## Supporting Information

# Endoplasmic Reticulum-Targeted Ratiometric NHCBorane Probe for Two-Photon Microscopic Imaging of Hypochlorous Acid 

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## 1. Synthetic Procedures.

## Compounds 1 and 2

Compounds $\mathbf{1}^{\mathrm{S} 1}$ and $\mathbf{2}^{\mathrm{S} 2, \mathrm{~S} 3}$ were prepared according to published procedures.

## Synthesis of 3

$2(1 \mathrm{~g}, 5.52 \mathrm{mmol})$ and $\mathrm{CH}_{3} \mathrm{I}(3.1 \mathrm{~mL}, 2.20 \mathrm{mmol})$ in $\mathrm{CH}_{3} \mathrm{CN}(15 \mathrm{~mL})$ were stirred under reflux for 12 h . After cooling to room temperature, the solvents were removed under vacuum, and the residue was purified by silica gel column chromatography, using $\mathrm{CH}_{2} \mathrm{Cl}_{2} / \mathrm{CH}_{3} \mathrm{OH}(20 / 1, \mathrm{v} / \mathrm{v})$ as eluent to yield $\mathbf{3}$ a white solid ( $0.95 \mathrm{~g}, 4.82 \mathrm{mmol}$, $87 \%$ ). ${ }^{1}$ H NMR (DMSO- $\left.d_{6}, 300 \mathrm{MHz}\right) \delta(\mathrm{ppm}): 9.83$ (br, s, 1H), 8.61 (s, 2H), 8.25$8.20(\mathrm{~m}, 2 \mathrm{H}), 7.71-7.66(\mathrm{~m}, 2 \mathrm{H}), 4.16(\mathrm{~s}, 6 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR (DMSO- $\left.{ }_{d 6}, 75 \mathrm{MHz}\right) \delta(\mathrm{ppm}):$ 147.9, 131.6, 131.6, 128.9, 127.3, 111.4, 34.0 HR-MS (ESI) m/z calcd. for $\mathrm{C}_{13} \mathrm{H}_{13} \mathrm{~N}_{2}^{+}$ $[\mathrm{M}]^{+}=197.1073$, found: 197.1096.

## 2. Experimental procedures

Determination of the Fluorescence Quantum Yield. The quantum yield of the probe was determined according to following equation:

$$
\Phi_{\mathrm{x}}=\Phi_{\mathrm{s}} \mathrm{x}\left(\mathrm{D}_{\mathrm{x}} / \mathrm{D}_{\mathrm{s}}\right) \mathrm{x}\left(\mathrm{~A}_{\mathrm{s}} / \mathrm{A}_{\mathrm{x}}\right) \mathrm{x}\left(\eta_{\mathrm{x}} / \eta_{\mathrm{s}}\right)^{2}
$$

where $\Phi_{\mathrm{s}}$ is the quantum yield of the standard, D is the area under the emission spectra, $A$ is the absorbance at the excitation wavelength, and $\eta$ is the refractive index of the solvent used. x subscript denotes unknown, and s means standard. We chose Quinine sulfate ( $\Phi=0.54$ in 0.1 M of $\mathrm{H}_{2} \mathrm{SO}_{4}$, refractive index, $\mathrm{n}=1.33$ ) as the standard. ${ }^{\text {S4 }}$
Density functional theory (DFT) calculation. Time-dependent density functional theory (TD-DFT) calculations were performed with the Gaussian 09 program package. ${ }^{\text {S5 }}$ Ground state geometry was optimized at B3LYP/6-311++G level of theory, and excited state geometry was optimized at TD-B3LYP/6-311++G level of theory. Vertical transition energy at each geometry, which corresponds to absorption and emission maximum wavelength respectively, was calculated TD-B3LYP/6-311++G level of theory.
Two-Photon Fluorescence Microscopy. Two-photon fluorescence microscopy images of $\mathbf{4}$ were obtained with multiphoton microscopes (Leica TCS SP8 MP) with $\times 40$ oil objectives. Probe was excited 720 nm by a mode-locked titanium-sapphire laser source (Mai Tai HP) with an output power of 1.98 W , which corresponded to approximately $9.19 \times 10^{5} \mathrm{~W} \mathrm{~cm}^{-2}$ average power in the focal plane.
Cell Culture. The cells were cultured on glass bottomed dishes (NEST) for 2 days under a $5 \% \mathrm{CO}_{2}$ humidified atmosphere at $37{ }^{\circ} \mathrm{C}$ and. For imaging, the growth medium was replaced with serum-free medium, treated with $4(10 \mu \mathrm{M})$ and incubated for 30 min . The culture medium is DMEM (WelGene) supplemented with $10 \% \mathrm{FBS}$, streptomycin ( $100 \mu \mathrm{~g} \mathrm{~mL}$ ) and penicillin ( 100 units $\mathrm{mL}^{-1}$ ).

