

Supporting Information
for

**Alkoxy-Substituted Anthra[1,2-*c*:5,6-*c'*]bis([1,2,5]thiadiazole) (ATz):
A New Electron-Acceptor Unit in the Semiconducting Polymers for
Organic Electronics**

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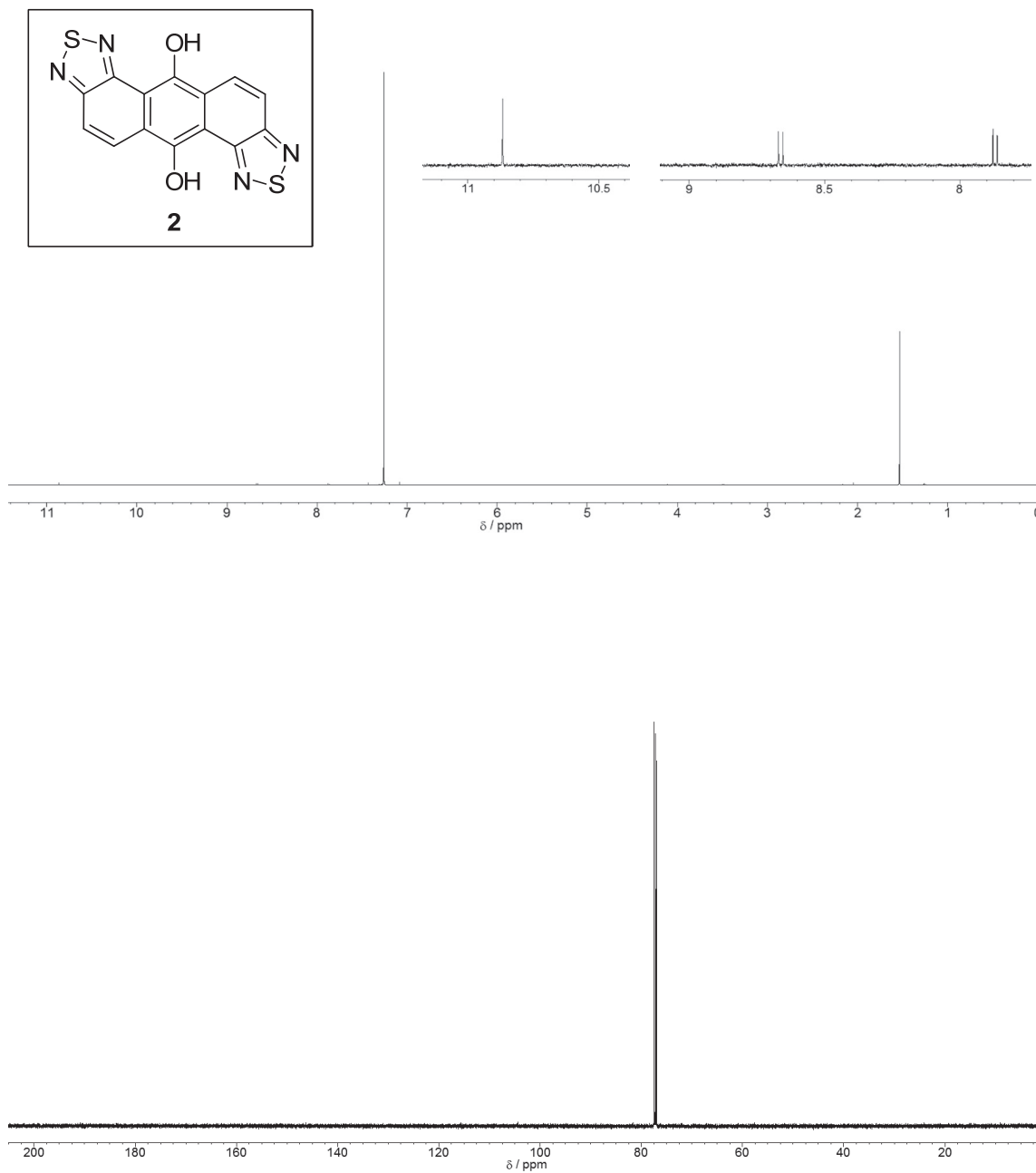
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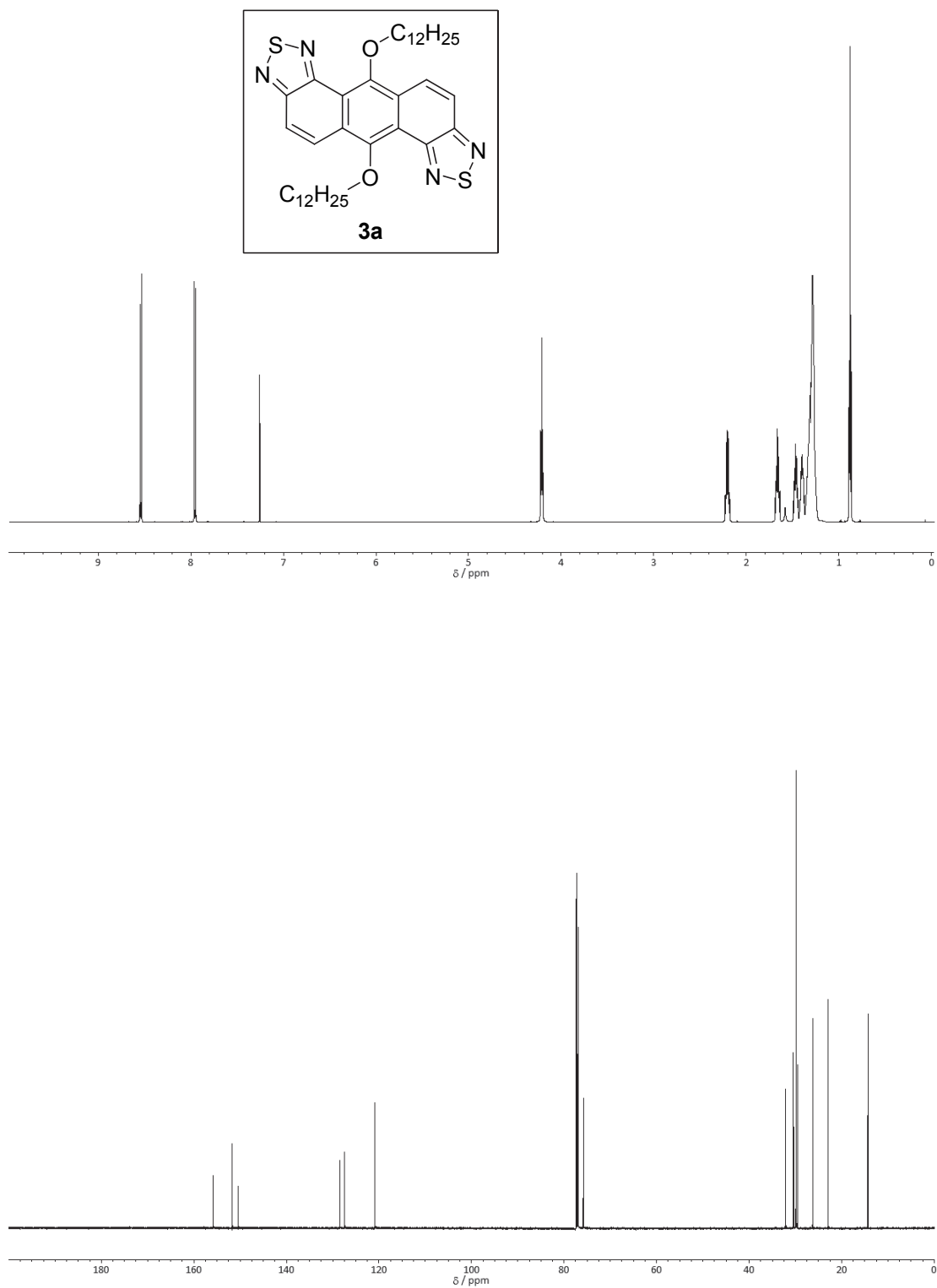
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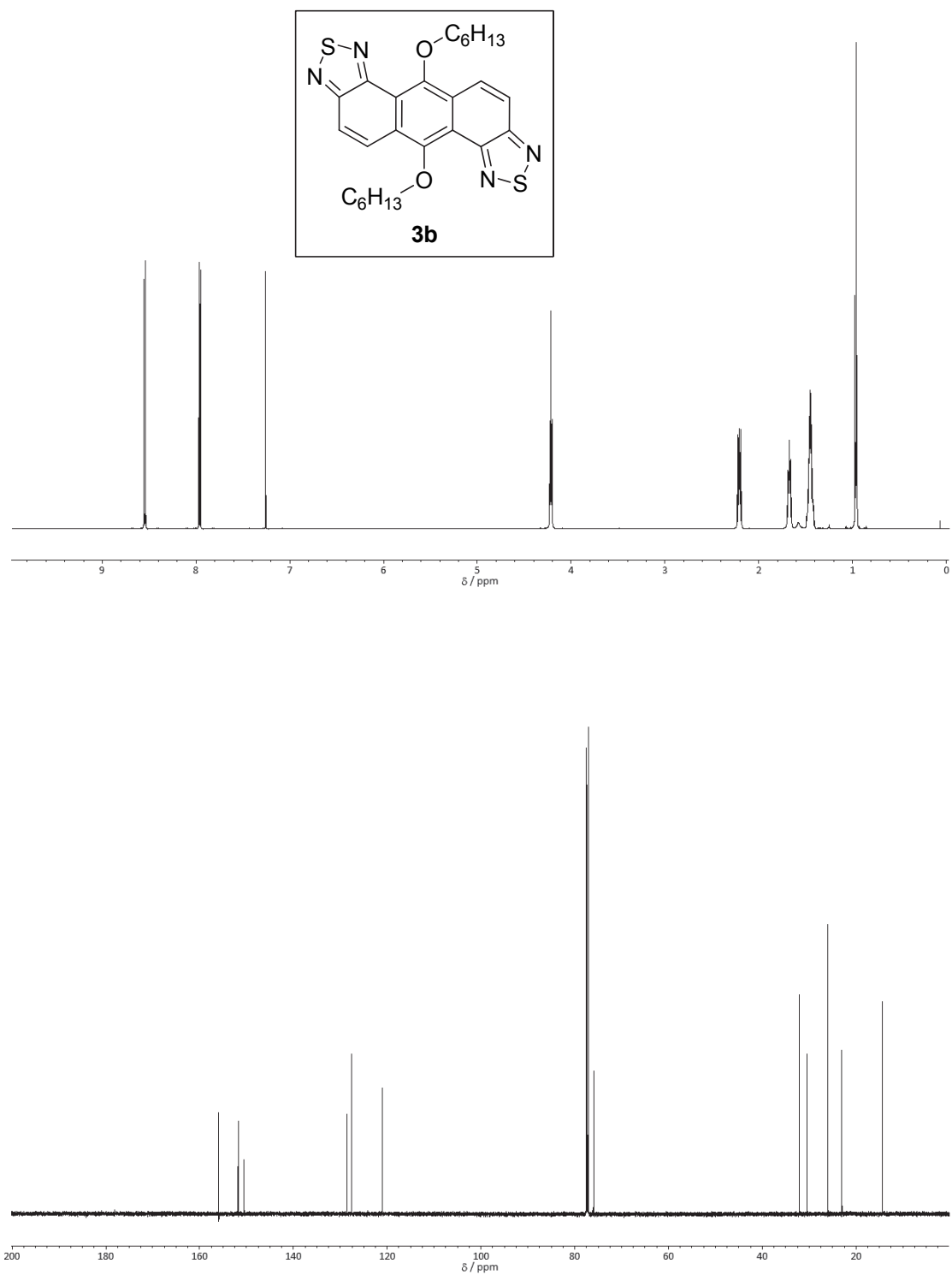
1. Copies of ^1H and $^{13}\text{C}\{^1\text{H}\}$ NMR Charts and NOESY Spectra for the New Compounds



