

Supporting Information

Allylic C-S Bond Construction through Metal-Free Direct

Nitroalkene Sulfonation

Xue Lei,[†] Lei Zheng,[†] Chuanxin Zhang,[†] Xiaodong Shi^{*,‡} and Yunfeng Chen^{*,†}

[†]School of Chemistry and Environmental Engineering, Wuhan Institute of Technology, Wuhan 430073, P. R. China.

[‡]The Department of Chemistry, University of South Florida, 4202 E. Fowler Avenue, Tampa, Florida 33620, United States

Corresponding Author : ^{*,†} E-mail: yfchen@wit.edu.cn ; ^{*,‡} E-mail: xmshi@usf.edu

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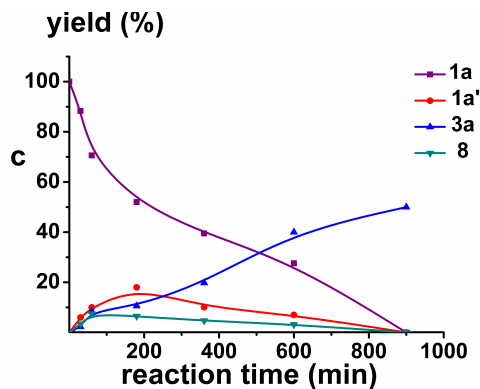
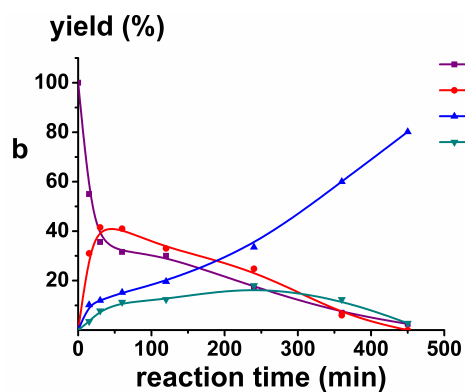
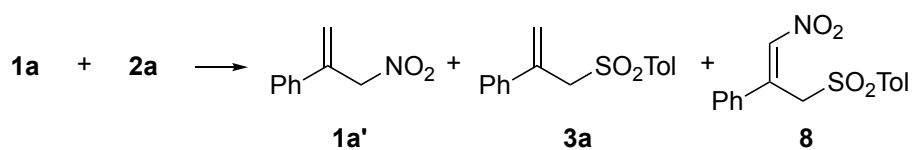
I. Other reactions conditions screening^a

entry	catalyst	additive	solvent	temp (°C)	time (h)	yield(%) ^b	
						3a	4a
1	AgNO ₃ (10 %)	K ₂ S ₂ O ₈ (2)	DMF	100	4	15 ^c	trace ^c
2	AgNO ₃ (10 %)	K ₂ S ₂ O ₈ (2)	DMF	100	4	30	trace
3	-	-	DMF	100	4	84	trace
4	-	-	Nitroethane	100	4	10	20
5	-	-	NMP	100	4	74	<5
6	-	-	toluene	100	4	0	0
7	-	-	n-Butylacetate	100	4	20	0
8	TBAI (10 %)	TBHP (2)	DMSO	100	4	20	trace
9	CuCl ₂ (10 %)	TBHP (2)	DMSO	100	4	trace	0
10	I ₂ (10 %)	K ₂ S ₂ O ₈ (2)	DMSO	80	4	89	trace
11	I ₂ (10 %)	DBTP (2)	DMSO	80	4	19	trace
12	I ₂ (10 %)	H ₂ O ₂ (2)	DMSO	80	4	50	trace
13	I ₂ (10 %)	-	DMSO	80	4	30	0
14	TBAI (20 %)	-	DMSO	100	4	75	0
15	I ₂ (10 %)	TBHP (1)	DMSO	80	4	85	trace
16	I ₂ (5 %)	TBHP (2)	DMSO	80	4	70	trace

^aReaction conditions: **1a** (0.3 mmol, 1 equiv), **2a** (1.5 equiv), catalyst (mol %), oxidant (equiv) in DMSO (3.0 mL), open air.

^bIsolated yield. ^cUnder argon protection.

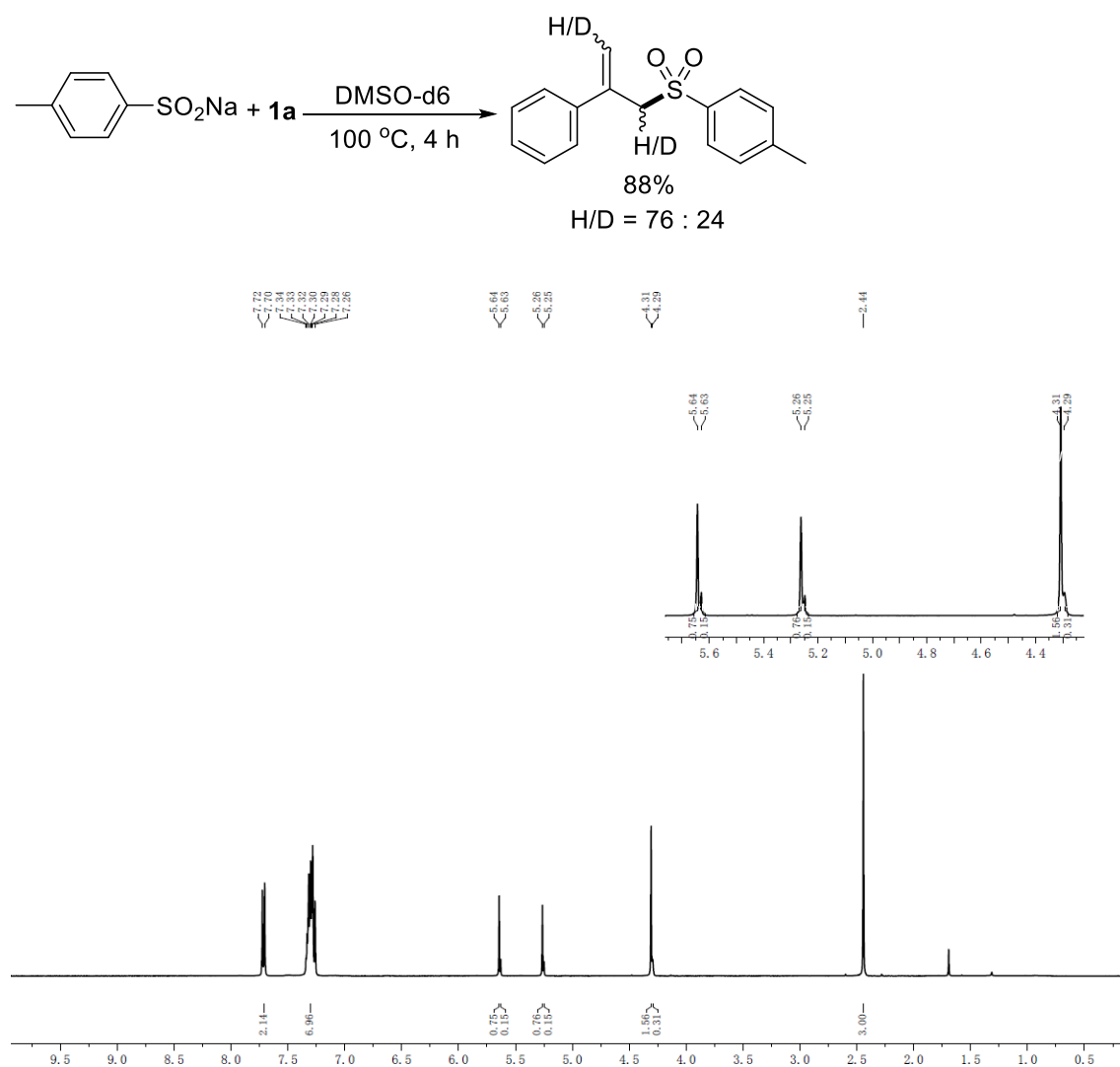
II. Reaction profile



Reaction conditions: nitroolefins (0.3 mmol), sodium sulfinates (0.45 mmol) (b) I_2 (0.03 mmol), DMSO, 80 °C, open air; (c) I_2 (0.3 mmol), DMSO, 80 °C, open air.

