Supporting Information

2 Source-specific health risk analysis on particulate trace elements: Coal combustion and traffic emission as major contributors in wintertime Beijing 3 Ru-Jin Huang*,†, Rui Cheng†, Miao Jing‡, Lu Yang†, Yongjie Li¹, Qi Chen§, Yang Chen⊥, Jin 4 5 Yan[†], Chunshui Lin^{†,¶}, Yunfei Wu[▽], Renjian Zhang[▽], Imad El Haddad[#], Andre S. H. Prevot[#], Colin D. O'Dowd¶, Junji Cao† 6 7 *Key Laboratory of Aerosol Chemistry and Physics, State Key Laboratory of Loess and 8 Quaternary Geology, Institute of Earth and Environment, Chinese Academy of Sciences, Xi'an 710061, China 9 ‡COE lab of Thermofisher Scientific Technology, Shanghai 201206, China 10 Department of Civil and Environmental Engineering, Faculty of Science and Technology, 11 12 University of Macau, Taipa 000000, Macau §State Key Joint Laboratory of Environmental Simulation and Pollution Control, College of 13 14 Environmental Sciences and Engineering, Peking University, Beijing 100871, China ¹Chongqing Institute of Green and Intelligent Technology, Chinese Academy of Sciences, 15 Chongqing 400714, China 16 School of Physics and Centre for Climate and Air Pollution Studies, Ryan Institute, National 17 University of Ireland Galway, University Road, Galway H91CF50, Ireland 18 19 ∇RCE-TEA, Institute of Atmospheric Physics, Chinese Academy of Sciences, Beijing 100029, 20 *Laboratory of Atmospheric Chemistry, Paul Scherrer Institute (PSI), 5232 Villigen, 21 22 Switzerland *corresponding authors, email: rujin.huang@ieecas.cn 23 24 25 26 The Supporting Information contains:

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Number of figures: 6 (Figure S1-S6)

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Sampling site description

- 32 The sampling site was in the Institute of Atmospheric Physics, Chinese Academy of Sciences,
- 33 located between the third and fourth ring roads in the northwest Beijing. It is within a
- 34 combination district of education, commercial, and residential units, a typical urban location
- 35 in Beijing.

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Reagents and standard reference materials

- Milli-Q water (18.2 M Ω cm⁻¹; Millipore, Massachusetts, USA) was used for preparation of
- reagents and standards. Analytical grade reagents of HNO₃ (JT Baker, Chemical Co.,
- 40 Phillipsburg, NJ, USA), HONH₃Cl (98.0%, Sigma-Aldrich, USA), CH₃COOH
- 41 (Sigma-Aldrich), CH₃COONH₄ (Sigma-Aldrich), perchloric acid solution (1.0 M,
- 42 Sigma-Aldrich), H₂O₂ (30 wt.% in H₂O, Aladdin, Shanghai, China), and HF (40 wt.% in H₂O,
- J&K., Beijing, China) were used for extraction and digestion. The multi-element standards
- 44 ([7697-37-2], in HNO₃, 20 µg mL⁻¹, SPEX CertiPrep, Avenue, Metuchen, USA) were
- 45 obtained from Inorganic Venture for instrumental calibration of Inductively coupled plasma
- mass spectrometry (ICP-MS). The standard GBW07406 (GSS-6) used for recovery analysis
- and quality control was purchased from the National Research Center for Geoanalysis, China.