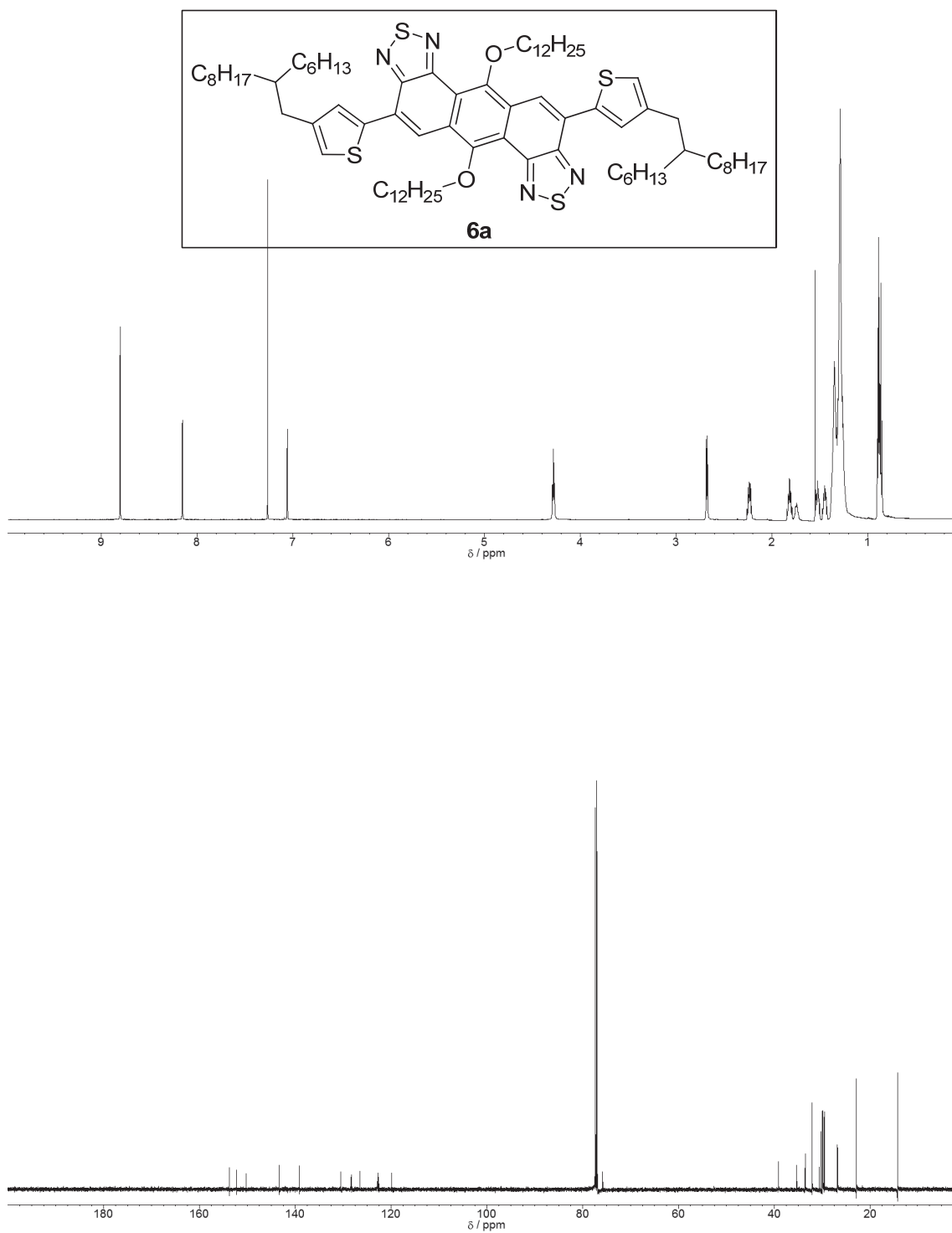
The ^1H and $^{13}\text{C}\{^1\text{H}\}$ NMR spectra of **2** (in CDCl_3).



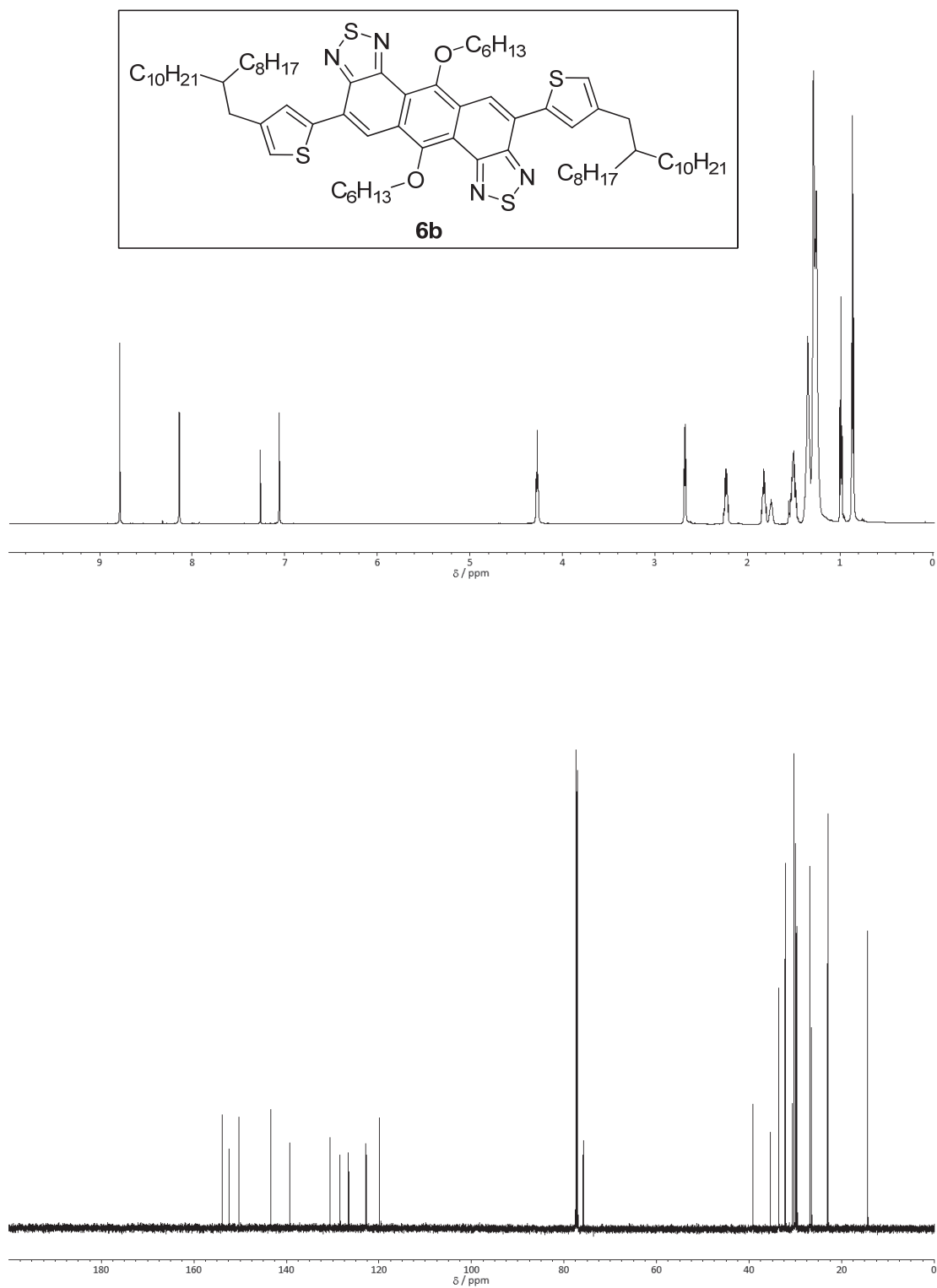
The ^1H and $^{13}\text{C}\{^1\text{H}\}$ NMR spectra of **3a** (in CDCl_3).



