

# *Supporting Information*

## Multicolor photoluminescence of a hybrid film via dual-emitting strategy of inorganic fluorescent Au nanocluster and organic room-temperature phosphorescent copolymer

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### **Contents:**

**Figure S1.** Preparation routine of poly-BrNpA.

**Figure S2.** The TEM spectra of AuNC@ histidine.

**Table S1.** The different mole proportions of triadic system, and calculated CIE coordinates from the PL spectra of the PVA composite films shown in Figure 2(c).

**Table S2.** Calculated CIE coordinates from the PL spectra changes under different humidity shown in Figure 2(e).

**Figure S3.** The excitation spectrum of poly-BrNpA,  $\lambda_{em}=580nm$ .

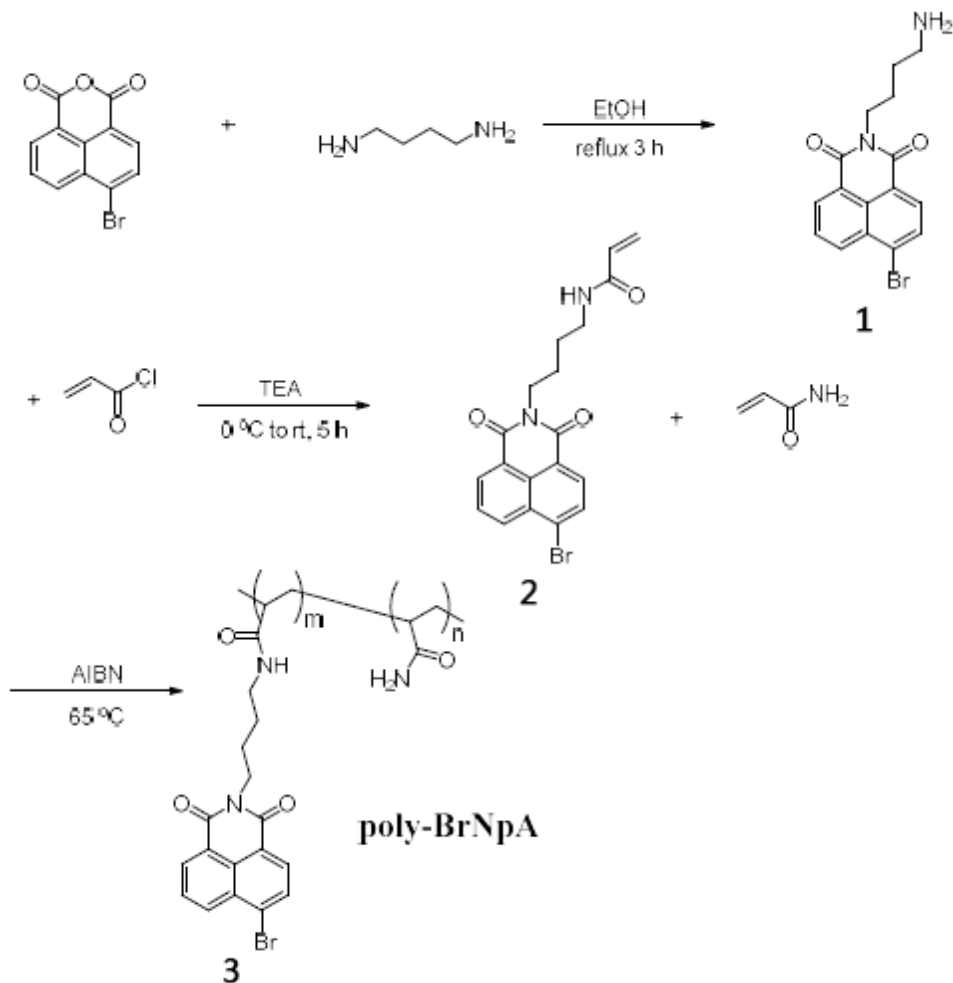
**Figure S4.** The absorption spectrum of poly-BrNpA/AuNC@histidine in different humidity.

