## Supporting Information

## In situ fabrication of Nickel/Molybdenum Carbide anchored N-doped graphene/CNT hybrid: An efficient (pre)catalyst for OER and HER

Debanjan Das, Saswati Santra and Karuna Kar Nanda\*

Materials Research Centre, Indian Institute of Science, Bangalore-560012, INDIA.

\* Corresponding Author: Karuna Kar Nanda, nanda@iisc.ac.in

Number of pages: 8, Number of figures: 5, Number of tables: 2

## **Content:**

SEM and TEM images of Ni-Mo <sub>x</sub> C/NC-100	Figure S1
SEM and TEM images of Ni-Mo <sub>x</sub> C/NC-50 and (b) Ni-Mo <sub>x</sub> C/NC-200	Figure S2
Morphological and XRD pattern of MoO3 and NiMoO4.xH2O nanobelts	Figure S3
SEM image and XRD pattern of Ni/NC-100	Figure S4
Nyquist plots	Figure S5
Comparison of HER activity	.Table S1
Comparison of OER activity	. Table S2



**Figure S1.** Large areal SEM (a), TEM (b) and HAADF-STEM (c) images of Ni-Mo<sub>x</sub>C/NC-100 that show the presence of CNTs along with graphitic carbon sheets.



**Figure S2.** SEM images (insets: TEM images) of (a) Ni-Mo<sub>x</sub>C/NC-50 and (b) Ni-Mo<sub>x</sub>C/NC-200.



**Figure S3**. Morphological and structural transformation of MoO<sub>3</sub> nanobelts upon thermal annealing with melamine under identical conditions applied to NiMoO<sub>4</sub>.xH<sub>2</sub>O nanobelts.

Note the absence of CNTs in the case of MoC/NC hybrid which further validates the notion that Ni nanoparticles are solely responsible for the growth of CNTs in Ni-Mo<sub>x</sub>C/NC hybrids (*see main text*). Also, note the visibly broadened XRD peaks of MoC/NC as compared to MoO<sub>3</sub> nanobelts signifying considerable miniaturization during the transformation process.



Figure S4. (a) SEM image and (b) XRD pattern of Ni/NC-100.



**Figure S5.** Nyquist plots at (a) HER at an overpotential of 170 mV and (b) OER at an overpotential of 370 mV in 1M KOH.

**Table S1.** HER activity of some of the recently reported molybdenum carbide-based electrocatalysts in alkaline medium

Catalyst	Electrolyte	η@J=10mA/cm <sup>2</sup>	<b>Tafel slope</b> (mV/dec)	Reference
MoCx octahedra	1М КОН	151	59	<i>Nat. Comm.</i> , <b>2015</b> , 6, 6512
Mo <sub>2</sub> C NPs	1M KOH	176	58	<i>J. Mater. Chem. A</i> , <b>2015</b> , 3, 8361
MoCN-3D	1М КОН	122	32.6	<i>NPG Asia</i> <i>Mater.</i> , <b>2016</b> , 8, e293
Mo <sub>2</sub> C-GNR	1M NaOH	217	64	ACS Sustainable Chem.Eng., <b>2016</b> , 4, 6313–6321
Mo <sub>2</sub> C NS	1М КОН	205	-	<i>Nanoscale</i> <b>2016</b> , <b>8</b> , 16251–16258
Mo <sub>2</sub> C nanocrystals@ N- doped CNTs	1М КОН	257	_	J. Mater. Chem. A, <b>2015</b> , 3, 5783
Mo <sub>2</sub> C@NPC	1M NaOH	144	47.5	<i>J. Mater. Chem.</i> <i>A</i> , <b>2017</b> , 5, 5178
m-Mo <sub>2</sub> C/G	1М КОН	128	56	ACS Appl. Mater. Interfaces, <b>2016</b> , 8, 18107–18118
Mo <sub>2</sub> C-C	1M KOH	169	66	<i>Nano Energy</i> , <b>2017</b> 32, 511–519
1D-Ni/Mo <sub>2</sub> C@C	1М КОН	179	101	<i>Chem. Sci.</i> , <b>2017</b> , 8, 968
Ni-Mo <sub>x</sub> C/NC-100	1M KOH	162	104.8	This work

**Table S2.** OER activity of some of the recently reported transition-metal carbide

 electrocatalysts in alkaline medium

Catalyst	Electrolyte	η @J=10 mA/cm <sup>2</sup>	Tafel slope (mV/dec)	Reference
Fe3C@NCNT/NP C	1М КОН	339	62	<i>Catal. Sci. Technol.,</i> <b>2016</b> , 6, 6365-6371
Fe <sub>3</sub> C@NG-800	0.1M KOH	361	62	ACS Appl. Mater. Interfaces, <b>2015</b> , 7, 21511–21520
Mo <sub>2</sub> C-C	0.1 M KOH	~550	-	<i>J. Mater. Chem. A</i> , <b>2016</b> , 4, 12583–12590
WC	1M NaOH	413	-	<i>Meeting Abstracts</i> <b>2016</b> , <i>MA2016-01</i> , 1449
Co/WC	1М КОН	417	-	<i>Meeting Abstracts</i> <b>2016</b> , <i>MA2016-01</i> , 1449
Mo <sub>2</sub> C-MWCNT	0.1M KOH	~410	-	<i>ChemCatChem</i> , <b>2015</b> , 7, 3911 – 3915
γ-MoC/Ni@NC	1М КОН	310	62.7	Nanoscale, <b>2017</b> ,9, 5583-5588
1D Ni/Mo <sub>2</sub> C@C	1М КОН	368	_	<i>Chem. Sci.</i> , <b>2017</b> , 8, 968-973
Ni-Mo <sub>x</sub> C/NC-100	1М КОН	328	74	This work