## Intermolecular carbonyl-carbonyl interactions in metal complexes

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## SUPPORTING INFORMATION



**Figure S1**. C=O····C=O dihedral as a function of the intermolecular O····C distance in crystal structures with short R-CO····CO contacts.



**Figure S2**. C=O···C angle as a function of the intermolecular O···C distance in crystal structures with short R-CO···CO contacts.





Pt

**Figure S3**. 0...C bond paths and associated BCPS and 2D electron density contours on the plane defined by the two atoms and the BCP involved in the 0...C=O contact for the three [*trans*-MBr<sub>2</sub>(CO)<sub>2</sub>] (M = Ni, Pd, Pt) complexes.

**Table 1**. Cartesian coordinates of the optimized structures of [*trans*-MBr<sub>2</sub>(CO)<sub>2</sub>] (M = Ni, Pd, Pt) complexes.

| Br | 2.379787  | 1.329317  | -0.028041 |
|----|-----------|-----------|-----------|
| Ni | 0.750574  | -0.317021 | 0.134160  |
| С  | 1.406569  | -1.069929 | -1.397916 |
| Br | -0.772333 | -2.070103 | 0.318801  |
| С  | 0.122979  | 0.450012  | 1.693137  |
| 0  | -0.221671 | 0.894580  | 2.665268  |
| 0  | -1.826875 | 1.637005  | -0.077068 |
| С  | -2.937319 | 1.404616  | -0.505446 |
| С  | -3.338246 | 1.798082  | -1.907445 |
| С  | -3.975657 | 0.691579  | 0.327506  |
| Н  | -3.715259 | 0.741709  | 1.382407  |
| Н  | -4.975033 | 1.097500  | 0.162162  |
| Н  | -3.994695 | -0.360272 | 0.026120  |
| Н  | -2.476636 | 2.172039  | -2.455629 |
| Н  | -3.780749 | 0.948458  | -2.433348 |
| Н  | -4.105247 | 2.576514  | -1.862463 |
| 0  | 1.811133  | -1.533836 | -2.339620 |

## Pd

| Br | 2.552737  | 1.299660  | -0.167724 |
|----|-----------|-----------|-----------|
| Pd | 0.698765  | -0.289272 | 0.162985  |
| С  | 1.445376  | -1.413960 | -1.237936 |
| Br | -1.101410 | -1.950989 | 0.496013  |
| С  | -0.005945 | 0.844761  | 1.611988  |
| 0  | -0.355170 | 1.468560  | 2.475676  |
| 0  | -2.073097 | 1.814383  | -0.149868 |
| С  | -3.094141 | 1.445760  | -0.690069 |
| С  | -3.255535 | 1.496155  | -2.191039 |
| С  | -4.255793 | 0.895991  | 0.102433  |
| Н  | -4.138997 | 1.132481  | 1.157611  |
| Н  | -5.209544 | 1.275525  | -0.269129 |
| Н  | -4.270356 | -0.191802 | -0.013827 |
| Н  | -2.323118 | 1.800912  | -2.660548 |
| Н  | -3.567868 | 0.521364  | -2.573828 |
| Н  | -4.044161 | 2.207012  | -2.452984 |
| 0  | 1.879596  | -2.064780 | -2.044179 |

| Br | -0.127130 | 2.459189  | 0.335279  |
|----|-----------|-----------|-----------|
| Pt | 0.682617  | 0.116618  | 0.107666  |
| С  | 1.788517  | 0.648348  | -1.380459 |
| Br | 1.511966  | -2.213139 | -0.124645 |
| С  | -0.379459 | -0.419450 | 1.662200  |
| 0  | -0.922538 | -0.716779 | 2.598738  |
| 0  | -2.795508 | -0.967687 | 0.254018  |
| С  | -3.688323 | -0.694470 | -0.519356 |
| С  | -3.638796 | 0.536653  | -1.392400 |
| С  | -4.906093 | -1.579593 | -0.653840 |
| Н  | -4.769560 | -2.496451 | -0.085202 |
| Н  | -5.785691 | -1.048087 | -0.279634 |
| Н  | -5.100606 | -1.813373 | -1.703205 |
| Н  | -2.817605 | 1.184000  | -1.093208 |
| Н  | -3.498537 | 0.232276  | -2.433964 |
| Н  | -4.584022 | 1.082343  | -1.348287 |
| 0  | 2.441492  | 0.959762  | -2.243193 |
|    |           |           |           |