

Modular Synthesis of β -Amino Boronate Peptidomimetics

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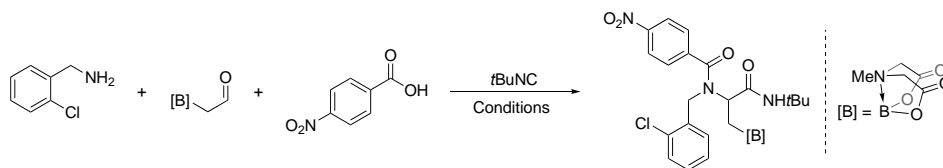
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Reaction Optimization

Table S1. Optimization experiments for the Ugi-reaction with α -borylaldehydes.

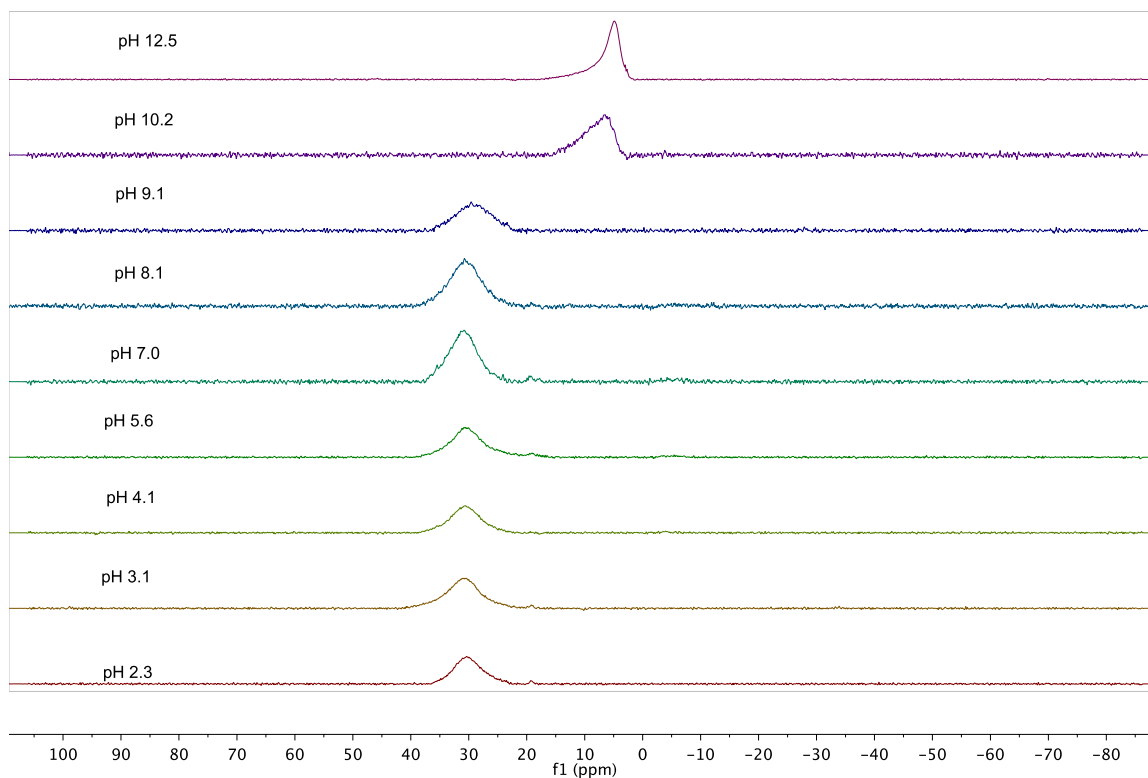


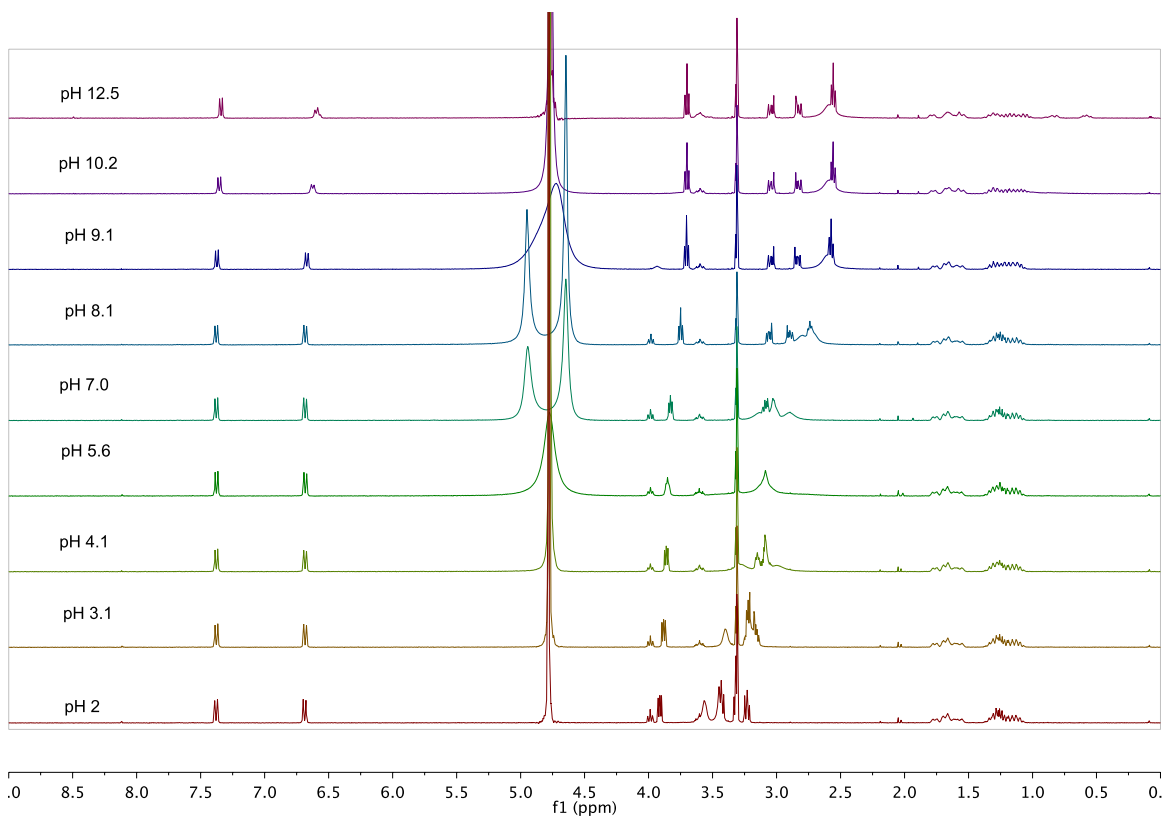
Entry	Solvent (M)	Conditions	T (°C)	Time (h)	Yield ^a
1	MeCN (1.0 M)	1.0 equiv amine and carboxylic acid, 1.0 equiv isocyanide	25	24	0%
2	MeCN (0.5 M)	1.0 equiv amine and carboxylic acid, 1.0 equiv isocyanide	25	24	50%
3	DMSO (0.2 M)	1.0 equiv of amine and carboxylic acid, 1.2 equiv isocyanide	120, μ wave	2	16%
