

Supporting Information for Crystal Growth & Design

**Coordination Frameworks Containing Magnetic Single Chain of
Imidazoledicarboxylate-Bridged Cobalt(II)/Nickel(II): Syntheses,
Structures, and Magnetic Properties**

Wenbo Wang,[†] Ruiying Wang,[‡] Lina Liu,[†] and Benlai Wu^{*†}

[†]College of Chemistry and Molecular Engineering, Zhengzhou University,
Zhengzhou 450001, P. R. China

[‡]School of Chemical Engineering, Henan Vocational College of applied technology,
Zhengzhou 450042, P. R. China

Table S1. Selected bond lengths (Å) and angles (°) for **1–5**.

1			
Co1–O5A	2.118(4)	Co1–O(5)	2.118(4)
Co1–N1A	2.120(4)	Co1–N1	2.120(4)
Co1–O1	2.129(3)	Co1–O1A	2.129(3)
Co2–O(6)	2.062(4)	Co2–O6B	2.062(4)
Co2–N2	2.147(4)	Co2–N2B	2.147(4)
Co2–O4B	2.149(4)	Co2–O4	2.149(4)
O5A–Co1–O5	180.0	O5A–Co1–N1A	92.3(2)
O5–Co1–N1A	87.7(2)	O5A–Co1–N1	87.7(2)
O5–Co1–N1	92.3(2)	N1A–Co1–N1	180.0
O5A–Co1–O1	88.8(2)	O5–Co1–O1	91.2(2)
N1A–Co1–O1	100.7(1)	N1–Co1–O1	79.3(1)
O5A–Co1–O1A	91.2(2)	O5–Co1–O1A	88.8(2)

N1A-Co1-O1A	79.3(1)	N1-Co1-O1A	100.7(1)
O1-Co1-O1A	180.0	O6-Co2-O6B	180.0(1)
O6-Co2-N2	91.0(2)	O6B-Co2-N2	89.0(2)
O6-Co2-N2B	89.0(2)	O6B-Co2-N2B	91.0(2)
N2-Co2-N2B	180.0(1)	O6-Co2-O4B	89.6(2)
O6B-Co2-O4B	90.5(2)	N2-Co2-O4B	101.3(1)
N2B-Co2-O4B	78.7(1)	O6-Co2-O4	90.5(2)
O6B-Co2-O4	89.6(2)	N2-Co2-O4	78.7(1)
N2B-Co2-O4	101.3(1)	O4B-Co2-O4	180.0(2)

2

Ni1-N1A	2.066(3)	Ni1-N1	2.066(3)
Ni1-O(1)	2.102(3)	Ni1-O1A	2.102(3)
Ni1-O5A	2.109(3)	Ni1-O5	2.109(3)
Ni2-O6	2.071(3)	Ni2-O6B	2.071(3)
Ni2-O4B	2.092(3)	Ni2-O4	2.092(3)
Ni2-N2B	2.104(3)	Ni2-N2	2.104(3)
N1A-Ni1-N1	180.0	N1A-Ni1-O1	99.3(1)
N1-Ni1-O1	80.7(1)	N1A-Ni1-O1A	80.7(1)
N1-Ni1-O1A	99.3(1)	O1-Ni1-O1A	180.0
N1A-Ni1-O5A	91.8(1)	N1-Ni1-O5A	88.2(1)
O1-Ni1-O5A	87.9(1)	O1A-Ni1-O5A	92.1(1)
N1A-Ni1-O5	88.2(1)	N1-Ni1-O5	91.8(1)
O1-Ni1-O5	92.1(1)	O1A-Ni1-O5	87.9(1)
O5A-Ni1-O5	180.0(1)	O6-Ni2-O6B	180.0(2)
O6-Ni2-O4B	89.5(1)	O6B-Ni2-O4B	90.5(1)
O6-Ni2-O4	90.5(1)	O6B-Ni2-O4	89.5(1)
O4B-Ni2-O4	180.0(1)	O6-Ni2-N2B	91.7(1)
O6B-Ni2-N2B	88.3(1)	O4B-Ni2-N2B	80.7(1)
O4-Ni2-N2B	99.3(1)	O6-Ni2-N2	88.3(1)

O6B-Ni2-N2	91.7(1)	O4B-Ni2-N2	99.3(1)
O4-Ni2-N2	80.7(1)	N2B-Ni2-N2	180.0(1)

3

Co1-N1A	2.126(4)	Co1-N1	2.126(4)
Co1-O1A	2.143(4)	Co(1)-O(1)	2.143(4)
Co1-N4B	2.155(6)	Co1-N3	2.158(7)
N1A-Co1-N1	179.7(2)	N1A-Co1-O1A	80.3(1)
N1-Co1-O1A	99.7(1)	N1A-Co1-O1	99.7(1)
N1-Co1-O1	80.3(1)	O1A-Co1-O1	176.6(2)
N1A-Co1-N4B	89.8(1)	N1-Co1-N4B	89.8(1)
O1A-Co1-N4B	91.7(1)	O1-Co1-N4B	91.7(1)
N1A-Co1-N3	90.3(1)	N1-Co1-N3	90.2(1)
O1A-Co1-N3	88.3(1)	O1-Co1-N3	88.32(10)
N4B-Co1-N3	180.0(1)		

4

Co1-N7A	2.088(7)	Co1-N4	2.120(7)
Co1-N1	2.129(6)	Co1-O4B	2.137(5)
Co1-N2B	2.142(6)	Co1-O1	2.170(6)
N7A-Co1-N4	92.1(3)	N7A-Co1-N1	96.4(3)
N4-Co1-N1	163.9(3)	N7A-Co1-O4B	90.5(2)
N4-Co1-O4B	98.1(3)	N7A-Co1-O1	92.4(2)
N1-Co1-O4B	95.5(2)	N7A-Co1-N2B	169.0(3)
N4-Co1-N2B	86.9(3)	N1-Co1-N2B	87.4(2)
O4B-Co1-N2B	78.7(2)	N4-Co1-O1	87.4(3)
N1-Co1-O1	78.6(2)	O4B-Co1-O1	173.7(2)
N2B-Co1-O1	98.5(2)		

5

Ni1-N4	2.055(8)	Ni1-N7A	2.080(7)
Ni1-N2B	2.094(7)	Ni1-N1	2.108(9)

