\mathrm{~s}, 1 \mathrm{H}), 7.73(\mathrm{~d}, J=7.7 \mathrm{~Hz}, 1 \mathrm{H}), 7.59(\mathrm{~d}, J=7.7 \mathrm{~Hz}, 1 \mathrm{H}), 7.44(\mathrm{t}, J=7.7 \mathrm{~Hz}, 1 \mathrm{H})$, $3.56(\mathrm{~s}, 1 \mathrm{H}) . \mathrm{MS}(\mathrm{ESI}): m / z: 162[\mathrm{M}+\mathrm{H}]^{+}$.

NMR Spectra of compounds $1 \mathrm{~S}, 2 \mathrm{~T}, \mathbf{3 G}, \mathbf{3 1 , 3 9}$
1S: ${ }^{1} \mathrm{H}\left(\mathrm{DMSO}-d_{6}\right),{ }^{13} \mathrm{C}$ (DMSO- $d_{6}$ )



2T: ${ }^{1} \mathrm{H}\left(\right.$ DMSO $\left.-d_{6}\right),{ }^{13} \mathrm{C}\left(\right.$ DMSO $\left.-d_{6}\right)$


## 3G: ${ }^{1} \mathrm{H}\left(\mathrm{CDCl}_{3}\right),{ }^{13} \mathrm{C}\left(\mathrm{DMSO}-\boldsymbol{d}_{6}\right)$




31: ${ }^{1} \mathrm{H}\left(\right.$ DMSO $\left.-d_{6}\right),{ }^{13} \mathrm{C}\left(\right.$ DMSO- $\left.d_{6}\right)$


39: ${ }^{1} \mathrm{H}\left(\mathrm{CDCl}_{3}\right),{ }^{13} \mathrm{C}\left(\right.$ DMSO- $\left.d_{6}\right)$



Biology


Figure S1. Concentration-response curves of 1S (A), 2T (B) and 3G (C) in Hek, Jurkat and BV-2 cell lines.

Hek WT






Hek shTRPC1


Figure S2. Effect of 2 T on the area under the curve, the peak amplitude, and the slope of the $\mathrm{Ca}^{2+}$ rise in wild-type and shTRPC1 Hek cells. Bottom histograms represent AUC, peak amplitude and initial rate. 2T and 3G were used at a concentration of $10 \mu \mathrm{M}$.


Figure S3. Effect of 2T (A) or 3G (B) (both at $10 \mu \mathrm{M}$ ) under basal conditions, with no $\mathrm{Ca}^{2+}$ emptying. Hek cells were monitored for 800 seconds, as depicted in the traces. Traces are average of 360 cells from 12 plates in four separate experimental days.
A


B


Figure S4. Concentration-response curves of Pyr3, Pyr6 and selected compounds in Hek cells.

| Gene Symbol | Mouse Primers | Human Primers |
| :---: | :---: | :---: |
| TRPC1 | FW 5'-GGACAGCCTCAGACATTCCA-3', RV 5'-CGGGCTAGCTCTTCATAATCA-3' | FW 5'-ATCTCTACCCAAGCCCCATG-3' RV 5'-ACATTATTAGAGCTGGACTGGC-3' |
| TRPC2 | FW 5'-AGAGGCAGAGCTGGAGTTCA-3' <br> RV 5'-GGGGAATTTGAGCTTTTTGTT-3' |  |
| TRPC3 | FW 5'-GATCAATGCCTACAAGGGACT-3' RV 5'-TGCATTGCATGGAGAGTTTC-3' | FW 5'-CTTACGGCCCTAGAGCTCAG-3' <br> RV 5'-CCAGCACACCCACTACAAAG-3' |
| TRPC4 | FW 5'-ACGCCATCAGAAAAGAGGTG-3' <br> RV 5'-CCAAGATGATGGGTGTGATG-3' | FW 5'-AAGAAGTCGTCGGAGCTGTT-3' RV 5'-TGTCTGGAGTGAATTCAGAGAAC-3' |
| TRPC5 | FW 5'-GAGGTGGTAGGAGCTGTGGA-3' <br> RV 5'-TGCCAACATAATGGGAGTGA-3' | FW 5'-CAGCGTGTATGTGGGTGATG-3' <br> RV 5'-AGAGAACTGCGTGTCCATCA-3' |
| TRPC6 | -------- | FW 5'-ATCTGCTCATGGACTCGGAG-3' <br> RV 5'-AACCTTCTCCССТТСТСACG-3' |
| TRPC7 | --- | FW 5'-CAAGCCTGCGTATTACACCC-3' <br> RV 5'-CTCGTTGAACATGTAGGCGG-3' |
| STIM1 | FW 5'-TCTGAAGAGTCTACCGAAGCAG-3' <br> RV 5'-TGGTAATTGAGGTCTTCCCTTAG-3' | FW 5'-GCCTCAGCCATAGTCACAGT-3' RV 5'-ATGTTACGGACTGCCTCGAA-3' |
| STIM2 | FW 5'-GACGGATGCGATCTGGTG-3' RV 5'-TTCAGTGAAGCAAGGTGGACT-3' | FW 5'-GACGGATGCGAGCTTGTG-3' RV 5'-AAGCATGGTGGACTCAGTGA-3' |
| Orail | FW 5'-CCTGGCGCAAGCTCTACTTA-3' RV 5'-TGCAGGCACTAAAGACGATG-3' | FW 5’-GACCTCGGCTCTGCTCTC-3' RV 5'-TGATCATGAGCGCAAACAGG-3' |
| Orai2 | FW 5'-CACTGTCCTGGAGGAAGCTC-3' RV 5'-GGGCTGAGGGTACTGGTACTT-3' | FW 5'- CCCTCCTCTCCGGCTTTG-3' RV 5'- TGATGAGGAGGGCGAACAG-3' |
| Orai3 | FW 5'-GAACCCGGAGGTGGACAG-3' <br> RV 5'-GCTGGAGGCTTTGAGCTTAG-3' | FW 5'-ACGTCTGCCTTGCTCTCG-3' <br> RV 5'-ACCATGAGTGCAAAGAGGTG-3' |

Table S1. List of primers used in the present manuscript to determine transcript levels

## Metabolic stability data

Hydrolytic stability of selected compounds in mouse plasma. All data are expressed as mean (SEM), $n=$ 3 in three independent experiments.