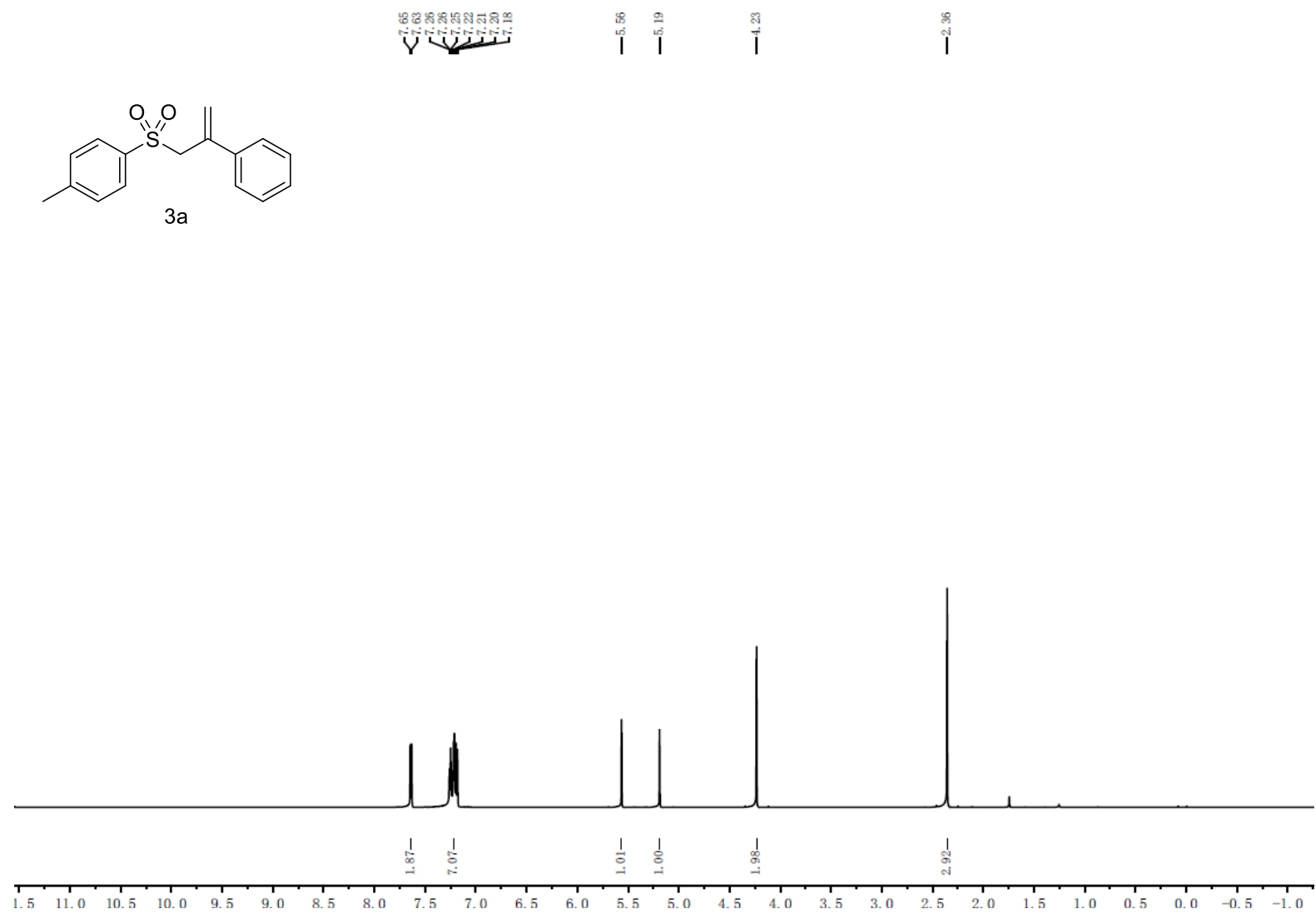
The time-dependent reaction profile of the reaction of **1a** with sodium *p*-tolylsulfinate was obtained using ^1H NMR. The Lewis base promoted the equilibrium between **1a** and **1a'** occurred within 40 min (condition b). However, 100% I_2 inhibited the formation of allylic compound **1a'** (condition c).

