

Supporting Information

Impact of peat fire on the soil and the export of dissolved organic carbon in tropical peat soil, Central Kalimantan, Indonesia

Kazuto Sazawa[†], Takatoshi Wakimoto[†], Noriko Hata[†], Shigeru Taguchi[†], Masami Fukushima[‡], Shunitz Tanaka[§], Yustiawati Yustiawati[§], M. Suhaemi Syawal^{||}, Daisuke Tanaka[†], and Hideki Kuramitz^{†,*}

[†] Department of Environmental Biology and Chemistry, Graduate School of Science and Engineering for Research, University of Toyama, Gofuku 3190, Toyama 930-8555, Japan

[‡] Laboratory of Chemical Resource, Division of Sustainable Resources Engineering, Faculty of Engineering, Hokkaido University, Sapporo 060-8628, Japan

[§] Division of Material Science, Graduate School of Environmental Earth Science, Hokkaido University, Sapporo, Hokkaido 060-0810, Japan

^{||} Research Center for Limnology, Indonesian Institutes of Sciences, Jl. Raya Jakarta-Bogor Km.46 Cibinong, Bogor 16911, Indonesia

*Corresponding Author Tel. & fax +81-76-445-6669, E-mail address kuramitz@sci.u-toyama.ac.jp

Table S1. The assay conditions (substrates, buffer and optimum pH) of these enzymes in this study. MUB: modified universal buffer; MB: mcllvaine buffer.

Enzymes	Substrate	Buffer	Optimum pH
Phenol oxidase (PO)	2,2'-azinobis-(3-ethylbenzothiazoline-6-sulfononic acid) diammonium salt	MUB	2.0
β -glucosidase (β -Glu)	<i>p</i> -NP- β -D-glucopyranoside	MB	5.0
β -xylosidase (β -Xyl)	<i>p</i> -NP- β -D-xylopyranoside	MUB	5.0
β -galactosidase (β -Gal)	<i>p</i> -NP- β -D-galactopyranoside	MB	5.0
α -mannosidase (α -Man)	<i>p</i> -NP- α -D-mannopyranoside	MB	5.0
<i>N</i> -acethyl-glucosaminidase (NAG)	<i>p</i> -NP- <i>N</i> -acetyl- β -D-glucosaminide	MB	5.0
Acid phosphatase (AcP)	<i>p</i> -NP- β -D-phosphate disodium salt hexahydrate	MUB	6.5
Alkaline phosphatase (ALP)	<i>p</i> -NP- β -D-phosphate disodium salt hexahydrate	MUB	11.0
Phosphodiesterase (PD)	Bis(<i>p</i> -NP)phosphate sodium salt	MUB	6.5

Table S2. Pysico-chemical properties of Indonesian peat soils.

