

## Supporting Information

### **Facile Synthesis and Superior Catalytic Activity of Nano-TiN@N-C for Hydrogen Storage in NaAlH<sub>4</sub>**

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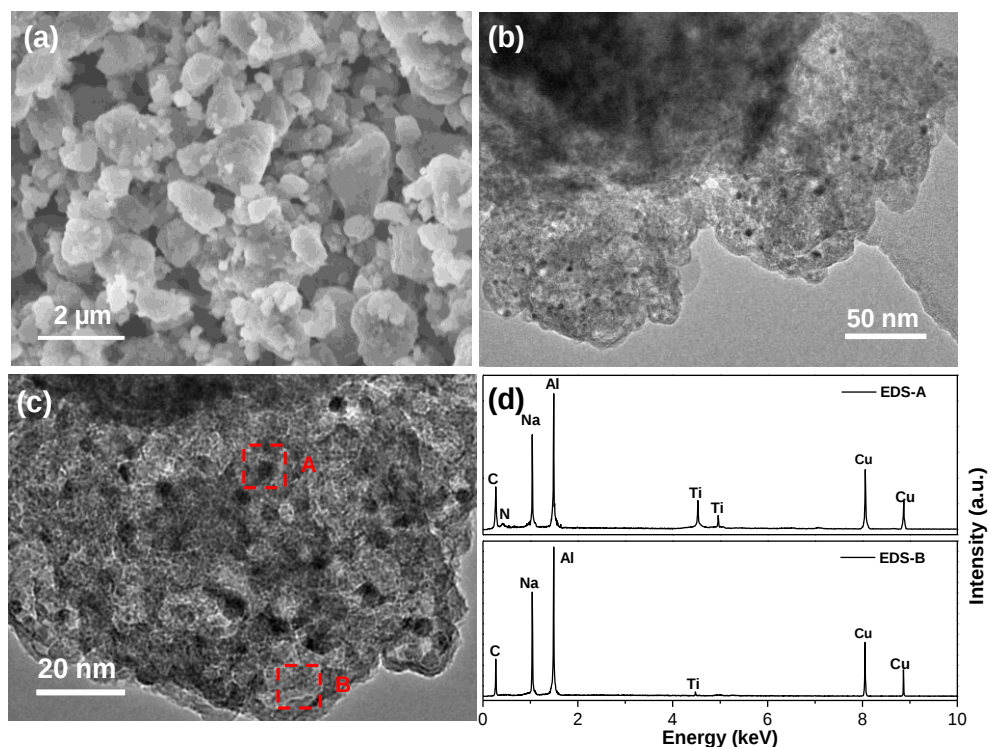
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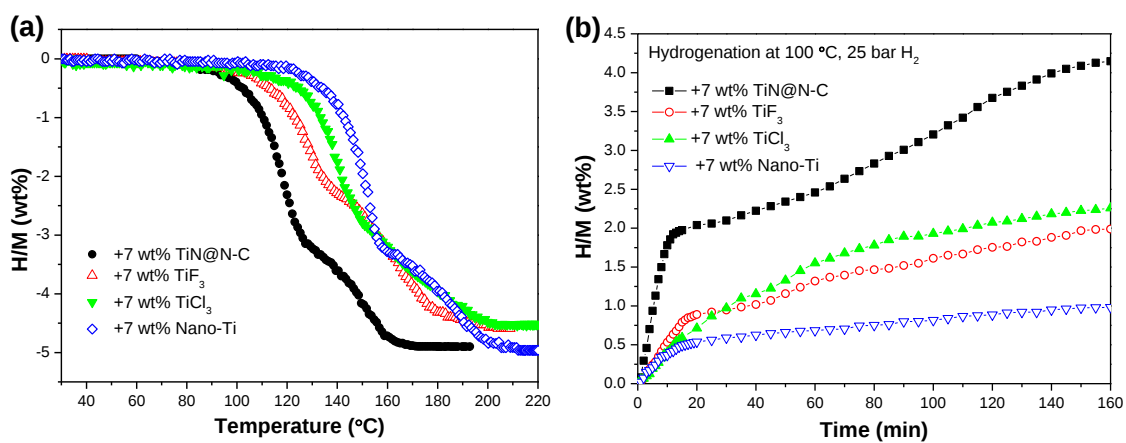
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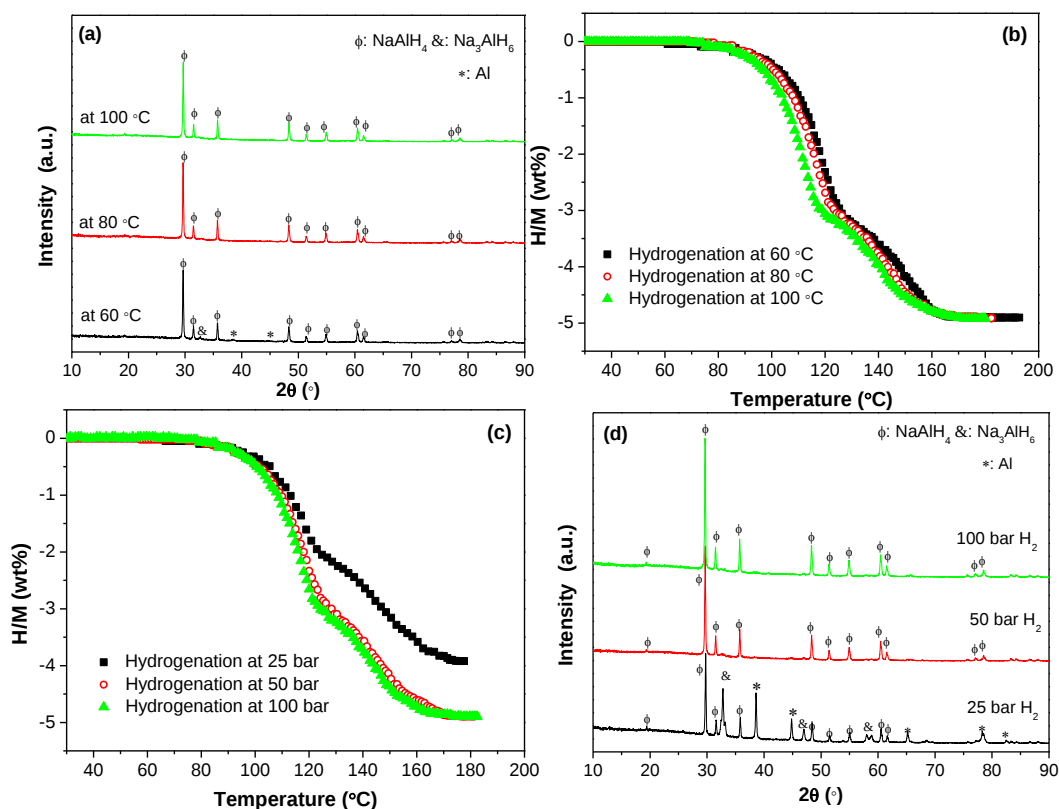
E-mail: luyh@zju.edu.cn (Y.H.L.), mselyf@zju.edu.cn (Y.F.L.).