Photostability. Photostability of $\mathbf{4}$ in Raw 264.7 cells was determined by monitoring the changes in TPEF intensity with time at three designated positions of 4 labeled cells. The digitized intensity was recorded with 2.00 sec intervals for the duration of 1 h using xyt mode. The TPEF intensities were collected at $380-600 \mathrm{~nm}$ upon excitation at 720 nm with femto-second pulses.
Co-localization Experiment. Co-localization experiments were performed by costaining with Raw 264.7 cells in the appropriate combinations of $10 \mu \mathrm{M}$ of $\mathbf{4}$ and each commercial organelle tracker $(1.0 \mu \mathrm{M})$ for 30 min . Excitation wavelengths for TPM and OPM are 720 nm and 552 nm , respectively, and the corresponding emissions were collected at $380-550 \mathrm{~nm}$ (4) and $600-650 \mathrm{~nm}$ (organelle trackers). The Pearson's colocalization coefficient $(A)$ was calculated using the AutoQuant X2 program.
Preparation and Imaging of Rat Hippocampus. Slices were prepared from the hippocampi of 14 days old male SD rat and cut into $400 \mu \mathrm{~m}$ thickness using a vibrating-blade microtome in DPBS (Gibco). Slices were incubated with 4 ( $100 \mu \mathrm{M}$, 1.5 h ) under a $5 \% \mathrm{CO}_{2}, 37{ }^{\circ} \mathrm{C}$ atmosphere for 1 h . Slices were then washed two times with DPBS and taken to glass-bottomed dishes. The TPM images were acquired at about 70-190 $\mu \mathrm{m}$ depth.
Determination of Octanol-PBS Partition Coefficient ( $\log \boldsymbol{P}_{\text {oct }}$ ). $20 \mu \mathrm{~L}$ of $\mathbf{4}$ solution in DMSO ( 20 mM ) was added to $5 \mathrm{~mL} n$-octanol. Then, this solution was added 5 mL of PBS buffer ( 10 mM PBS, pH 7.4 ). The resulting mixture was stirred in the dark for 10 min . The probe concentration in each layer was measured spectrophotometrically, using their molar extinction coefficients shown in Table S1. The $\log P_{\text {oct }}$ value was calculated by using $\log P_{\text {oct }}=\log [\text { Probe }]_{\text {oct }}-\log [\text { Probe }]_{\text {PBS }}$; where the $[\text { Probe }]_{\text {oct }}$ and [Probe] $]_{\text {PBS }}$ are the concentrations of the probe in $n$-octanol and PBS, respectively.
Cell viability. Cells were seeded in a 96-well plate with culture media. After 24 hr culture, cells were incubated with probe 4 for 24 hr and washed with DPBS. To identify cell viability, $0.5 \mathrm{mg} / \mathrm{ml}$ of MTT (Sigma) media was added to the cells for 4 hr , and the produced formazan was dissolved in 0.1 ml of dimethylsulfoxide (DMSO) and read at OD 650 nm with a Spectramax Microwell plate reader.

## 3. Spectroscopic Properties and Fluorescence Assays in Solution.



Figure S1. UV-vis (a) and fluorescence (b) spectra of probe $4(10 \mu \mathrm{M})$ before and after the addition of $\mathrm{OCl}^{-}(100 \mu \mathrm{M})$, superimposed with the spectra of compound $\mathbf{3}$.

Table S1. Lowest transition wavelength of $\mathbf{3}$ and 4.

| Molecule | Absorption <br> maximum (nm) | Oscillator <br> strength at <br> ground state <br> geometry | Emission <br> maximum (nm) | Oscillator <br> strength at <br> excited state <br> geometry |
| :--- | :--- | :--- | :--- | :--- |
| $\mathbf{3}$ | 384.04 | 0.0435 | 466.29 | 0.0357 |
| $\mathbf{4}$ | 313.67 | 0.0704 | 347.37 | 0.0759 |

3


4


Figure S2. Highest occupied molecular orbital (HOMO) and lowest unoccupied molecular orbital (LUMO) of $\mathbf{3}$ and 4.


