

Supplementary Information

Metal-Organic Framework-Derived Hollow $\text{CoS}_x\text{@MoS}_2$ Microcubes as Superior Bifunctional Electrocatalysts for Hydrogen Evolution and Oxygen Evolution Reactions

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Number of SI pages: 14

Number of SI figures: 11

Number of SI tables: 2

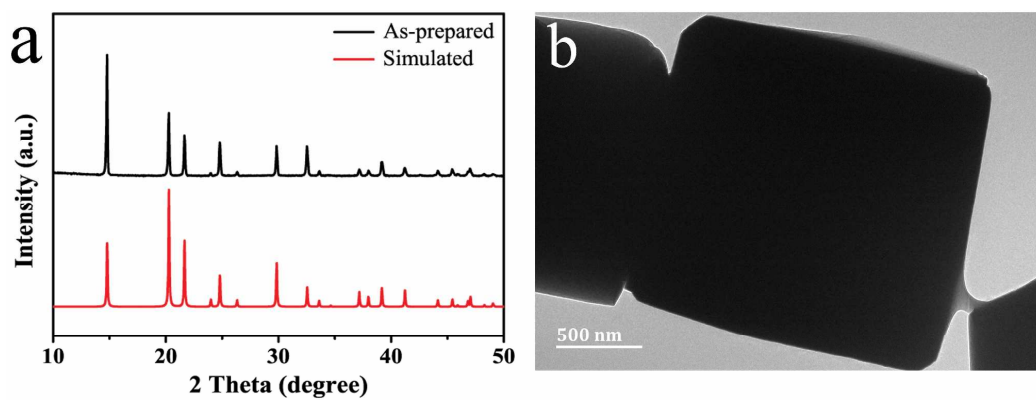


Figure S1. PXRD pattern (a) and TEM image (b) of Co-MOF.

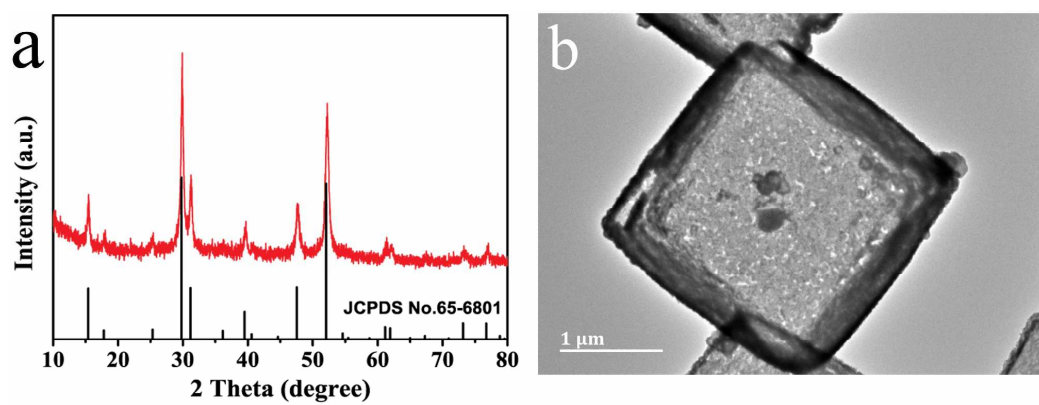


Figure S2. PXRD pattern (a) and TEM image (b) of Co₉S₈.

