

Supporting Information:

Single Molybdenum Atom Anchored on N-Doped Carbon as a Promising Electrocatalyst for Nitrogen Reduction into Ammonia at Ambient Conditions

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Figures

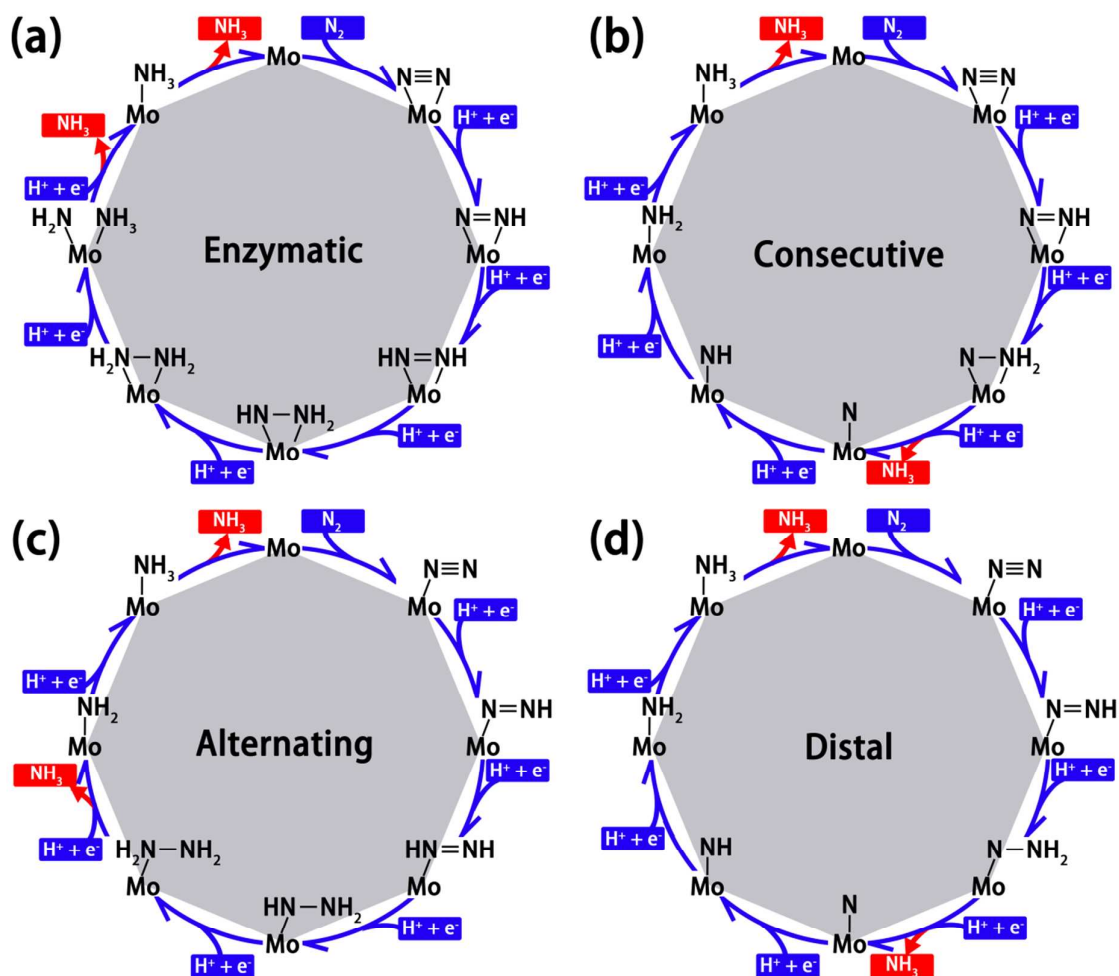


Figure S1. Schematic depiction of (a) enzymatic, (b) consecutive, (c) alternating and (d) distal mechanisms for N_2 reduction to NH_3 on $\text{Mo}_1\text{-N}_1\text{C}_2$.

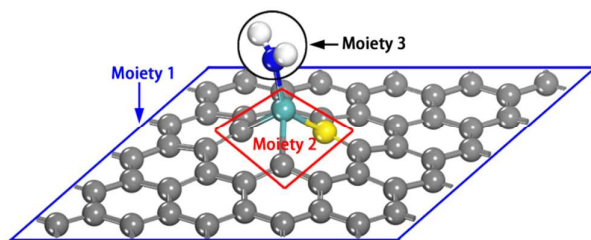


Figure S2. Definition of three moieties of N_xH_y adsorbed $\text{Mo}_1\text{-N}_1\text{C}_2$ by using NH_2 as prototype.