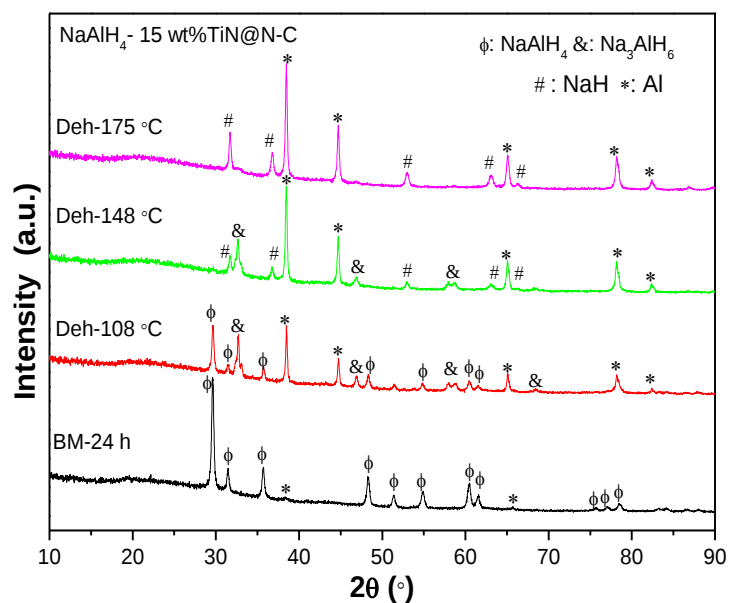
**Figure S1** SEM image (a), TEM images (b, c) and EDS patterns (d) of prepared  $\text{NaAlH}_4$ -7 wt% TiN@N-C sample.



