

Crystal Chemistry of Vanadium-Bearing Ellestadite Waste Forms

Yanan Fang,^{,1} Sam J. Page,² Gregory J. Rees,² Maxim Avdeev,³ John V. Hanna,² Tim J. White¹*

¹ School of Materials Science & Engineering, Nanyang Technological University, 50 Nanyang Avenue 639798, Singapore

² Department of Physics, University of Warwick, Gibbet Hill Road, Coventry CV4 7AL, United Kingdom

³ Australian Nuclear Science and Technology Organisation, Locked Bag 2001, Kirrawee DC New South Wales 2232, Australia

Table S1. Lattice parameters and crystallographic data for hexagonal $\text{Ca}_{10}(\text{SiO}_4)_x(\text{VO}_4)_{6-2x}(\text{SO}_4)_x\text{Cl}_2$, $0.19 \leq x \leq 3$.

Single phase		Miscibility gap						Single phase	
<i>x</i>		0.90 (average)		1.47 (average)		2.14 (average)			
	0.44	0.19	2.44	0.45	2.62	0.77	2.45	2.50	3.0
Space group	$P6_3/m$	$P6_3/m$	$P6_3/m$	$P6_3/m$	$P6_3/m$	$P6_3/m$	$P6_3/m$	$P6_3/m$	$P6_3/m$
Wt	99.4%	68.6%	31.4%	52.8%	47.1%	10.7%	84.3%	100%	95.6%
<i>a</i> (Å)	10.0541(4)	10.1132(3)	9.749(1)	10.0583(7)	9.6967(8)	10.0177(4)	9.7333(6)	9.7245(4)	9.6691(3)
<i>c</i> (Å)	6.8223(3)	6.8004(4)	6.865(1)	6.8256(9)	6.8803(1)	6.7744(4)	6.8750(8)	6.8722(4)	6.8531(3)
<i>V</i> (Å ³)	597.23(6)	602.34(5)	565.10(2)	598.03(1)	560.25(1)	595.00(1)	564.06(1)	562.81(6)	554.88(4))
φ(deg)	15.6 ₈	13.4 ₆	18.5 ₂	15.4 ₈	20.2 ₂	16.0 ₃	20.0 ₉	19.9 ₂	20.4 ₂
<i>R</i> _{wp}	0.076	0.072	0.072	0.078	0.078	0.075	0.075	0.079	0.052
<i>R</i> _p	0.059	0.056	0.056	0.061	0.061	0.058	0.058	0.062	0.040
<i>R</i> _b	0.023	0.026	0.014	0.024	0.016	0.014	0.023	0.028	0.032
Ca(1), 4<i>f</i> (1/3, 2/3, <i>z</i>)									
<i>z</i>	0.0039(9)	0.001(1)	-0.009(3)	0.0022(2)	0.0023(2)	0.0073(1)	0.0002(1)	0.006(1)	0.0022(1)
<i>B</i> (Å ²)	1.85(7)	1.06(2)	0.50(2)	0.43(3)	0.91(2)	0.87(7)	0.82(7)	0.51(7)	2.04(5)
Ca(2), 6<i>h</i> (<i>x</i>, <i>y</i>, 1/4)									