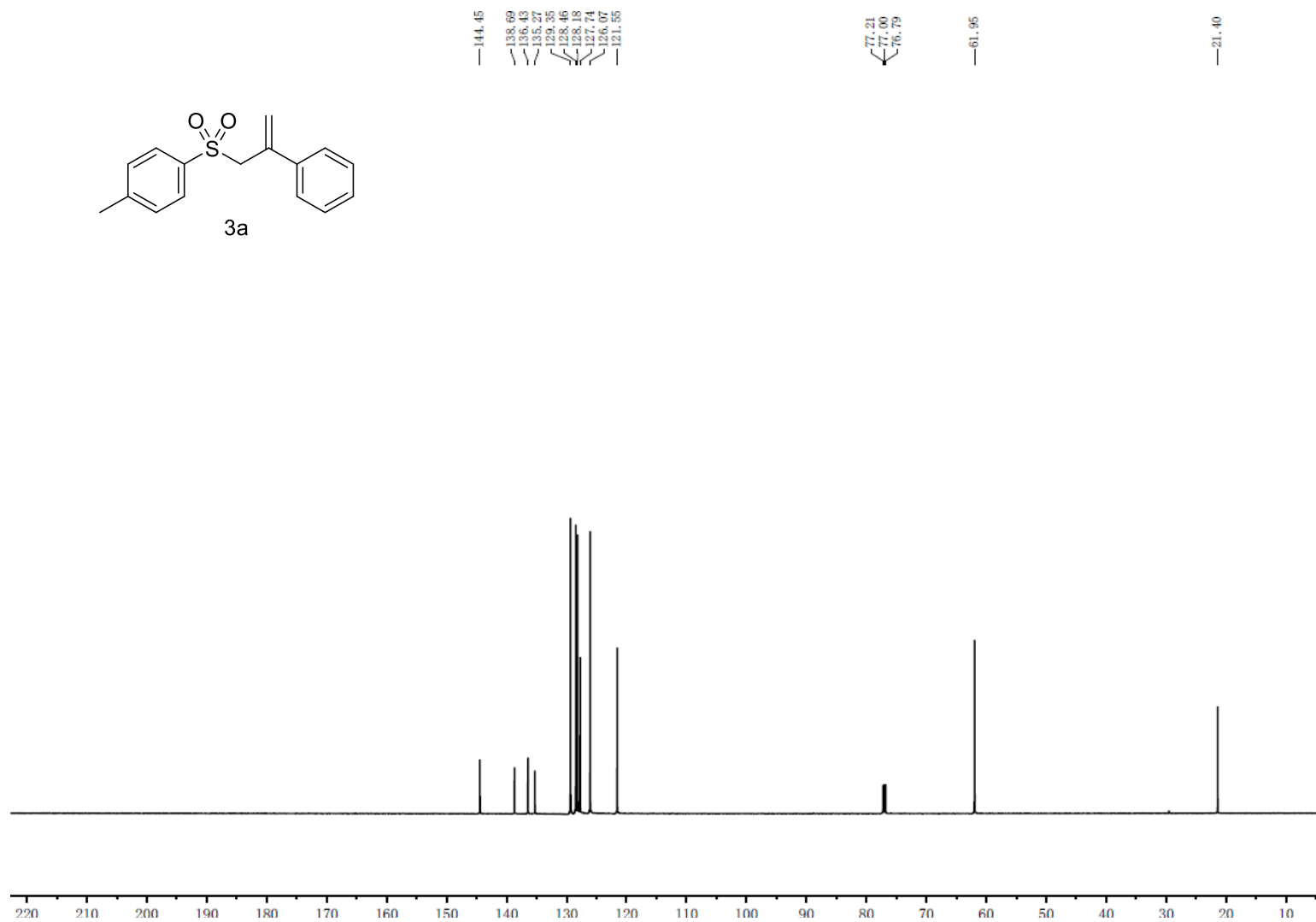
III. Deuterium-Labeling Experiments

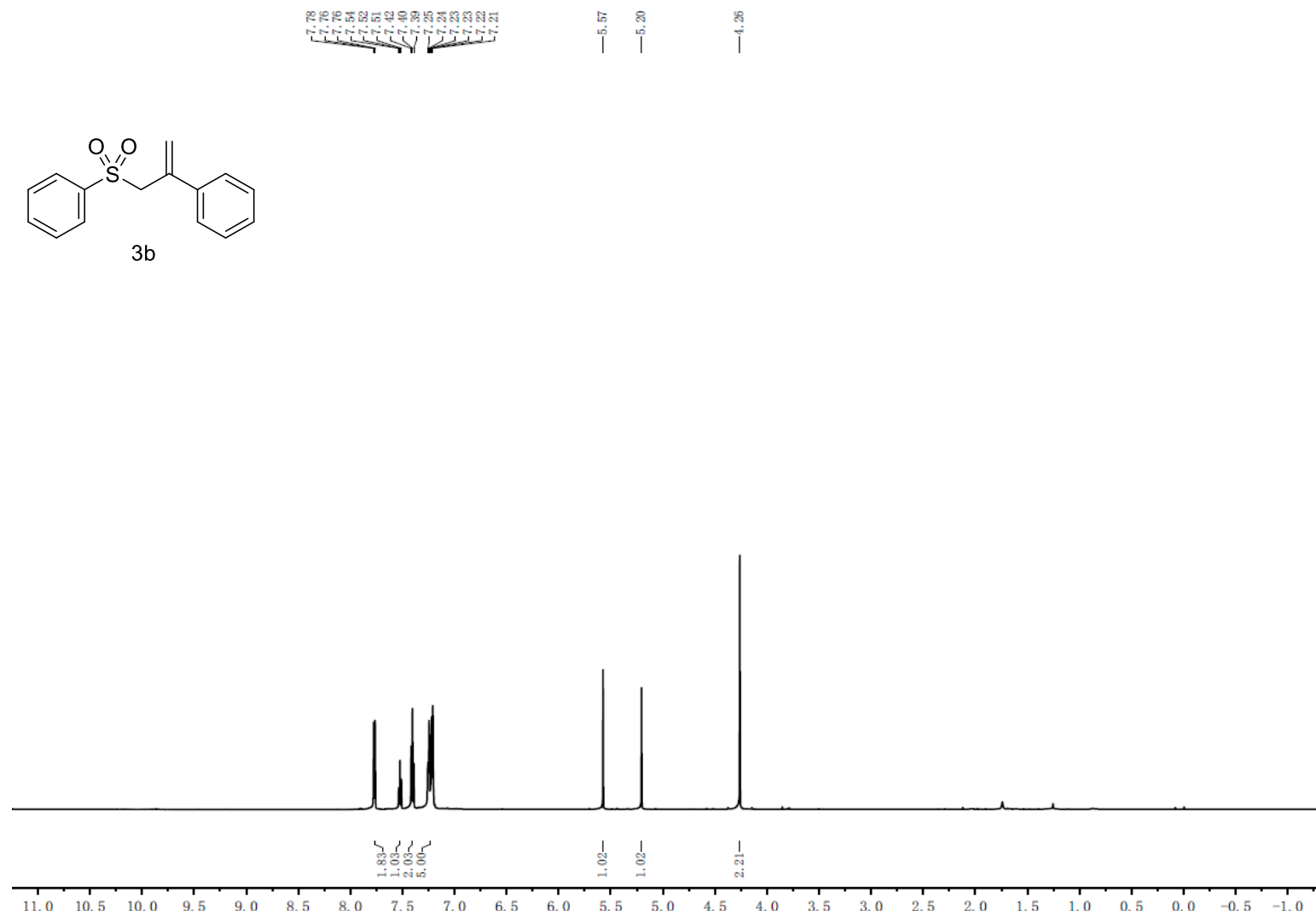


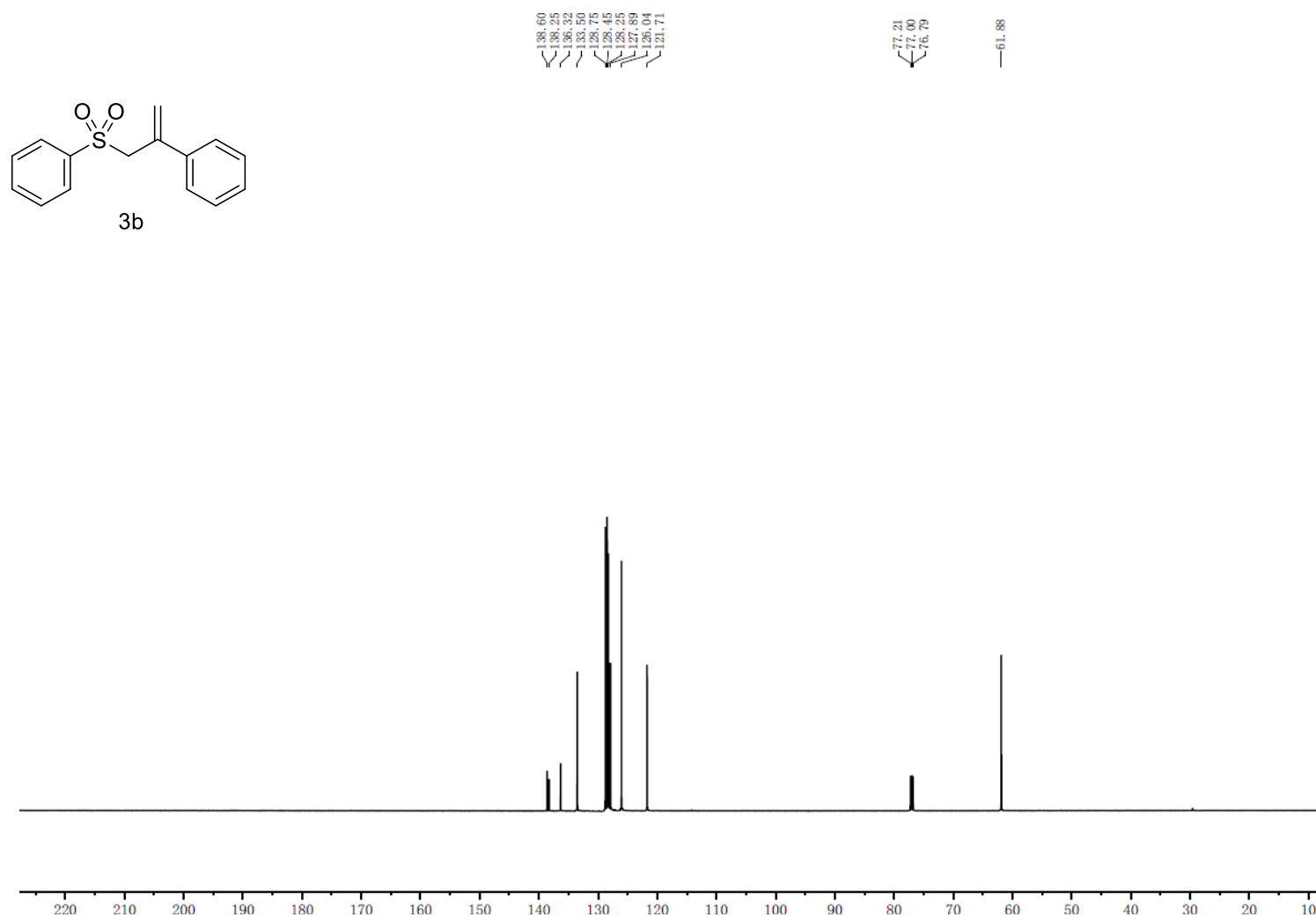
The reaction of **1a** and sodium *p*-tolylsulfate in DMSO- d_6 was also performed, and the ^1H NMR spectrum of the corresponding product revealed that D-substituted product formed, the ratio of H and D was 76:24.