Sequential extraction procedure

- 49 A punch of filter (47 mm diameter) was used for each sample. Half of the punch was used for
- 50 the sequential extraction and the other half for total digestion. After each extraction step (see
- Table S1), the extract was separated from the residues by centrifuging the mixture at 4500
- 52 rpm for 15 min. The supernatant was carefully transferred to a Teflon beaker. The residual
- filter was rinsed with fresh extraction solution and then centrifuged again, and finally the
- 54 supernatant was decanted into the same Teflon beaker. The combined supernatants were
- 55 heated and purged to 1-2 mL, then diluted into 5 mL with 2% HNO₃. The extract was then
- 56 filtered through a 0.22 μm PTFE syringe filter (Sterlitech Corp., Kent, WA, USA) and stored
- 57 at 4 °C until ICP-MS analysis.
- 58 We used a four-step extraction procedure for the chemical fractionation of trace elements in
- PM_{2.5}. In the fourth step, quartz filter can digest in the reagent mixture of HNO₃+H₂O₂+HF to
- 60 give high extraction efficiency for the 'residual fraction' of trace elements, while Teflon filter
- 61 cannot digest leading to potential underestimate. We also tested the extraction efficiencies
- 62 using the same extraction procedure on Quartz and Teflon filter samples taken simultaneously.
- 63 Our results show that extraction efficiencies of Teflon filter samples were only about 50% of
- 64 that of quartz filter samples. A number of previous studies also used quartz filters for metal
- analysis in PM.^{1,2} As for the background issue, the data reported in this study were corrected
- 66 for the values from blank samples. Those elements of high background concentrations in
- quartz filters (e.g., K, Na, Mg, Al, Ca) were not discussed in our study.

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Quality assurance / quality control

The recovery of certified materials

- 71 The certified material (GSS-6) was used to assess the accuracy of this method. The recoveries
- for total concentrations of the fourteen elements ranged from 81% to 112%, supporting that
- 73 the extraction and measurement methods, as part of the quality control/quality assurance

protocol, were sufficient and effective. The detailed results of recoveries are listed in Table S2.

Internal check of the recovery of this method

Internal extraction efficiency of the sequential extraction was examined by comparing the sum of the concentration extracted in each fraction (i.e., F1, F2, F3, and F4) with the total concentration (TC) measured in the digested solution for each element. The recoveries ranged from 78% to 123% for these 14 elements, demonstrating the high extraction efficiency and reliability of this method (see Table S3).

ICP-MS analysis

The ICP-MS (iCAP Q, ThermoFisher Scientific, Waltham, MA, USA) was optimized daily. A seven-point calibration curve (i.e., 1, 2, 5, 10, 20, 50, 100 μg L⁻¹) was established for each targeted element, and the regression coefficients for all elements were >0.999. For the analysis, ¹⁰³Rh and ¹⁸⁵Re were added as internal standards at a concentration of 10 μg L⁻¹ in 2% HNO₃. A spiked sample was analyzed for every 10 samples. Blank filter samples were extracted and analyzed following the same procedures. All data reported here were corrected for the field blanks. The accuracy was estimated by analyzing the reference material GBW07406 (GSS-6). The differences between the measured and certified values ranged from -19% to 12% for the fourteen elements, demonstrating a good accuracy of the method.

PMF analysis

The EPA PMF is one of the receptor models that the US EPA's Office of Research and Development has developed. PMF was used to identify and quantify the main sources of these elements. The PMF receptor model has been widely used for PM_{2.5} source apportionment. In this study, the concentrations and uncertainties of fourteen trace elements were included in the PMF 5.0. If the concentration is less than or equal to the method detection limit (MDL) provided, the uncertainty (Unc) is calculated using the following equation ^{3,4}

$$Unc = \frac{5}{6} \times MDL \tag{1}$$

If the concentration is greater than the *MDL*, the uncertainty is then calculated as the detection limits (MDL) and a relative error (5%) summed in quadrature,

Unc =
$$\sqrt{(Error\ Fraction \times concentration)^2 + (MDL)^2}$$
 (2)

The model is a multivariate factor analysis and descriptive model, providing a solution that minimizes an objective function Q based on uncertainty of each measurement. In this study, the PMF solutions from 3 to 6 factors were examined, and the 4-factor solution is selected as it provides the most interpretable profiles and minimal Q value. In Figure S4 we show the correlation (R^2) of each element with each source identified by PMF. From the figure, we can see that relatively high correlation coefficients are obtained between the source markers and a specific source. For example, coal combustion is closely correlated with As (R^2 =0.83), Cd (R^2 =0.91) and Pb (R^2 =0.87). Traffic-related emission is highly correlated with Ni (R^2 =0.90) and Ba (R^2 =0.90). Oil combustion shows a high R^2 value with V (0.86) and dust source is