<i>x</i>	0.2671(5)	0.2688(9)	0.251(2)	0.2721(1)	0.2499(1)	0.2687(1)	0.2549(1)	0.2595(7)	0.2600(7)
<i>y</i>	0.0223(6)	0.0201(9)	-0.016(2)	0.0257(1)	0.0122(2)	0.0200(3)	0.0054(1)	0.0033(9)	0.0004(8)
<i>B</i> (Å ²)	1.85(7)	1.06(2)	0.50(2)	0.43(3)	0.91(2)	0.87(7)	0.82(7)	0.51(7)	2.04(5)
Si/S, 6<i>h</i> (<i>x</i>, <i>y</i>, 1/4)									
<i>x</i>	0.4057(2)	0.4060(2)	0.4070(2)	0.4055(2)	0.4072(2)	0.4046(1)	0.4070(1)	0.407(1)	0.4030(4)
<i>y</i>	0.3755(4)	0.3756(3)	0.3744(1)	0.3760(2)	0.3746(4)	0.3733(1)	0.3746(8)	0.374(1)	0.3728(4)
<i>B</i> (Å ²)	0.80(4)	1.39(2)	0.86(2)	1.39(2)	1.38(4)	0.87(7)	1.16(3)	0.82(6)	2.04(5)
Occ.	Si	0.073	0.032	0.408	0.075	0.437	0.128	0.408	0.42
	S	0.073	0.032	0.408	0.075	0.437	0.128	0.408	0.42
	V	0.853	0.936	0.184	0.85	0.126	0.744	0.184	0.16
O(1), 6<i>h</i> (<i>x</i>, <i>y</i>, 1/4)									
<i>x</i>	0.3491(2)	0.3543(4)	0.3436(7)	0.3499(6)	0.3299(6)	0.3438(4)	0.3338(5)	0.3368(3)	0.3361(3)
<i>y</i>	0.5042(2)	0.5090(4)	0.4956(8)	0.5060(6)	0.4881(6)	0.4960(3)	0.4883(5)	0.4885(3)	0.4880(3)
<i>B</i> (Å ²)	0.80(4)	1.39(2)	0.86(2)	1.39(2)	1.38(4)	0.87(7)	1.16(3)	0.82(6)	2.20(3)
O(2), 6<i>h</i> (<i>x</i>, <i>y</i>, 1/4)									
<i>x</i>	0.5986(2)	0.6008(4)	0.5935(7)	0.5998(6)	0.5958(6)	0.6005(8)	0.5960(5)	0.5929(3)	0.5886(3)
<i>y</i>	0.4648(4)	0.4662(7)	0.4659(1)	0.4659(1)	0.4761(1)	0.4701(3)	0.4674(8)	0.4696(3)	0.4672(5)
<i>B</i> (Å ²)	0.80(4)	1.39(2)	0.86(2)	1.39(2)	1.38(4)	0.87(7)	1.16(3)	0.82(6)	2.20(3)
O(3), 12<i>i</i> (<i>x</i>, <i>y</i>, <i>z</i>)									
<i>x</i>	0.3521(3)	0.3530(4)	0.3493(1)	0.3508(8)	0.3507(8)	0.3521(3)	0.3481(6)	0.3487(4)	0.3538(3)
<i>y</i>	0.2611(2)	0.2618(3)	0.2750(1)	0.2587(5)	0.2675(5)	0.2618(2)	0.2650(5)	0.2659(3)	0.2665(3)
<i>z</i>	0.0541(2)	0.0527(4)	0.0572(5)	0.0563(6)	0.0627(6)	0.0053(2)	0.0655(5)	0.0661(3)	0.0657(3)
<i>B</i> (Å ²)	0.80(4)	1.39(2)	0.86(2)	1.39(2)	1.38(4)	0.87(7)	1.16(3)	0.82(6)	2.20(3)