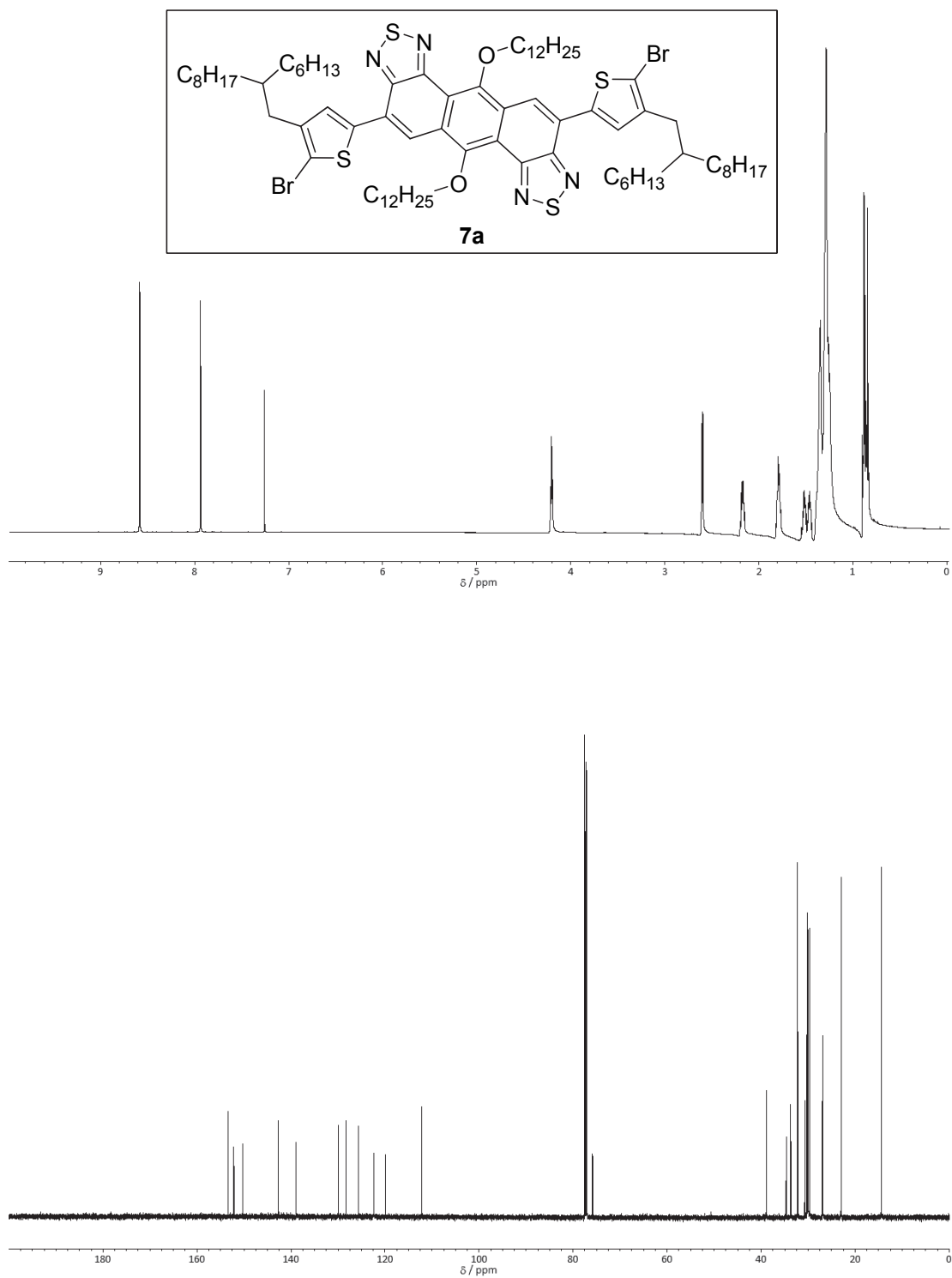
The ^1H and $^{13}\text{C}\{^1\text{H}\}$ NMR spectra of **3b** (in CDCl_3).



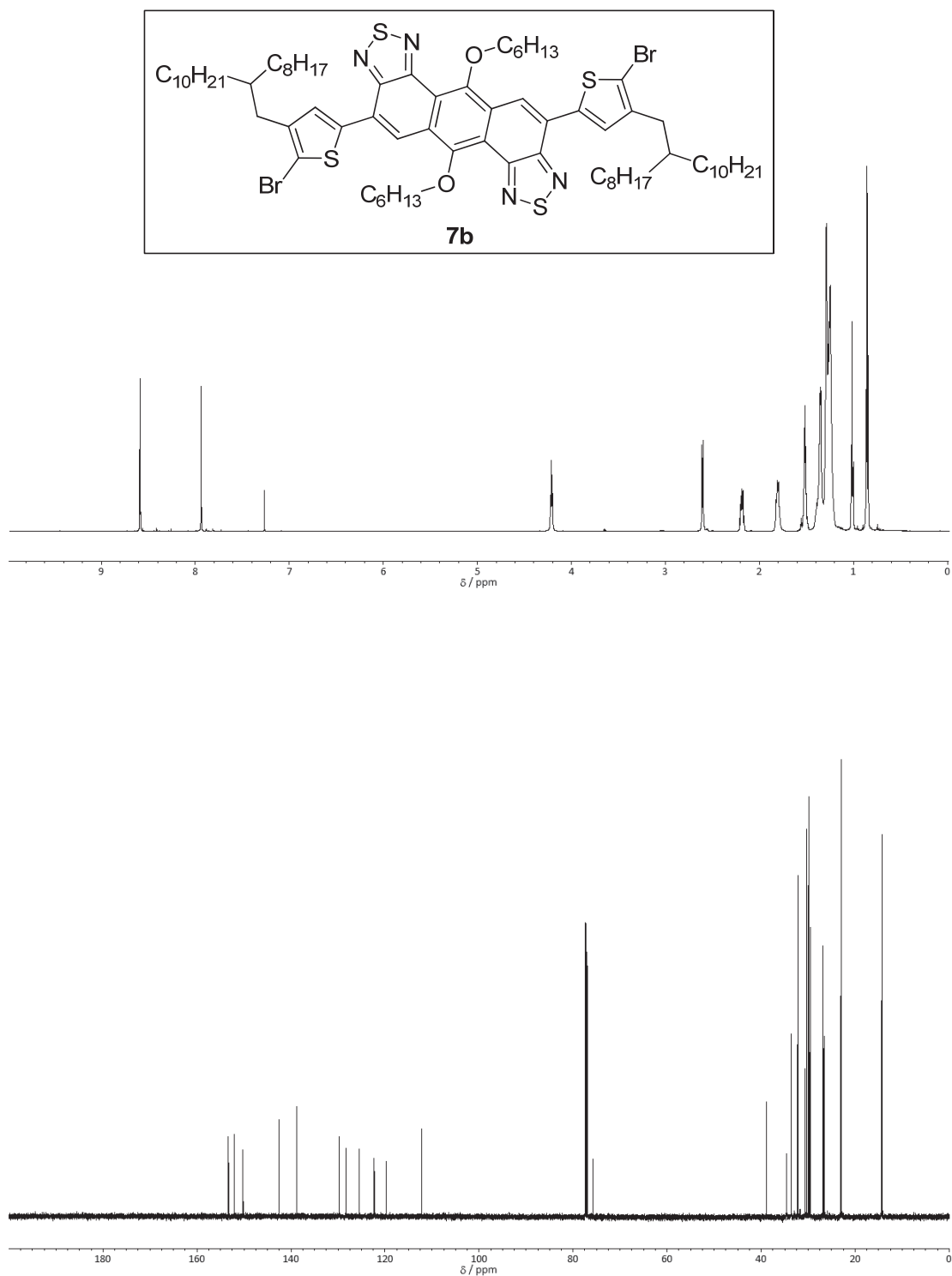
The ^1H and $^{13}\text{C}\{^1\text{H}\}$ NMR spectra of **6a** (in CDCl_3).