highly correlated with Ti (0.96) and Fe (0.81). Figure S5a, b and c show the source profiles resolved from the PMF model with 3, 5 and 6 factors, respectively. In the 3-factor solution (Figure S5a), factors 1 and 3 are likely traffic-related emission and coal combustion, respectively. However, factor 2 is a mixture of multiple sources, explaining a major fraction of the variability of both Ti and V, which typically arise from distinct sources: crustal material and oil combustion, respectively. The 4-factor solution enabled the separation of these two sources. In the 5-factor solution (Figure S5b), factor 2, factor 3 and factor 5 could be identified as dust, oil combustion and coal combustion, respectively. However, factor 1 and factor 4 seem to be a split from traffic-related emissions, potentially related to different, but unknown processes. Similar to the 5-factor solution, in the 6-factor solution (Figure S5c), coal combustion was split into factor 4 and factor 6 while traffic-related emission was split into factor 2 and factor 5. Therefore, the 4-factor solution was chosen as the optimal representation of our dataset. Figure S5d shows the value of goodness-of-fit parameter Q corresponding to the solution of different number of factors. It indicates that as the number of factors increases from 3 to 6, the Q/Q_{exp} value decreases. Considering the source profiles resolved from PMF analysis and the Q/Q_{exp} values, we chose the 4-factor solution.

Health risk assessment

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According to the study by Volckens and Leith, the deposition efficiency of particles of size *i* that penetrate into the lung can be calculated as follows:⁵

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$$Ei = -0.081 + 0.23 \cdot \ln(d_p)^2 + 0.23 \cdot \sqrt{d_p}$$
 (3)

Ei is the deposition fraction of particles of size i that penetrate into the lung, which is a complicated function of several deposition mechanisms that include impaction, interception, sedimentation, diffusion, and electric force. In this interpolation equation Ei depends on the aerodynamic particle diameter (dp) and can be used to estimate deposition to the lungs for particles between 0.01 and 10 μ m in diameter.⁵

The winter days from 1 Nov 2013 to 31 Jan 2014 in Beijing were classified into three periods according to the PM_{2.5} concentrations: low pollution days ($\leq 75 \,\mu g \, m^{-3}$, 49 days), moderate pollution days ($75-170 \,\mu g \, m^{-3}$, 30 days), and severe pollution days ($\geq 170 \,\mu g \, m^{-3}$, 13 days). The average concentrations of bioavailable fraction (exchangeable fraction (F1) and reducible fraction (F2)) of particulate-bound trace elements were used for calculation of exposure-point concentration (C). The exposure-point concentration (C) of individual elements in winter 2014 in Beijing was calculated following Eq. (4),

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$$C = \frac{(F1+F2)_{clean} \times 49 + (F1+F2)_{moderate} \times 30 + (F1+F2)_{severe} \times 13}{90}$$
 (4)

where the values of F1 and F2 are the average concentrations in exchangeable fraction and reducible fraction for each measured element during low, moderate and severe pollution days from 1st to 25th January 2014, respectively.

The carcinogenic and non-carcinogenic risks by airborne metals via direct inhalation of PM_{2.5} were calculated using US Environmental Protection Agency (US EPA) human health risk assessment models (US EPA 2009),⁶ which mainly involve exposure assessment and risk characterization. The methodology has been used in previous studies.^{1,2}

Sensitive local residents were divided into two groups (i.e., children and adults). The

inhalation exposure concentration (EC), hazard quotient (HQ) for non-cancer risk, and 155 carcinogenic risks (CR) by a toxic element in PM_{2.5} were calculated following Eqs. (5-7) 156

$$EC = \frac{C \times Ei \times ET \times EF \times ED}{ATn}$$
 (5)

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$$EC = \frac{C \times Ei \times ET \times EF \times ED}{ATn}$$

$$EC = \frac{EC}{RfC \times 1000 \ \mu g \ mg^{-1}}$$
(5)

$$CR = IUR \times EC \tag{7}$$

where ET is the exposure time (24 h/day), EF is the exposure frequency (180 day/year), ED is the exposure duration (6 years for children and 24 years for adults), ATn is the averaging time (for non-carcinogens ATn = ED \times 365 day/year \times 24 h/day, and for carcinogens ATn = 70 year \times 365 day/year \times 24 h/day), RfC is the inhalation reference concentration (mg m⁻³) and IUR is the inhalation unit risk (µg m⁻³)⁻¹). The RfC, IUR and default values for exposure were obtained from the user's guide and technical background document for the U.S. EPA region 9 regional screening level tables (US EPA, 2013). Elements that induce non-carcinogenic but toxic effects are As, Cd, Co, Cr (VI), Mn, Ni and V. The RfC is used for the non-carcinogenic risk characterization. Elements that induce carcinogenic effects include As, Cd, Cr (VI), Ni, Co, and Pb. The IUR is used for the carcinogenic risk characterization.