Cl *									
x	0.017(7)	0.011(6)	0	0.019(7)	0	0.011(8)	0	0	0
y	-0.020(7)	-0.029(4)	0	-0.030(6)	0	-0.023(9)	0	0	0
z	0.3183(1)	0.3223(1)	0.358(4)	0.287(5)	0.353(2)	0.404(4)	0.327(4)	0.381(2)	0.425(1)
B (Å ²)	1.85(7)	1.06(2)	1.06(2)	0.43(3)	0.91(2)	0.87(7)	0.82(7)	0.51(7)	2.04(5)
Occ	0.144(2)	0.154(3)	0.50	0.137(5)	0.50(4)	0.146(2)	0.50	0.488(9)	0.416(2)

$R_p = (\sum_i |y_i - y_{ci}|) / \sum_i y_i$, $R_{wp} = [(\sum_i w |y_i - y_{ci}|^2) / \sum_i w |y_i|^2]^{1/2}$; *Cl in V-rich phases at the site of 12i (x, y, z), while at 4e (0, 0, z) in (Si/S)-rich phases.

Table S2. Bond lengths for $\text{Ca}_{10}(\text{SiO}_4)_x(\text{VO}_4)_{6-2x}(\text{SO}_4)_x\text{Cl}_2$.

Composition x	0.19	0.44	0.45	0.77	2.44	2.45	2.50	2.62	3.0
Bond length (Å)									
Ca(1)-O(1) $\times 3$	2.407(1)	2.402(2)	2.402(1)	2.411(5)	2.416(1)	2.445(5)	2.421(6)	2.440(4)	2.438(3)
Ca(1)-O(2) $\times 3$	2.462(1)	2.473(3)	2.468(1)	2.502(4)	2.474(6)	2.454(2)	2.516(7)	2.500(8)	2.466(3)
< Ca(1)-O >	2.434	2.437	2.435	2.456	2.445	2.450	2.468	2.47	2.452
Ca(2)-O(1)	3.407(7)	3.318(2)	3.389(3)	3.224(2)	2.827(2)	2.836(8)	2.866(9)	2.746(1)	2.893(4)
Ca(2)-O(2)	2.327(1)	2.345(3)	2.294(1)	2.274(6)	2.387(2)	2.340(8)	2.241(8)	2.301(9)	2.283(4)
Ca(2)-O(3) $\times 2$	2.330(5)	2.354(2)	2.374(9)	2.320(1)	2.223(1)	2.315(4)	2.331(4)	2.289(4)	2.582(3)
Ca(2)-O(3) $\times 2$	2.534(7)	2.495(2)	2.451(1)	2.511(8)	2.837(2)	2.640(9)	2.572(8)	2.706(5)	2.358(3)
< Ca(2)-O >	2.577	2.56	2.555	2.527	2.555	2.514	2.485	2.506	2.509
B-O(1)	1.673(5)	1.653(3)	1.660(7)	1.621(5)	1.581(9)	1.586(9)	1.554(3)	1.609(7)	1.544(5)
B-O(2)	1.707(4)	1.681(2)	1.694(5)	1.699(2)	1.575(6)	1.593(3)	1.571(3)	1.587(5)	1.545(1)
B-O(3)	1.671(2)	1.667(1)	1.671(4)	1.644(3)	1.571(5)	1.569(6)	1.562(2)	1.573(4)	1.545(2)
Average	1.680	1.667	1.674	1.652	1.574	1.579	1.562	1.585	1.545

Table S3. Bond lengths and angles of $\text{Ca}_{10}(\text{VO}_4)_6\text{Cl}_2$.

<Ca-O> (Å)	O(1)	O(2)	O(3)	O(4)	O(5)	O(6)	average
Ca(1)	2.310(1)	2.347(5)	2.320(2)	2.414(4)	2.443(1)	2.502(1)	2.389
Ca(2)	2.448(1)	2.388(5)	2.507(2)	2.523(4)	2.412(1)	2.421(2)	2.450
<V-O> (Å)							
V(1)	O(1)	O(4)	O(7)	O(10)			
	1.691(1)	1.711(3)	1.692(3)	1.684(6)			1.694
V(2)	O(2)	O(5)	O(8)	O(11)			
	1.695(3)	1.682(2)	1.711(1)	1.709(8)			1.699
V(3)	O(3)	O(6)	O(9)	O(12)			
	1.703(2)	1.720(2)	1.700(3)	1.695(1)			1.704
<O-V-O> (°)							
V(1)	O(1)-V(1)-O(4)	O(1)-V(1)-O(7)	O(1)-V(1)-O(10)	O(4)-V(1)-O(7)	O(4)-V(1)-O(10)	O(7)-V(1)-O(10)	
	106.97(1)	110.88(1)	111.83(2)	107.58(2)	115.83(3)	103.65(3)	109.45
V(2)	O(2)-V(2)-O(5)	O(2)-V(2)-O(8)	O(2)-V(2)-O(11)	O(5)-V(2)-O(8)	O(5)-V(2)-O(11)	O(8)-V(2)-O(11)	
	108.18(7)	116.33(4)	119.21(2)	101.67(5)	105.30(6)	104.31(3)	109.16
V(3)	O(1)-V(1)-O(4)	O(1)-V(1)-O(7)	O(1)-V(1)-O(10)	O(4)-V(1)-O(7)	O(4)-V(1)-O(10)	O(7)-V(1)-O(10)	
	108.91(5)	113.28(3)	118.59(3)	108.21(8)	101.01(2)	105.80(4)	109.30

Table S4. Variation in the metaprism twist angle of calcium apatite as a function of average crystal radius.

