

Supporting Information

Identification of active sites over Fe₂O₃-based architecture:

The promotion effect of H₂SO₄ erosion synthetic protocol

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Catalytic performance evaluation

The performance of catalysts for the NH₃-SCR of NO was carried out in a fixed bed glass reactor with inner diameter of 6 mm under steady state reaction conditions. The composition of gas mixture was: 600 ppm NO, 600 ppm NH₃, 3 vol% O₂, 300 ppm SO₂ (when used), 10 vol% H₂O (when used) and N₂ as the balance. The flow rate of the feeding gases was 600 mL min⁻¹. The catalysts (0.5 g) were pressed into blocks, crushed and sieved with 40–60 meshes before used. The concentrations of NO were continually monitored by a testo 340 flue gas analyzer. N₂ selectivity and outlet N₂O concentration were measured by Thermo Scientific Antaris IGS analyzer.

To calculate the NO conversion ratio, we use the next formula:

$$\text{NO Conversion (\%)} = \frac{[\text{NOx}]_{\text{in}} - [\text{NOx}]_{\text{out}}}{[\text{NOx}]_{\text{in}}} \times 100 \%$$

Additional Figures and Data

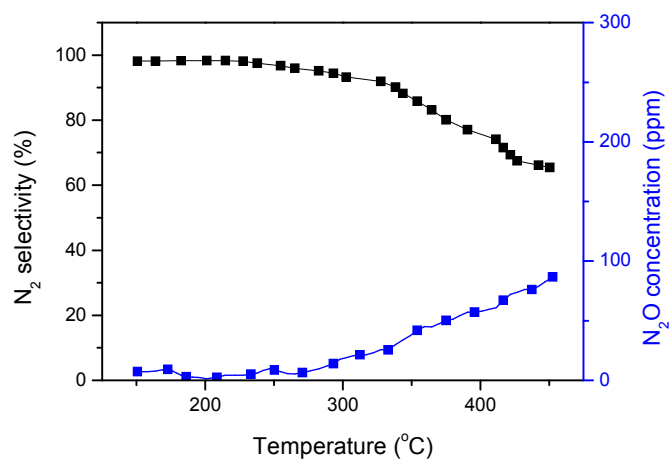


Figure S1. N₂ selectivity and outlet N₂O concentration versus temperature data over parent Fe₂O₃. Reactant feed contains 600 ppm of NO, 600 ppm of NH₃, 3 vol% O₂, balanced with N₂.

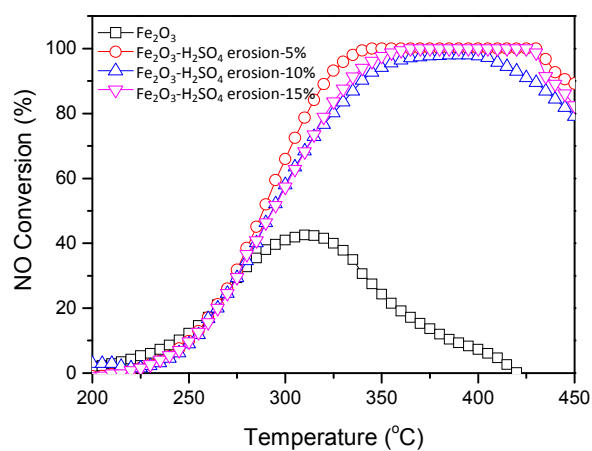


Figure S2. NO conversion versus temperature data over parent Fe₂O₃ and Fe₂O₃-H₂SO₄ erosion samples. Reactant feed contains 600 ppm of NO, 600 ppm of NH₃, 3 vol% O₂, balanced with N₂.