Ni1-O1	2.11(1)	Ni1-O4B	2.12(1)
N4-Ni1-N7A	91.8(4)	N4-Ni1-N2B	95.3(3)
N7A-Ni1-N2B	166.5(2)	N4-Ni1-N1	169.5(2)
N7A-Ni1-N1	87.2(4)	N2B-Ni1-N1	87.9(4)
N4-Ni1-O1	89.6(4)	N7A-Ni1-O1	96.5(3)
N2B-Ni1-O1	95.0(3)	N1-Ni1-O1	80.1(3)
N4-Ni1-O4B	93.3(4)	N7A-Ni1-O4B	87.7(3)
N2B-Ni1-O4B	80.5(3)	N1-Ni1-O4B	97.1(3)
O1-Ni1-O4B	174.9(1)		

Symmetry codes: A = $2 - x, -y, 1 - z$; B = $1 - x, 1 - y, 1 - z$ for **1**. A = $-x, 2 - y, 1 - z$; B = $1 - x, 1 - y, 1 - z$ for **2**. A = $x, 1/2 - y, 1 - z$; B = $-1 + x, y, z$ for **3**. A = $1 - x, 1 - y, 1 - z$; B = $3/2 - x, -1/2 + y, 3/2 - z$ for **4**. A = $-x, 1 - y, -z$; B = $1/2 - x, 1/2 + y, 1/2 - z$; C = $1 - x, -y, -z$ for **5**.

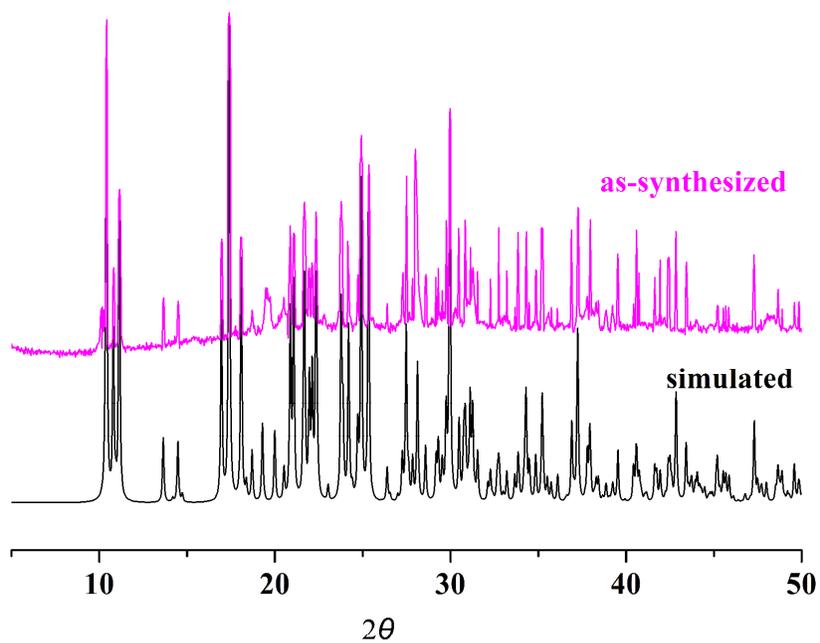


Figure S1. As-synthesized and simulated PXRD patterns of compound **1**.

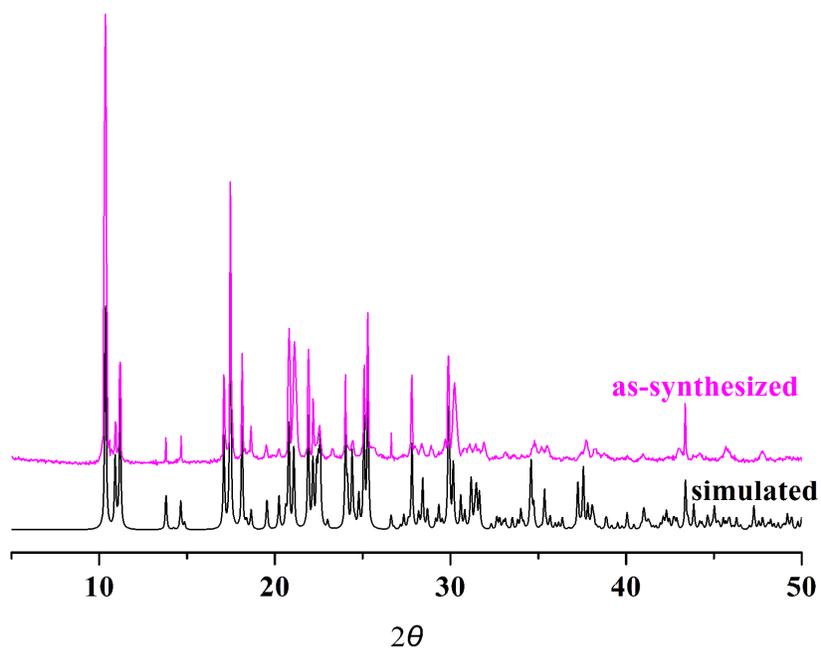


Figure S2. As-synthesized and simulated PXR D patterns of compound 2.

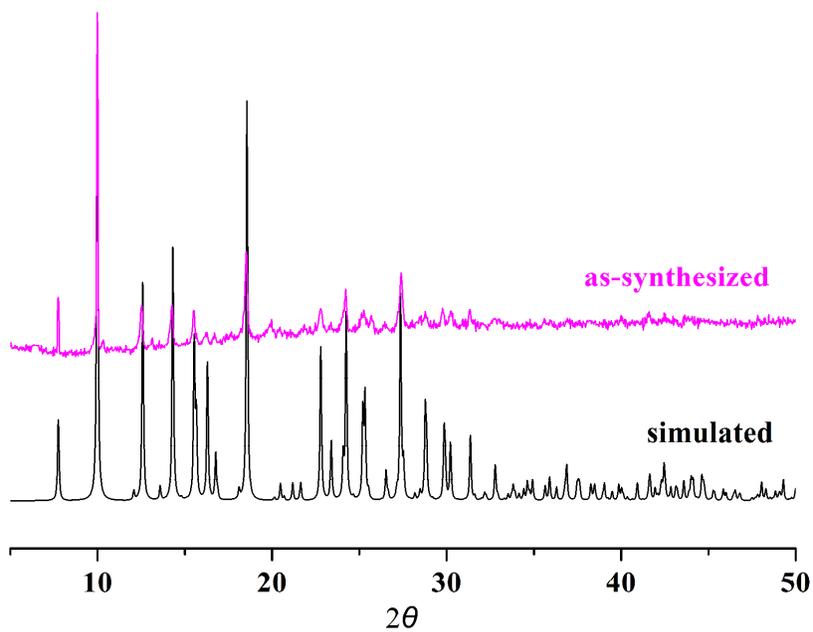


Figure S3. As-synthesized and simulated PXR D patterns of compound 3.