The carcinogenic risk (CR) is the probability of an individual developing any type of cancer from lifetime exposure to carcinogenic hazards; the acceptable risk range is 1.0×10^{-6} according to the US EPA risk management guidelines (2009).6 The sum of hazard quotient (HQ) below the precautionary level of 1 suggests that there is no significant risk of non-carcinogenic effects, while their values above 1 indicate that there is a chance of non-carcinogenic effects occurring, with a probability that tends to increase as the value of HQ increases (US EPA 2013).⁷

The carcinogenic and non-carcinogenic health risks of the four emission sources (oil combustion, dust, traffic-related and coal combustion) were assessed. The sum of carcinogenic risk (CR) of carcinogenic elements is used to assess the total carcinogenic risk of each emission source. For non-carcinogenic risks, the sum of hazard quotient (HQ) of non-carcinogenic elements is used to assess the overall non-carcinogenic effects of emission source.^{1,2} The carcinogenic risk and non-carcinogenic risk of each source (CR_s and HO_s) is calculated using Eqs. (8) and (9), respectively,

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$$CR_s = \sum (CR_i \times RC_i)$$
 (i= As, Cd, Cr (VI), Ni, Co, and Pb) (8)

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$$HQ_s = \sum (HQ_i \times RC_i) \quad (i=As, Cd, Co, Cr (VI), Mn, Ni \text{ and } V)$$
 (9)

RCi is the relative contribution of emission source to individual elements resolved from 186 source profiles (Figure S3). 187

The fraction (F_{car(s)}) of CR_s of each source in the total carcinogenic risk is calculated using Eq. 188

(10), while the fraction (F_{non(s)}) of HQ_s of each source in the total non-carcinogenic risk is 189

calculated using Eq. (11), 190

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$$F_{car(s)} = \frac{CR_s \times RC_s}{\sum (CR_s \times RC_s)}$$
 (S= oil combustion, dust, traffic-related and coal combustion) (10)

 $F_{\text{non(s)}} = \frac{HQ_s \times RC_s}{\sum (HQ_s \times RC_s)} \quad \text{(S= oil combustion, dust, traffic-related and coal combustion) (11)}$

RCs is contribution of each emission source to total mass of measured elements in source apportionment (Figure 3a in main text).

According to the guideline provided by U.S. EPA, Cr (VI) and Cr (III) are classified as Group A (human carcinogens) and Group D (not classifiable for human carcinogenicity), respectively. However, the total concentrations of Cr were obtained by the sequential

atmosphere was reported to be about 1:6, the concentration of Cr (VI) was approximated to be

extraction method in this study. As the concentration ratio of Cr (VI) to Cr (III) in the

one seventh of the total Cr concentration for the carcinogenic risk calculation.^{8,9}

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Table S1. Chemical fractions, reagents and operational conditions during the sequential extraction procedure in this study.

extraction procedure in this	study.		
Metal fraction	Reagent	Experimental conditions	
Fraction 1. Soluble and	15 mL Milli-Q water	agitation on a shaker at room	
exchangeable metals	(pH=7.4)	temperature for 3 hours	
Fraction 2. Carbonates,	10 mL 0.25 M NH ₂ OH HCl	agitation on a shaker at room	
oxides and reducible	at pH=2.0	temperature for 5 hours	
metals			
		agitation on a shaker at 95 °C	
Fraction 3. Bound to	$7.5 \text{ mL } H_2O_2 \ 30 \ \% \ + \ 7.5$	until near dryness + agitation on	
organic matter, oxidisable	$mL\ H_2O_2\ 30\%\ +\ 15\ mL\ 2.5$	a shaker at 95 °C until near	
and sulphidic metals	M NH ₄ AcO at pH=3.0	dryness + agitation on a shaker at	
		room temperature for 90 min	
Enaction 4 Decident	4I IINO + 2I II O +	10 min, 500 W, 130 °C; 5 min,	
Fraction 4. Residual	4 mL HNO ₃ + 2 mL H ₂ O ₂ +	500 W, 150 °C; 5 min, 500 W,	
metals	0.2 mL HF	180 °C; 15 min, 500 W, 200 °C	

Table S2. The recovery of each element measured in standard sample GBW07406 (GSS-6) following the same extraction procedure.