Composition	Average crystal radius (Å)	Twist angle ϕ (°)	Reference
$\text{Ca}_{10}(\text{PO}_4)_6\text{F}_2$	1.143	23.3	⁵¹
$\text{Ca}_{10}(\text{PO}_4)_6(\text{OH})_2$	1.146	23.2	⁵²
$\text{Ca}_{9.82}(\text{SiO}_4)_3(\text{SO}_4)_3\text{Cl}_{1.64}$	1.158	20.4	this study
$\text{Ca}_{10}(\text{PO}_4)_6\text{Cl}_2$	1.166	19.1	⁵²
$\text{Ca}_{10}(\text{CrO}_4)_6(\text{OH})_2$	1.171	17.8	⁵³
$\text{Ca}_{10}(\text{PO}_4)_6\text{Br}_2$	1.173	16.3	⁵²
$\text{Ca}_{10}(\text{AsO}_4)_6\text{Cl}_2$	1.189	13.0	⁵⁴
$\text{Ca}_{10}(\text{VO}_4)_6\text{Cl}_2$	1.192	11.6*	this study
$\text{Ca}_4\text{Pb}_6(\text{AsO}_4)_6\text{Cl}_2$	1.214	5.2	⁵⁵

* The twist angle of $\text{Ca}_{10}(\text{VO}_4)_6\text{Cl}_2$ is the average value of three twist angles in a triclinic $P-1$ structure.

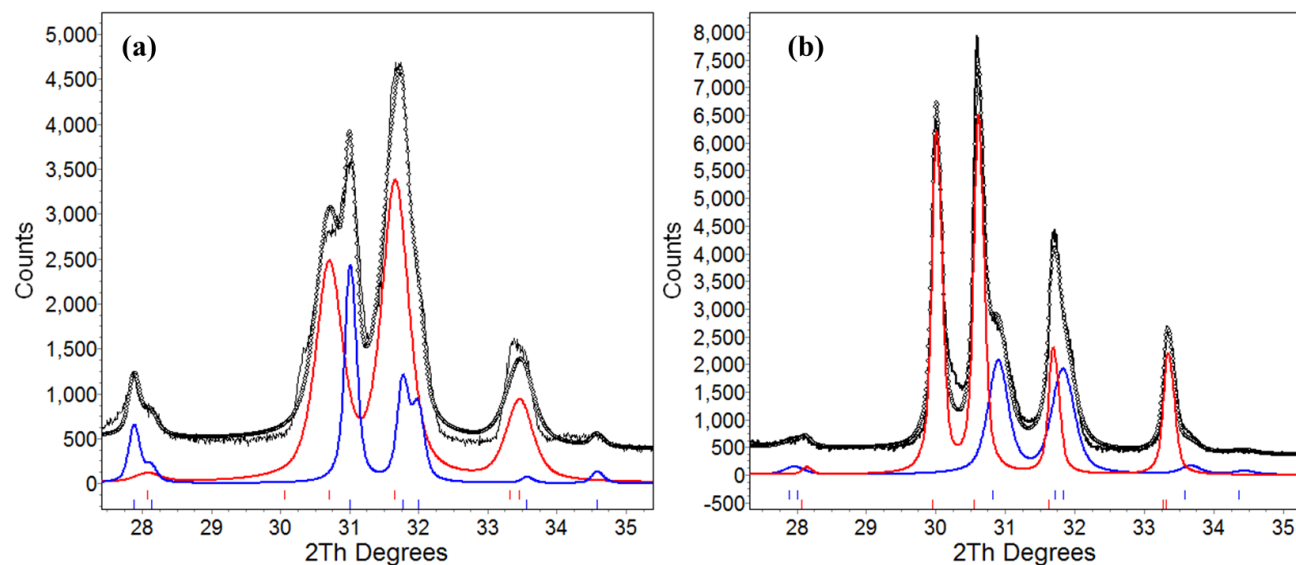


Figure S1. Rietveld fit between calculated (black dot) and experimental (black line) diffraction profiles for $\text{Ca}_{10}(\text{SiO}_4)_x(\text{VO}_4)_{6-2x}(\text{SO}_4)_x\text{Cl}_2$, with (a) nominal $x = 1.5$ and (b) $x = 1.0$. At these compositions immiscibility is reflected by co-existing ellestadite phase pairs. The contributions of V-rich phase (red line) and (Si/S)-rich phase (blue line) are shown.