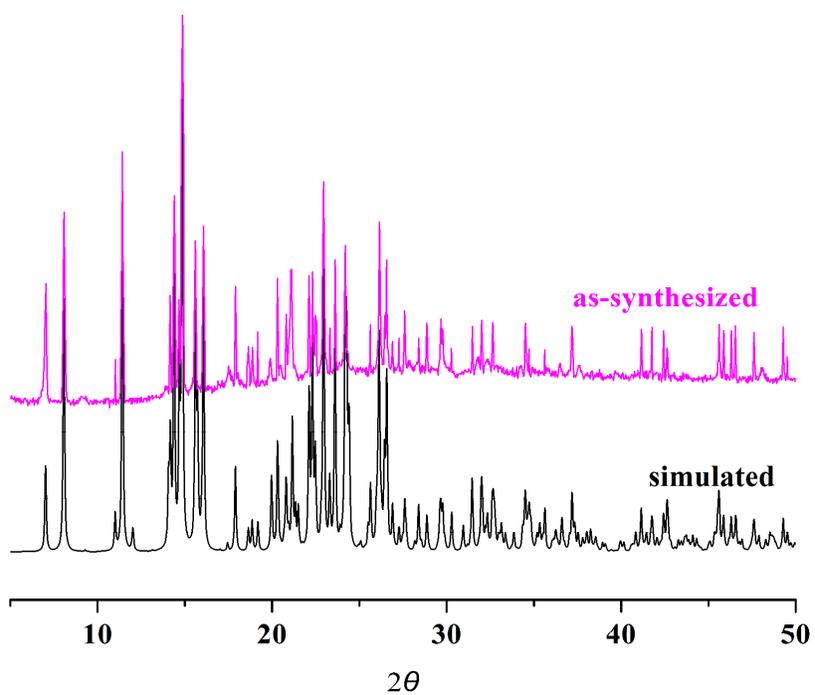


Figure S4. As-synthesized and simulated PXRD patterns of compound 4.

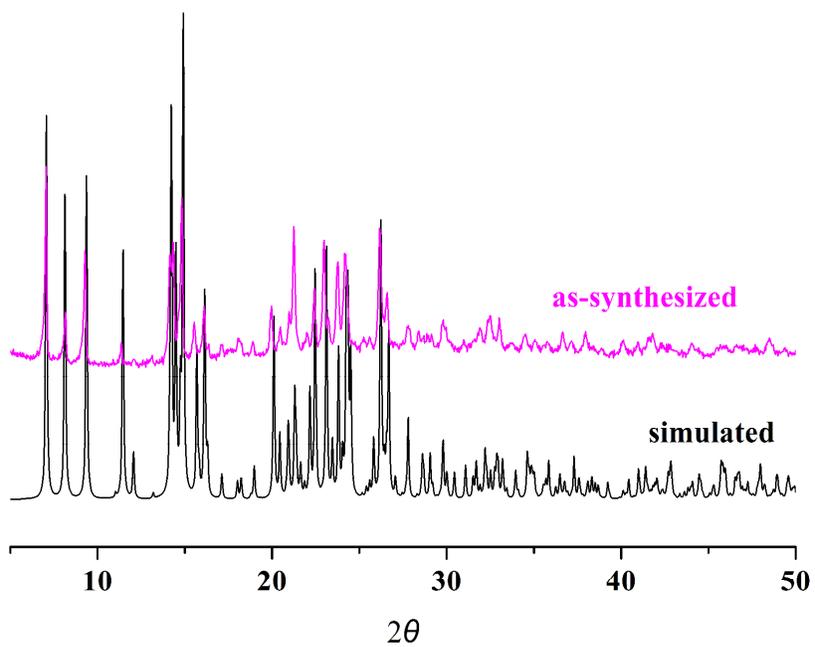


Figure S5. As-synthesized and simulated PXRD patterns of compound 5.

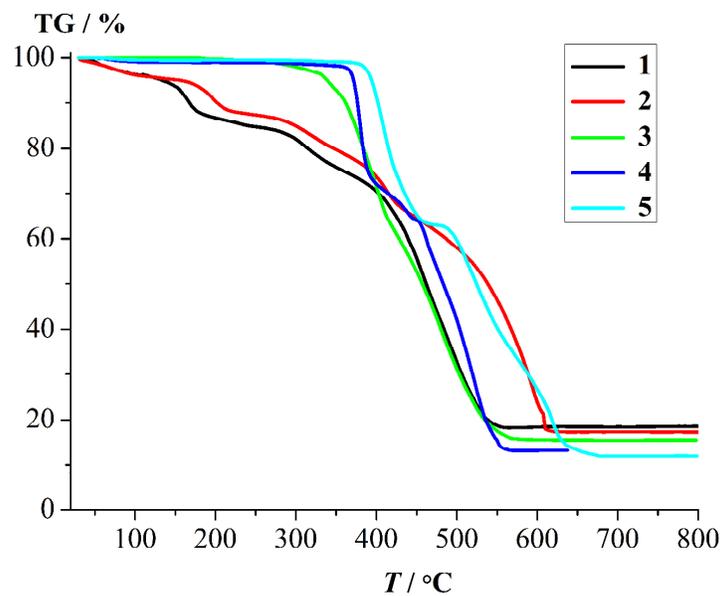


Figure S6. TGA curves of compounds 1–5.

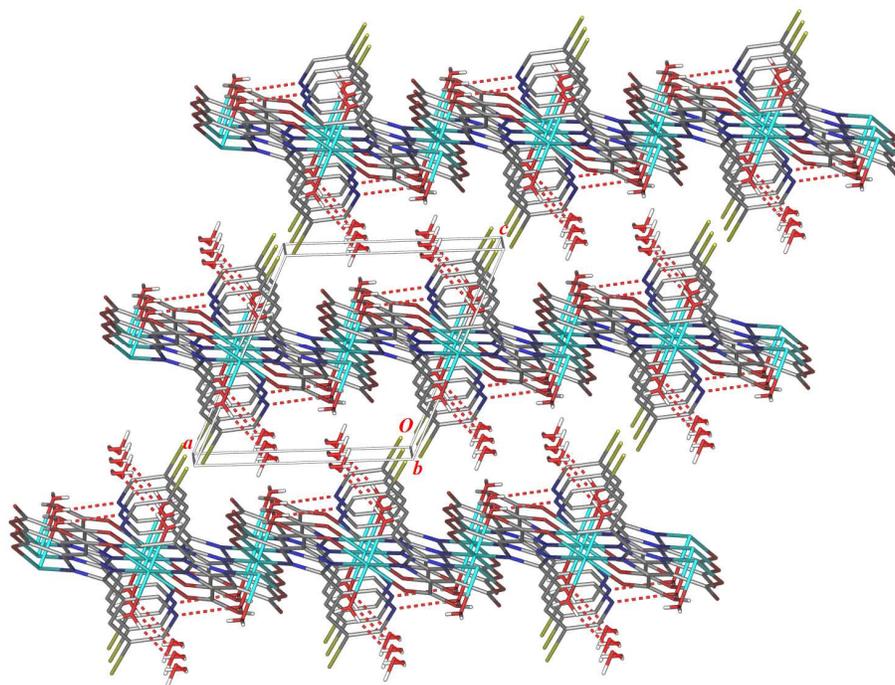


Figure S7. View of the stacking of 2D supramolecular frameworks in **1**, showing lattice water molecules being anchored between layers through hydrogen-bonding interactions.

