

Supplementary Information

VO₂(M)@SiO₂/Poly(N-isopropylacrylamide) Hybrid Nanothermochromic Microgels for Smart Window

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1.1 The X-ray diffraction pattern of VO₂(M) and VO₂(M)@SiO₂

We verified the crystalline phase of VO₂ in nanoparticles (NPs) of VO₂(M) and VO₂(M)@SiO₂ by X-ray diffraction (XRD), respectively. As shown in Fig. S1, all the diffraction peaks of the VO₂(M)@SiO₂ sample corresponded to those of monoclinic VO₂(M) (JCPDS no.043-1051), without a broad reflection peak from $2\theta=20^\circ$ to $2\theta=30^\circ$ which can be assigned to non-uniformity of the SiO₂ shell. The unchanged XRD pattern also confirmed the chemical stability of VO₂(M) is stable during modifying process.

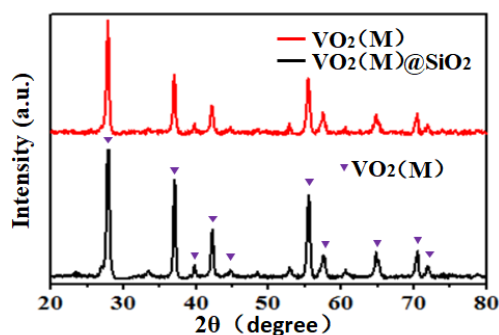


Figure S1. XRD results of the VO₂(M) NPs and VO₂(M)@SiO₂ NPs.

1.2 Effect of Surfactant Type on VO₂(M) Dispersion

Figs. S2 show the stability of solvent prepared with two kinds of surfactant: sodium dodecyl sulfate (SDS) and poly(vinylpyrrolidone) (PVP). It can be seen that all the VO₂ NPs subsided in both SDS systems with two different contents of VO₂ NPs (Figs. S2a and b), while VO₂ NPs were all well dispersed in the existence of PVP polymer (Figs. S2c and d).

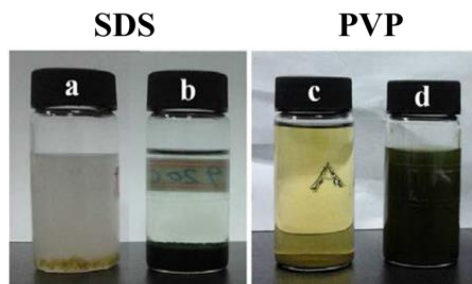


Figure S2. Composite microgels synthesized with different surfactants and varying amounts of VO₂(M) nanoparticles: (a) SDS, 0.05 wt% VO₂(M); (b) SDS, 0.1 wt% VO₂(M); (c) PVP, 0.05 wt% VO₂(M); (d) PVP, 0.1 wt% VO₂(M).

1.3 The transparent-opaque transition of the smart window

Before or after phase transition, all the hybrid microgels (samples I, II and III) were kept transparent and homogeneous, without any precipitation phenomenon. At room temperature, the hybrids were transparent, and brownish due to the existence of VO₂(M) NPs (left column of Fig. S3). The color of films changed with VO₂(M) content: the higher the VO₂(M) content was, the deeper the brownish color was.

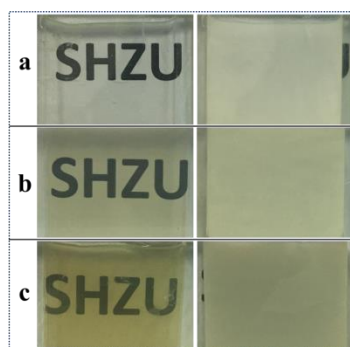


Figure S3. Pictures of different hybrid microgel samples (a) sample I, (b) sample II, (c) sample III at 20 °C (left column) and 50 °C (right column).

1.4 The phase transition behavior of VO₂ and VO₂@SiO₂

The pure VO₂(M) nanoparticles were synthesized through a “heating-up” process. According to the DSC (Differential scanning calorimeter) results, the phase transition began at 60 °C and ended at 100 °C. The latent heat during the insulator-to-metal transition (IMT) of VO₂(M) NPs is around 43 J g⁻¹. This high crystal quality will achieve strong solar energy regulation ability. The transition temperature of the nanoparticle remained unchanged after being coated by SiO₂ (VO₂(M)@SiO₂, solid lines).

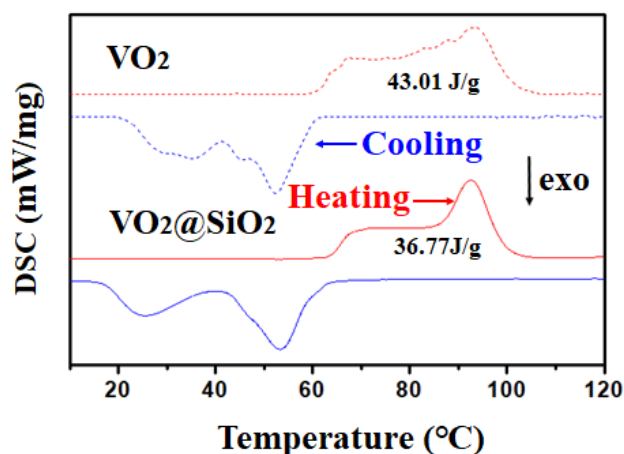


Figure S4. DSC curves as function of temperature upon heating (red lines) or cooling (blue lines) for VO₂ (dashed lines) and VO₂@SiO₂ (solid lines). The latent heat of VO₂ and VO₂@SiO₂ are shown as the inserts.