μg L-1	Certified value	Measured value	Recovery %
Ti	1756.0	1615.5	92.5
V	52.0	45.8	88.1
Cr	30.0	31.5	105.3
Mn	580.0	551.0	95.9
Fe	24400.0	23668.0	97.8
Co	3.0	3.2	104.9
Ni	21.2	22.5	106.7
Cu	156.0	142.0	91.7
Zn	38.8	40.0	103.4
As	88.0	75.7	86.3
Sr	15.6	12.6	81.9
Cd	0.15	0.06	112.1
Ba	47.2	51.0	108.8
Pb	125.6	113.0	90.8

Table S3. The internal recovery of each element measured in $PM_{2.5}$ samples using the same extraction procedure.

Element	Average Internal
Liement	Recovery %
Ti	78.4 ± 12.8
V	81.4 ± 13.4
Cr	123.3 ± 19.0
Mn	93.5 ± 15.0
Fe	85.1 ± 14.5
Co	77.5 ± 8.7
Ni	101.3 ± 9.7
Cu	92.8 ± 9.3
Zn	120.0 ± 27.7
As	84.4 ± 14.7
Sr	80.3 ± 11.7
Cd	101.1 ± 26.2
Ba	104.3 ± 9.5
Pb	120.3 ± 14.6

Table S4. The concentrations of trace elements during low pollution days, moderate pollution days, severe pollution days and entire sampling period. These abbreviations, such as L, M and S represent the concentrations of particulate-bound trace elements in low pollution days, moderate pollution days and severe pollution days, respectively.

ng m ⁻³	Low	Moderate	Severe	Full period	M/L	S/L
Ti	51.6 ± 14.7	59.0 ± 5.8	66.6 ± 20.4	56.6 ± 13.4	1.14	1.3
V	2.2 ± 0.7	4.7 ± 1.6	5.4 ± 2.0	3.7 ± 2.0	2.1	2.5
Cr	12.5 ± 3.1	17.1 ± 4.3	20.3 ± 1.2	15.4 ± 5.1	1.4	1.6
Mn	22.6 ± 6.6	46.1 ± 13.1	77.8 ± 21.1	40.9 ± 22.1	2.0	3.4
Fe	536.7 ± 146.8	946.1 ± 315.3	1080.3 ± 322.6	771.8 ± 297.7	1.8	2.0
Co	0.6 ± 0.1	0.9 ± 0.2	1.3 ± 0.4	0.8 ± 0.3	1.5	2.2
Ni	52.6 ± 16.6	59.2 ± 17.4	67.4 ± 9.4	57.8 ± 17.6	1.1	1.3
Cu	35.2 ± 6.4	59.1 ± 10.6	77.4 ± 20.6	53.5 ± 19.5	1.7	2.2
Zn	73.9 ± 19.6	188.4 ± 68.4	427.9 ± 73.1	183.6 ± 138.0	2.5	5.8
As	4.5 ± 1.1	9.5 ± 2.8	32.9 ± 13.0	11.0 ± 9.4	2.1	7.3
Sr	14.0 ± 4.3	17.6 ± 7.4	20.4 ± 8.0	16.4 ± 6.9	1.3	1.5
Cd	0.8 ± 0.3	2.0 ± 0.8	5.6 ± 2.3	2.0 ± 1.3	2.5	7.0
Ba	87.8 ± 25.2	119.8 ± 23.0	129.1 ± 29.3	109.4 ± 27.6	1.4	1.5
Pb	65.3 ± 20.1	147.8 ± 46.9	284.8 ± 94.4	138.1 ± 69.9	2.3	4.4

Table S5. The calculated EC, RfC and IUR.