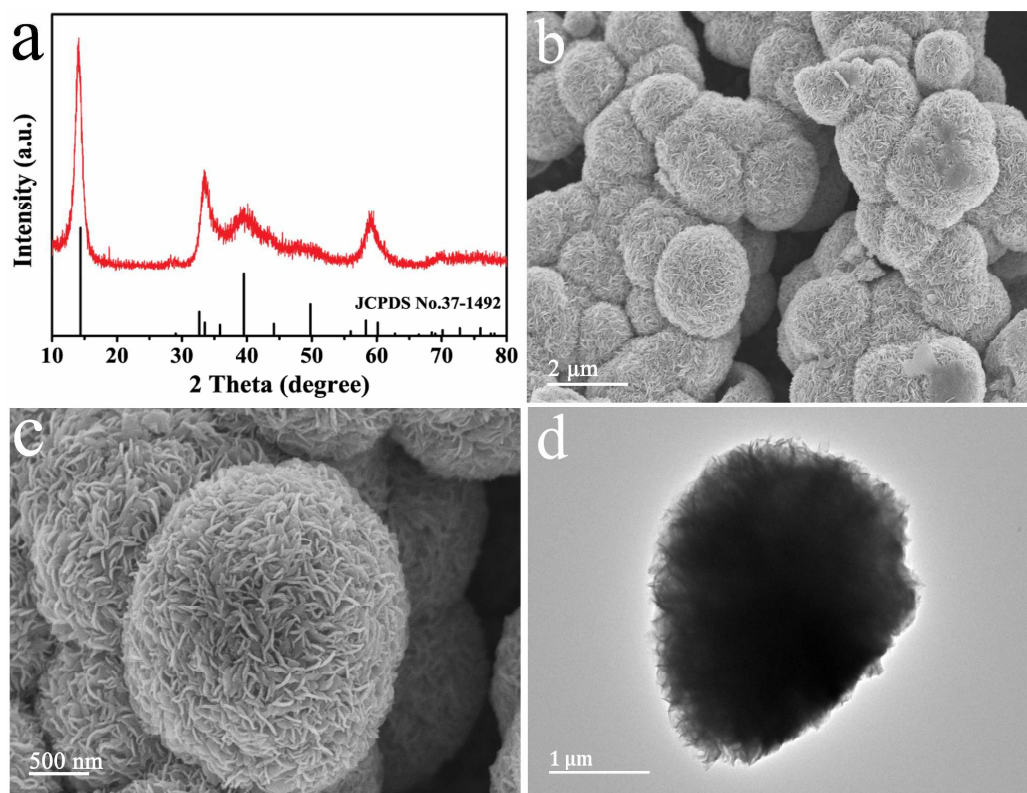


Figure S3. PXRD pattern (a), FESEM (b, c) and TEM images (d) of MoS₂.

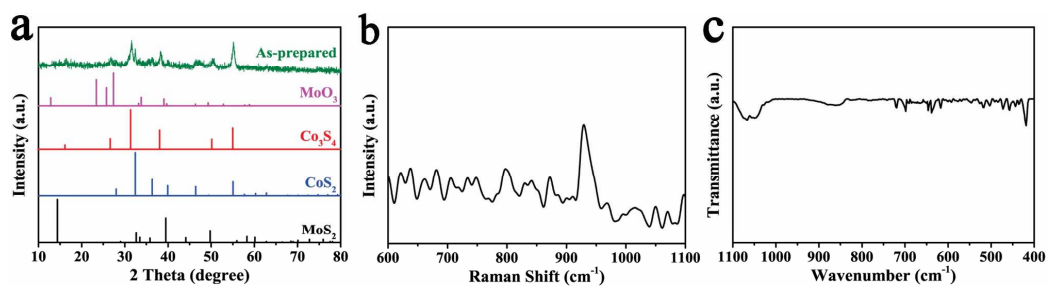


Figure S4. PXRD pattern (a), Raman spectrum (b) and FTIR spectrum (c) of $\text{CoS}_x@\text{MoS}_2$.