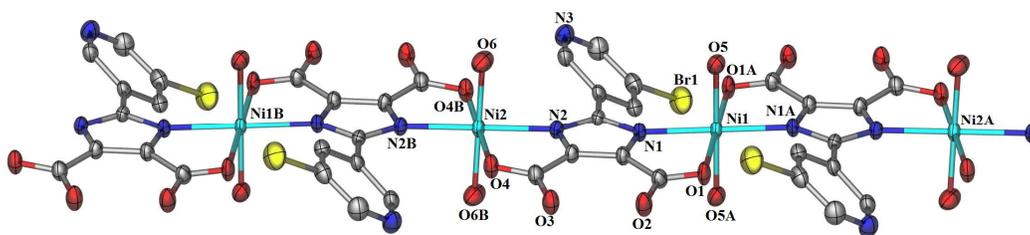


Figure S8. View of the typically linear chain structure in **2** formed by the imidazoledicarboxylates of $(HL)^{2-}$ ligands bis-chelating two independent Ni^{II} , showing the coordination environments of Ni1 and Ni2. Symmetry codes: (A) $-x, 2 - y, 1 - z$; (B) $1 - x, 1 - y, 1 - z$.

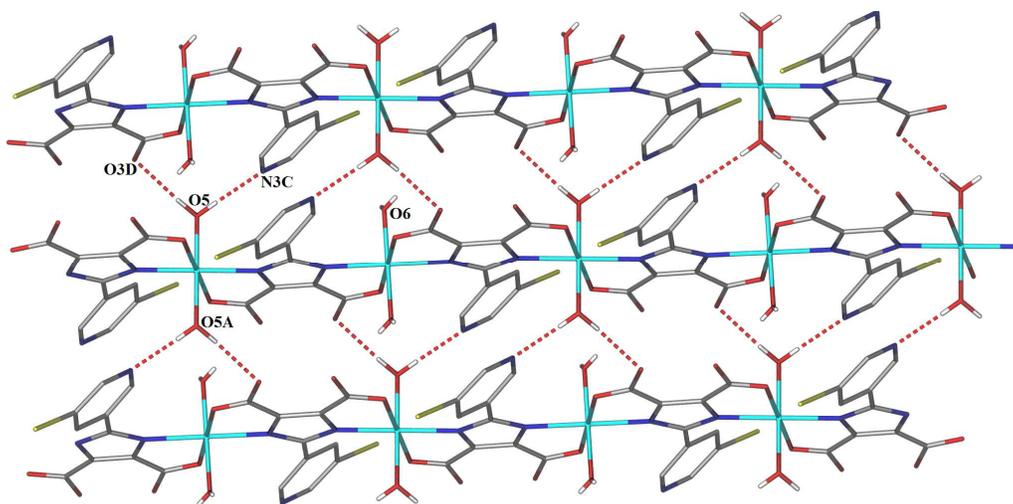


Figure S9. View of the 2D supramolecular framework of **2** formed by interchain hydrogen-bonding interactions. Symmetry codes: (A) $-x, 2 - y, 1 - z$; (C) $-1 + x, y, z$; (D) $-x, 1 - y, 1 - z$.

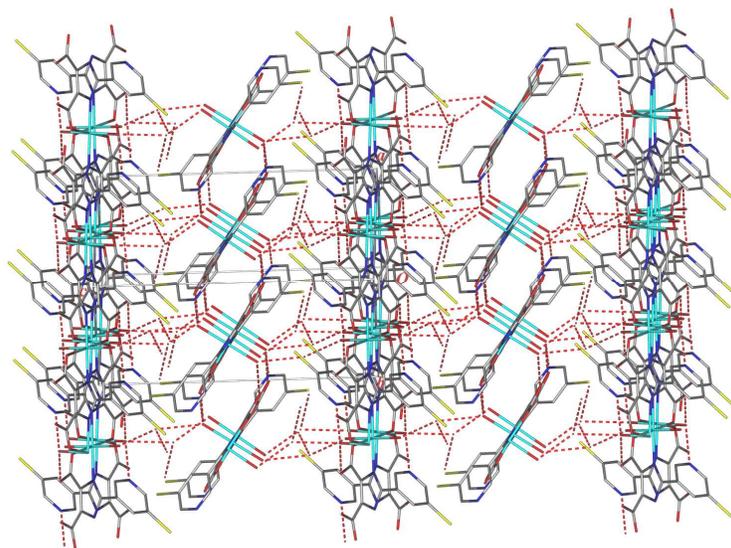


Figure S10. View of the stacking of 2D supramolecular frameworks in **2**, showing lattice water molecules being anchored between layers through hydrogen-bonding interactions.

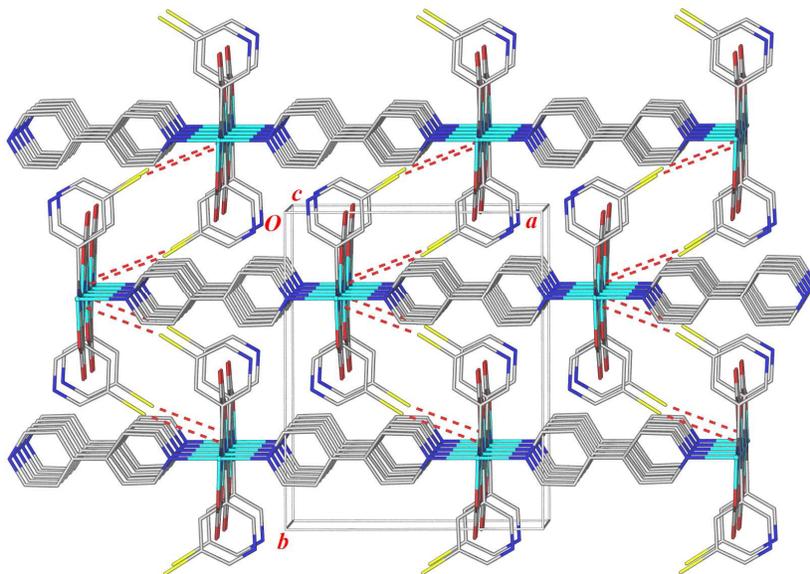


Figure S11. View of the 3D supramolecular framework of **3** formed through interlayered Br $\cdots\pi$ interactions.