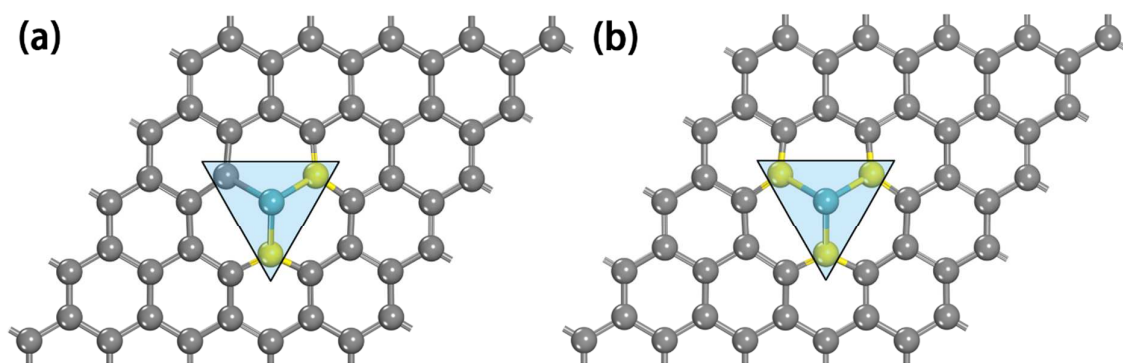


Figure S3. Structures of (a) Mo-N₂C₁ and (b) Mo-N₃ active centers. Gray, cyan and yellow balls represent the C, Mo and N atoms, respectively.

