

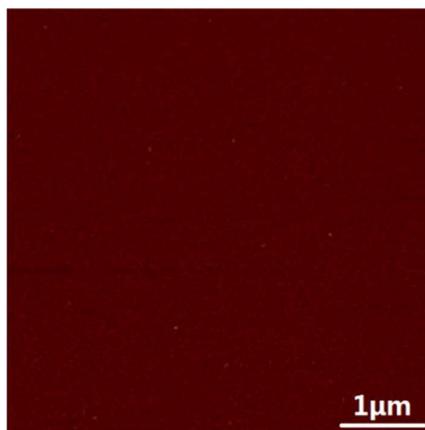
## Supporting Information

Bio-inspired synthesis of Au nanostructures templated from amyloid  $\beta$  peptide  
assembly with enhanced catalytic activity

*Yonghai Feng,<sup>†</sup> Huijie Wang,<sup>†</sup> Jie Zhang,<sup>†</sup> Yongxiu Song,<sup>†</sup> Minjia Meng,<sup>‡</sup> Jianli Mi,<sup>†</sup>  
Hengbo Yin,<sup>‡</sup> Lei Liu,<sup>\*†</sup>*

<sup>†</sup>Institute for Advanced Materials, School of Materials Science and Engineering,  
Jiangsu University, Zhenjiang, China, 212013

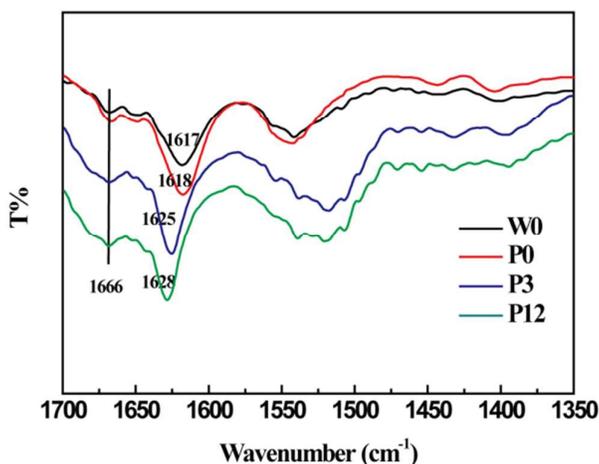
<sup>‡</sup>School of Chemistry and Chemical Engineering, Jiangsu University, Zhenjiang,  
China, 212013



**Figure S1.** AFM image of P0 monomer.

### FTIR analysis

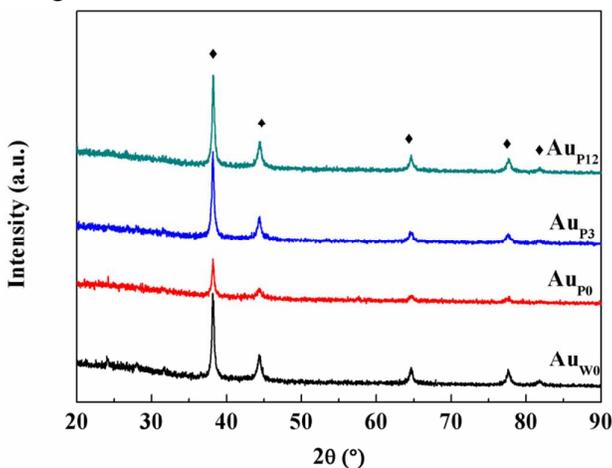
For the P0 and W0 solution, the peaks at 1618 nm and 1617 nm of the FTIR spectra, respectively, could be ascribed to the disordered structure of  $A\beta_{25-35}$  monomer. For the P3 and P12 solution, the peak at 1625 nm and 1628 nm of the FTIR spectra, respectively, could be ascribed to the ordered  $\beta$ -sheet structure of  $A\beta_{25-35}$  nanofibrils. The result indicated the secondary structures of the  $A\beta_{25-35}$  monomer and  $A\beta_{25-35}$  nanofibrils were different.