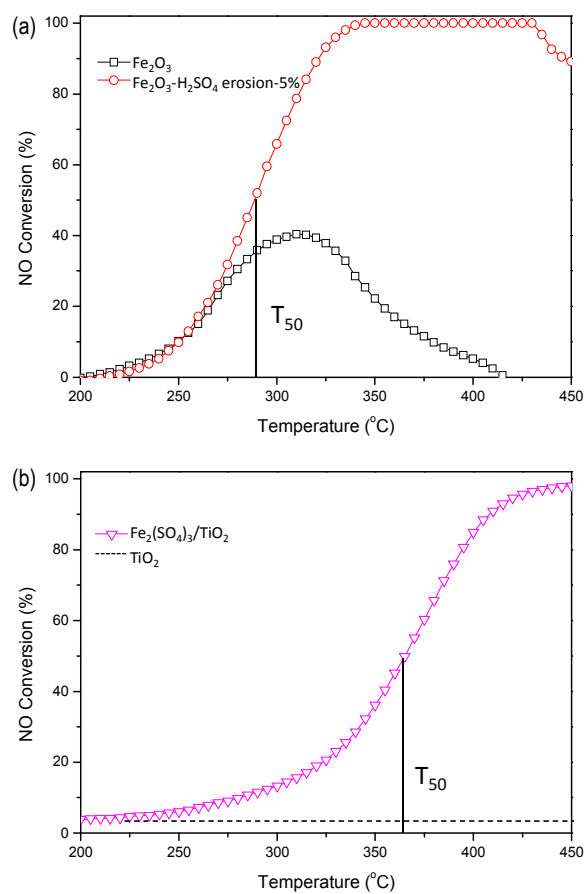


Figure S3. NO conversion versus temperature data over (a) $\text{Fe}_2\text{O}_3\text{-H}_2\text{SO}_4$ erosion-5%, Fe_2O_3 samples and (b) $\text{Fe}_2(\text{SO}_4)_3/\text{TiO}_2$, TiO_2 samples. Reactant feed contains 600 ppm of NO, 600 ppm of NH_3 , 3 vol% O_2 , balanced with N_2 . T_{50} (temperature of 50% conversion) of $\text{Fe}_2\text{O}_3\text{-H}_2\text{SO}_4$ erosion-5% sample is much lower than that of $\text{Fe}_2(\text{SO}_4)_3/\text{TiO}_2$, despite that these two catalyst has identical $\text{Fe}_2(\text{SO}_4)_3$ loading amount.

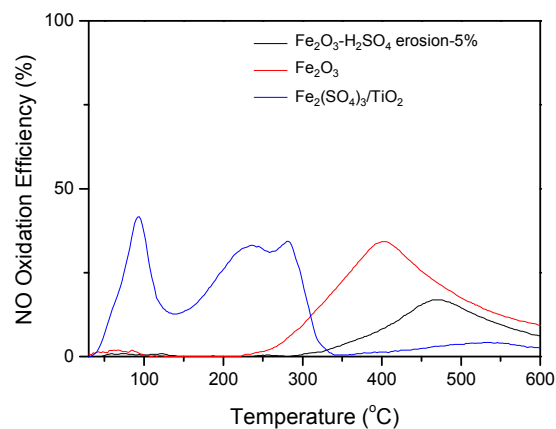


Figure S4. The NO probe molecule temperature-programmed oxidation (TPO) of as-prepared $\text{Fe}_2\text{O}_3\text{-H}_2\text{SO}_4$ erosion sample, Fe_2O_3 and $\text{Fe}_2(\text{SO}_4)_3/\text{TiO}_2$. Reactant feed contains 600 ppm of NO, 3 vol% of O_2 , balanced with N_2 .

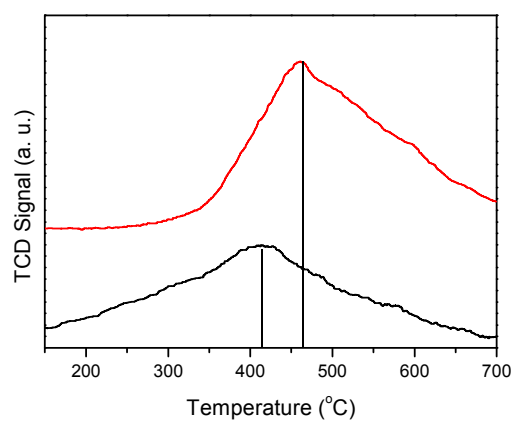


Figure S5. The NH_3 temperature-programmed desorption (TPD) of Fe_2O_3 (black) and $\text{Fe}_2\text{O}_3\text{-H}_2\text{SO}_4$ erosion-5% sample (red).