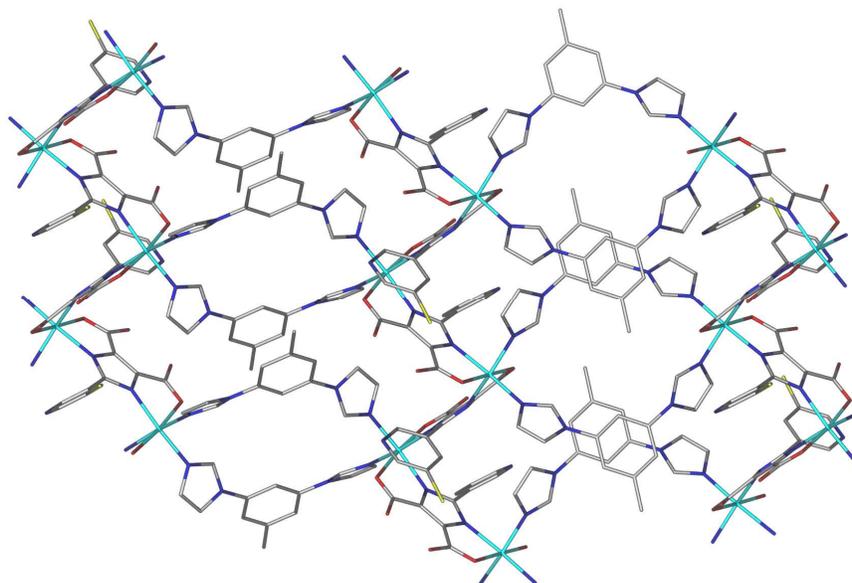


Figure S12. Helically constructed mesolayer in **4**, presenting an architecture with 6^3 -topology as simplified the two doubly-bridged pbim ligands as one spacer.

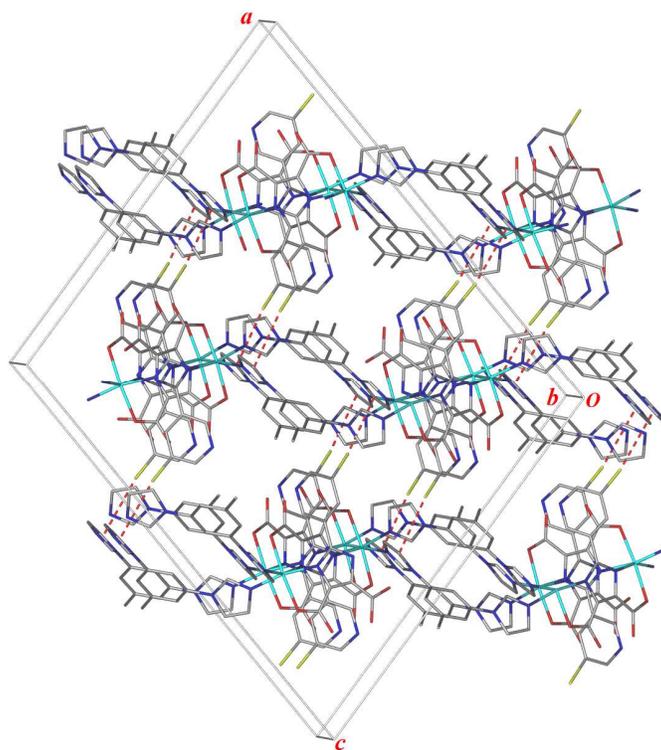


Figure S13. View of the 3D supramolecular framework of **4** formed through interlayered $\text{Br} \cdots \pi$ interactions.

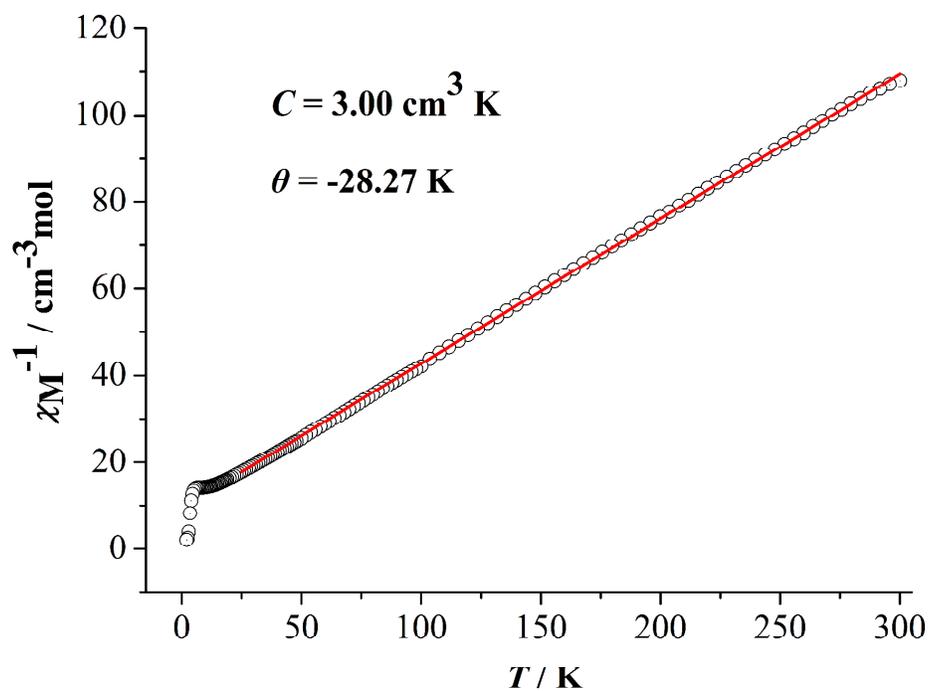


Figure S14. χ_M^{-1} vs T plot with the theoretical fit (—) for 1.