1.5 The optical performance of Sample I, II and III and pure PNIPAm microgel

Optical properties between pure PNIPAm microgel and hybrids (Sample I, II and III) are shown in Table S1.

Table S1. Comparison of the optical properties for Sample I, II, III with PNIPAm microgel.

Sample	$T_{\text{lum,low}}$ (%)	$T_{\text{lum,high}}$ (%)	ΔT_{lum} (%)	$T_{\text{lum,average}}$ (%)	ΔT_{IR} (%)	ΔT_{sol} (%)
PNIPAm	67.0	1.6	65.3	34.3	29.4	64.9
Sample I	72.7	1.7	71.0	37.2	26.7	64.9
Sample II	59.7	1.3	58.4	30.5	23.4	55.3
Sample III	75.6	1.3	74.3	38.4	22.4	62.7

1.6 TEM image of hybrids with 4 g/L PVP content

As shown in Fig. S5, there existed a thick gel layer which was caused by the high content of PVP. Therefore, a high PVP concentration will influence the hybrid structure and make the structure change from microgel to large-sized hydrogel to some extent. The large-size hydrogel shows a long response time and a relatively small reduction of transmission in both visible and IR ranges after phase transition, as reported by the literatures (Zhou, Y; Cai, Y.-F.; Hu, X.; Long, Y. Temperature-responsive hydrogel with ultra-large solar modulation and high luminous transmission for “smart window” applications. *J. Mater. Chem. A* **2014**, 2, 13550-13555.). Thus a high PVP concentration is not favorable for achieving good optical performance of the hybrid microgel.

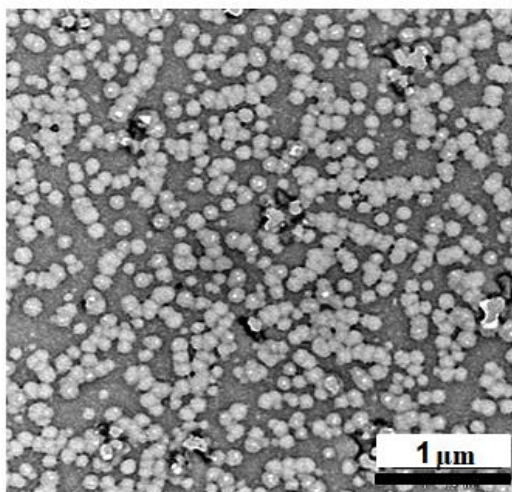


Figure S5. TEM image of hybrids obtained from 4 g/L PVP content.

1.7 Comparison of optical performance of previously reported results and my hybrid sample

As shown in Table, comparing with the result of given reviews, our composite film shows the excellent solar modulation properties (62.7%) with high luminous transmittance (75.6%). But we also have some short comings, the high solar modulation properties sacrificed the luminous transmittance at high temperature, which means that we need extra energy to keep visible transmittance for views.

Table S2. Comparison of our hybrid sample with other thermochromic materials (“-” means data is not available).

Category	Sub-category	Material content	Modulated solar spectrum (nm)	$T_{lum}(\%)$	$\Delta T_{sol}(\%)$	Ref.
Unhybridized system	VO ₂	VO ₂	700-2500	45.6	22.3	1
		W+Zr-doped VO ₂	700-2500	56.4	12.3	2
	Hydrogel	PNIPAm	250-2200	70.7	25.5	3
		Modified cellulose	250-2200	67.4	25.7	4
		Modified starch	250-2200	87.7	40.1	5
	Metamaterial	Kiri-Kirigami metamaterials	200-2200	-	-	6
	Perovskite	Halide perovskite	400-700	-	-	7
	Liquid crystal	-	400-2200	-	-	8
VO ₂ -based hybrid system	The other material is non-thermal-responsive	VO ₂ /PU	700-2500	45.6	22.3	1
		VO ₂ /Si-Al gel	700-2500	59.1	12	9
		VO ₂ /PDMS	700-2500	85	-	10
		VO ₂ /TiO ₂	700-2500	61.2	14.6	11
	The other material is thermal-responsive	VO ₂ /PNIPAm	250-2500	62.6	34.7	12
		VO ₂ /HPC	250-2500	56	36	13
		VO ₂ /NLETS	500-2500	71	18.2	14
		Our composite VO ₂ @SiO ₂ /PNIPAm	250-2500	75.6	62.7	This work

1.8 References

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