4	DMSO (1.0 M)	1.0 equiv of amine and carboxylic acid, 1.2 equiv isocyanide	100, μ wave	3	29%
5	MeCN (0.1 M)	1.2 equiv amine and carboxylic acid, 2.0 equiv isocyanide	60	24	75%

^aNMR yield of the desired product.

pH Titration

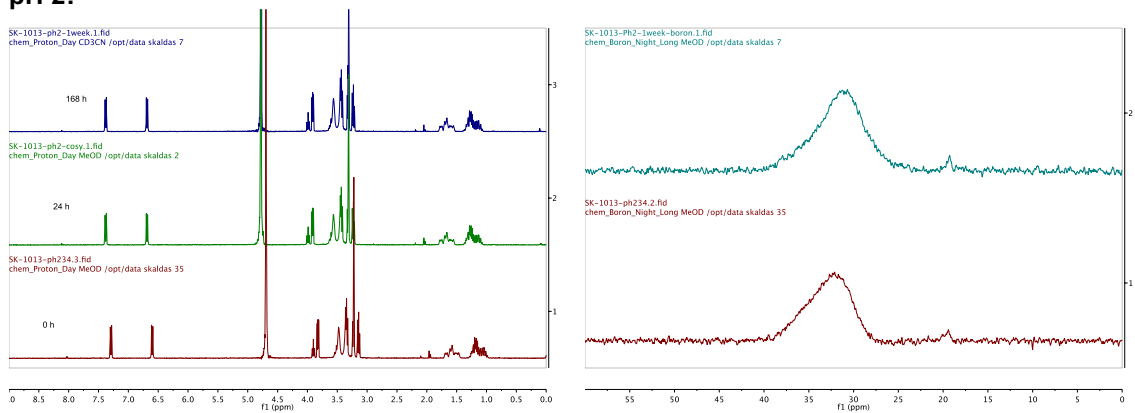
General Procedure: To a solution compound **3b** (20. mg, 0.05 mmol) in 2.0 ml of CD₃OD was added 0.6 ml of HEPES buffer (0.10 M in D₂O) and was left to stir for 5 min. The pH was adjusted to ~3 by dropwise addition of 2.0 M HClO₄ in D₂O. A 200 μ L aliquot was taken for ¹H and ¹¹B NMR. The pH was increased by ~1 pH unit by addition of 2.0 M NaOH(aq) in D₂O until a pH of ~12.5 was reached. At each pH, a 200 μ L aliquot was taken for NMR. This experiment was repeated in duplicate.