	Site code	Location	Year	Sample depth (cm)	pH (H ₂ O)	pH (KCl)	EC (mS cm ⁻¹)	Particle density (g cm ⁻³)	Moisture content (%)	Ignition loss (%)
Unburned site (UB)	UB1	2° 19' 48.50" S, 113° 54' 18.10" E	2010	30-50	3.25	2.13	2.70	0.89	78.4	70.8
			2011	0-20	No data	No data	No data	No data	81.7	No data
			2011	30-50	3.41	2.01	1.66	1.05	86.0	81.0
			2012	0-20	No data	No data	No data	No data	77.4	No data
			2012	30-50	3.09	1.93	0.35	1.22	81.8	81.4
	UB2	2° 20' 18.11" S, 113° 53' 47.37" E	2010	30-50	3.05	2.15	1.72	0.83	80.2	83.8
			2011	0-20	No data	No data	No data	No data	73.1	No data
			2011	30-50	3.04	2.15	0.92	0.83	81.2	81.3
	UB3	2° 21' 8.60" S, 114° 2' 12.90" E	2010	30-50	3.15	2.04	0.82	0.80	79.0	76.9
			2011	0-20	No data	No data	No data	No data	80.9	No data
			2011	30-50	3.08	2.18	0.75	1.01	79.1	86.6
			2012	0-20	No data	No data	No data	No data	75.7	No data
			2012	30-50	3.46	1.91	0.46	1.09	78.1	80.7
Burned site (B)	B1	2° 20' 50.60" S, 114° 2' 17.90" E	2010	30-50	3.31	2.14	0.89	0.95	76.4	70.7
			2011	0-20	No data	No data	No data	No data	39.0	No data
			2011	30-50	2.77	1.80	0.76	1.24	74.0	75.6
			2012	0-20	No data	No data	No data	No data	29.5	No data
			2012	30-50	3.71	1.85	0.13	1.20	83.2	78.9
	B2	2° 20' 47.50" S, 114° 0' 59.80" E	2010	30-50	3.39	1.89	0.88	0.91	73.1	72.7
			2011	0-20	No data	No data	No data	No data	31.9	No data
			2011	30-50	2.86	1.76	0.50	1.16	79.9	71.2
			2012	0-20	No data	No data	No data	No data	26.9	No data
			2012	30-50	3.45	1.69	0.26	1.27	70.9	73.2
	B3	2° 18' 28.90" S, 114° 1' 36.20" E	2010	30-50	3.06	1.89	0.84	1.02	78.9	69.4
			2011	0-20	No data	No data	No data	No data	40.0	No data
			2011	30-50	2.89	1.88	0.67	1.29	82.1	78.4
			2012	0-20	No data	No data	No data	No data	29.5	No data
			2012	30-50	3.49	1.75	0.27	1.28	81.2	72.9
	B4	2° 17' 41.40" S, 114° 1' 58.00" E	2010	30-50	2.95	1.95	3.42	1.06	74.9	61.2
			2011	0-20	No data	No data	No data	No data	45.2	No data
			2011	30-50	2.75	1.84	0.68	0.93	65.9	79.8
			2012	0-20	4.00	1.92	0.29	1.24	20.5	71.1
			2012	30-50	3.45	1.78	0.29	1.37	80.9	74.4
	B5	2° 16' 47.30" S, 113° 55' 14.40" E	2011	0-20	5.81	5.55	4.34	1.58	42.7	46.1
			2011	30-50	3.26	2.40	1.20	1.20	77.9	81.6
			2012	0-20	4.80	3.00	0.13	1.39	55.6	67.1
			2012	30-50	3.92	2.22	0.16	1.33	77.9	82.3

Table S3. The elemental composition and atomic ratio (H/C, O/C and C/N) of the unburned soil (UB1 and 3), burned soil (B1– 5) and the thermally treated samples.