PYR3


31


15


39


30


43


Metabolic stability data of selected compounds in mouse liver S9 (MLS9) and mouse liver microsomes (MLM) fractions.


## LC-ESI-MS ${ }^{\mathrm{n}}$ analysis of $\mathbf{1 S}$ and its metabolites.





LC-ESI-MS ${ }^{\mathrm{n}}$ analysis of $\mathbf{3 1}$ and its metabolites.



$31[\mathrm{M}+\mathrm{H}]^{+}=486$



LC-ESI-MS ${ }^{\mathrm{n}}$ analysis of $\mathbf{3 9}$ and its metabolites.






## Chromatographic methods

## 1) LC-UV method for purity and metabolic stability determination

A Shimadzu HPLC system (Shimadzu, Kyoto, Japan), consisting of two LC-10AD Vp module pumps, an SLC-10A Vp system controller, an SIL-10AD Vp autosampler, and a DGU-14-A on-line degasser were used for the analysis. The SPD-M10Avp photodiode array detector was used to detect the analytes. LC-Solution 1.24 software was used to process the chromatograms.

- Column: Phenomenex Kinetex $X B$ C18, $150 \times 4.6 \mathrm{~mm}$ (5 $\mu \mathrm{m}$ d.p.) protected with a SecurityGuard ${ }^{\circledR}$ (Torrance, CA, USA).
- Eluant:

A: $0.5 \%$ trifluoroacetic acid in water.
B: $0.5 \%$ trifluoroacetic acid in acetonitrile.

- Flow rate: $1 \mathrm{~mL} / \mathrm{min}$.
- Injection volume: $20 \mu \mathrm{~L}$.
- Wavelength: 249 nm .
- Gradient program: $0 \min [\mathrm{~B} \%=40 \%], 8 \mathrm{~min}[\mathrm{~B} \%=90 \%], 9.5 \mathrm{~min}[\mathrm{~B} \%=95 \%], 10 \mathrm{~min}[\mathrm{~B} \%=40 \%]$, $15 \mathrm{~min}[B \%=40 \%]$.


## 2) LC-ESI-MS methods for metabolite characterization and pharmacokinetic analysis

A Thermo Finningan LCQ Deca XP plus system equipped with a quaternary pump, a Surveyor AS autosampler, a Surveyor photodiode array detector and a vacuum degasser was used for LC-MS analyses (Thermo Electron Corporation, Waltham, MA). Data were acquired and processed using Xcalibur ${ }^{\circledR}$ software.

- Column: Phenomenex Synergi Polar $150 \times 3.0 \mathrm{~mm}$ ( $2 \mu \mathrm{~m}$ d.p.) protected with a SecurityGuard ${ }^{\circledR}$ kept at $35^{\circ} \mathrm{C}$ (Torrance, CA, USA).
- Eluant: A: $0.2 \%$ formic acid in water, B: acetonitrile.
- Flow rate: $0.2 \mathrm{~mL} / \mathrm{min}$.
- Injection volume: $5 \mu \mathrm{~L}$.
- Gradient program for metabolite characterization: $0 \mathrm{~min}[B \%=30 \%], 9 \mathrm{~min}[B \%=80 \%], 9.5 \mathrm{~min}$ $[\mathrm{B} \%=80 \%], 10.5 \mathrm{~min}[\mathrm{~B} \%=30 \%], 15 \mathrm{~min}[\mathrm{~B} \%=30 \%]$.
- Gradient program for pharmacokinetic study: $0 \mathrm{~min}[\mathrm{~B} \%=30 \%], 7 \mathrm{~min}[\mathrm{~B} \%=80 \%], 10.5 \mathrm{~min}$ $[\mathrm{B} \%=80 \%], 11 \mathrm{~min}[\mathrm{~B} \%=30 \%], 15 \mathrm{~min}[\mathrm{~B} \%=30 \%]$.

The operating conditions of the ion trap mass spectrometer were as follows: positive mode, spray voltage, 5.30 kV ; source current, $80 \mu \mathrm{~A}$; capillary temperature, $300^{\circ} \mathrm{C}$; capillary voltage, 21.00 V ; tube lens offset, 5.00 V ; multipole 1 offset, -5.75 V ; multipole 2 offset, -8.50 V ; sheath gas flow $\left(\mathrm{N}_{2}\right), 40$ Auxiliary Units. Negative mode: spray voltage, 3.40 kV ; source current, $80 \mu \mathrm{~A}$; capillary temperature, $300{ }^{\circ} \mathrm{C}$; capillary voltage, -5.00 V ; tube lens offset, -50.00 V ; multipole 1 offset, 8.75 V ; multipole 2 offset, 17.00 V ; sheath gas flow $\left(\mathrm{N}_{2}\right), 40$ Auxiliary Units. Data was acquired in full-scan and product ion scan modes ( $\mathrm{MS}^{\mathrm{n}}$ ) using mass scan range $m / z$ 105-900. The collision energy was optimized at 28-35\%.