**Table S3.** Hydrogen bonding between the components.

**Figure S5.** The luminescence photos of dual-emission films of different proportions, conditions as follows: aqueous solution under daily light, aqueous solution under  $\lambda=365$  nm UV irradiation, drying film under daily light, drying film under  $\lambda=365$  nm UV irradiation (from up to down).

**Figure S6.** The luminescence photos of dual-emission PAM films of different proportions.

**Figure S7.** The luminescence photos of dual-emission PVP films.



**Figure S1.** Preparation routine of poly- BrNpA.

**Synthesis of 2-(4-aminobutyl)-6-bromo-1H-benzo [de]isoquinoline-1,3(2H)-dione. (1)**  $^1\text{H}$  NMR (400 MHz,  $\text{D}_2\text{O}$ ,  $\delta$ ) 8.24 (d,  $J = 8.5$  Hz, 1H, Ar H), 8.19 (d,  $J = 7.4$  Hz, 1H, Ar H), 7.91 (d,  $J = 7.9$  Hz, 1H, Ar H),

7.82 (d, J = 7.9 Hz, 1H, Ar H), 7.64 (t, J = 7.9 Hz, 1H, Ar H), 3.97 (t, J = 6.5 Hz, 2H, CH<sub>2</sub>), 3.06 (t, J = 6.9 Hz, 2H, CH<sub>2</sub>), 1.81-1.66 (m, 4H, CH<sub>2</sub>). <sup>13</sup>C NMR (100 MHz, DMSO-d<sub>6</sub>, δ) 162.97, 162.92, 132.71, 131.67, 131.43, 131.05, 129.78, 129.25, 128.88, 128.25, 122.67, 121.89, 38.55, 24.71. HRMS (ESI) (m/z): [M+H]<sup>+</sup> calcd for [C<sub>16</sub>H<sub>16</sub>BrN<sub>2</sub>O<sub>2</sub>]<sup>+</sup>, 347.0395; found, 347.0384.

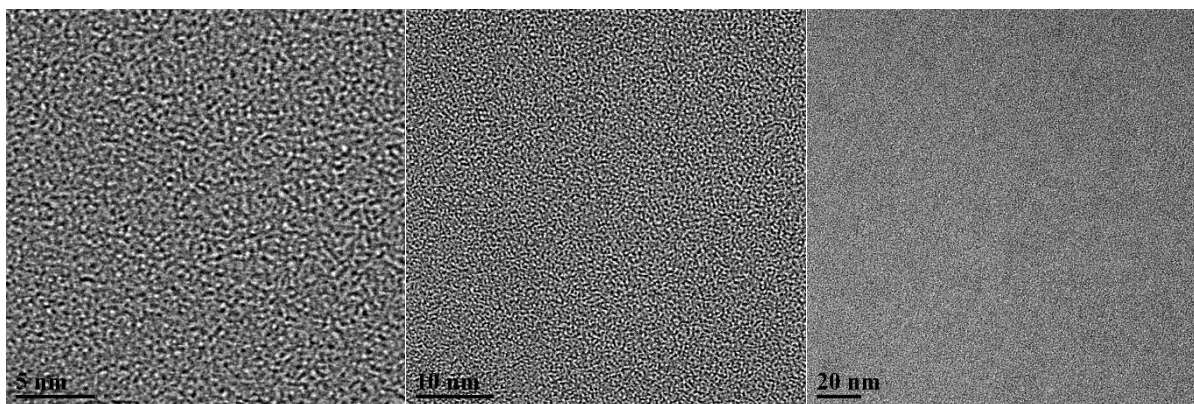
**Synthesis of N-(4-(6-bromo-1, 3-dioxo-1 H-benzo [de]isoquinolin-2(3H)-yl) butyl) acrylamide (2) and poly-BrNpA. (3)**