Sample code	year	Elemental composition (%)						Atomic ratio		
		C	H	N	O	S	Ash	H/C	O/C	C/N
UB1 (subsurface)	2010	29.5	4.7	0.7	46.3	0.1	18.7	1.9	1.2	48
	2011	51.9	5.2	1.2	41.4	0.0	0.3	1.2	0.6	49
	2012	51.6	5.4	1.6	40.9	0.0	0.4	1.3	0.6	39
UB3 (subsurface)	2010	36.4	5.6	1.3	51.9	0.2	4.7	1.8	1.1	33
	2011	52.3	5.5	1.3	40.2	0.0	0.7	1.2	0.6	48
	2012	51.3	5.6	1.7	41.0	0.0	0.4	1.3	0.6	36
B1 (subsurface)	2010	47.8	5.6	1.0	42.2	0.2	3.2	1.4	0.7	55
	2011	57.0	5.2	0.6	36.8	0.0	0.5	1.1	0.5	114
	2012	54.2	5.1	0.8	39.6	0.0	0.3	1.1	0.6	82
B2 (subsurface)	2010	43.8	5.1	0.6	46.2	0.1	4.2	1.4	0.8	81
	2011	53.9	4.6	0.6	40.2	0.0	0.8	1.0	0.6	112
	2012	51.9	4.8	0.9	38.0	0.0	4.3	1.1	0.6	67
B3 (subsurface)	2010	47.0	5.3	0.8	45.1	0.2	1.7	1.3	0.7	70
	2011	54.8	4.7	0.6	39.4	0.0	0.5	1.0	0.5	103
	2012	54.8	4.9	0.8	38.9	0.0	0.5	1.1	0.5	79
B4 (subsurface)	2010	53.2	5.4	0.9	38.5	0.2	1.8	1.2	0.5	70
	2011	54.3	5.5	0.9	36.9	0.0	2.4	1.2	0.5	74
	2012	54.7	4.8	0.8	38.8	0.0	0.8	1.1	0.5	78
B5 (surface)	2011	16.8	1.7	0.5	13.4	0.0	67.7	1.2	0.6	41
	2012	55.5	4.6	2.1	33.5	0.0	4.4	1.0	0.5	31
B5 (subsurface)	2011	45.6	4.1	0.6	32.0	0.0	17.7	1.1	0.5	83
	2012	55.6	5.3	0.9	36.6	0.0	1.6	1.1	0.5	71
Thermally treated samples										
90 °C 60 min		53.7	5.0	1.3	39.7	0.0	0.3	1.1	0.6	50
120 °C 60 min		53.7	5.2	1.3	39.2	0.0	0.6	1.2	0.6	48
150 °C 60 min		53.2	5.1	1.3	39.9	0.0	0.5	1.1	0.6	48
175 °C 60 min		55.1	5.0	1.5	38.1	0.0	0.4	1.1	0.5	44
180 °C 60 min		56.0	5.0	1.6	37.1	0.0	0.4	1.1	0.5	42
190 °C 60 min		57.0	4.8	1.6	36.2	0.0	0.4	1.0	0.5	42
200 °C 1 min		54.2	5.2	1.4	38.9	0.0	0.3	1.1	0.5	44
200 °C 5 min		54.6	5.1	1.5	38.6	0.0	0.3	1.1	0.5	44
200 °C 30 min		58.8	4.5	1.6	34.6	0.0	0.5	0.9	0.4	42
200 °C 60 min		60.9	4.0	1.8	33.0	0.0	0.3	0.8	0.4	39
200 °C 120 min		61.5	3.9	1.9	32.2	0.0	0.6	0.8	0.4	38
250 °C 1 min		60.4	4.5	1.7	33.0	0.0	0.4	0.9	0.4	42
250 °C 5 min		60.5	4.2	1.7	33.1	0.0	0.5	0.8	0.4	42
250 °C 30 min		62.8	4.1	1.8	30.9	0.0	0.4	0.8	0.4	40
250 °C 60 min		63.2	3.6	2.1	30.8	0.0	0.5	0.7	0.4	36
250 °C 120 min		63.3	3.2	2.1	30.7	0.0	0.7	0.6	0.4	35
350 °C 60 min		64.2	3.3	2.3	29.4	0.0	0.9	0.6	0.3	33
480 °C 60 min		62.3	3.1	3.4	28.9	0.0	2.3	0.6	0.4	21

Table S4. The assignments for the peak identified in the Py-TMAH-GC/MS analyses of soils. C: Compounds derived from *p*-hydroxyphenol (cinnamyl) structures; G: Compounds derived from guaiacyl structures; S: Compounds derived from syringyl structures; O: Other non-lignin derived compounds; F: Fatty acids methyl esters.