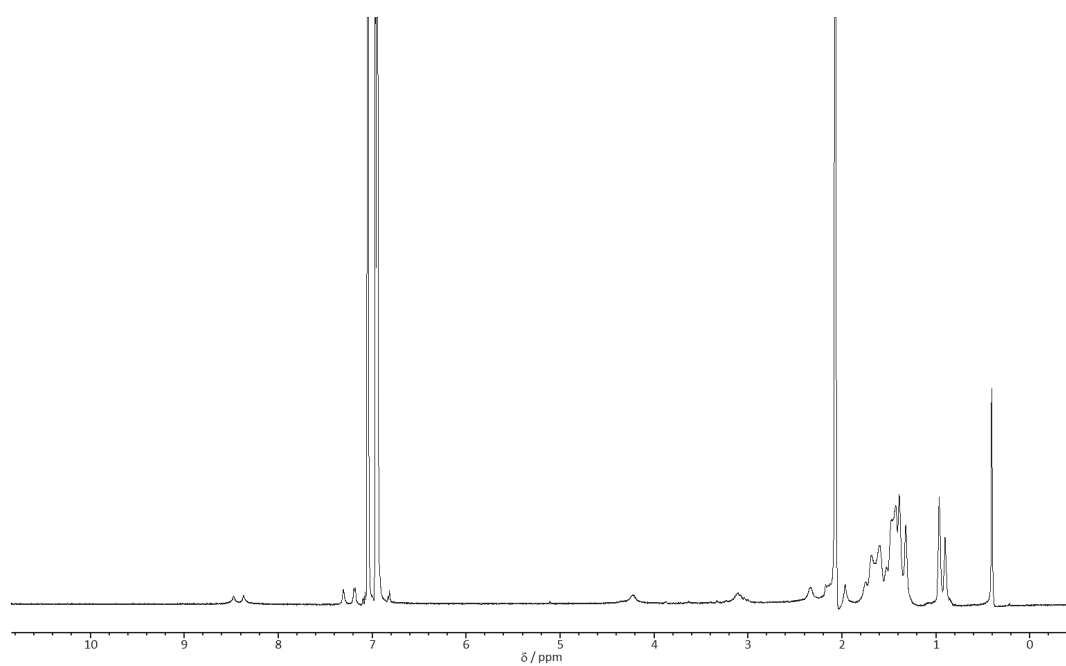
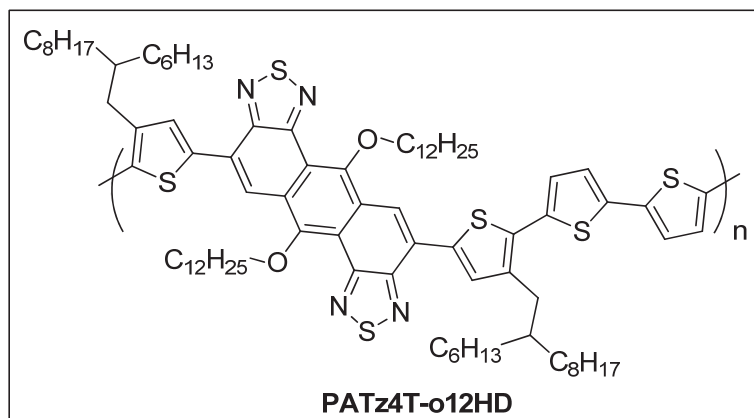
The ^1H and $^{13}\text{C}\{^1\text{H}\}$ NMR spectra of **6b** (in CDCl_3).



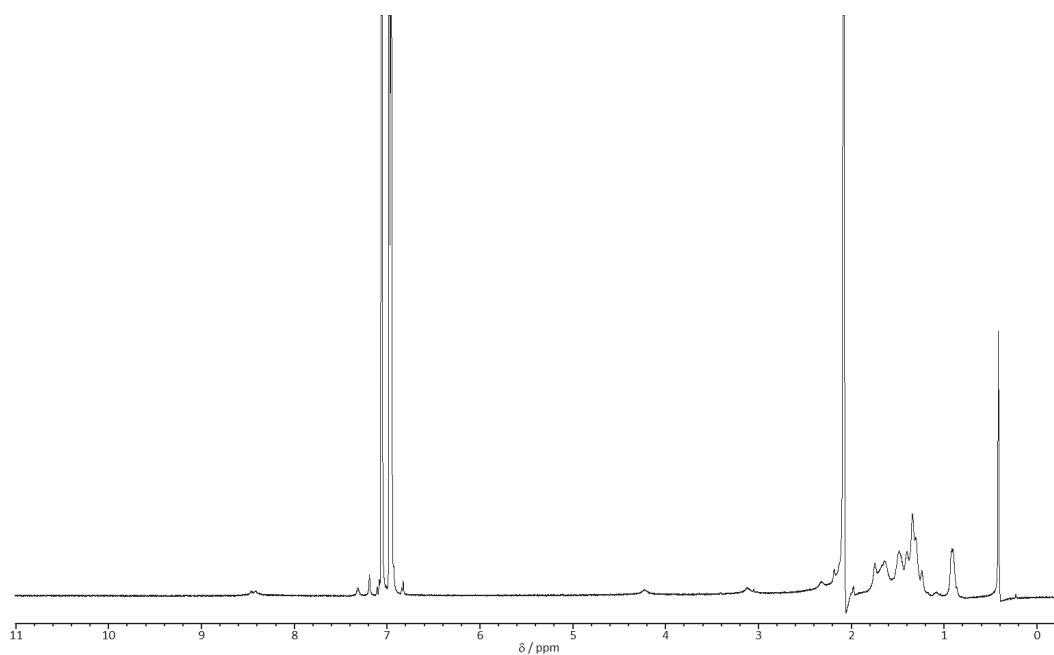
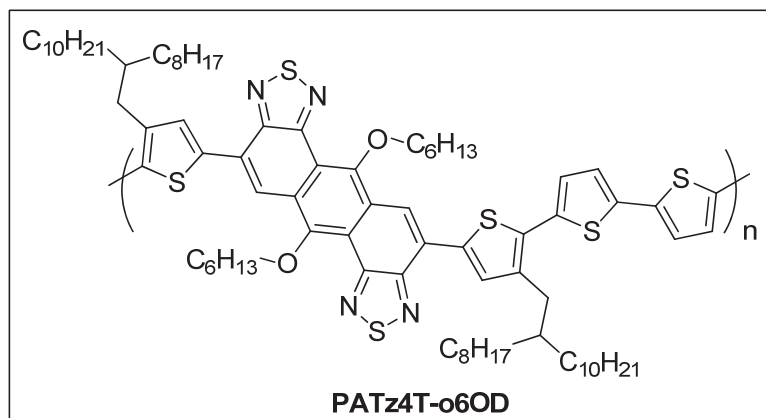
The ^1H and $^{13}\text{C}\{^1\text{H}\}$ NMR spectra of **7a** (in CDCl_3).