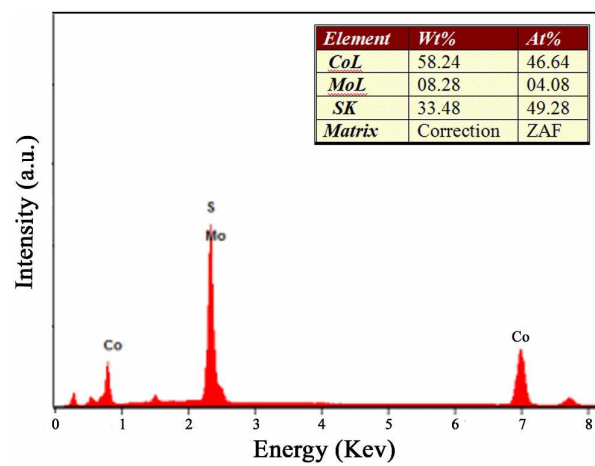


Figure S5. EDX spectrum of hollow $\text{CoS}_x\text{@MoS}_2$ microcubes.

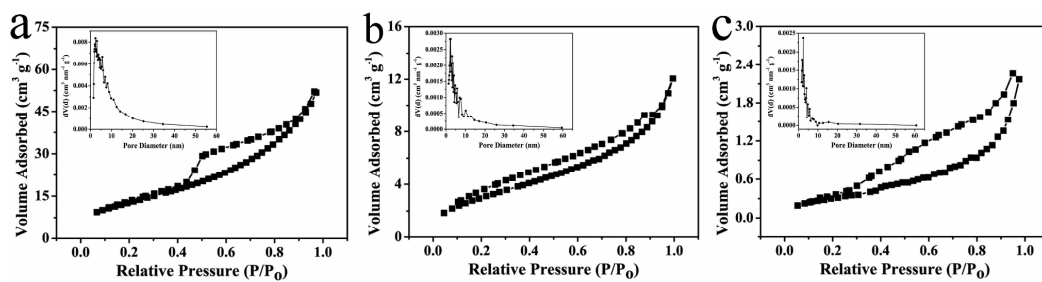


Figure S6. Nitrogen adsorption–desorption isotherms of $\text{CoS}_x@\text{MoS}_2$ (a), Co_9S_8 (b), MoS_2 (c) and corresponding pore size distributions (insets).