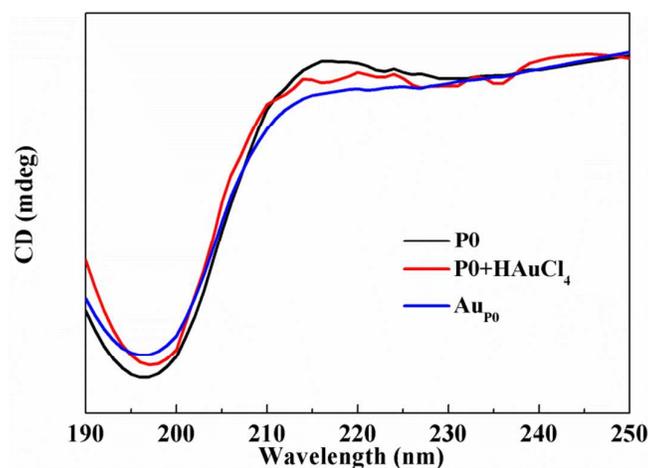
**Figure S2.** FTIR spectra of W0, P0, P3, and P12 solution.

### XRD analysis

Five peaks in the XRD spectra of  $Au_{W0}$ ,  $Au_{P0}$ ,  $Au_{P3}$ , and  $Au_{P12}$  appearing at  $2\theta = 38.18^\circ$ ,  $44.39^\circ$ ,  $66.58^\circ$ ,  $77.55^\circ$ , and  $81.72^\circ$  were indexed as the (111), (200), (220), (311), and (222) planes of the face centered-cubic (fcc) gold (JCPDS 04-0784), respectively, indicating that metallic Au were formation after  $NaBH_4$  reduction.



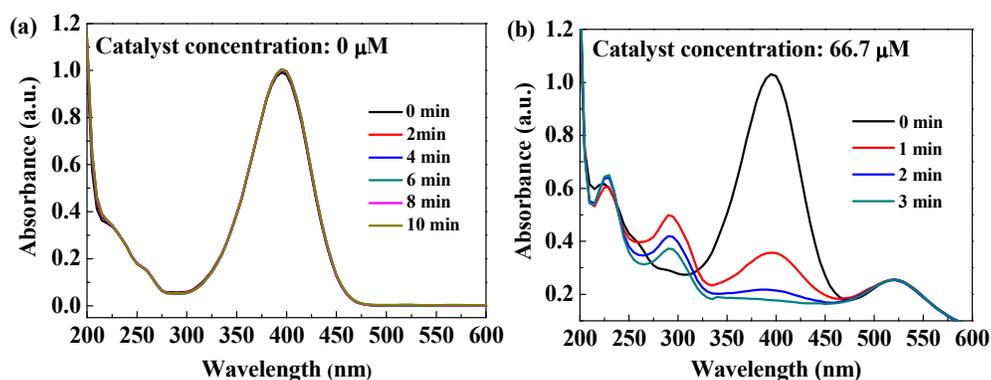
**Figure S3.** XRD spectra of  $Au_{P0}$ ,  $Au_{P3}$ ,  $Au_{P12}$ , and  $Au_{W0}$ . ♦  $Au^0$ .