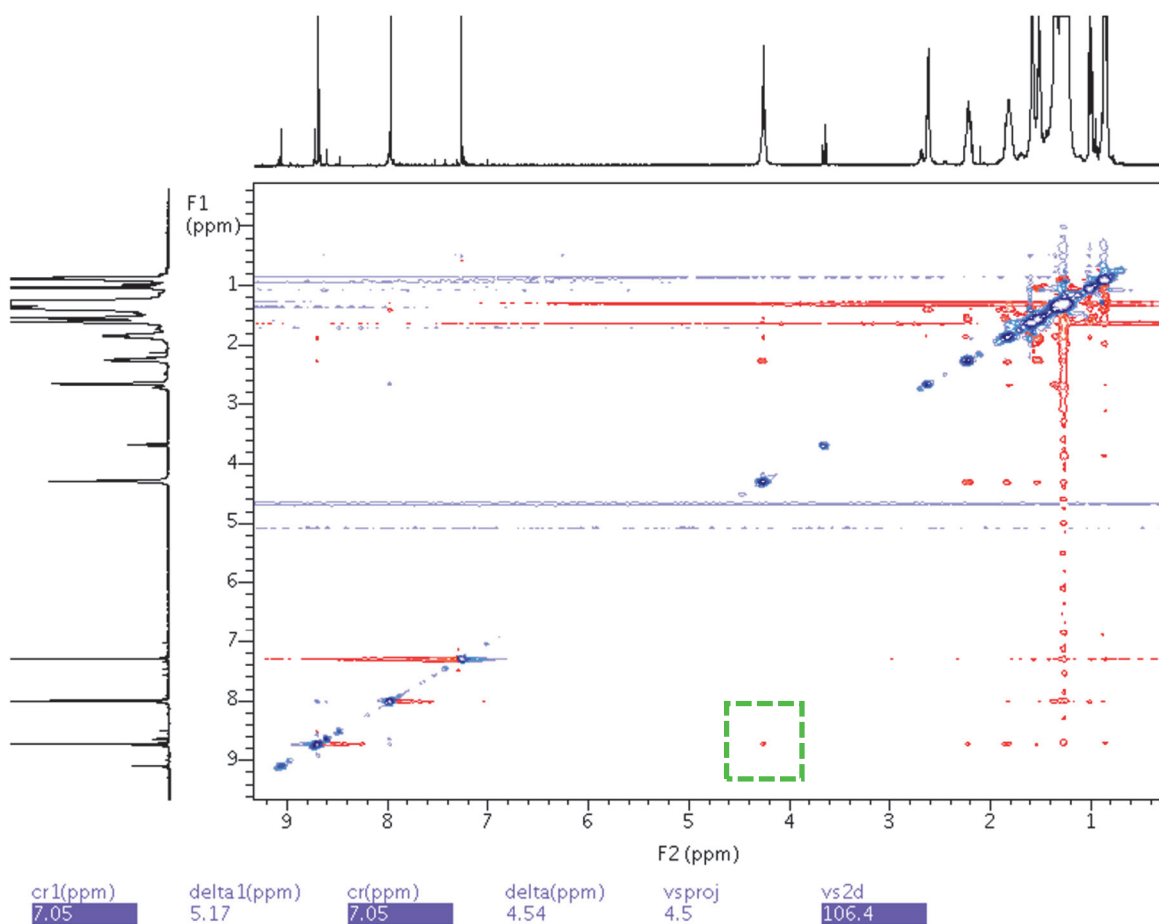
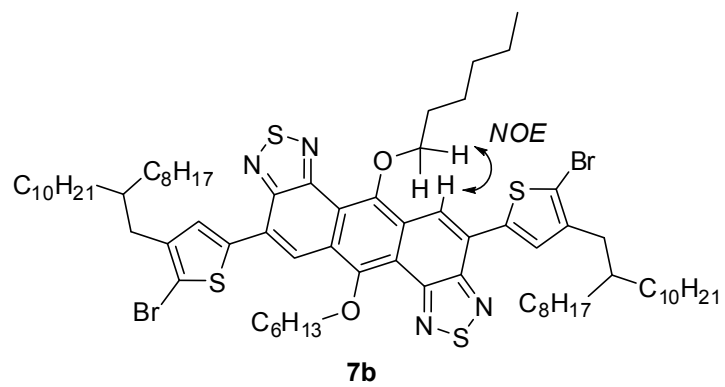
The ^1H and $^{13}\text{C}\{^1\text{H}\}$ NMR spectra of **7b** (in CDCl_3).

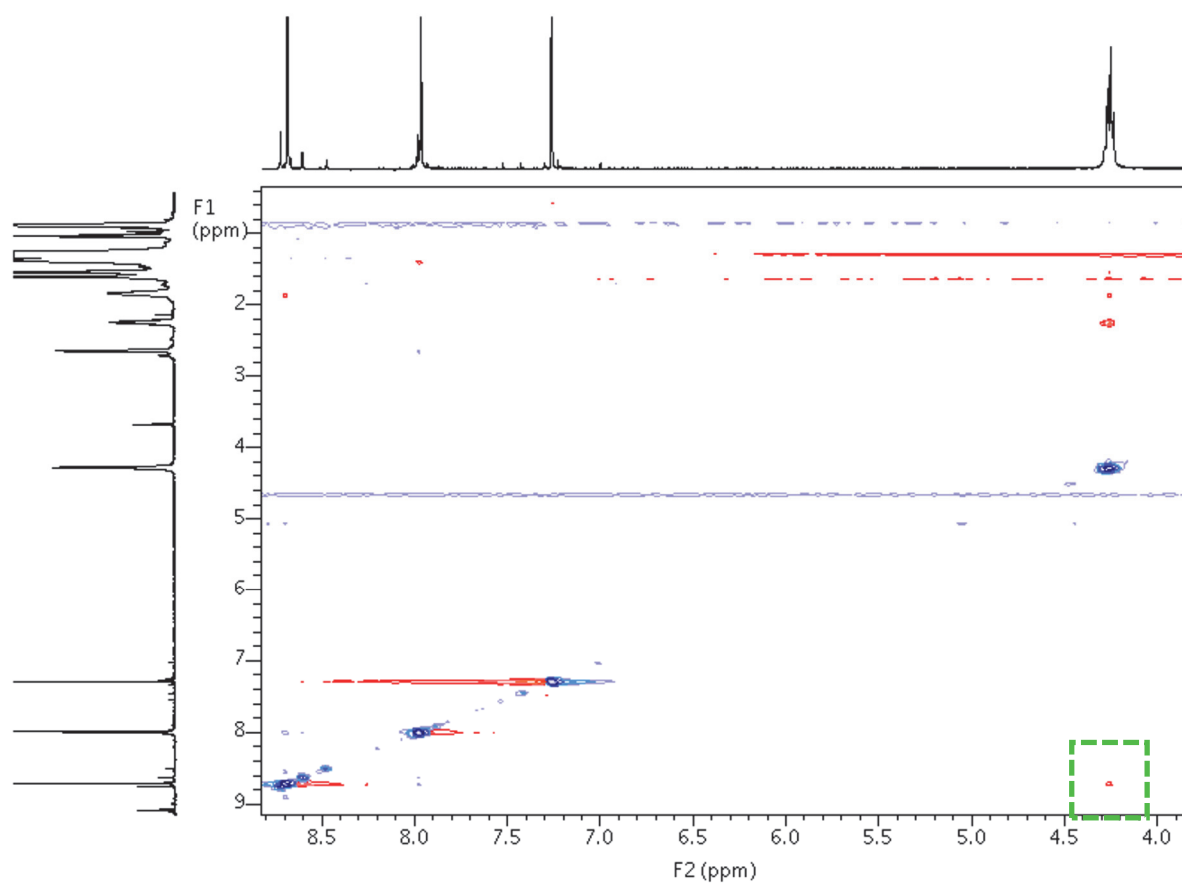


The ^1H NMR spectra of **PATz4T-o12HD** (in toluene- d_8 , 80 °C).



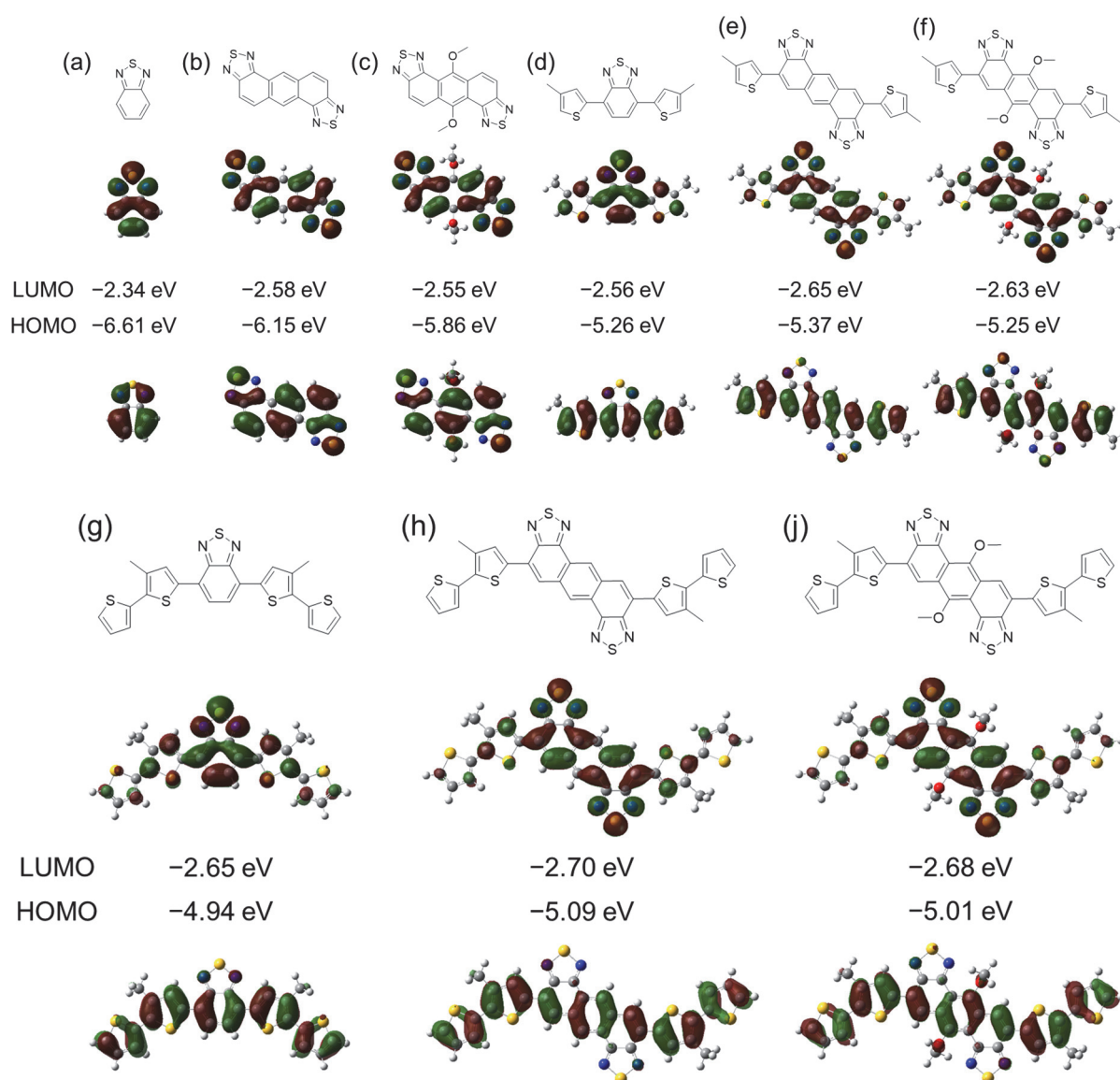
The ^1H NMR spectra of **PATz4T-o6OD** (in toluene- d_8 , 80 °C).