**(2):** <sup>1</sup>H NMR (400 MHz, DMSO-d<sub>6</sub>, δ) 8.50 (d, J = 7.2 Hz, 1H, Ar H), 8.47 (d, J = 8.4 Hz, 1H, Ar H), 8.26 (d, J = 7.9 Hz, 1H, Ar H), 8.15 (d, J = 7.9 Hz, 1H, Ar H), 8.09 (s, 1H, NH), 7.94 (s, 1H, Ar H), 6.18 (dd, J = 17.1, 10.1 Hz, 1H, CH), 6.03 (dd, J = 17.1, 2.1 Hz, 1H, CH), 5.54 (dd, J = 10.1, 2.1 Hz, 1H, CH), 4.01 (t, J = 7.1 Hz, 2H, CH<sub>2</sub>), 3.15 (dd, J = 12.7, 6.6 Hz, 2H, CH<sub>2</sub>), 1.70 – 1.59 (m, 2H, CH<sub>2</sub>), 1.50 (dd, J = 14.9, 7.1 Hz, 2H, CH<sub>2</sub>). <sup>13</sup>C NMR (100 MHz, DMSO-d<sub>6</sub>, δ) 164.50, 162.89, 162.84, 132.59, 131.84, 131.58, 131.34, 130.96, 129.75, 129.13, 128.78, 128.24, 124.88, 122.70, 121.92, 38.40, 26.73, 25.21. HRMS (ESI) (m/z): [M+H]<sup>+</sup> calcd for [C<sub>19</sub>H<sub>18</sub>BrN<sub>2</sub>O<sub>3</sub>]<sup>+</sup>, 401.0501; found, 401.0505.

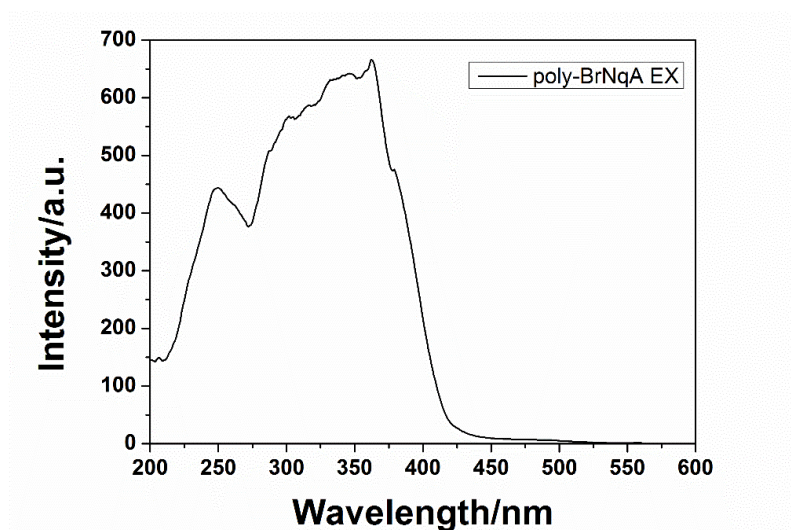
**(3):** <sup>1</sup>H NMR (400 MHz, D<sub>2</sub>O, δ) 8.20-6.91 (m, broad, 5H, aromatic protons), 2.38-2.07 (broad, principal chain protons), 1.82-1.40 (broad, principal chain protons). GPC (H<sub>2</sub>O): Mn (PDI) = 4.6 kDa (6.23).

**Preparation of poly-BrNpA-AuNC@histidine-PAM composite.** The composite was prepared by mixing the PAM (0.14g, Mw=400000-800000), polymer **3** (0.001 g, 0.0025 mmol) and AuNC@histidine (0-1400μl, 8.4mg/ml) in DI water, and then stirring for 15 minutes at room temperature. Adopt a centrifuge method to avoid bubbles. Then rest for 1 hour. Such obtained PAM solution are smeared on glass to prepare PAM films by natural withering.