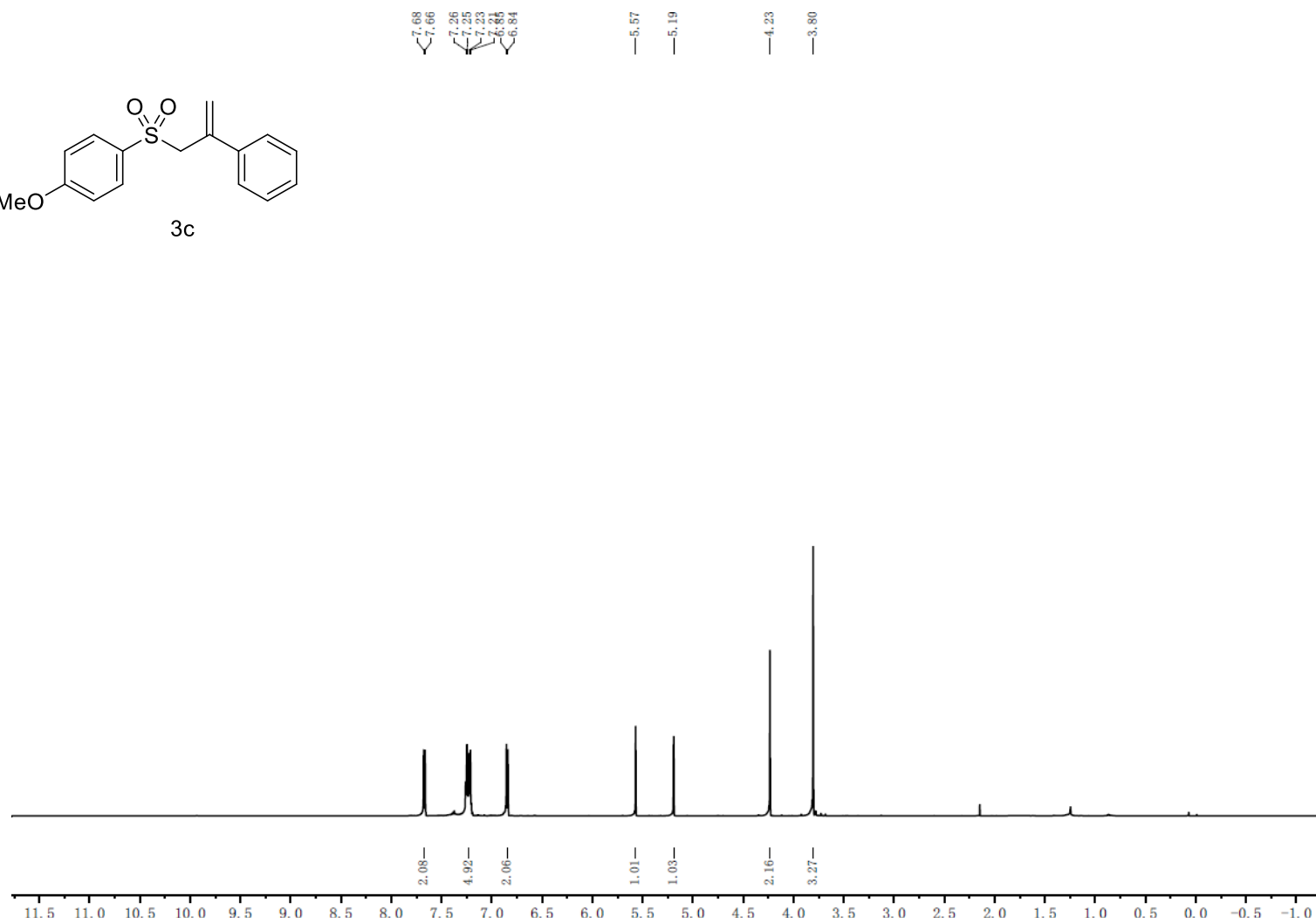
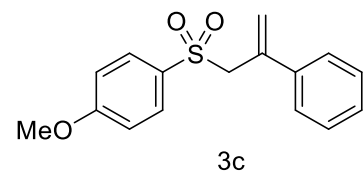
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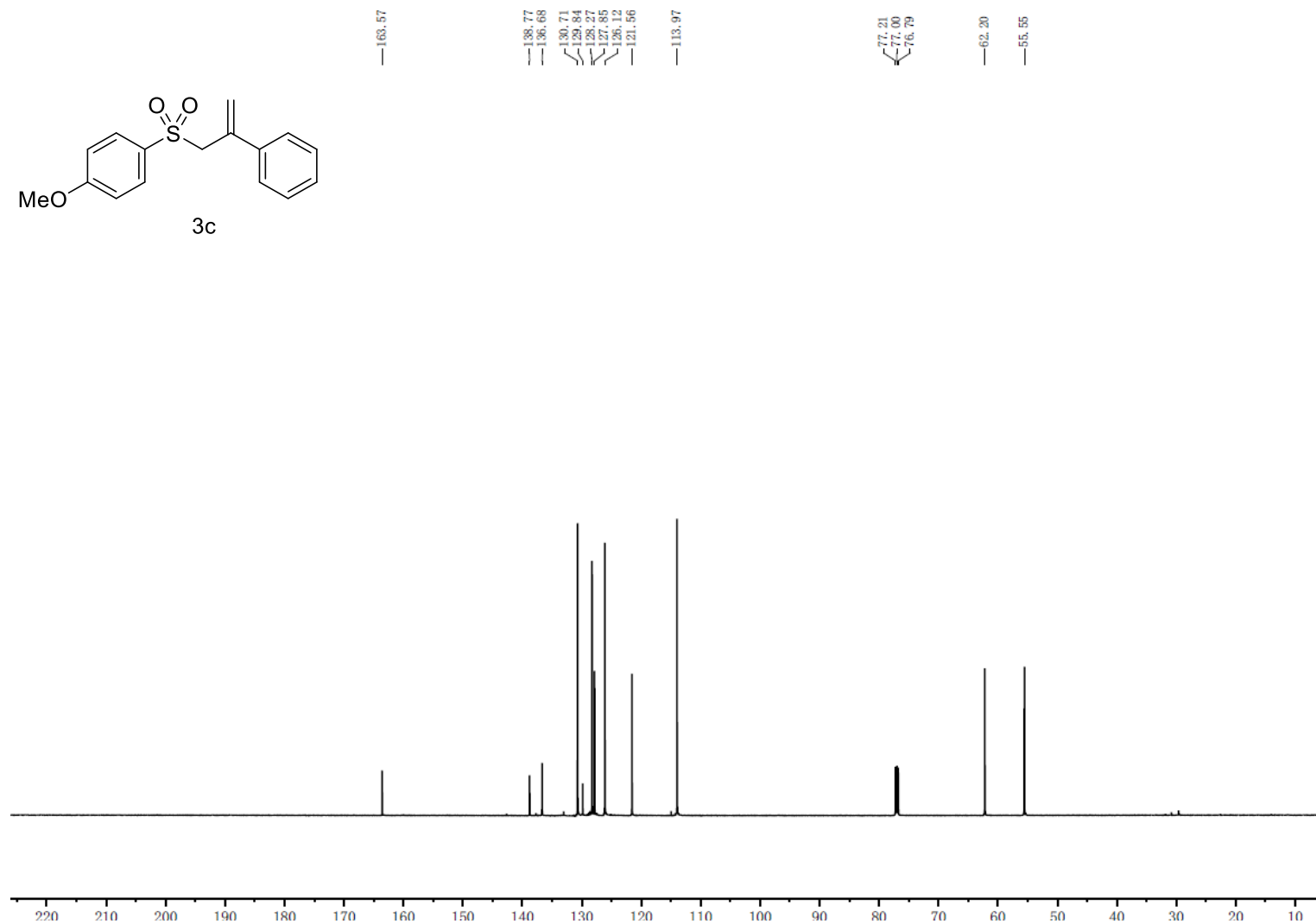
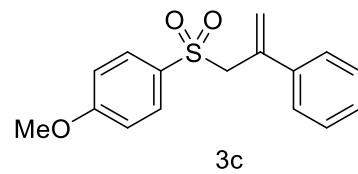