NOESY spectra of **7b** (in CDCl₃, rt).

2. DFT Calculations of Model Compound of PATz4T Polymers



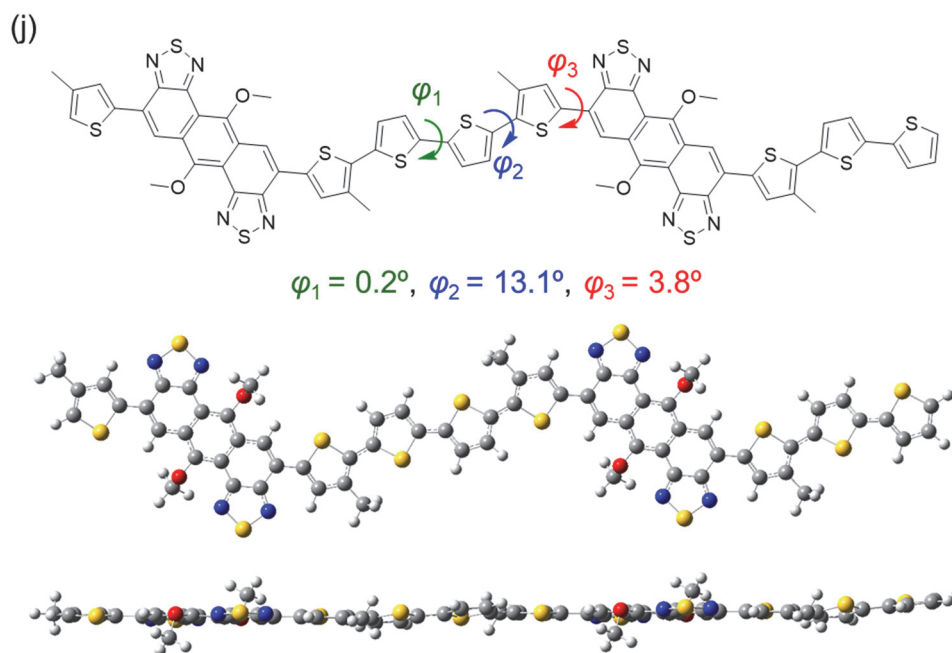


Figure S1. Geometry of HOMO and LUMO orbitals for (a) **BTz**, (b) **ATz**, (c) **ATzoMe**, (d) **BTz2T**, (e) **ATz2T**, (f) **ATz2ToMe**, (g) **BTz4T**, (h) **ATz4T**, and (i) **ATz4ToMe** calculated by using DFT at B3LYP/6-31G (d). (j) The optimized dimer structure of model compound for **PATz4T-oR¹R²**. All alkyl groups are replaced with a methyl group for simplicity.

3. GPC Charts of PATz4T Polymers

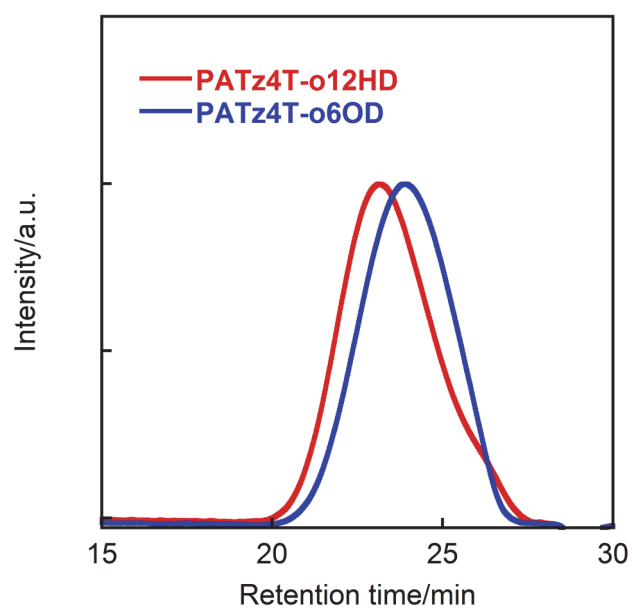


Figure S2. GPC Charts of PATz4T polymers.

4. Physicochemical Properties of BTz Derivatives

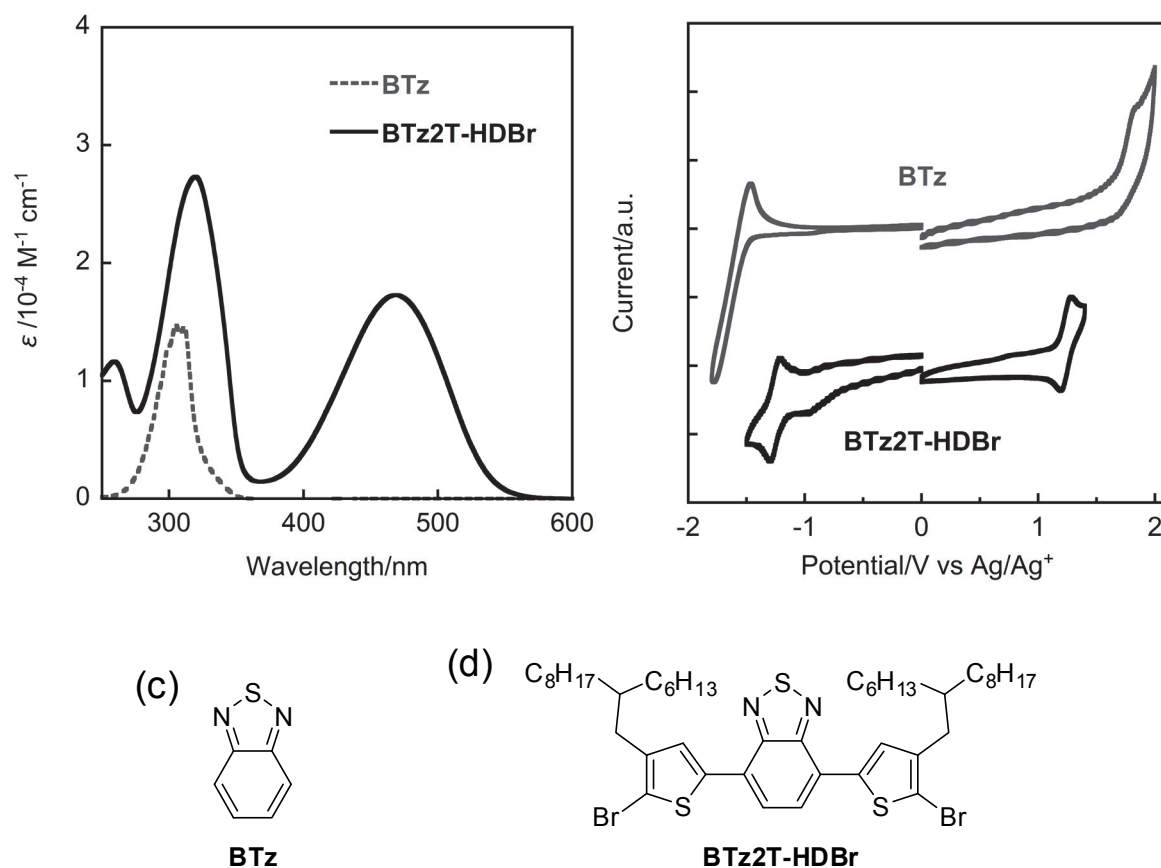


Figure S3. (a) UV-vis absorption spectra of BTz derivatives in CH₂Cl₂ solution and (b) cyclic voltammograms of BTz derivatives in CH₂Cl₂ solution with 0.1 M TBAPF₆ (Fc/Fc⁺: $E_{1/2} = +0.44$ V). Chemical structures of (c) parent BTz and (d) BTz2T-HDBr.