Figure S3. Ratiometric calibration curve of the $\mathbf{4}$ probe in the presence of $\mathrm{OCl}^{-}$, plotting the fluorescence intensity ratio $\left(F_{450} / F_{361}\right)$ against the $\mathrm{OCl}^{-}$concentration $\left(\lambda_{\text {ex }}\right.$ $=326 \mathrm{~nm}$; RT, $10: 90 \mathrm{CH}_{3} \mathrm{CN}$-Aq. PBS, $10 \mathrm{mM}, \mathrm{pH} 7.4$ ).


Figure S4. Calculation of the $\mathrm{OCl}^{-}$detection limit of probe $\mathbf{4}\left(1 \mu \mathrm{M} ; \mathrm{F}=\mathrm{I}_{361}\right)$. The detection limit was taken as concentration corresponding to the Y -axis intercept of the linear regression, where $\log \left[\mathrm{OCl}^{-}\right]=-5.446$, or $\left[\mathrm{OCl}^{-}\right]=3.6 \mu \mathrm{M}$.


Figure S5. pH effect on fluorescence intensity ratio of $\mathbf{4}(10 \mu \mathrm{M})$ in the absence and presence of $\mathrm{OCl}^{-}(100 \mu \mathrm{M})$ with excitation at 326 nm .


Figure S6. Time courses of 4 intensity at $450 \mathrm{~nm}(10 \mu \mathrm{M})$ in the absence and presence of $\mathrm{OCl}^{-}(30 \mu \mathrm{M}$ and $100 \mu \mathrm{M})$ with excitation at 326 nm .


Figure S7. ESI-MS spectrum after the reaction of $\mathbf{B 4}(10 \mu \mathrm{M})$ with $\mathrm{OCl}^{-}(100 \mu \mathrm{M})$.


Figure S8. Cytotoxicity of probe 4.


Figure S9. (a) TPM image of 4 in Raw 264.7 cells and (b) corresponding relative fluorescence intensity from A-C. Fluorescence intensity was recorded for 1 h with 2 sec intervals. The TPM image was acquired at $380-600 \mathrm{~nm}$ upon excitation at 720 nm . Scale bar $=50 \mu \mathrm{~m}$.


Figure S10. (a and e) TPM and (b and f) OPM images of (a-c) HeLa and (e-g) RKO cells co-labeled with $4(10 \mu \mathrm{M})$ and ER-Trackers Red ( $1.0 \mu \mathrm{M}$ ). (c and g) Merged images. (d and h) Line profile of fluorescence intensity obtained from corresponding cells images. Excitation wavelengths for TPM and OPM are 720 nm and 552 nm , respectively, and the corresponding emissions were collected at $380-550 \mathrm{~nm}$ (4) and $600-650 \mathrm{~nm}$ (ER-Tracker Red). Scale bars = (a) 25 and (e) $20 \mu \mathrm{~m}$.

Table S2. Molar extinction coefficient and $\log P_{\text {oct }}$ for $\mathbf{4}$ in $n$-octanol and PBS buffer.

| Probe | Solvent | $\varepsilon\left(10^{-4} \mathrm{M}^{-1} \mathrm{~cm}^{-1}\right)$ | $\log P_{\text {oct }}$ |
| :--- | :--- | :--- | :--- |
|  | $n$-octanol | 1.38 |  |
|  | PBS | 0.71 | $1.04 \pm 0.02$ |
|  |  |  |  |



Figure S11. Pseudocolored ratiometric TPM images of Raw 264.7 cells labeled with 4 $(10 \mu \mathrm{M})$ for 30 min . (a) Control image. (b-e) Cells pretreated with NaOCl for $100 \mu \mathrm{M}$, $200 \mu \mathrm{M}, 500 \mu \mathrm{M}$ and 1 mM for 30 min respectively and then incubated with 4. (f) Average $F_{\text {green }} / F_{\text {blue }}$ ratios in the TPM images. Excitation wavelength for 4 is 720 nm and TPM images were obtained at $380-430 \mathrm{~nm}$ (blue) and 450-600 nm (green). Scale bars $=20 \mu \mathrm{~m}$.


Figure S12. ${ }^{1} \mathrm{H}$ NMR of $\mathbf{3}$.


Figure S13. ${ }^{13} \mathrm{C}$ NMR of 3 .


Figure S14. ESI mass spectrum of 3 .


Figure S15. ${ }^{1} \mathrm{H}$ NMR of 4


Figure S16. ${ }^{13} \mathrm{C}$ NMR of 4


Figure S17. ${ }^{11}$ B NMR of 4


Figure S18. FAB spectrum of 4.

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