No	Compound Name	Origin	No.	Compound Name	Origin
1	Phenol	O	18	3,4,5-Trimethoxybenzoic acid, methyl ester	S
2	1,2,3-Trimethylbenzene	O	19	1,6-Anhydro- β -D-glucopyranose	O
3	4-Methoxytoluene	C	20	3,4-Dimethoxybenzoic acid, methyl ester	G
4	2-Methylphenol	O	21	4,8-Dimethyl-1-nonanol	O
5	4-Methoxybenzaldehyde	C	22	3-(3,4-dimethoxyphenyl)-2-Propenoic acid, methyl ester	G
6	4-Methoxyphenol	O	23	11-Octadecenoic acid, methyl ester	F
7	Benzoic acid, methyl ester	O	24	Hexadecanoic acid, methyl ester	F
8	4-Methylphenol	O	25	17-methyl-Octadecanoic acid, methyl ester	F
9	2-Methoxy-4-methylphenol	G	26	12,15-Octadecadienoic acid, methyl ester	F
10	2,3-Dimethoxytoluene	G	27	15-Tetracosenoic acid, methyl ester	F
11	3,4-Dimethoxytoluene	G	28	17-Octadecenoic acid, methyl ester	F
12	3,4-Dimethoxystyrene	G	29	Tetracosanoic acid, methyl ester	F
13	4-Methoxybenzoic acid, methyl ester	C	30	Octadecanoic acid, methyl ester	F
14	1,3,5-Trimethoxybenzene	O	31	Pentacosanoic acid, methyl ester	F
15	3-Allyl-6-methoxyphenol	O	32	Hexacosanoic acid, methyl ester	F
16	Dodecanoic acid, methyl ester	F	33	Octacosanoic acid, methyl ester	F
17	1-(3,4-dimethoxyphenyl)-ethanone	G	34	Ursolic acid, methyl ester	O