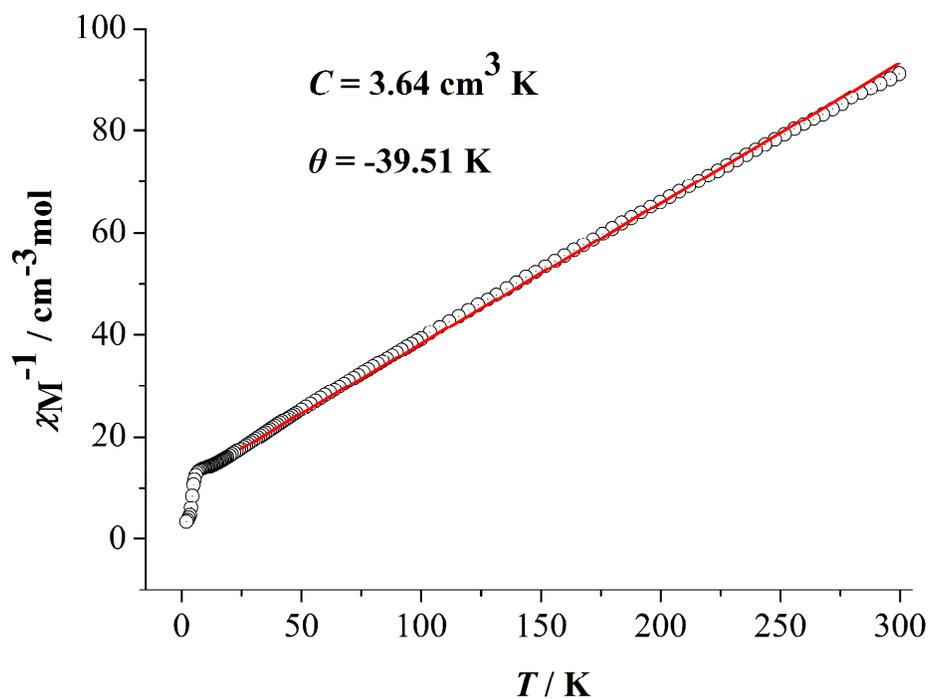


Figure S15. χ_M^{-1} vs T plot with the theoretical fit (—) for 3.

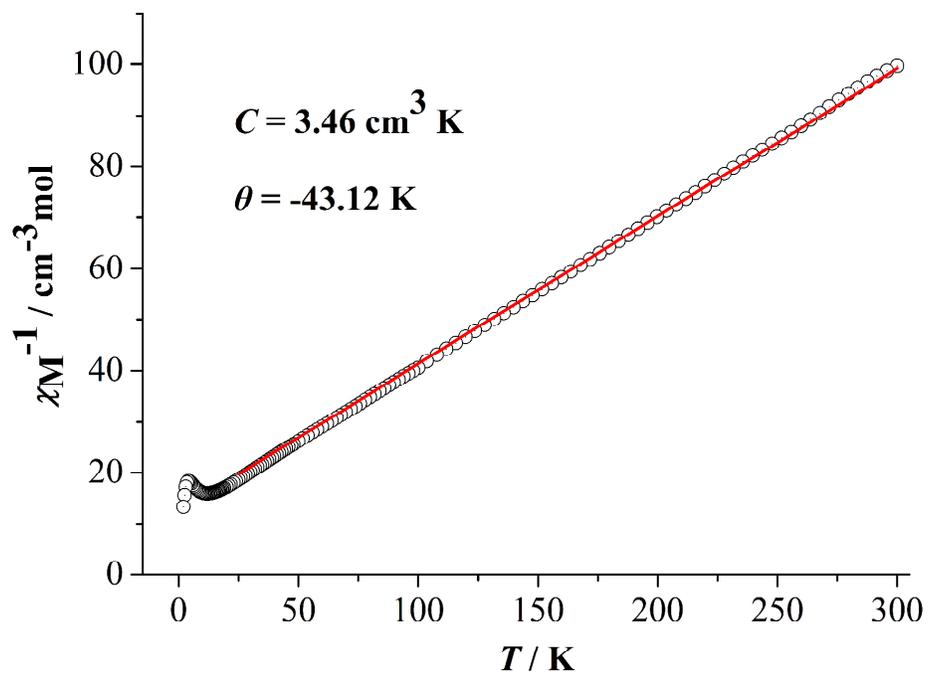


Figure S16. χ_M^{-1} vs T plot with the theoretical fit (—) for 4.

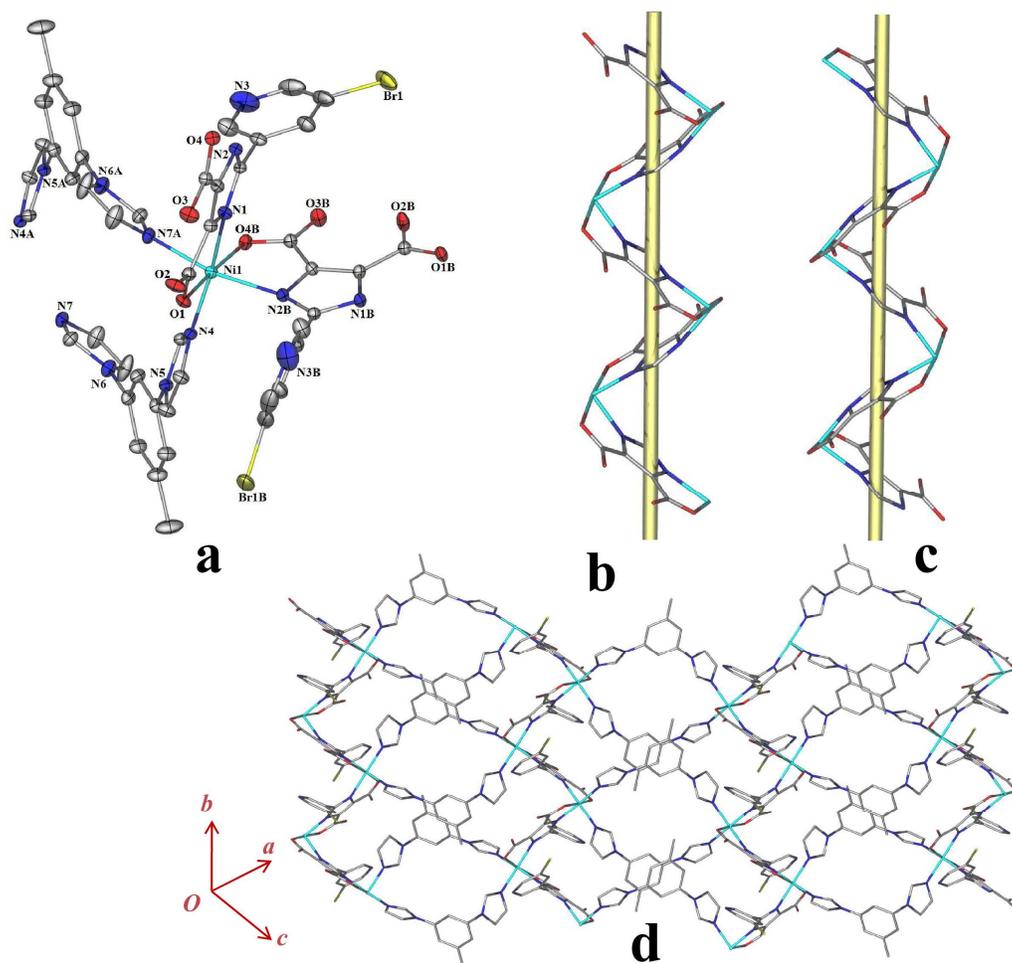


Figure S17. View of the coordination environment of Ni^{II} (a), *b* axially extended left- and right-handed helices constructed by the imidazolecarboxylates of (HL)²⁻ ligands bis-chelating Ni^{II} ions with 5-bromo-pyridin-3-yl groups being omitted for clarity (b and c), and helically constructed mesolayer in **5** (d). Symmetry code: (A) $-x, 1 - y, -z$; (B) $1/2 - x, 1/2 + y, 1/2 - z$.