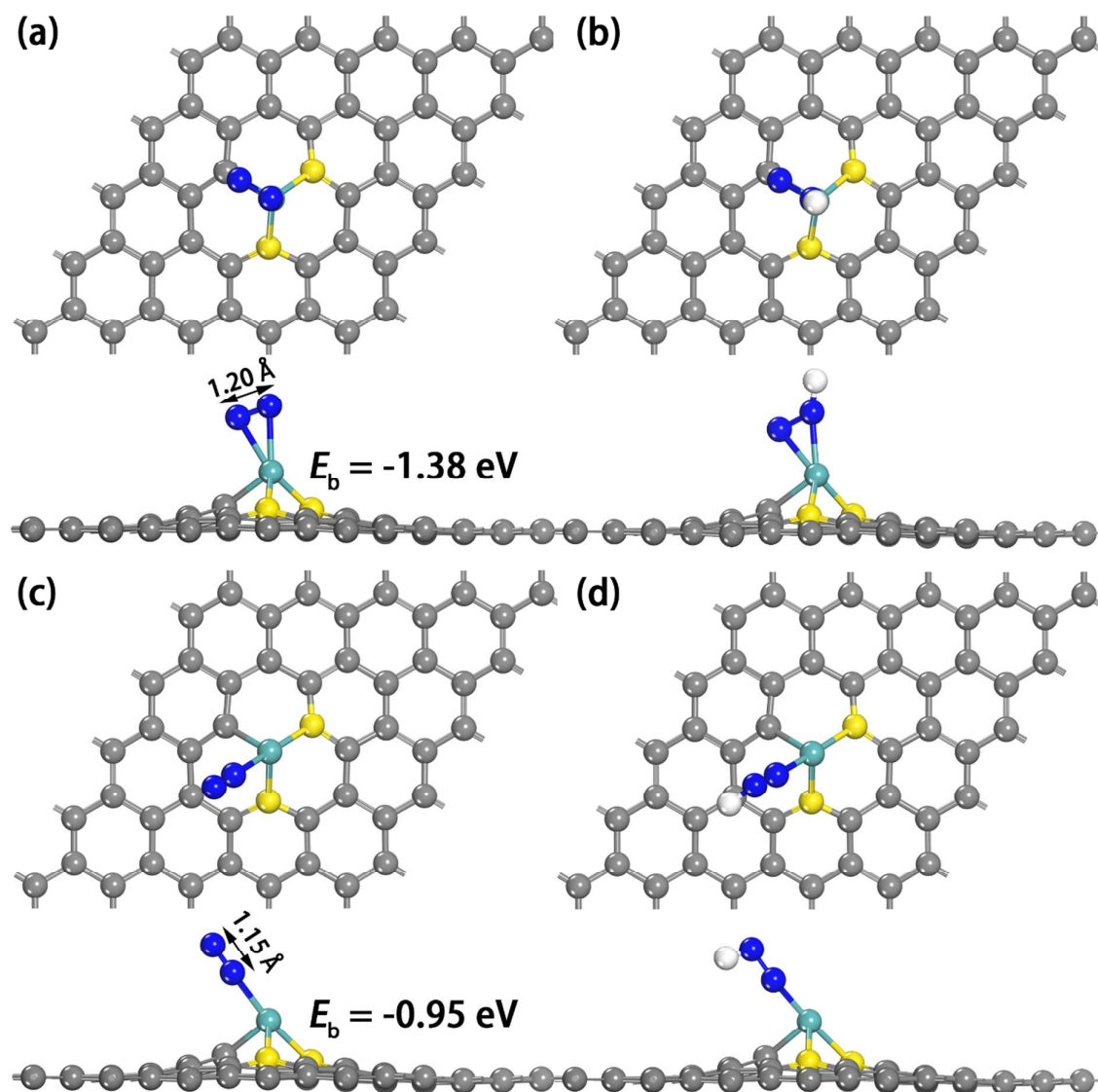


Figure S4. Top and side views of the adsorption of N₂ and N₂H via the (a, b) side-on and (c, d)

end-on patterns on the Mo-N₂C₁ site. Gray, cyan, yellow and blue balls represent the C, Mo, doped N and adsorbed N atoms, respectively.

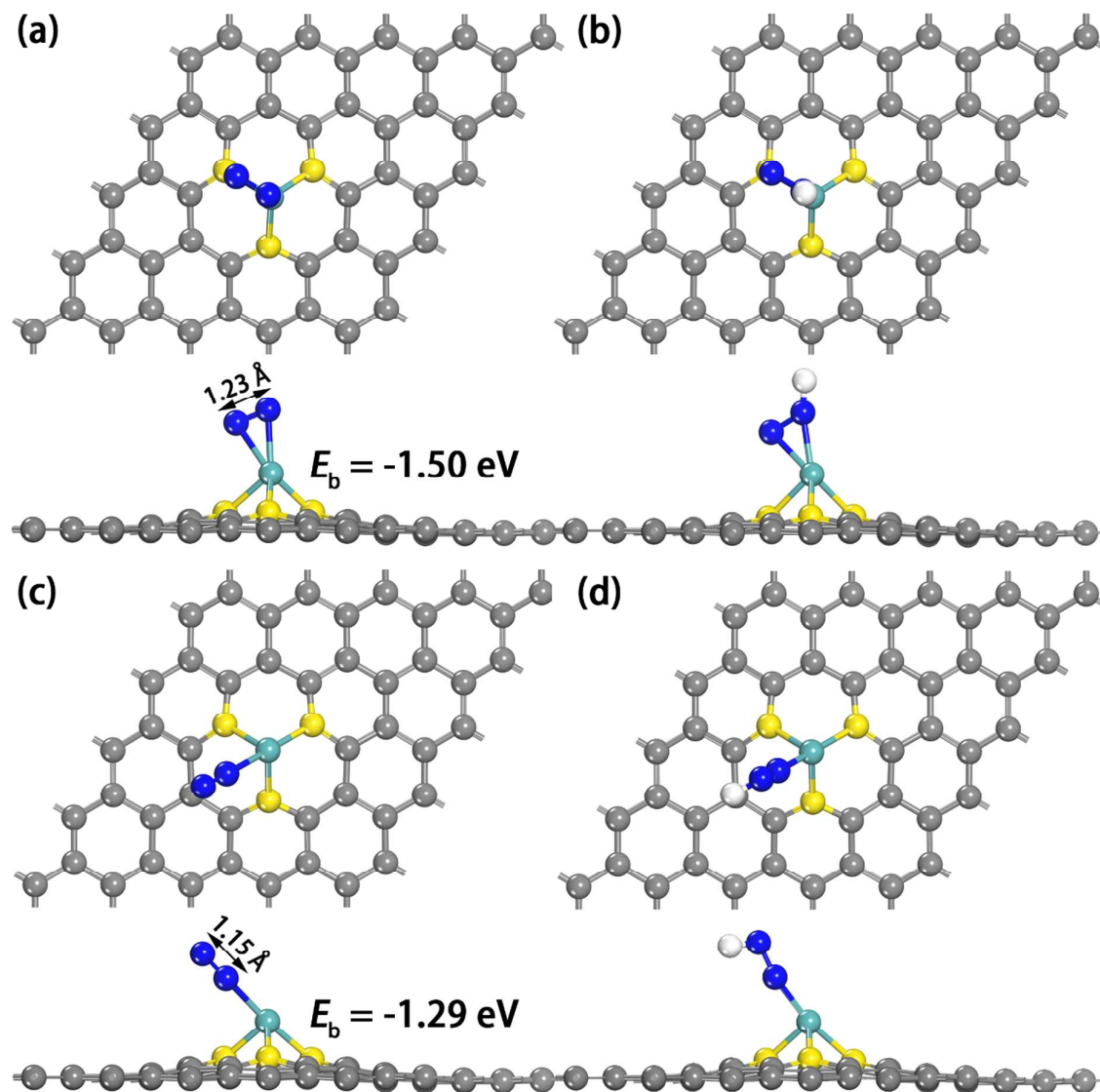


Figure S5. Top and side views of the adsorption N₂ and N₂H via the (a, b) side-on and (c, d) end-on patterns on the Mo-N₃. Gray, cyan, yellow and blue balls represent the C, Mo, doped N and adsorbed N atoms, respectively.