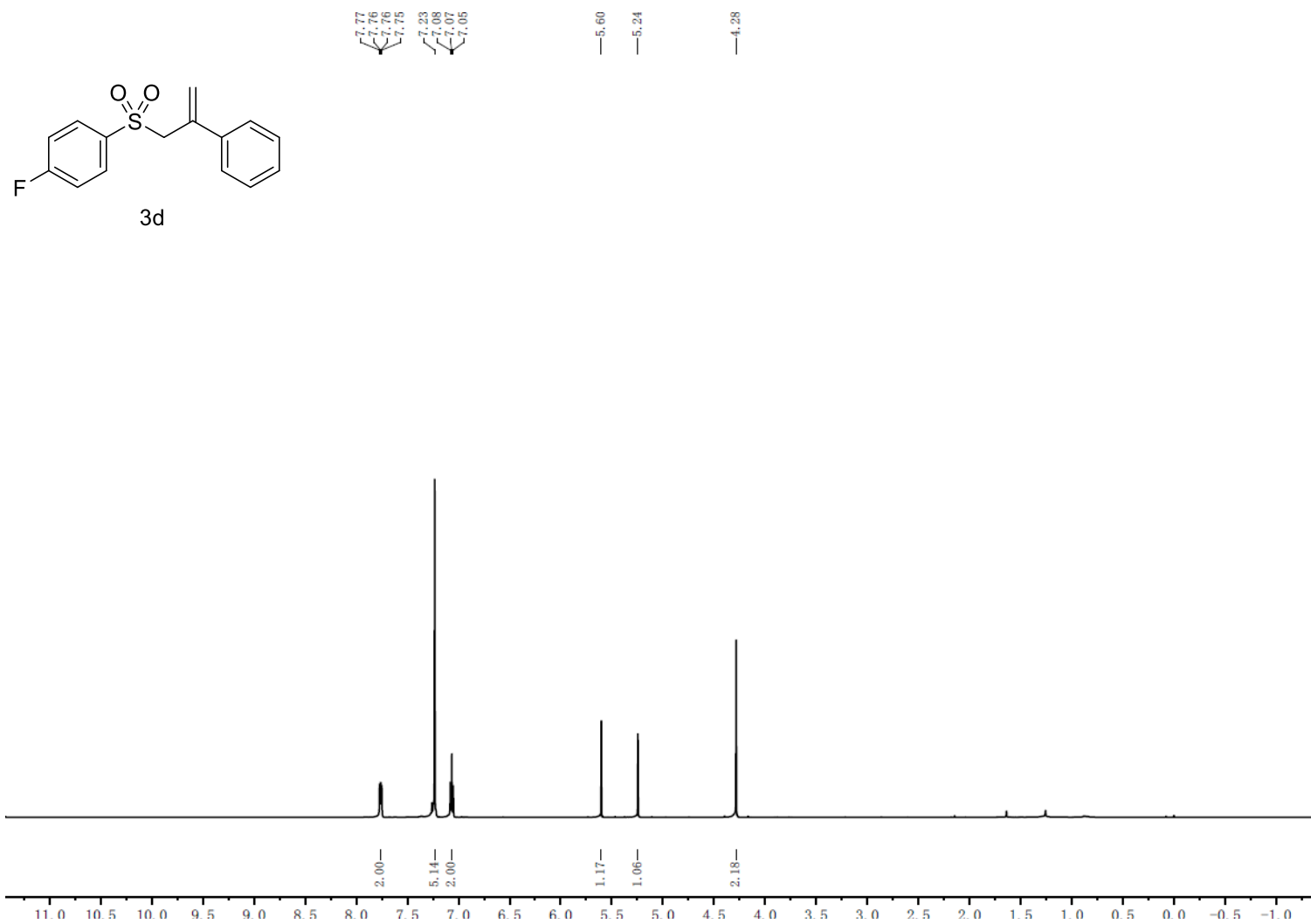


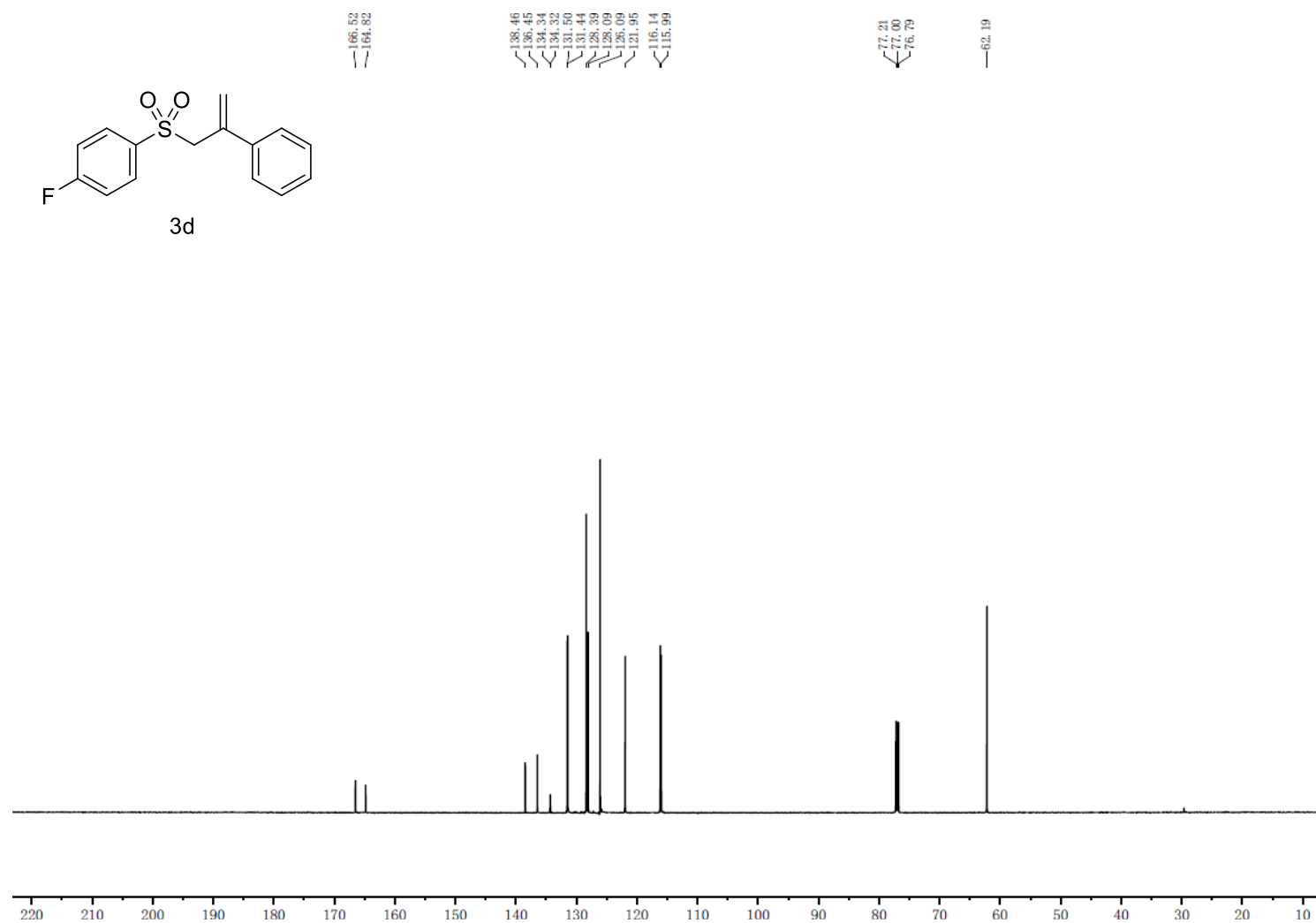


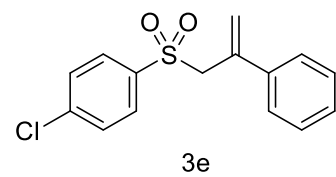




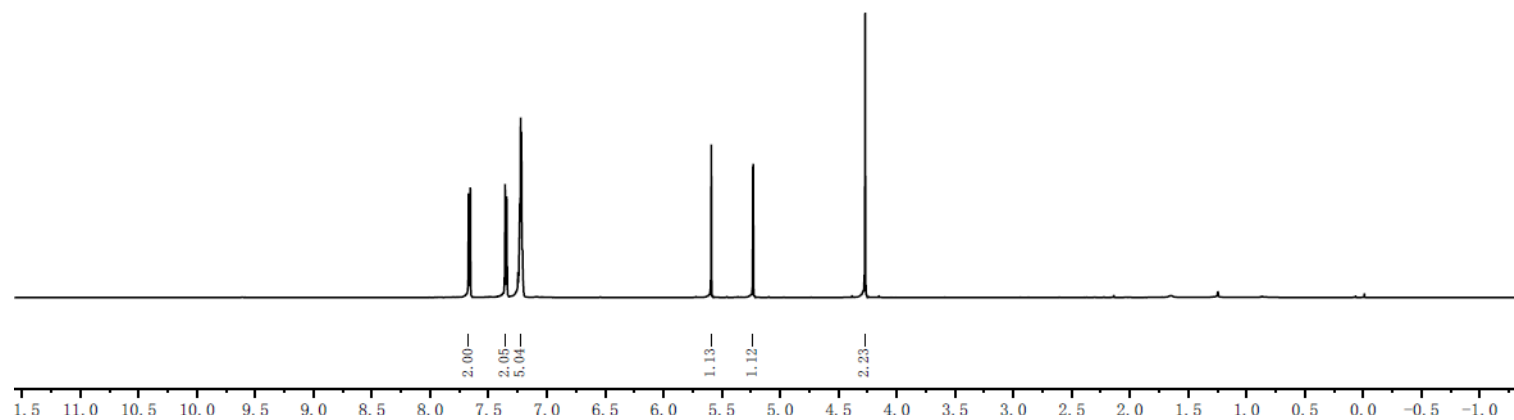