Element	Concentration (ng m ⁻³)			RFC	IUR
	(ng m ⁻³)	Children	Adults	- (mg m ⁻³)	(µg m ⁻³) ⁻¹
As (Inorganic)	7.53E+00	2.61E+00	2.61E+00	1.50E-05	_
Cd (Diet / water)	1.87E+00	6.48E-01	6.48E-01	1.00E-05	
Co	5.01E-01	1.74E-01	1.74E-01	6.00E-06	
Cr (VI)	4.84E-01	1.68E-01	1.68E-01	1.00E-04	
Mn (Diet)	3.10E+01	1.08E+01	1.08E+01	5.00E-05	
Ni (refinery dust)	2.60E+01	9.03E+00	9.03E+00	1.40E-05	
V	2.39E+00	8.30E-01	8.30E-01	1.00E-04	
As (Inorganic)	7.53E+00	2.24E-01	8.96E-01		4.30E-03
Cd (Diet / water)	1.87E+00	5.55E-02	2.22E-01		1.80E-03
Co	5.01E-01	1.49E-02	5.96E-02		9.00E-03
Cr (VI)	4.84E-01	1.44E-02	5.76E-02		8.40E-02
Ni (refinery dust)	2.60E+01	7.74E-01	3.10E+00		2.40E-04
Pb (acetate)	1.06E+02	3.16E+00	1.26E+01		8.00E-05

Table S6. Carcinogenic and non-carcinogenic risks resulted from trace elements via inhalation exposure to $PM_{2.5}$ in winter 2014 in Beijing.

Metal	Carcinogenic (CR)		Non-Carcinogenic (HQ)		
Metai -	Children	Adults	Children	Adults	
As (Inorganic)	9.64E-07	3.85E-06	1.74E-01	1.74E-01	
Cd (Diet / water)	9.99E-08	4.00E-07	6.48E-02	6.48E-02	
Co	1.34E-07	5.37E-07	2.90E-02	2.90E-02	
Cr (VI)	1.21E-06	4.84E-06	1.68E-03	1.68E-03	
Mn (Diet)			2.15E-01	2.15E-01	
Ni (refinery dust)	1.86E-07	7.43E-07	6.45E-01	6.45E-01	
V			8.30E-03	8.30E-03	
Pb (acetate)	2.53E-07	1.01E-06			

Sum	2.85E-06	1.14E-05	1.14E+00	1.14E+00

Table S7. Carcinogenic and non-carcinogenic risks resulted from four emission sources via
 inhalation exposure to PM_{2.5} in winter 2014 in Beijing.

Sources	Carcinogenic (CR)		Non-Carcinogenic (HQ)		
Sources	Children	Adults	Children	Adults	
Oil combustion	4.55E-08	1.82E-07	6.52E-02	6.52E-02	
Traffic-related	1.21E-06	4.84E-06	6.89E-01	6.89E-01	
Coal combustion	1.33E-06	5.31E-06	3.29E-01	3.29E-01	
Dust	2.62E-07	1.05E-06	5.48E-02	5.48E-02	

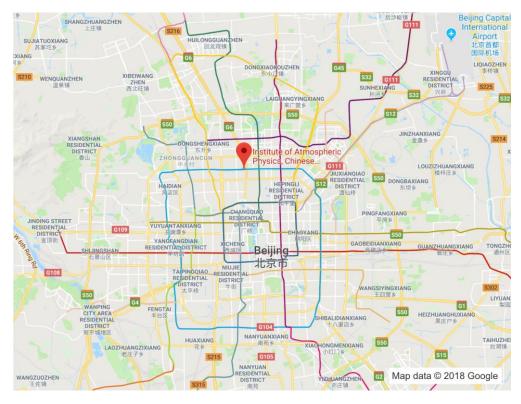
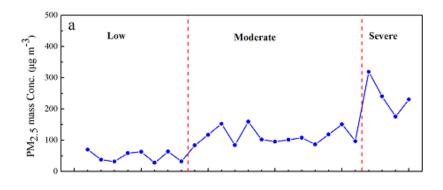


Figure S1. The sampling site in Beijing (http://www.google.com).



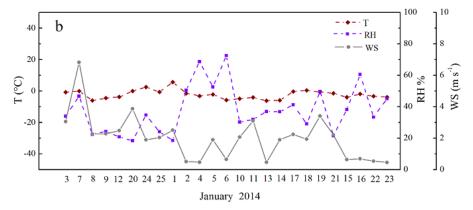


Figure S2. Daily PM_{2.5} mass concentrations (a) and meteorological data (b) (T = temperature, RH = relative humidity and WS = wind speed) during the sampling period.