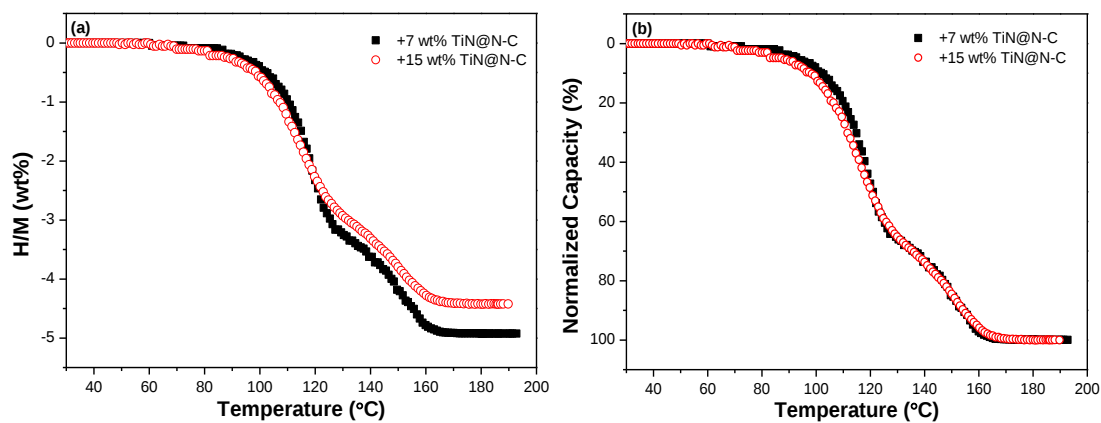
**Figure S2** Volumetric release (a) and isothermal hydrogenation curves (b) of  $\text{NaAlH}_4$  doped with different catalysts.



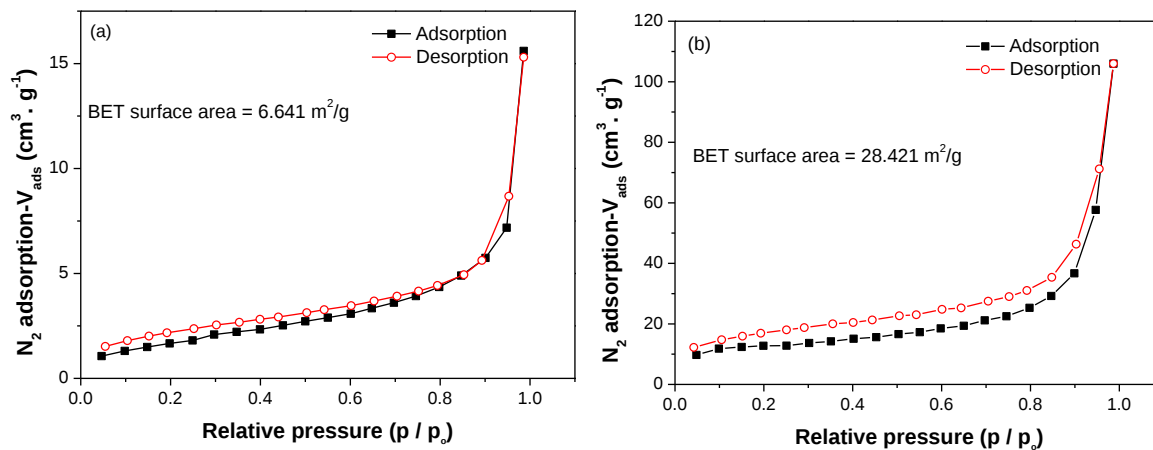
**Figure S3** XRD profiles (a, d) and re-dehydrogenation curves (b, c) of  $\text{NaAlH}_4$ -7 wt%  $\text{TiN@N-C}$  sample hydrogenated isothermally at various conditions.



**Figure S4** XRD patterns of 15wt%  $\text{TiN@N-C}$ -containing  $\text{NaAlH}_4$  samples dehydrogenated at different stages.



**Figure S5** Volumetric release (a) and normalized volumetric release curves (b) of NaAlH<sub>4</sub> doped with 7 wt% and 15 wt% TiN@N-C.



**Figure S6.** N<sub>2</sub> sorption isotherms of the NaAlH<sub>4</sub>-7 wt% TiC@N-C samples after ball milling (a) and 25 cycles (b).