5. Output Curves and Optimization of PATz4T-Based OFETs

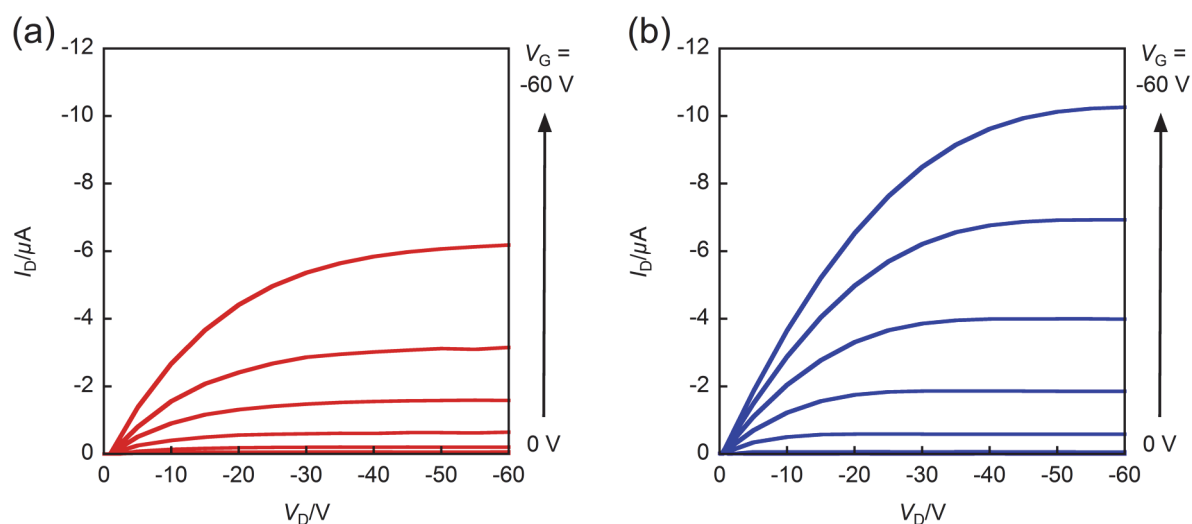


Figure S4. The output curves of (a) **PATz4T-o12HD-** and (b) **PATz4T-o6OD-**based OFET devices.

Table S1. OFET Performances of PATz4T-based Devices^a

PATz4T	SAM^b	T_{anneal}/°C^c	μ_{FET}/cm² V⁻¹ s⁻¹^d	V_{th}/V	I_{on/off}
-o12HD	ODTS	—	2.74 × 10 ⁻⁴ (2.67 × 10 ⁻⁴)	+5 (+5)	10 ⁴ ~10 ⁶
	ODTS	200	4.59 × 10 ⁻⁴ (3.91 × 10 ⁻⁴)	-8 (-5)	10 ¹ ~10 ²
	ODTS	250	1.18 × 10 ⁻³ (7.21 × 10 ⁻⁴)	-3 (+5)	10 ¹ ~10 ⁴
	FOTS	250	0.025 (0.021)	-17 (-15)	10 ³ ~10 ⁴
-o6OD	ODTS	—	1.77 × 10 ⁻³ (1.36 × 10 ⁻³)	-8 (-2)	10 ² ~10 ⁴
	ODTS	200	6.02 × 10 ⁻³ (4.44 × 10 ⁻³)	-11 (-8)	10 ³ ~10 ⁴
	ODTS	250	2.99 × 10 ⁻³ (2.23 × 10 ⁻³)	-8 (-2)	10 ³ ~10 ⁶
	FOTS	250	0.032 (0.023)	-11 (-6)	10 ³ ~10 ⁶

^aAverage values are shown in parentheses. ^bSAM = self-assembled monolayer. ODTS = *n*-octadecyltriethoxysilane. FOTS = 1*H*,1*H*,2*H*,2*H*-perfluorooctyltriethoxysilane. ^cThermal annealing temperature. ^dExtracted from transfer curves in saturation regime.

6. Photovoltaic Properties of PATz4T-Based OPVs

Table S2. Solar Cell Performances of Polymer/PC₆₁BM-based Devices

PATz4T	p/n ratio ^a	solvent ^b	$J_{sc}/\text{mA cm}^{-2}$	V_{oc}/V	FF	PCE _{max} (PCE _{avg})/%
-o12HD	1:1	DCB	5.77	0.84	0.55	2.64 (2.07)
	1:2	DCB	2.99	0.86	0.46	1.19 (1.14)
	1:1	DCB+1 vol% DIO	6.02	0.82	0.56	2.76 (2.58)
	1:1	DCB+2 vol% DIO	5.95	0.81	0.71	3.44 (3.34)
	1:1	CB+1 vol% DIO	7.51	0.80	0.46	2.75 (2.12)
	1:1	CB+1 vol% DIH	6.22	0.81	0.72	3.63 (3.14)
	1:1	DCB+ 1 vol% DIH	6.39	0.82	0.73	3.81 (3.52)
-o6OD	1:1	CB	9.53	0.84	0.67	5.36 (5.04)
	1:2	CB	10.67	0.83	0.64	5.66 (5.42)
	1:2	DCB	9.24	0.83	0.71	5.42 (5.17)
	1:3	DCB	11.03	0.82	0.57	5.20 (5.10)
	1:2	CB+1 vol% DIO	10.48	0.80	0.52	4.31 (4.14)
	1:2	CB+1 vol% DPE	10.75	0.79	0.48	4.10 (3.90)
	1:2	CB+1 vol% CN	9.55	0.83	0.56	4.40 (4.29)

^aWeight ratios of polymers and PC₆₁BM. ^bCB = chlorobenzene. DCB = *o*-dichlorobenzene. DIO = 1,8-diiodooctane. DPE = diphenyl ether. DIH = 1,6-diiodohexane. CN = 1-chloronaphthalene.

7. SCLC Hole Mobility of PATz4T-Based Hole-Only Devices

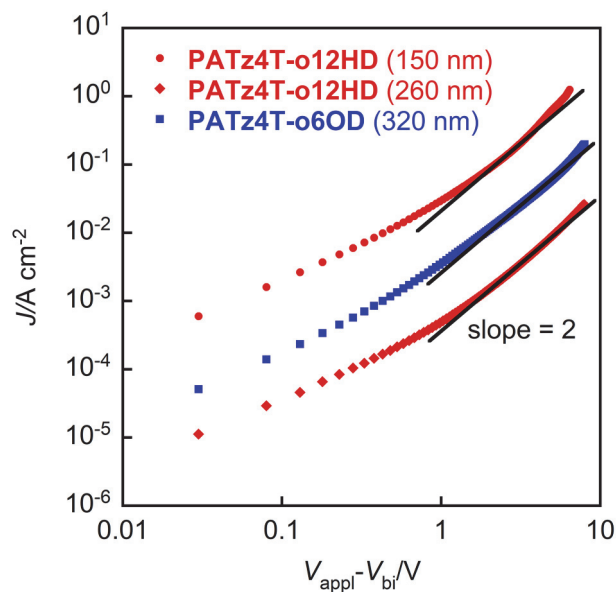


Figure S5. J - V characteristics of **PATz4T**-based hole-only devices with configuration of ITO/(PEDOT:PSS)/(polymer:PC₆₁BM)/MoO₃/Al. Blend ratios of polymer and PC₆₁BM are 1:1 for **PATz4T-o12HD** and 1:2 for **PATz4T-o6OD**. The thickness of active layer is 150 nm or 260 nm for **PATz4T-o12HD**, and 320 nm for **PATz4T-o6OD**. No SCLC region was observed in **PATz4T-o6OD**-based hole-only devices due to its high current density.

7. AFM Images of PATz4T films

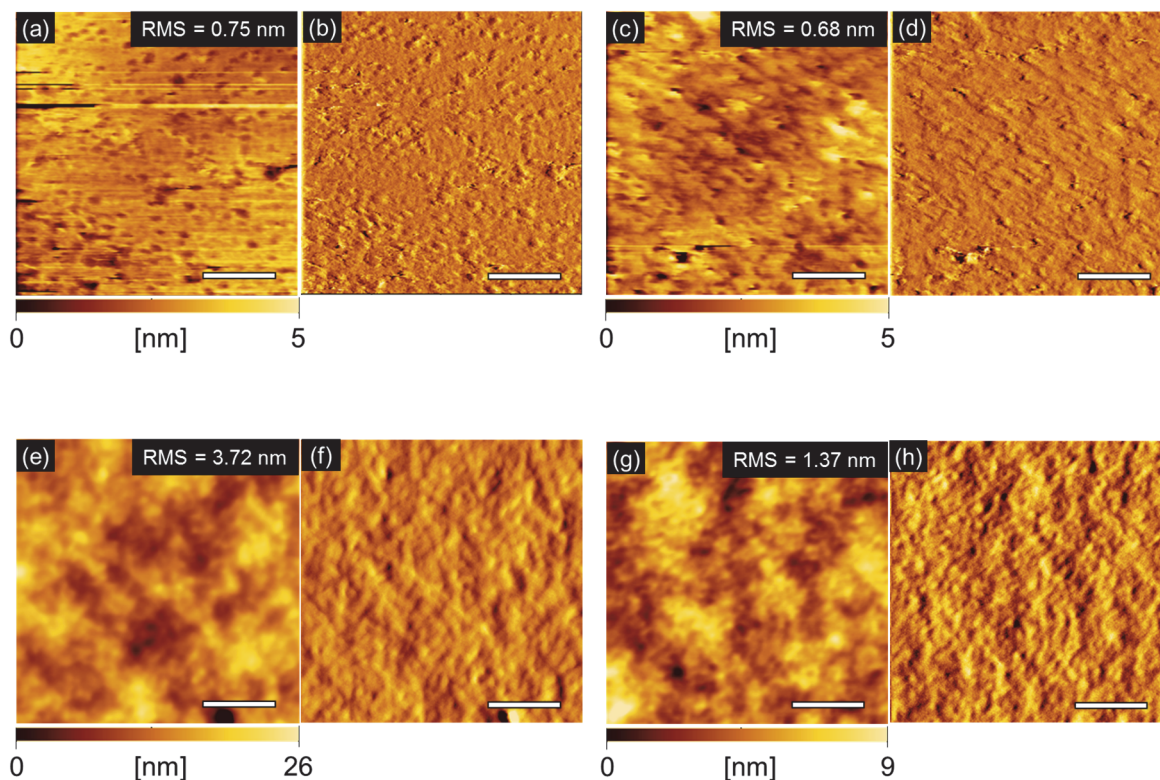


Figure S6. Topological (a,c,e,g) and error-signal (b,d,f,h) images ($2 \times 2 \mu\text{m}$) of (a-d) **PATz4T-oR¹R²** films on FOTS-treated n^+ -Si/SiO₂ substrate annealed at 250 °C, and (e-h) **PATz4T-oR¹R²/PC₆₁BM** blended films on ITO/ZnO; (a,b,e,f) **PATz4T-o12HD**, and (c,d,g,h) **PATz4T-o6OD**. Blend ratios of polymer and PC₆₁BM are 1:1 for **PATz4T-o12HD** and 1:2 for **PATz4T-o6OD**. The scale bar is 500 nm.