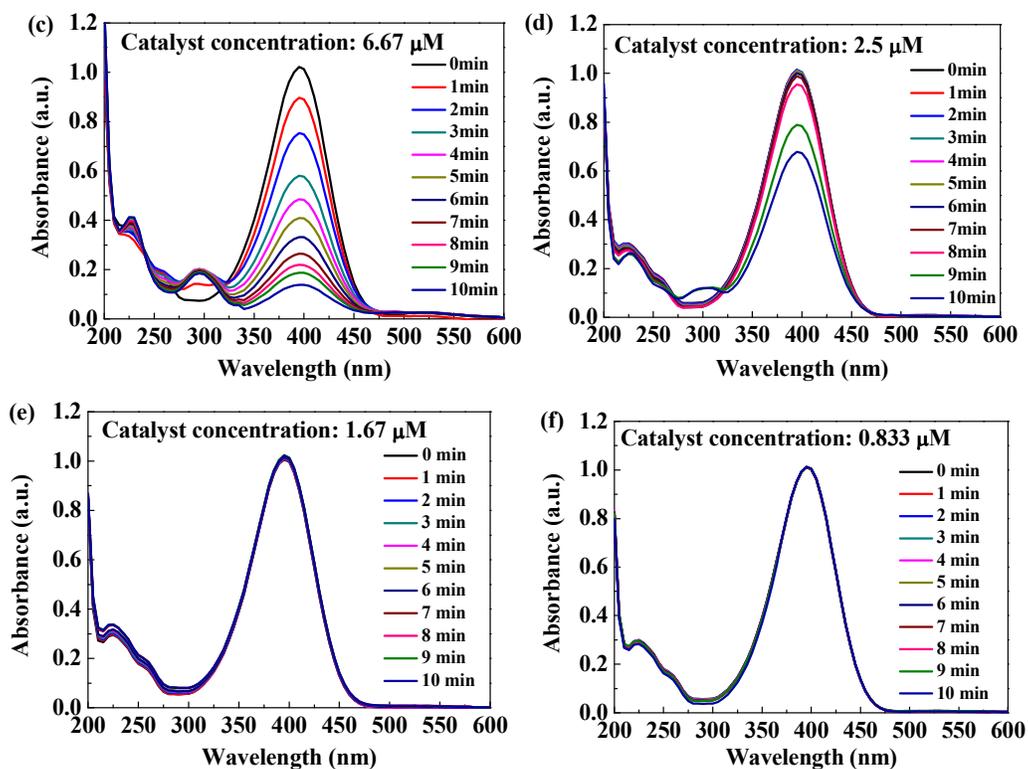


**Figure S4.** CD spectra of P0, P0+HAuCl<sub>4</sub>, and Au<sub>P0</sub> solution.

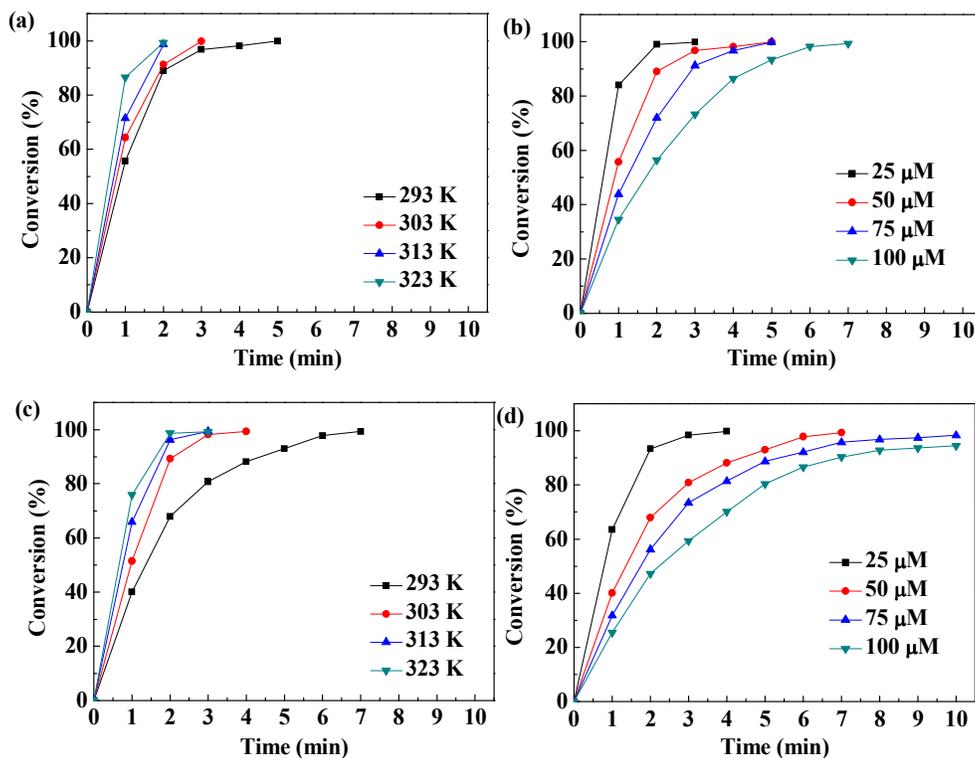
### Limit of catalyst concentration

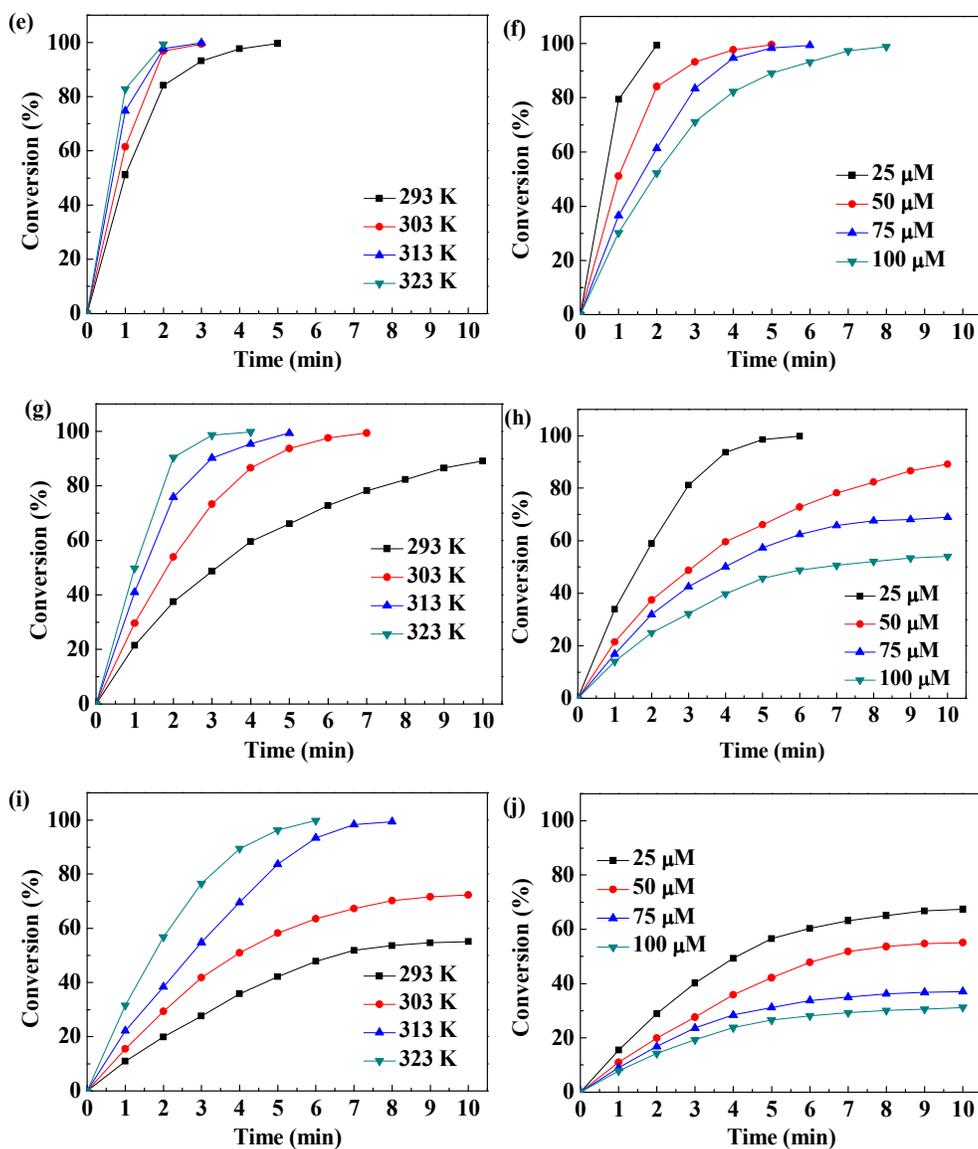
The limit of catalyst concentration for the catalytic reduction of 4-nitrophenol is shown in Figure S5. Without the addition of Au catalyst, the reduction of 4-nitrophenol could hardly proceed (Figure S5a). However, after the addition of Au, the reduction of 4-nitrophenol can proceed significantly (Figure S5b). When we lower the Au concentration from 66.7 to 2.5  $\mu\text{M}$ , the conversion of 4-nitrophenol became very slow, but still proceeded with the conversion of 33.6% for 10 min (Figure S5d). Further decreasing the catalyst concentration to less than 1.67  $\mu\text{M}$ , the conversion of 4-nitrophenol could hardly proceed (Figure S5e and f), meaning that the limit of the catalyst concentration was about 1.7  $\mu\text{M}$ . The result indicated that the A $\beta$  templated-Au nanostructures are highly reactive catalysts.





**Figure S5.** Absorption peak at 400 nm of 4-nitrophenol vs catalyst concentration. Reaction condition: Au<sub>P0</sub> as catalyst, Au concentration, 66.7–0.833  $\mu\text{M}$ ; 4-nitrophenol concentration, 50  $\mu\text{M}$ ; reaction temperature, 293 K.





**Figure S6.** Effects of reaction temperature and 4-nitrophenol concentration on the conversions of 4-nitrophenol catalyzed by Au<sub>P0</sub> (a, b), Au<sub>P3</sub> (c, d), Au<sub>P12</sub> (e, f), Au<sub>W0</sub> (g, h), and Au<sub>0</sub> (i, j) nanostructures, respectively. Reaction conditions: (a, c, e, g, I, k, m, and o) 4-nitrophenol concentration, 50 μM; catalyst, 63.3 μM; (b, d, f, h, j, n, and p) reaction temperature, 293–323 K; reaction temperature, 293 K; catalyst, 63.3 μM; 4-nitrophenol concentration 25–100 μM.