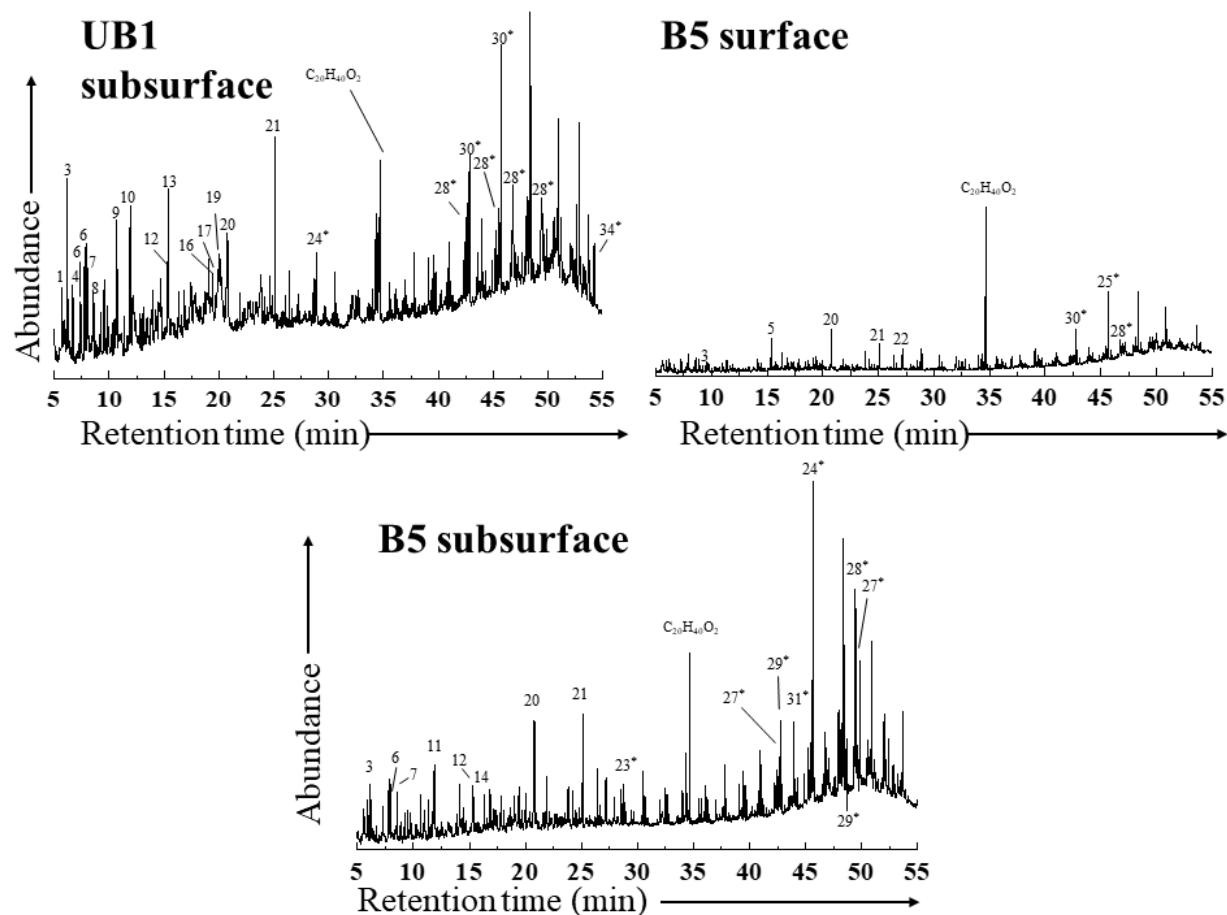


Figure S1. Pyrograms total ion chromatograms from Indonesian peat soils collected from UB1 (subsurface), B5 (surface and subsurface) in 2011. Number on the peaks corresponds to the pyrolysis compounds listed in Table S3. Asterisks represent Fatty acids. $C_{20}H_{40}O_2$: nonadecanoic acid methyl ester in acetone internal standard (ISTD).