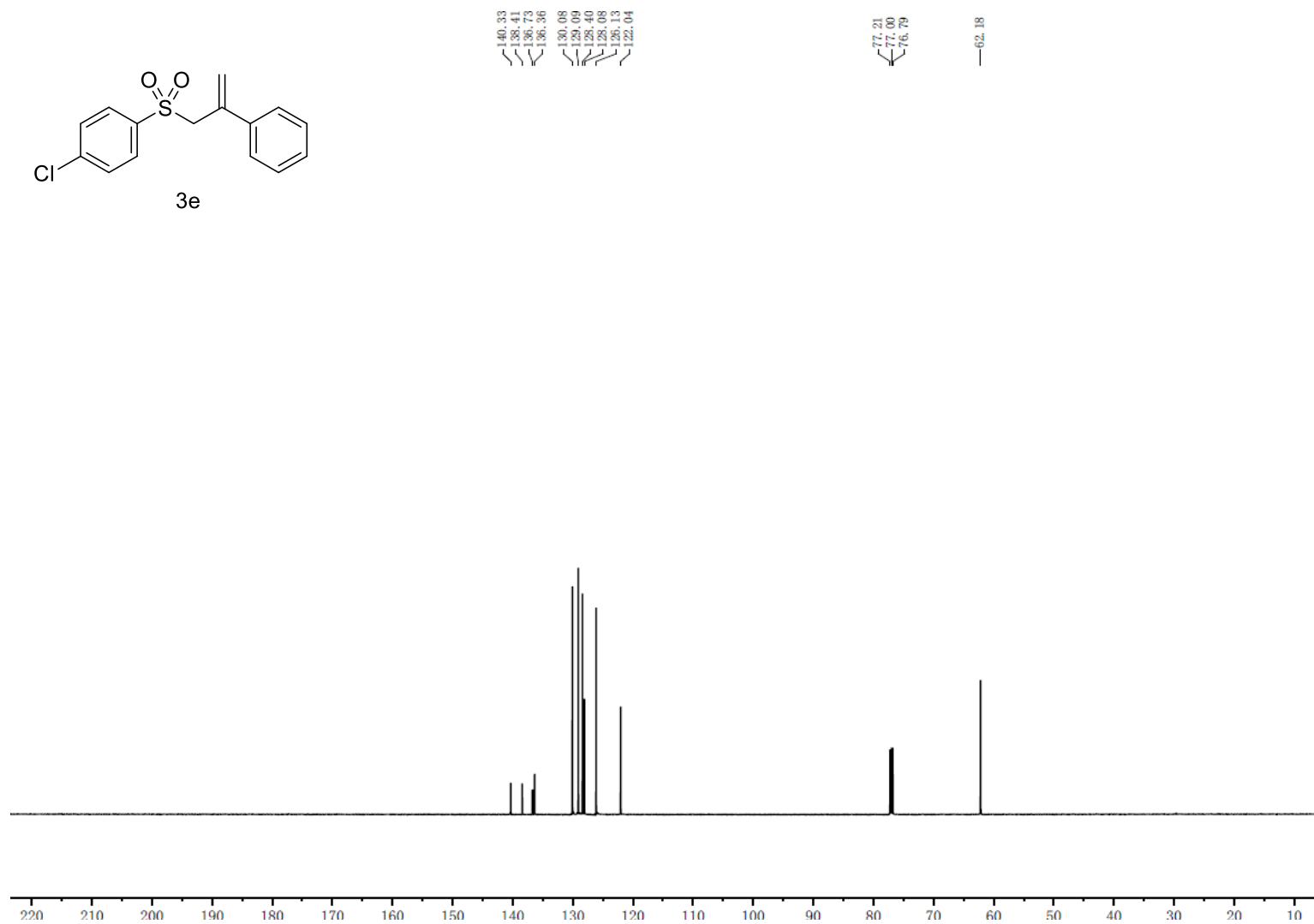
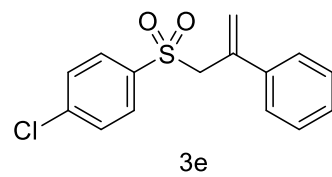


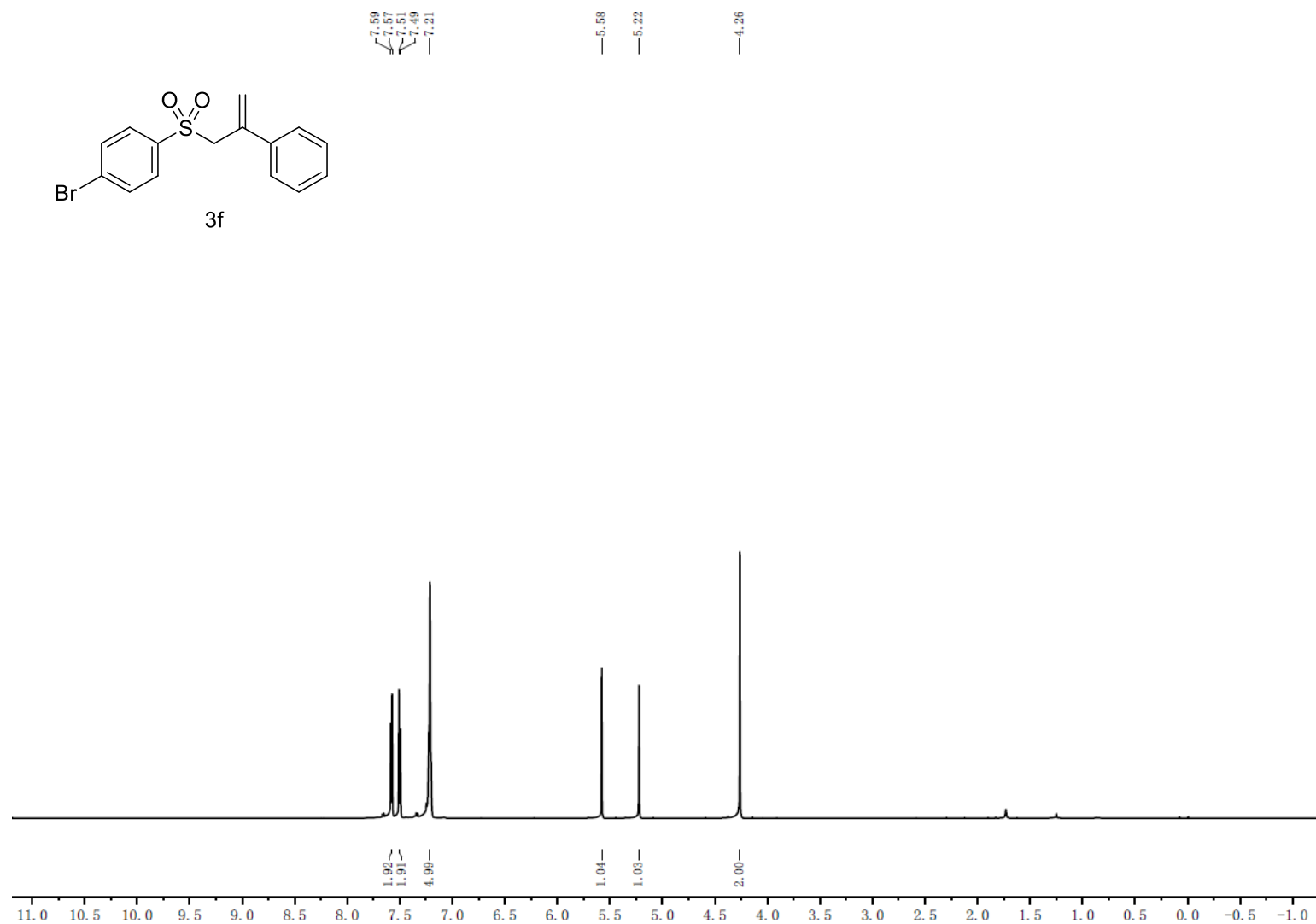
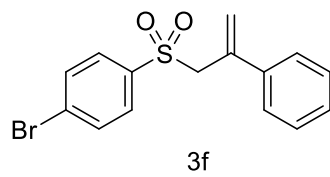


