

Supporting Information

Isobutane dehydrogenation over InPtSn/ZnAl₂O₄ catalysts: Effect of indium promoter

Jianfeng Liu,^{†,‡} Wei Zhou,[†] Dongyu Jiang,^{*,‡} Wenhui Wu,[‡] Changxi Miao,[‡] Yue Wang,[†] and Xinbin Ma^{*,†}

[†]Key Laboratory for Green Chemical Technology of Ministry of Education,
Collaborative Innovation Centre of Chemical Science and Engineering, School of
Chemical Engineering and Technology, Tianjin University, Tianjin 300072, China

[‡]State Key Laboratory of Green Chemical Engineering and Industrial Catalysis,
Sinopec Shanghai Research Institute of Petrochemical Technology, Shanghai 201208,
China

^{*}Corresponding author: Prof. Xinbin Ma, Dr. Dongyu Jiang

E-mail: xbma@tju.edu.cn; jiangdy.sshy@sinopec.com

Fax: +86-22-87401818

Tel: +86-22-27409248

Contents

Tables

Table S1 Textural properties of xInPtSn/ ZnAl₂O₄ catalysts.

Samples	S _{BET} (m ² /g)	Vp(cm ³ /g)	D _{pore} (nm)
PtSn/ZA	84.9	0.14	6.9
0.2InPtSn/ZA	85.1	0.14	6.8
0.4InPtSn/ZA	84.7	0.14	6.8
0.6InPtSn/ZA	85.1	0.15	7.1
0.8InPtSn/ZA	85.0	0.15	7.0
1.0InPtSn/ZA	85.1	0.15	7.1

Table S2 Summary of XPS results xInPtSn/ ZnAl₂O₄ catalysts.

Samples	Binding energy Sn3d _{5/2} (eV)
PtSn/ZA	485.4 (29%); 486.2 (44%); 487.1 (27%)
0.4InPtSn/ZA	485.5 (18%); 486.5 (50%); 487.5 (32%)
0.6InPtSn/ZA	485.5 (23%); 486.4 (44%); 487.4 (33%)
1.0InPtSn/ZA	485.7 (28%); 486.5 (40%); 487.4 (32%)

Table S3 Amount of coke on the xInPtSn/ ZnAl₂O₄ catalysts after isobutane dehydrogenation

Samples	Coke amount (%)
PtSn/ZA	1.48
0.4InPtSn/ZA	1.21
1.0InPtSn/ZA	2.68

Figures

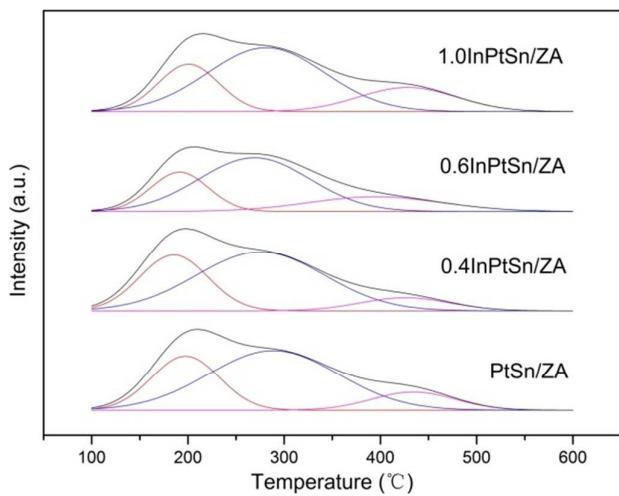


Figure S1 NH₃-TPD profiles of xInPtSn/ZnAl₂O₄ catalysts.