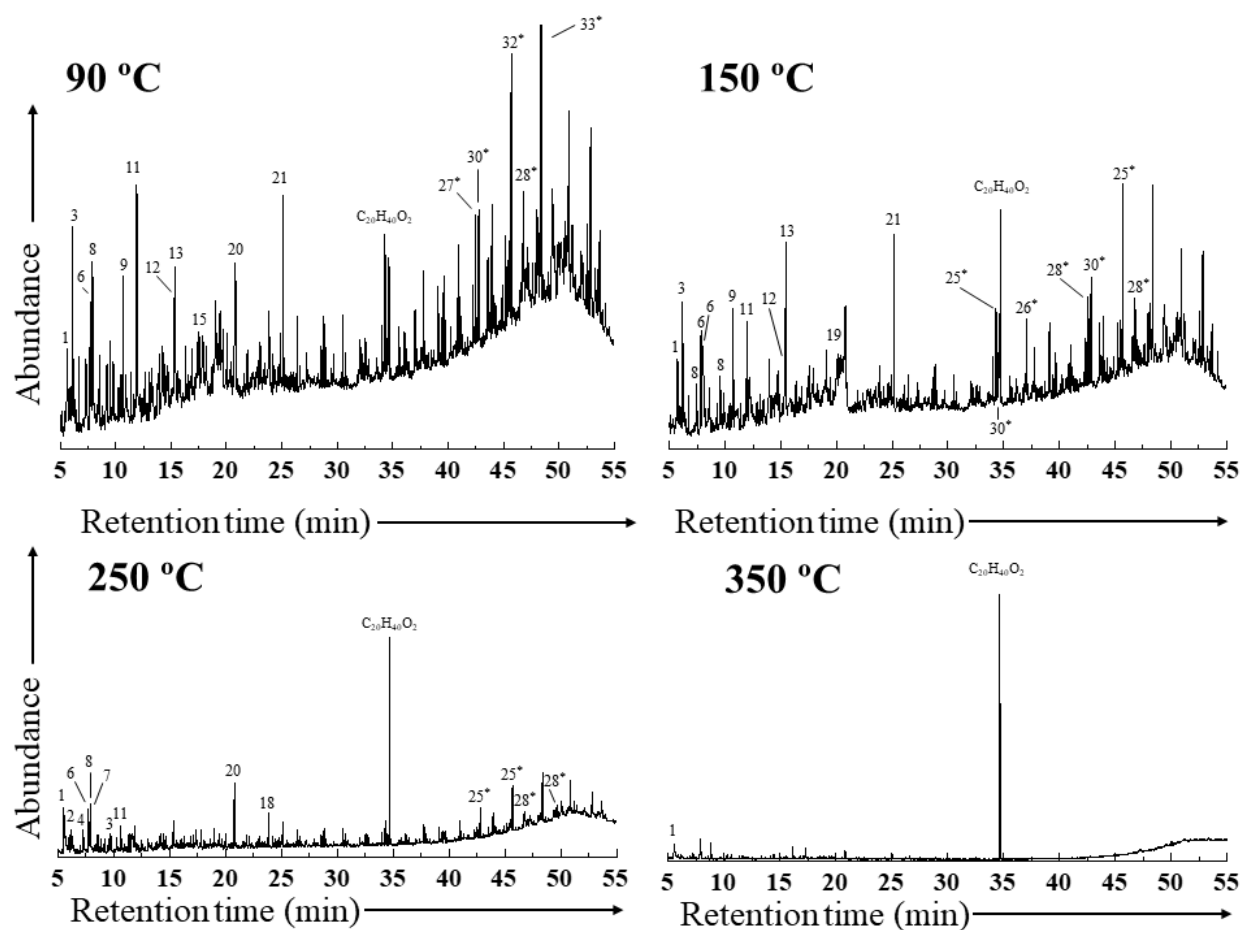


Figure S2. Pyrograms total ion chromatograms from thermally treated samples (90 °C, 150 °C, 250 °C and 350 °C for 60 min). Number on the peaks corresponds to the pyrolysis compounds listed in Table S3. Asterisks represent Fatty acids. $C_{20}H_{40}O_2$: nonadecanoic acid methyl ester in acetone internal standard (ISTD).

Table S4. The assignments for the peak identified in the Py-TMAH-GC/MS analyses of soils. C: Compounds derived from *p*-hydroxyphenol (cinnamyl) structures; G: Compounds derived from guaiacyl structures; S: Compounds derived from syringyl structures; O: Other non-lignin derived compounds; F: Fatty acids methyl esters.

No	Compound Name	Origin	No.	Compound Name	Origin
1	Phenol	O	18	3,4,5-Trimethoxybenzoic acid, methyl ester	S
2	1,2,3-Trimethylbenzene	O	19	1,6-Anhydro- β -D-glucopyranose	O
3	4-Methoxytoluene	C	20	3,4-Dimethoxybenzoic acid, methyl ester	G
4	2-Methylphenol	O	21	4,8-Dimethyl-1-nonanol	O
5	4-Methoxybenzaldehyde	C	22	3-(3,4-dimethoxyphenyl)-2-Propenoic acid, methyl ester	G
6	4-Methoxyphenol	O	23	11-Octadecenoic acid, methyl ester	F
7	Benzoic acid, methyl ester	O	24	Hexadecanoic acid, methyl ester	F
8	4-Methylphenol	O	25	17-methyl-Octadecanoic acid, methyl ester	F
9	2-Methoxy-4-methylphenol	G	26	12,15-Octadecadienoic acid, methyl ester	F
10	2,3-Dimethoxytoluene	G	27	15-Tetracosenoic acid, methyl ester	F
11	3,4-Dimethoxytoluene	G	28	17-Octadecenoic acid, methyl ester	F
12	3,4-Dimethoxystyrene	G	29	Tetracosanoic acid, methyl ester	F
13	4-Methoxybenzoic acid, methyl ester	C	30	Octadecanoic acid, methyl ester	F
14	1,3,5-Trimethoxybenzene	O	31	Pentacosanoic acid, methyl ester	F
15	3-Allyl-6-methoxyphenol	O	32	Hexacosanoic acid, methyl ester	F
16	Dodecanoic acid, methyl ester	F	33	Octacosanoic acid, methyl ester	F
17	1-(3,4-dimethoxyphenyl)-ethanone	G	34	Ursolic acid, methyl ester	O