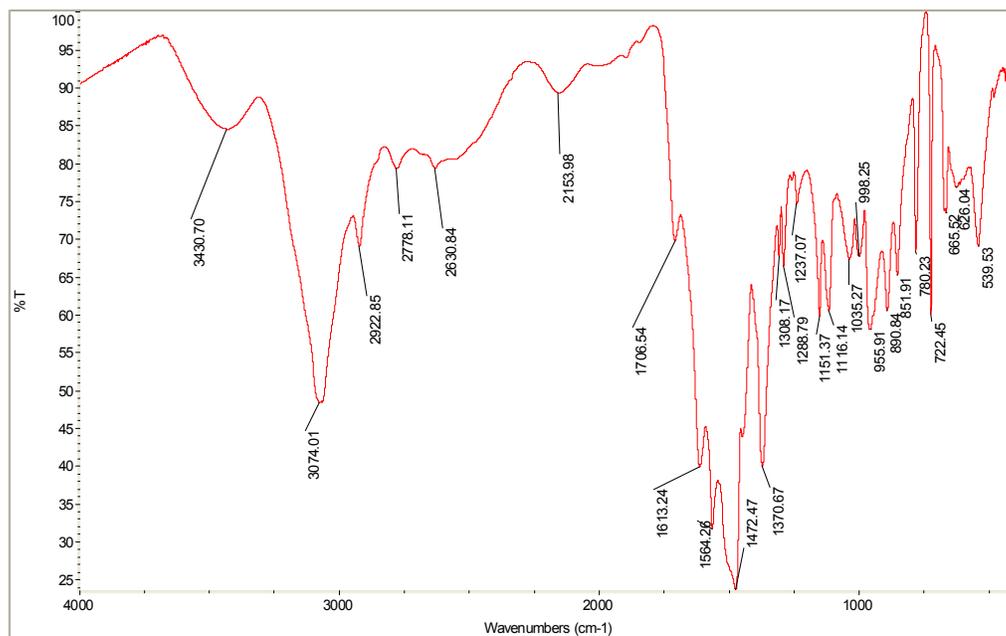


Figure S18. IR spectrum of free ligand H₃L.

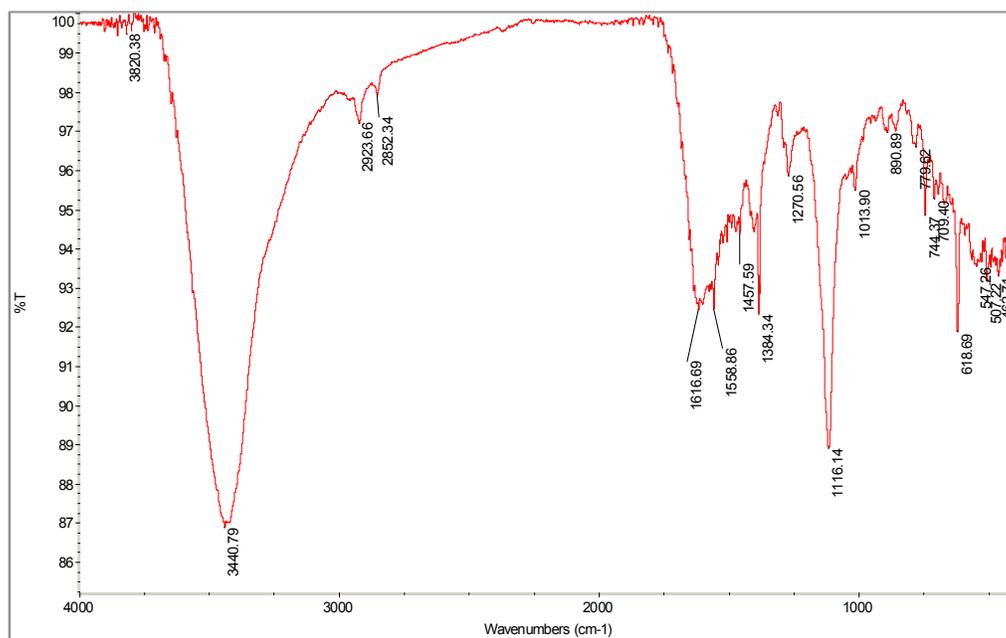


Figure S19. IR spectrum of compound 1.

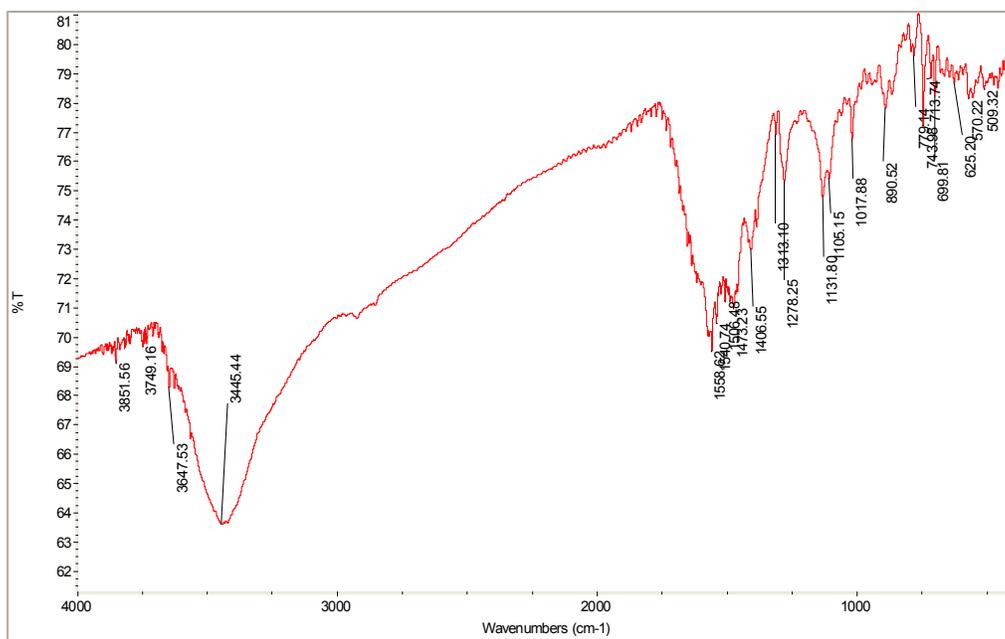


Figure S20. IR spectrum of compound 2.