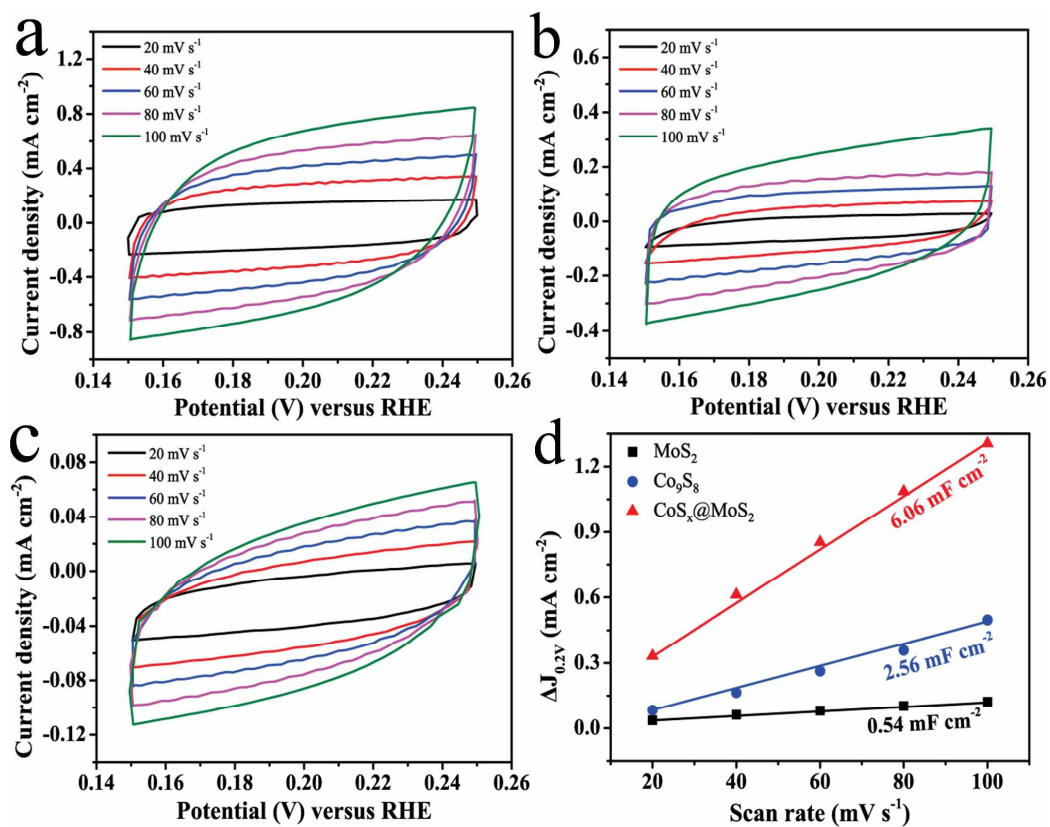


Figure S7. Cyclic voltammograms of CoS_x@MoS₂ (a), Co₉S₈ (b) and MoS₂ (c) measured at various scan rates in 0.5 M H₂SO₄, corresponding linear slopes at 0.2 V vs RHE (d). The double-layer capacitance (C_{dl}) is equivalent to half the slope.