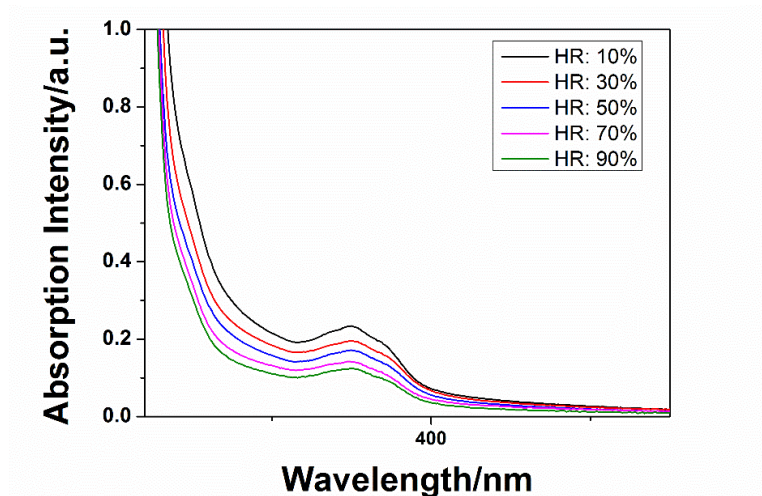
**Preparation of poly-BrNpA-AuNC@histidine-PVP composite.** The composite was prepared by mixing the PVP (0.14g, Mw=50000-60000), polymer **3** (0.001 g, 0.0025 mmol) and AuNC@histidine (100μl, 8.4mg/ml) in DI water, and then stirring for 15 minutes at room temperature. Adopt a centrifuge method to avoid bubbles. Then rest for 1 hour. Such obtained PVP solution are smeared on glass to prepare PVP films by natural withering.



**Figure S2.** The TEM spectra of AuNC@ histidine.



**Figure S3.** The excitation spectrum of poly-BrNpA,  $\lambda_{em}$  =580nm; (phosphorescence mode; excitation slit = 10 nm; emission slit = 10 nm; Measurements were carried out in the amorphous solid state.)



**Figure S4.** The absorption spectrum of poly-BrNpA/AuNC@histidine in different humidity.

**Table S1.** The different mole proportions of triadic system, and calculated CIE coordinates from the PL spectra of the PVA composite films shown in Figure 2(c).

No	poly-BrNpA /mg	AuNC@histidine/ $\mu$ l	PVA/mg	H <sub>2</sub> O/ $\mu$ l	Ratio a'/b' (w/w)	CIE coordinates
1	1	0	140	1400	1:0	(0.4415, 0.3784)
2	1	100	140	1300	1:0.84	(0.3629, 0.3285)
3	1	200	140	1200	1:1.68	(0.331, 0.32)
4	1	300	140	1100	1:2.52	(0.3227, 0.3062)
5	1	400	140	1000	1:3.36	(0.3001, 0.2931)
6	1	500	140	900	1:4.2	(0.2875, 0.2975)
7	1	600	140	800	1:5.06	(0.2785, 0.2895)
8	1	700	140	700	1:5.88	(0.276, 0.2849)
9	1	800	140	600	1:6.72	(0.274, 0.2837)
10	1	900	140	500	1:7.56	(0.2707, 0.2783)
11	1	1000	140	400	1:8.4	(0.2714, 0.2766)
12	1	1100	140	300	1:9.24	(0.2636, 0.2752)
13	1	1200	140	200	1:10.08	(0.2646, 0.2728)
14	1	1300	140	100	1:10.92	(0.2516, 0.2653)
15	0	1400	140	0	0:11.76	(0.2277, 0.2565)