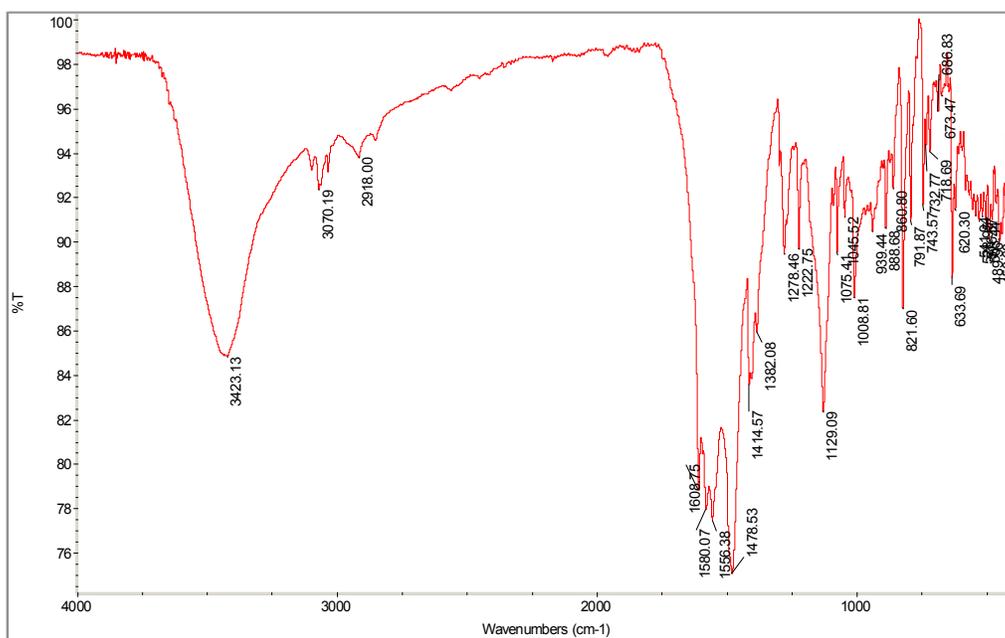


Figure S21. IR spectrum of compound 3.

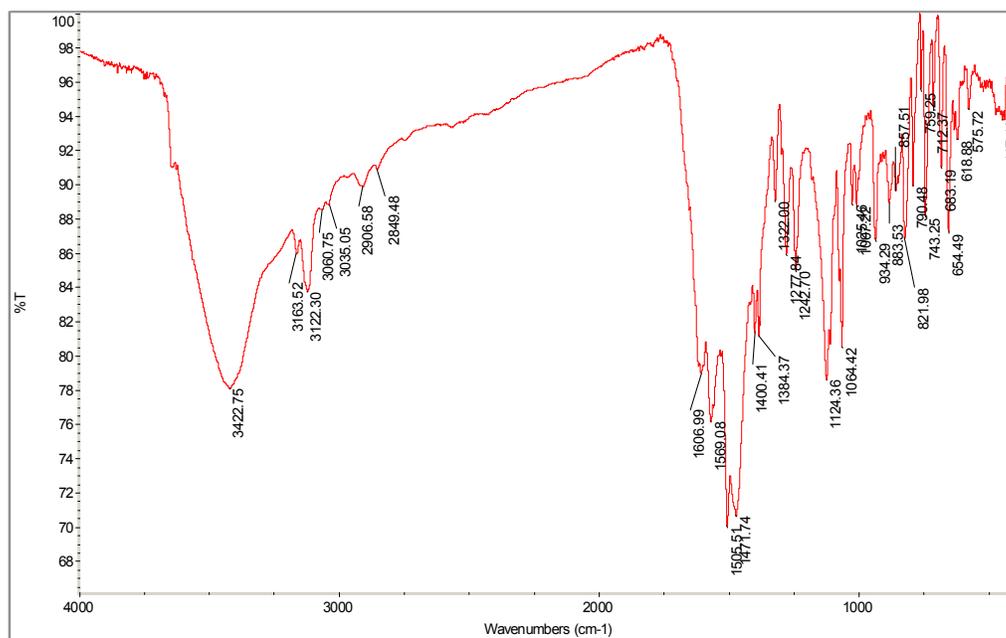


Figure S22. IR spectrum of compound 4.

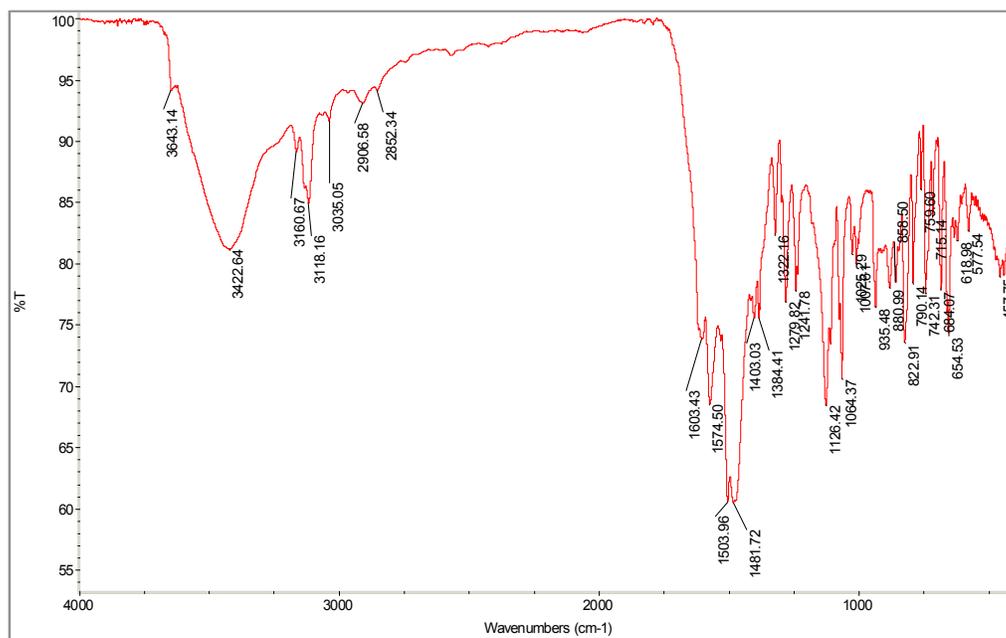


Figure S23. IR spectrum of compound 5.