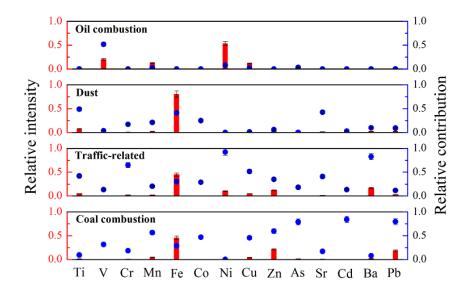
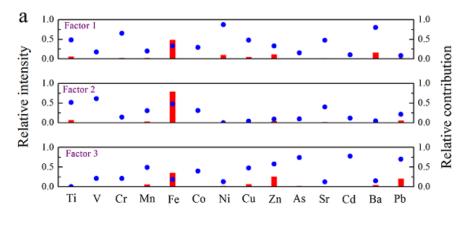
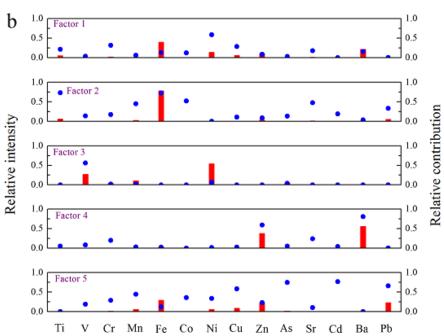


Figure S3. Source profiles resolved from the PMF model. The bars (left y axis) represent the relative intensity of element to each factor in ng m⁻³, circles (right y axis) represent the fraction of the total predicted concentration for a given element.

Figure S4. The correlations of each element with each source identified by the PMF model. The bars (left y axis) represent the R^2 (factors vs elements), while circles (right y axis) represent the contribution of the total predicted concentration for a given element





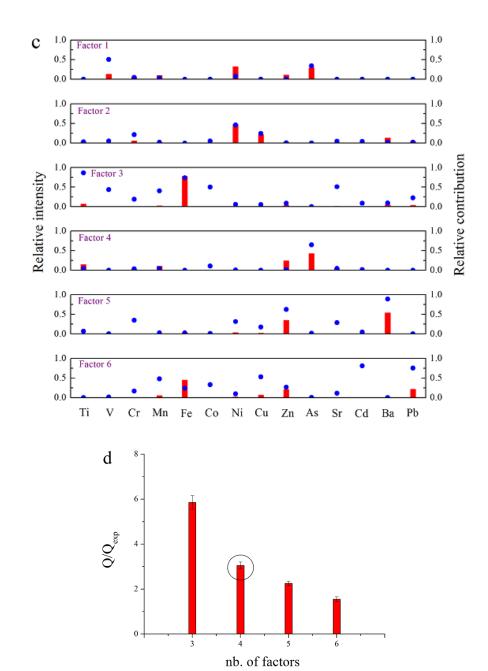
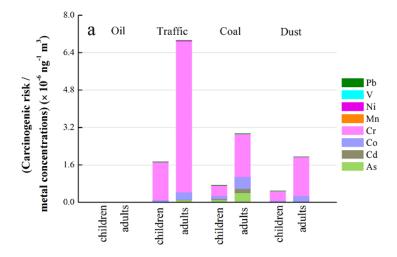


Figure S5. Source profiles resolved from the PMF model with 3 factors (a), 5 factors (b) and 6 factors (c), respectively. The bars (left y axis) represent the relative intensity of element to each factor in ng m⁻³, circles (right y axis) represent the fraction of the total predicted concentration for a given element. The value of goodness-of-fit parameter Q (d) corresponded to the source profiles of different number of factors.





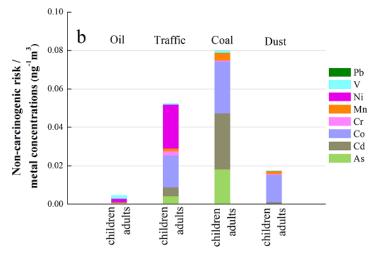


Figure S6. The normalized health risk of toxic elements (As, Cd, Co, Cr (VI), Mn, Ni, V and Pb) in four emission sources.

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