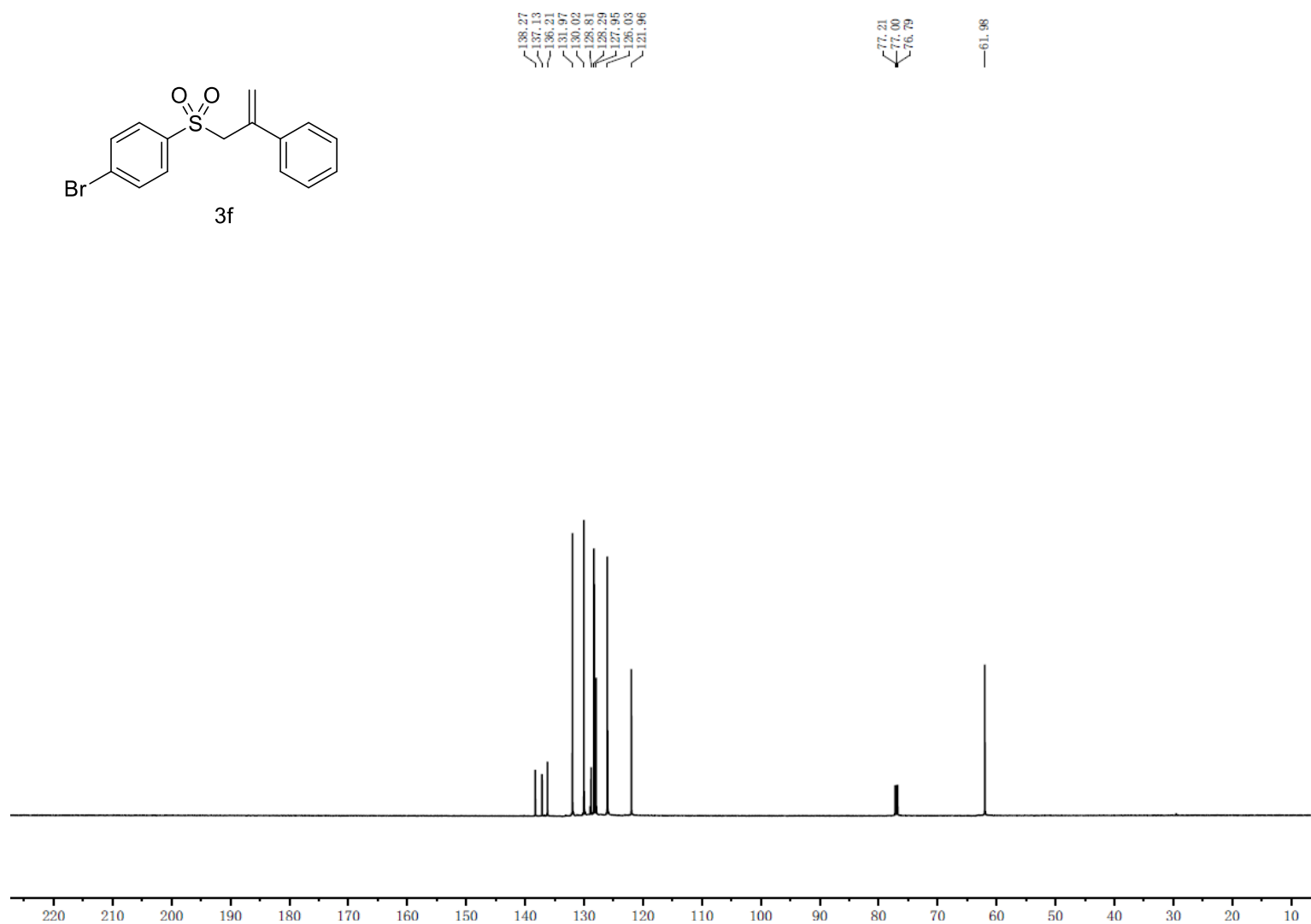
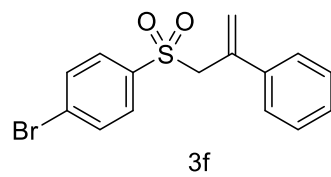