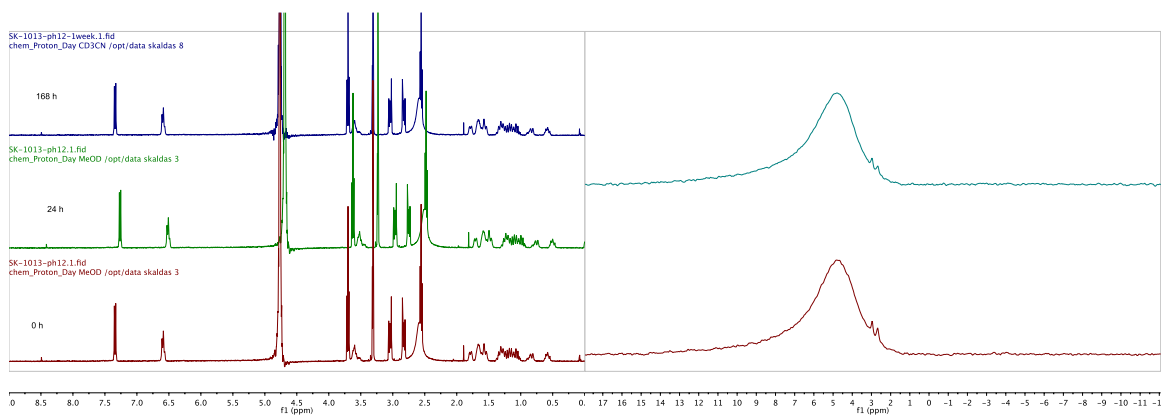


After 1 week at pH 2 and pH 12 in CD_3OD there was no observable decomposition by proton or boron NMR.

pH 2:

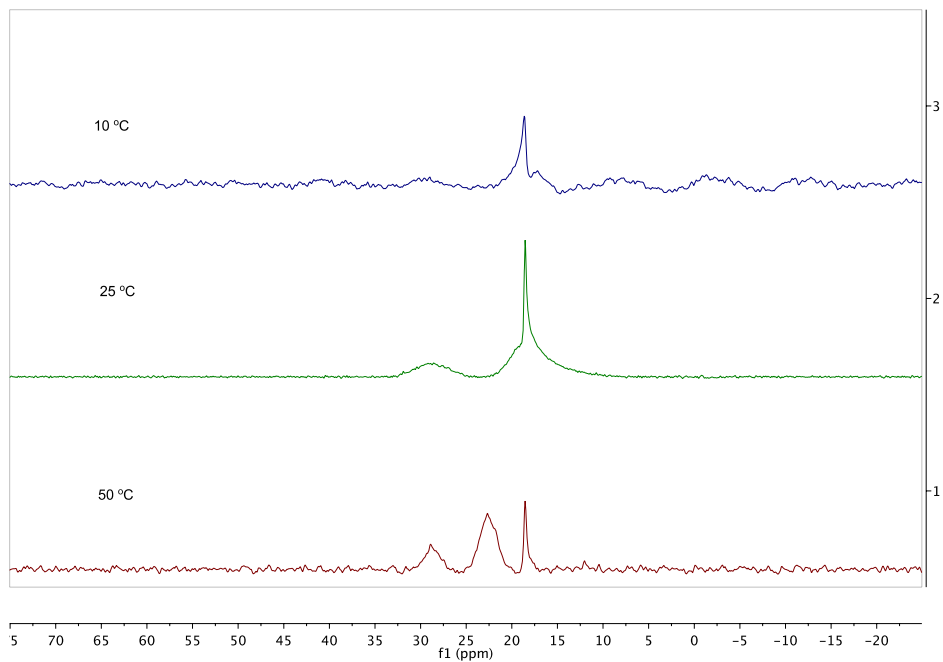
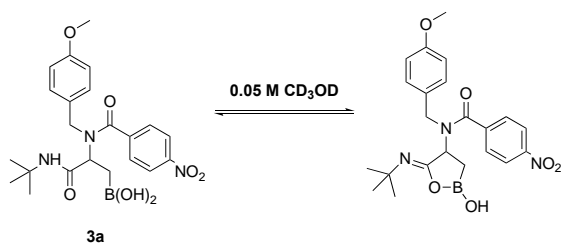


pH 12:



Oxaborolane Study

General Procedure: Compound **3a** was dissolved in deuterated methanol (0.05 M) and ^{11}B NMR (128 MHz) analysis was carried out at 10 °C, 25 °C, and 50 °C.



NMR Spectra