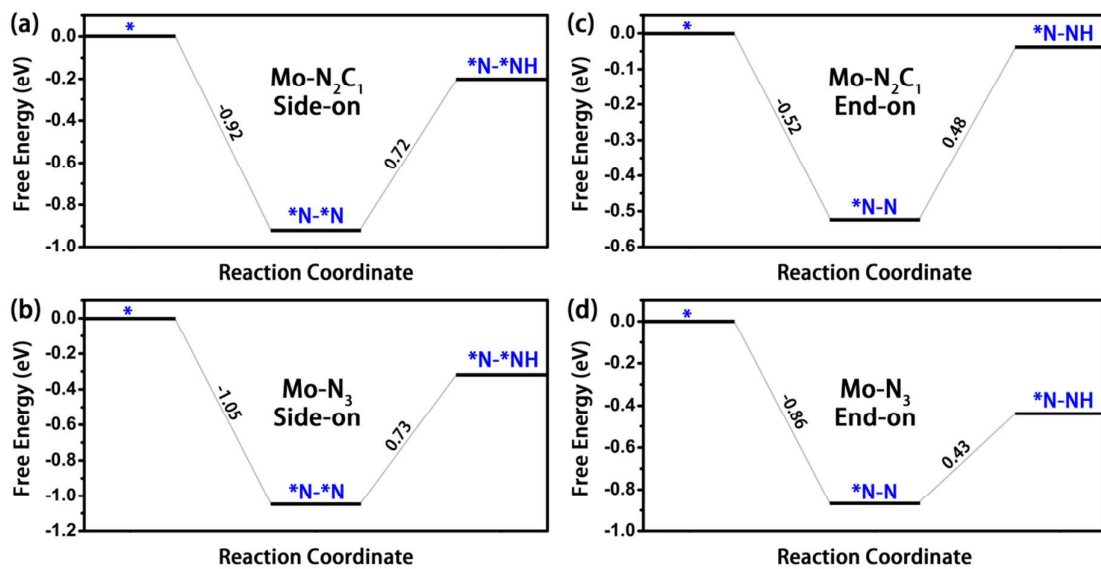


Figure S6. Calculated free energy diagrams for N_2 adsorption and protonation of side-on adsorbed N_2 on (a) Mo- N_2C_1 and (b) Mo- N_3 sites as well as of end-on adsorbed N_2 on (c) Mo- N_2C_1 and (d) Mo- N_3 sites Mo₁/N-C.

Tables

Table S1. Calculated zero point energies and entropy of different adsorption species, where the * denotes the adsorption site. Therefore, $*N\equiv N$ and $*N\equiv N$ represent the side-on and end-on adsorption configurations, respectively.

Adsorption Species	E_{ZPE} (eV)	TS (eV)
N_2	0.15	0.58
$*N\equiv N$	0.19	0.15
$*N=NH$	0.49	0.13
$*NH=NH$	0.79	0.16
$*NH-*NH_2$	1.14	0.17
$*NH_2-*NH_2$	1.34	0.22
$*NH_2-*NH_3$	1.66	0.30
$*N-*NH_2$	0.83	0.15
$*N$	0.08	0.06
$*NH$	0.35	0.09
$*NH_2$	0.65	0.14
$*NH_3$	1.02	0.16
$*N\equiv N$	0.20	0.18
$*N=NH$	0.49	0.16
$*NH=NH$	0.80	0.24
$*NH-NH_2$	1.13	0.19
$*NH_2-NH_2$	1.49	0.25
$*N-NH_2$	0.82	0.18
NH_3	0.58	0.56

Table S2. Calculated adsorption energies of N_2 on various SACs that have been synthesized. For $Cu_1/N-C$, $Pd_1/N-C$ and $Pt_1/N-C$, only the end-on configurations can be gained.

SACs	Adsorption Energies of N_2 (eV)	
	Side-on	End-on
$Mo_1/N-C$	-1.19	-1.18
$Cu_1/N-C$		-0.58
$Pd_1/N-C$		-0.58
$Pt_1/N-C$		-0.40