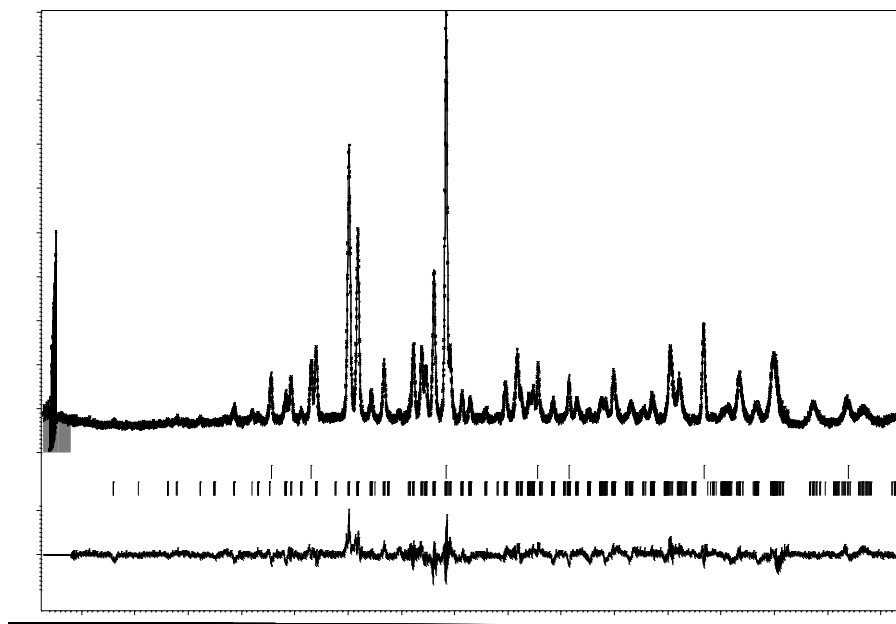


Figure S2. Neutron diffraction Rietveld profiles of $\text{Ca}_{10}(\text{VO}_4)_6\text{Cl}_2$. The calculated weight percentage of internal standard CaF_2 and $\text{Ca}_{10}(\text{VO}_4)_6\text{Cl}_2$ are 20.09(1)% and 79.91(1)% respectively. The differences between observed and calculated profile is shown.

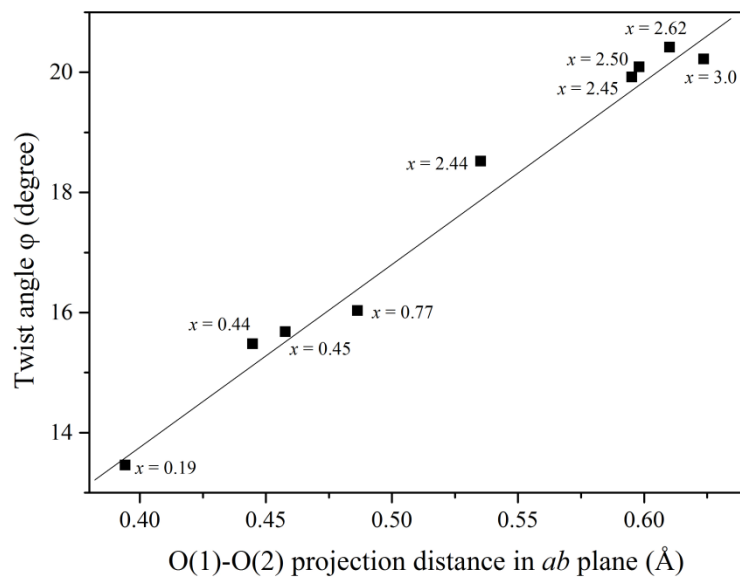


Figure S3. The relationship between metaprism twist angle ϕ of $\text{Ca}_{10}(\text{SiO}_4)_x(\text{VO}_4)_{6-2x}(\text{SO}_4)_x\text{Cl}_2$ and the O(1)-O(2) projection distance in ab plane.