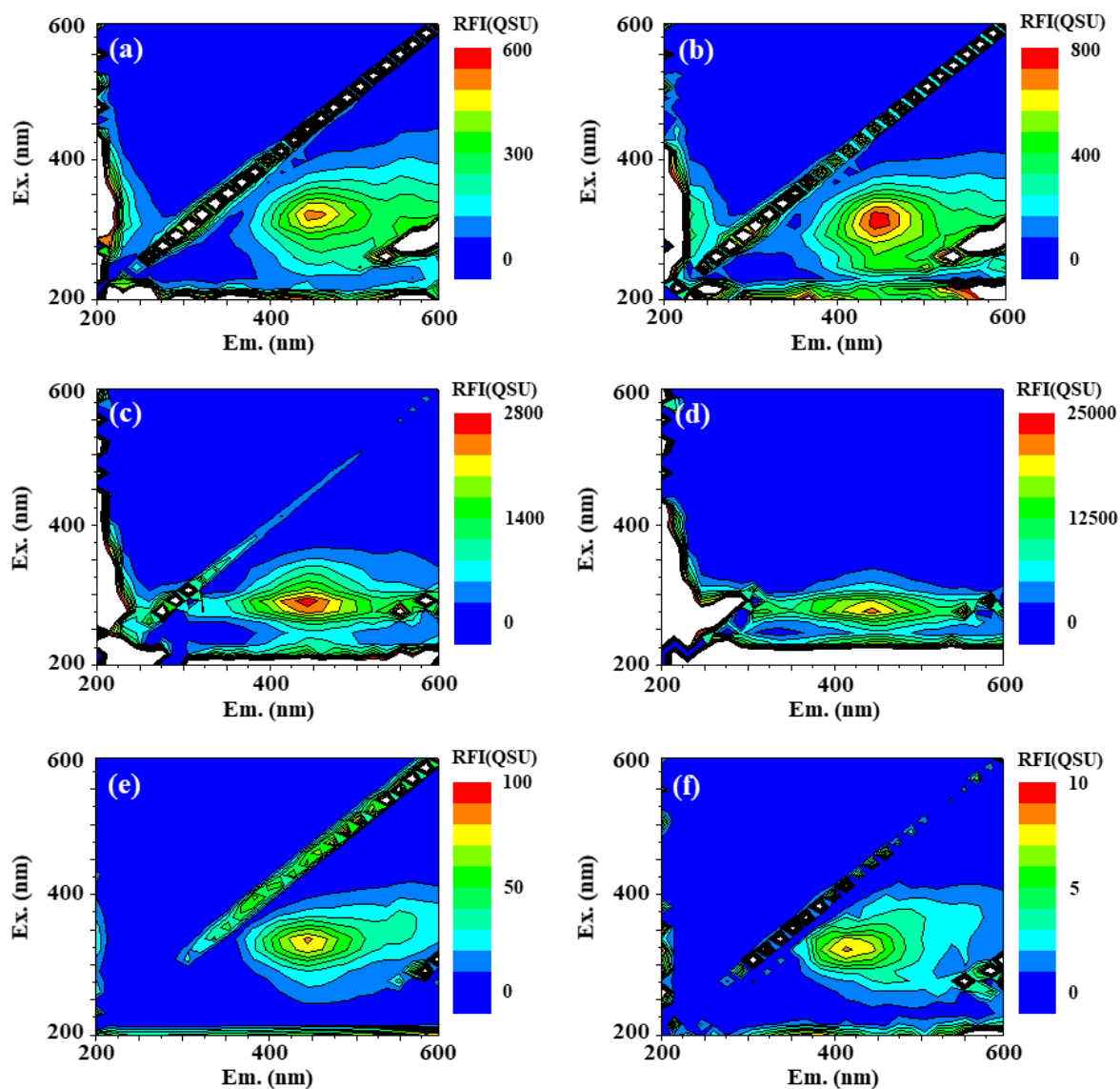


Figure S3. The 3DEEM fluorescence spectrum of water-extracted from (a) UB1 at subsurface in 2011 and thermally treated samples ((b) 120 °C, (c) 150 °C (d) 175 °C (e) 200 °C (f) 250 °C for 60min).