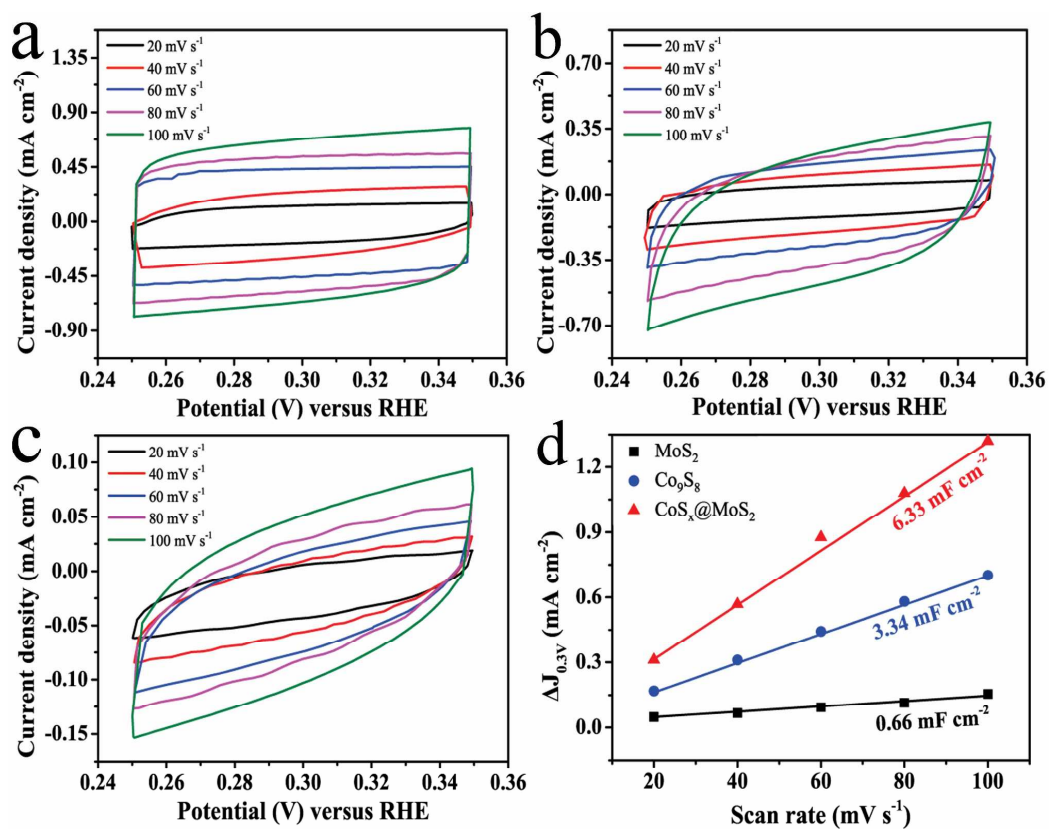


Figure S8. Cyclic voltammograms of CoS_x@MoS₂ (a), Co₉S₈ (b) and MoS₂ (c) measured at various scan rates in 1 M KOH, corresponding linear slopes at 0.3 V vs RHE (d). The double-layer capacitance (C_{dl}) is equivalent to half the slope.

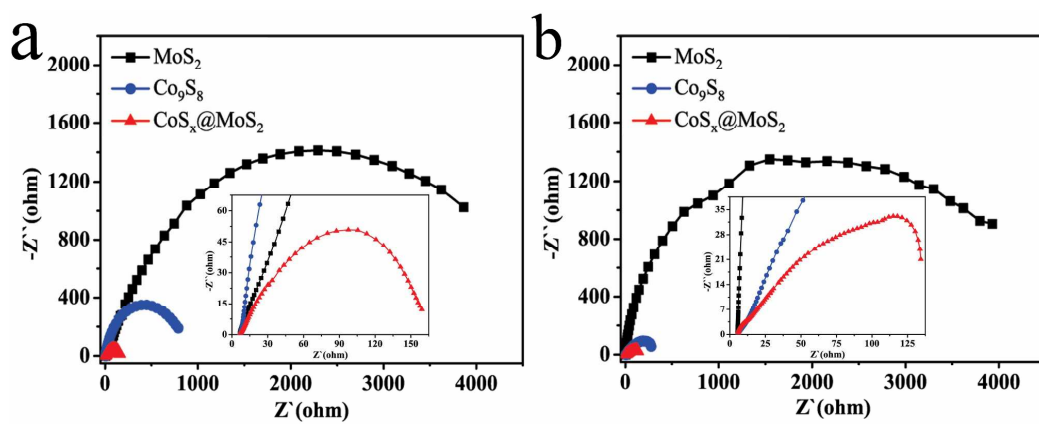


Figure S9. Nyquist plots of $\text{CoS}_x@\text{MoS}_2$, Co_9S_8 and MoS_2 for HER collected at -0.15 V vs RHE (a), for OER collected at 1.5 V vs RHE (b).

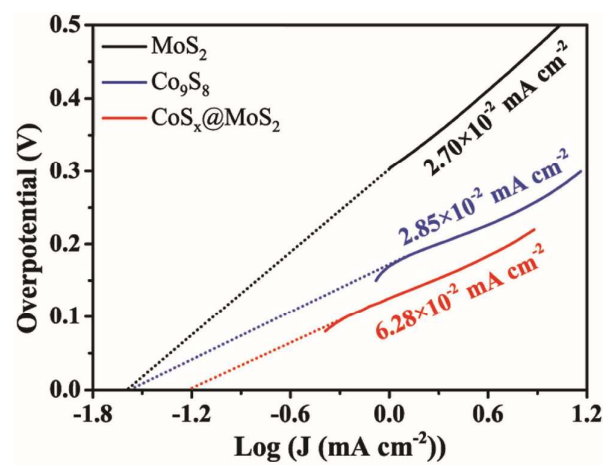


Figure S10. Exchange current densities of MoS₂, Co₉S₈ and CoS_x@MoS₂ for HER.

