## **Supporting Information**

## Nanocellulose Structured Paper-based Lithium Metal Batteries

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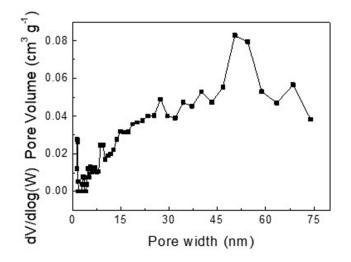
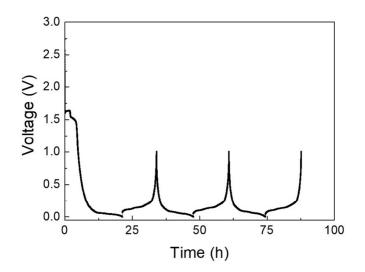


Figure S1. The pore size distribution for the CCP electrode.



**Figure S2.** The cycling curves corresponding to the three-cycle pretreatment of the Li-CCP cell. This process involved a cycling of the cell at a current density of  $0.125 \text{ mA cm}^{-2}$  in a voltage window between 0.01 and 1.0 V (vs. Li<sup>+</sup>/Li).

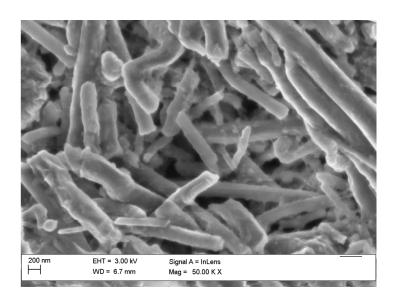
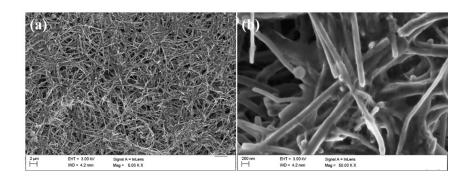
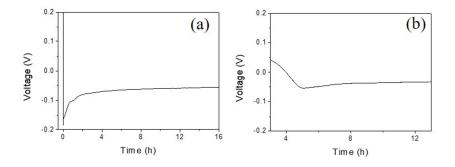


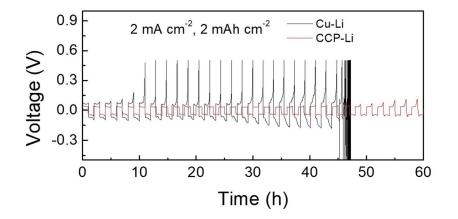
Figure S3. High resolution cross-section SEM image of the CCP electrode obtained after deposition of 8 mAh  $cm^{-2}$  Li.



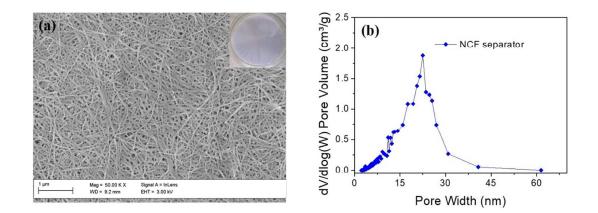
**Figure S4.** SEM images of the CCP electrode obtained after oxidizing the previously deposited Li coating, (a) low magnification image, (b), high magnification image.



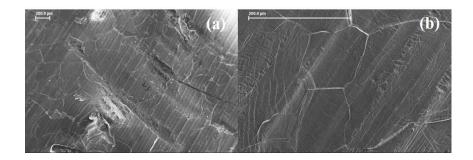
**Figure S5.** The potential as a function of the time during Li deposition on a (a) Cu and (b) CCP electrode versus the Li counter/reference electrode using a current density of 0.5 mA cm<sup>-2</sup>. The shape of the chronopotentiogram for the Cu electrode indicates the presence of a nucleation overpotential whereas the positive potential seen initially for the CCP electrode may be explained by underpotential deposition as explained in the main text.



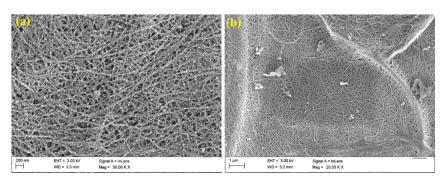
**Figure S6.** The cell voltage as a function of time for repeated Li metal deposition and oxidation using a fixed charge of 2 mAh cm<sup>-2</sup> and a current density of 2 mA cm<sup>-2</sup> for the Li-Cu|Li and Li-CCP|Li cells, respectively. For the Li-CCP or Li-Cu electrode, lithium metal was first pre-plated using a charge of 6 mAh cm<sup>-2</sup> at a current density of 0.5 mA cm<sup>-2</sup>.



**Figure S7.** (a) SEM image of the NCF separator with a photo of the NCF separator as an inset. (b) The pore size distribution for the NCF separator.



**Figure S8.** SEM images of the pristine Li metal foil. The image in a) was recorded with a lower degree of magnification than that used in the b) image.



**Figure S9.** SEM images ((a) high magnification and (b) low magnification) depicting a NCF separator recovered from a NCF-based Li|Li cell after 300 cycles at 0.5 mA cm<sup>-2</sup>, 1 mAh cm<sup>-2</sup> in a LP 40 electrolyte.

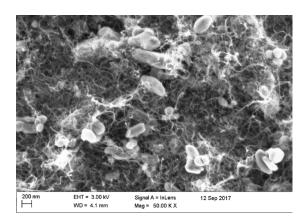
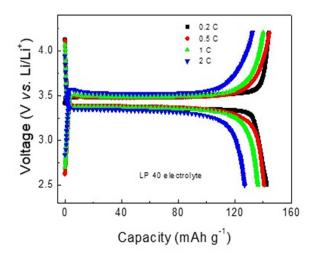
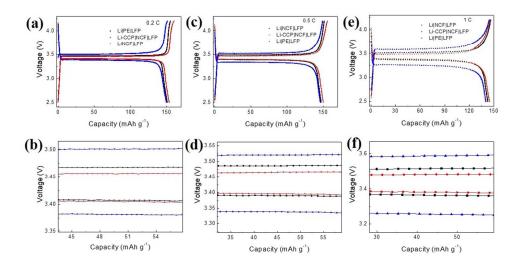


Figure S10. High resolution SEM image of LFP paper cathode.



**Figure S11.** Galvanostatic voltage profiles recorded for different cycling rates using a Li/LFP cell containing a PE separator and a LP 40 electrolyte.



**Figure S12.** Galvanostatic voltage profiles for different full cells obtained at a rate of (a, b) 0.2 C, (c, d) 0.5 C and (e, f) 1 C, respectively. The differences between the charge and discharge voltages are more clearly seen in the (b, d and f) figures which show magnifications of the plateau regions for the different full cells.