Calulation of metaprism twist angle (ϕ)

Metaprism twist angle is calculated by

$$\phi = \arccos\left(\frac{d_1^2 + d_2^2 - d_3^2}{2 \times d_1 \times d_2}\right)$$

where d_1 , d_2 and d_3 is the distance between the projections of Ca(1) and O(1), Ca(1) and O(2), and O(1) and O(2) in ab plane (Fig. S2). They are calculated from the cell parameters and fractional coordinates of Ca(1), O(1) and O(2) with the expression

$$d_1^2 = (a x_{O1} \cos 30^\circ - a x_{Ca1} \cos 30^\circ)^2 + [(b y_{O1} - a x_{O1} \sin 30^\circ) - (b y_{Ca1} - a x_{Ca1} \sin 30^\circ)]^2,$$

$$d_2^2 = (a x_{O2} \cos 30^\circ - a x_{Ca1} \cos 30^\circ)^2 + [(b y_{O2} - a x_{O2} \sin 30^\circ) - (b y_{Ca1} - a x_{Ca1} \sin 30^\circ)]^2,$$

$$d_3^2 = (a x_{O1} \cos 30^\circ - a x_{O2} \cos 30^\circ)^2 + [(b y_{O1} - a x_{O1} \sin 30^\circ) - (b y_{O2} - a x_{O2} \sin 30^\circ)]^2.$$

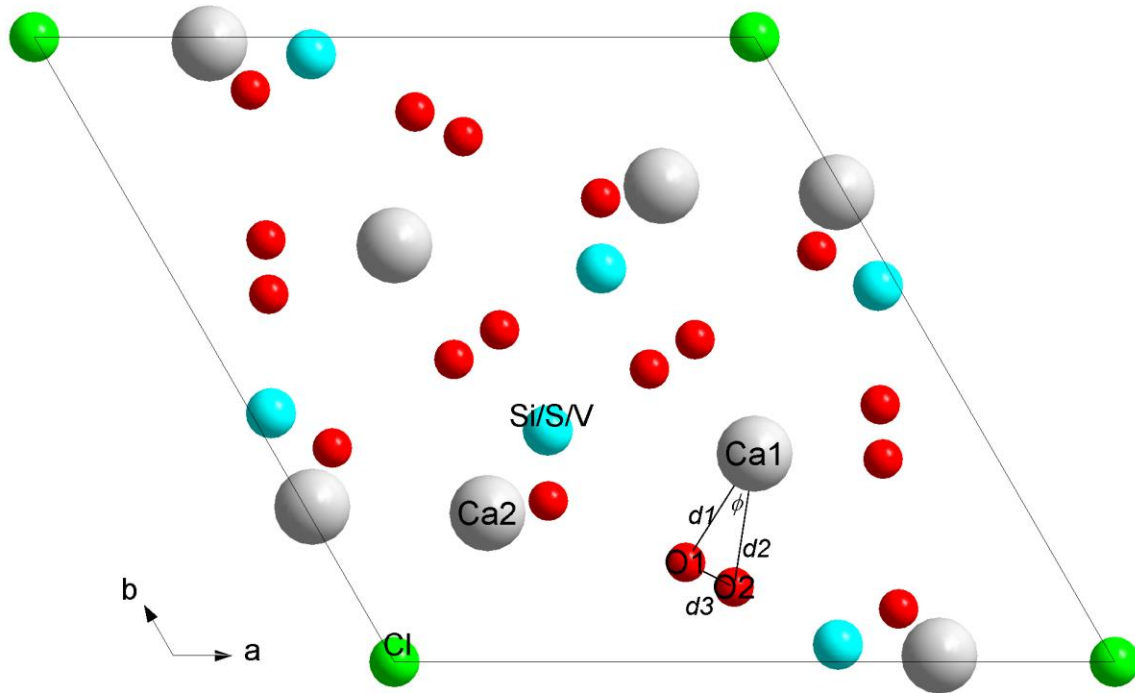


Figure S4. The projection of ellestadite unit cell in ab plane.