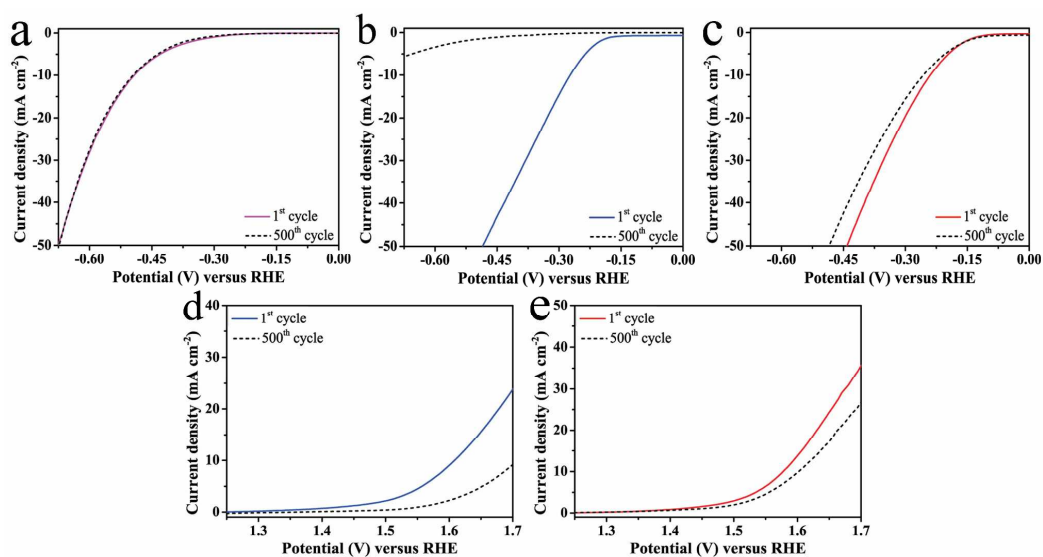


Figure S11. Stability tests of MoS₂ (a), Co₉S₈ (b) and CoS_x@MoS₂ (c) for HER. Stability tests of Co₉S₈ (d) and CoS_x@MoS₂ (e) for OER.

Table S1. Comparison of HER activities of CoS_x@MoS₂ with other reported electrocatalysts.

Catalyst	η_{10} (mV)	Tafel slope (mV dec ⁻¹)	Electrolyte	Ref.
Co ₉ S ₈ @MoS ₂ /CNFs	190	110	0.5 M H ₂ SO ₄	Adv. Mater. 2015, 27, 4752-4759
MoO ₃ -MoS ₂ /FTO	310	50-60	0.5 M H ₂ SO ₄	Nano lett. 2011, 11, 4168-4175
Hollow Co ₃ S ₄	400	97	0.5 M H ₂ SO ₄	Chem. Mater. 2017, 29, 5566–5573
MoS ₂ microboxes	475	134	0.5 M H ₂ SO ₄	Energy Environ. Sci. 2014, 7, 3302-3306
MoS ₂	276	96	0.5 M H ₂ SO ₄	J. Mater. Chem. A, 2017, 5, 22654–22661
CoS _x @MoS ₂	239	103	0.5 M H ₂ SO ₄	This work

Table S2. Comparison of OER activities of $\text{CoS}_x\text{@MoS}_2$ with other reported electrocatalysts.

Catalyst	η_{10} (mV)	Tafel slope (mV dec ⁻¹)	Electrolyte	Ref.
$\text{Co}_9\text{S}_8\text{@MoS}_2/\text{CNFs}$	430	61	1 M KOH	Adv. Mater. 2015, 27, 4752-4759
$\text{CoS-Co(OH)}_2\text{@}\alpha\text{MoS}_2\text{+}$ x/NF	380	68	1 M KOH	Adv. Funct. Mater. 2016, 26, 7386
$\text{PNG-NiCo}_2\text{O}_4$	349	156	0.1 M KOH	ACS Nano 2013, 7, 10190-10196
Co_3S_4 nanosheets	363	90	0.1 M KOH	ACS Nano 2014, 8, 10909-10919
CoNC/CNF	430	89	1 M KOH	J. Mater. Chem. A, 2017, 5, 23898–23908
$\text{CoS}_x\text{@MoS}_2$	347	147	1 M KOH	This work