8. 1D Profiles and 2D Images of GIWAXS Measurement of PATz4T and blended films

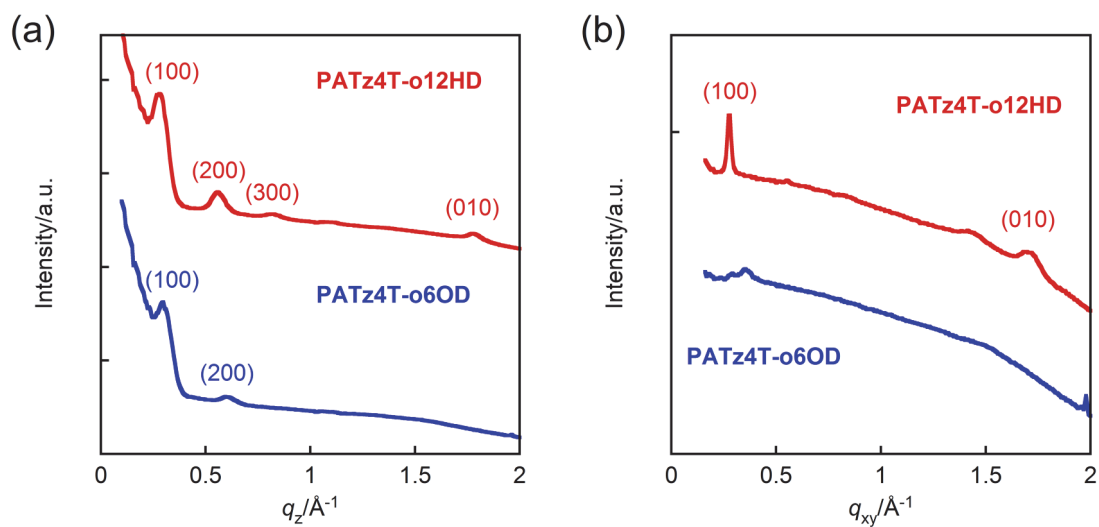


Figure S7. 1D profiles of GIWAXS images of **PATz4T** films annealed at 250 °C on FOTS-treated n^+ -Si/SiO₂ substrate; (a) out-of-plane and (b) in-plane.

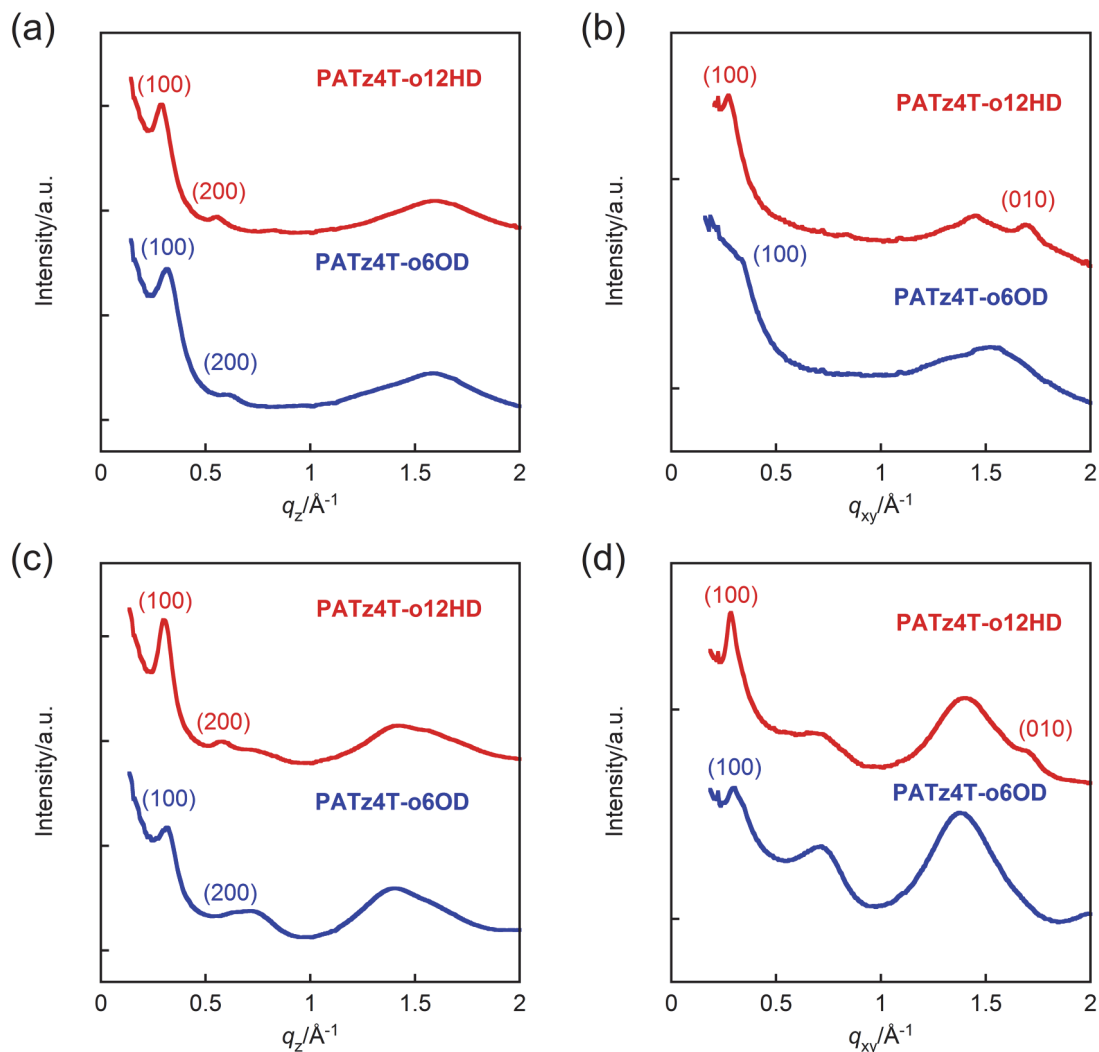


Figure S8. 1D profiles of GIWAXS images of pure **PATz4T** films (a,b) and blended films with PC₆₁BM (c,d) on ZnO-coated ITO substrate; (a,c) out-of-plane and (b,d) in-plane. Blend ratios of polymer:PC₆₁BM are 1:1 for **PATz4T-o12HD** and 1:2 for **PATz4T-o6OD**, respectively.

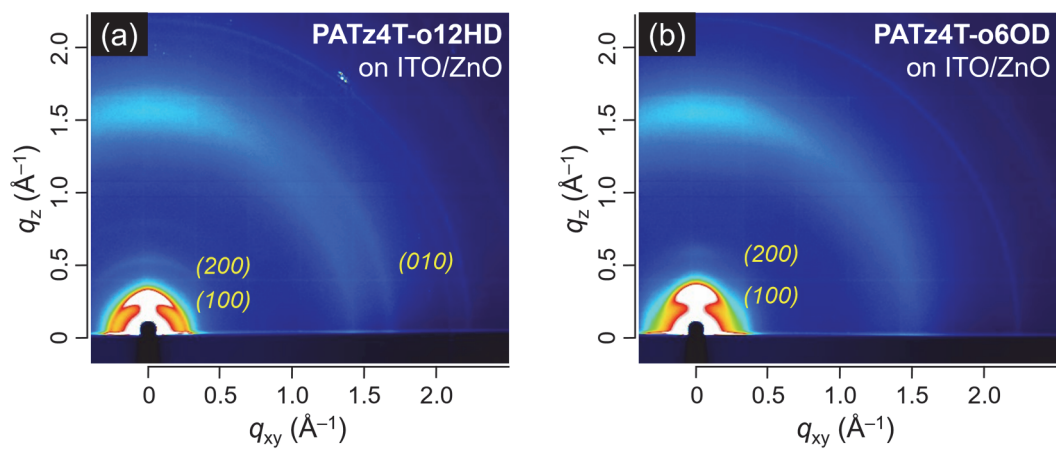


Figure S9. 2D GIWAXS images of pure **PATz4T** films on ZnO-coated ITO substrate; (a) **PATz4T-o12HD** and (b) **PATz4T-o6OD**.