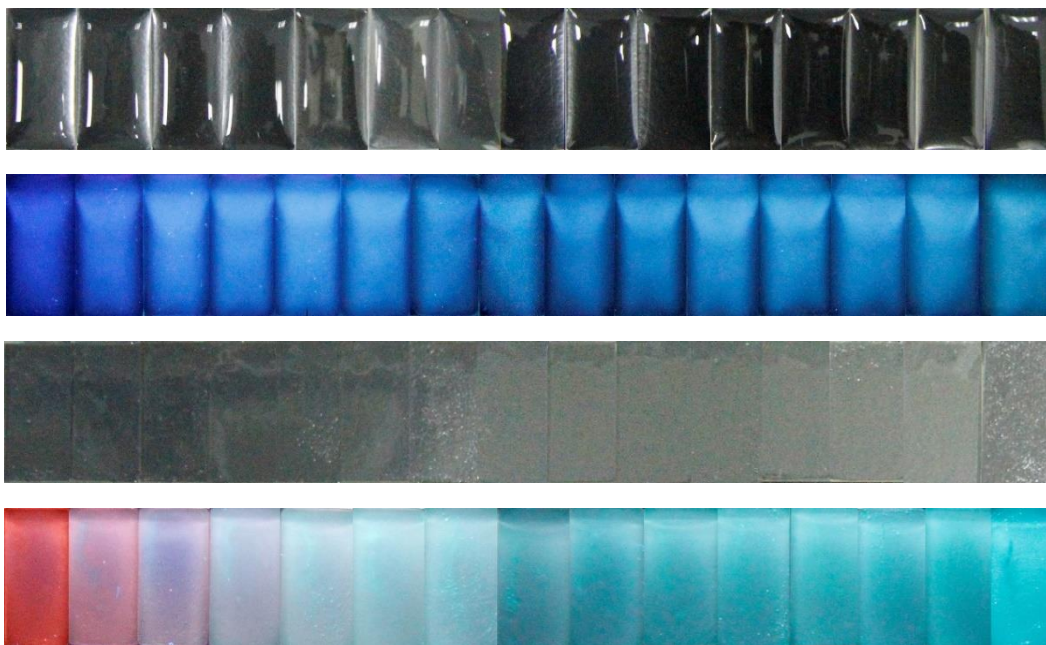
**Table S2.** Calculated CIE coordinates from the PL spectra changes under different humidity shown in

Figure 2(e).

No	humidity	CIE coordinates
1	10%	(0.3961, 0.3296)
2	20%	(0.3698, 0.3163)
3	30%	(0.3639, 0.3116)
4	40%	(0.3336, 0.3101)
5	50%	(0.3136, 0.3074)
6	60%	(0.2992, 0.297)
7	70%	(0.2885, 0.2941)
8	80%	(0.2725, 0.2868)
9	90%	(0.2541, 0.2814)

**Table S3.** Hydrogen bonding between the components.

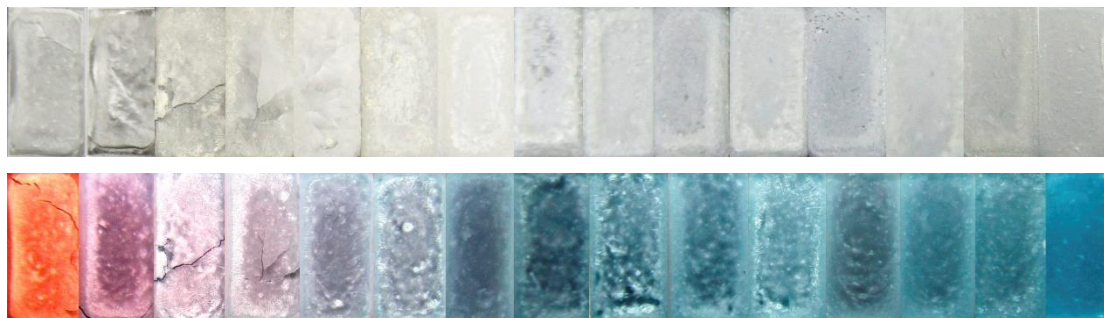
Polymer	AuNC@ histidine	BrNqA	PAM
PVA	✓	✓	✓
PVP	✓		✓
PAM	✓	✓	✓



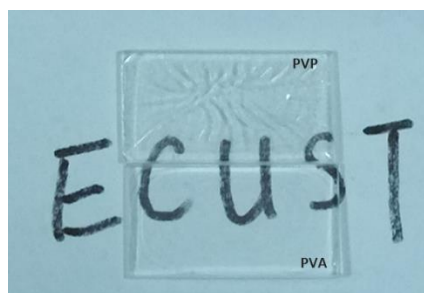
**Figure S5.** The luminescence photos of dual-emission films of different proportions in PVA, conditions as follows: aqueous solution on the quartz plates under daily light, aqueous solution on the quartz plates



under  $\lambda=365$  nm UV irradiation, dry film under daily light, dry film under  $\lambda=365$  nm UV irradiation (from top to down).



**Figure S6.** The luminescence photos of dual-emission PAM films of different proportions in dry film under daily light (top) and dry film under  $\lambda=365$  nm UV irradiation (down)



**Figure S7.** The luminescence photos of dual-emission PVP films in dry film under daily light.