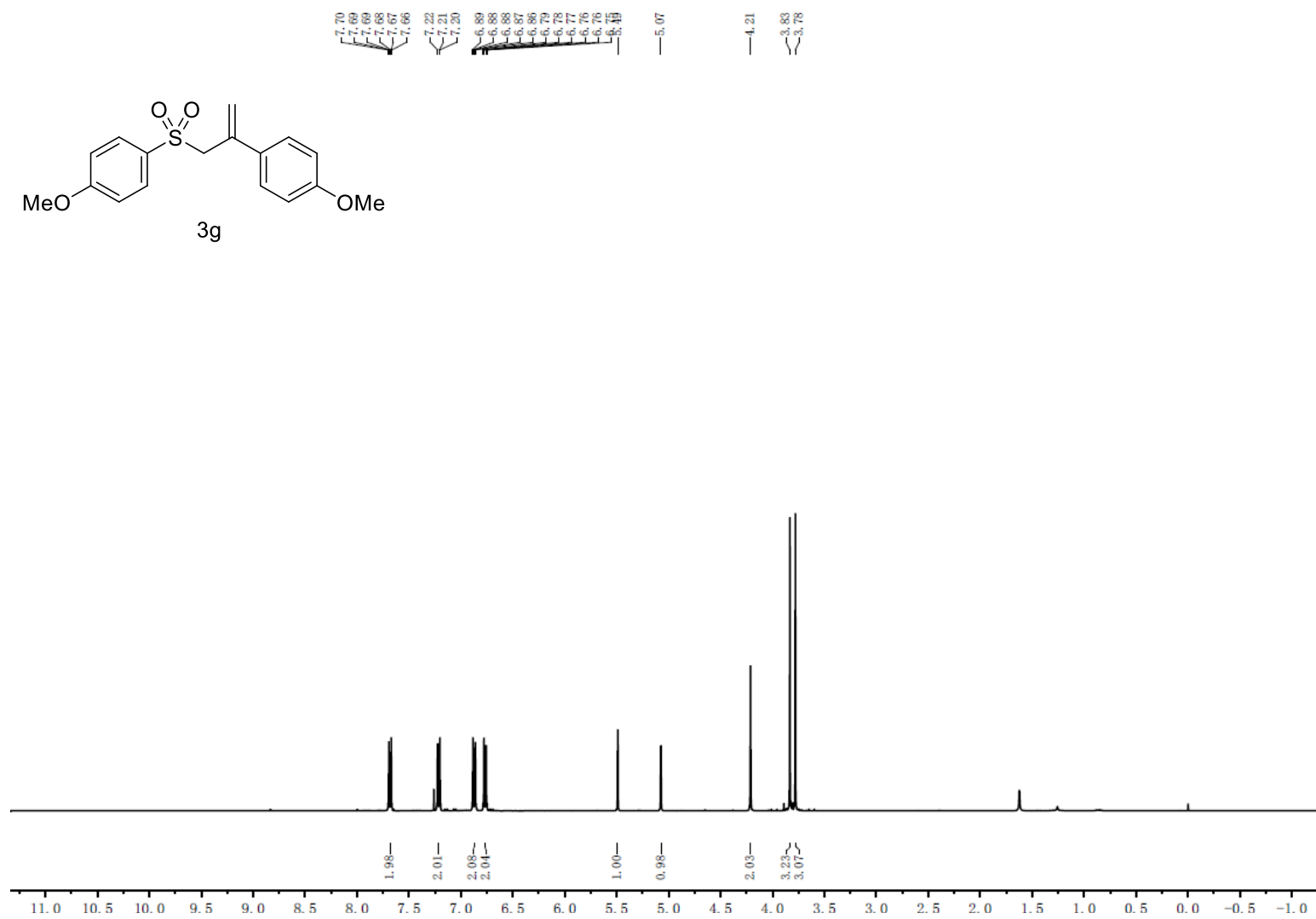
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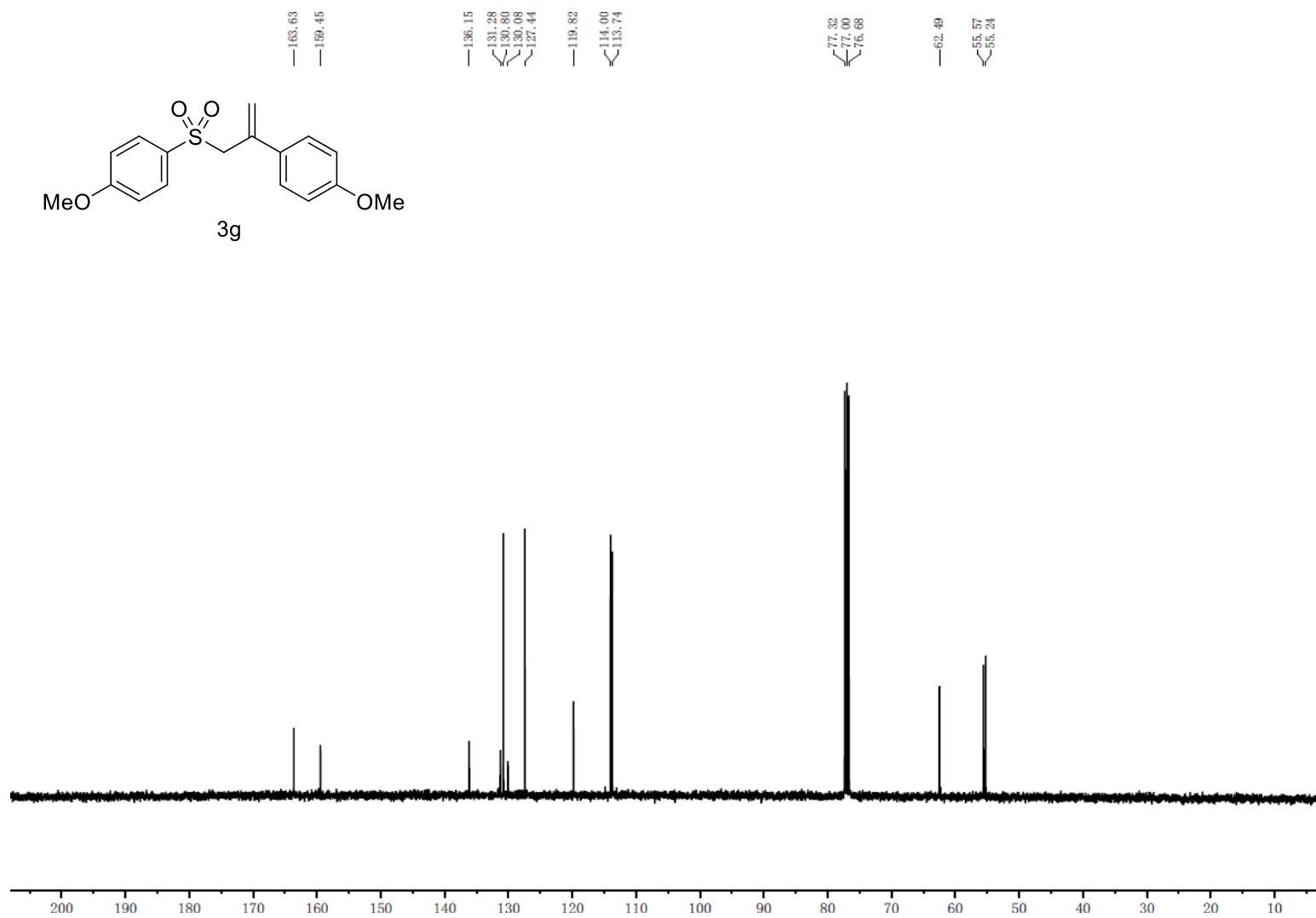


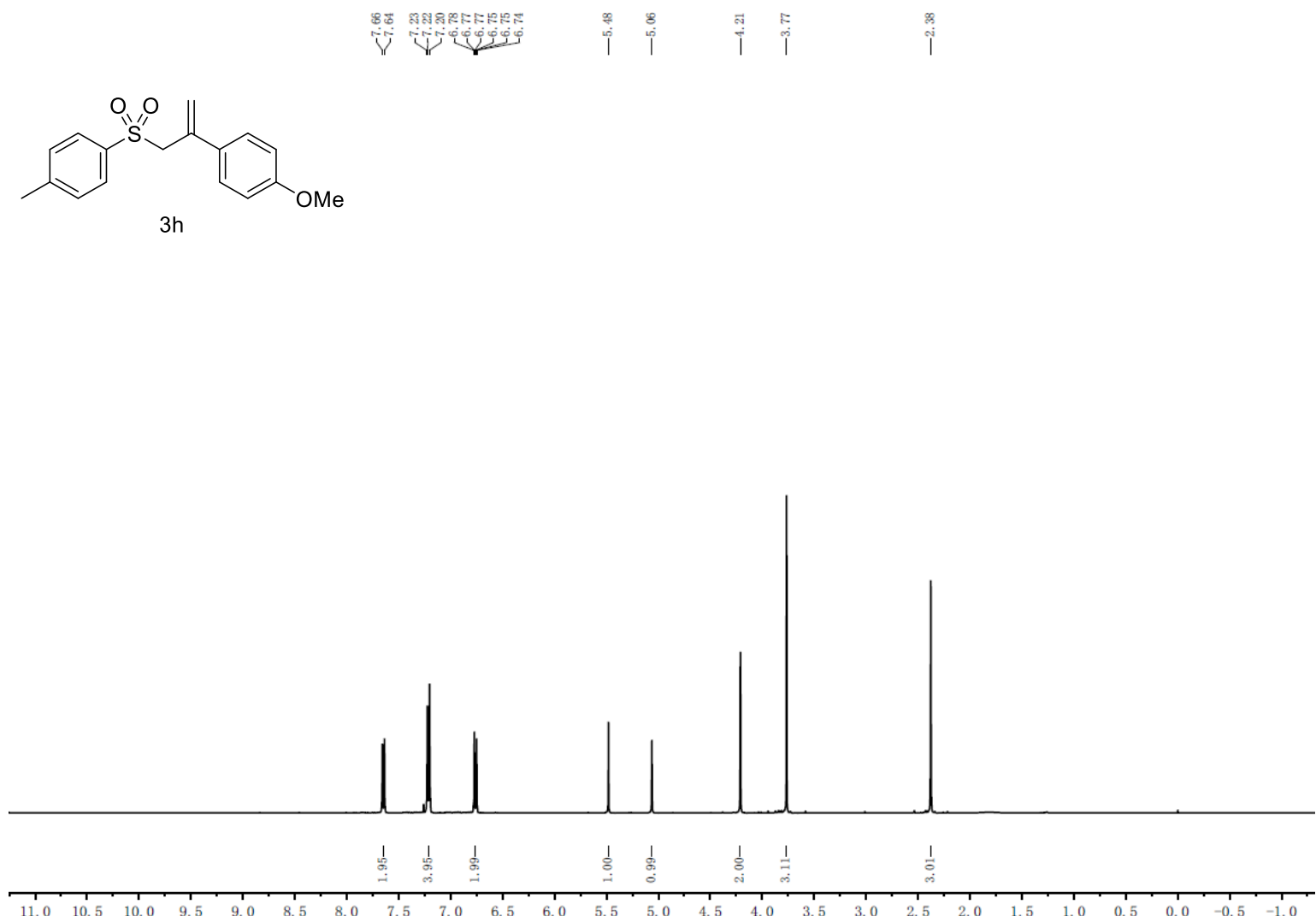


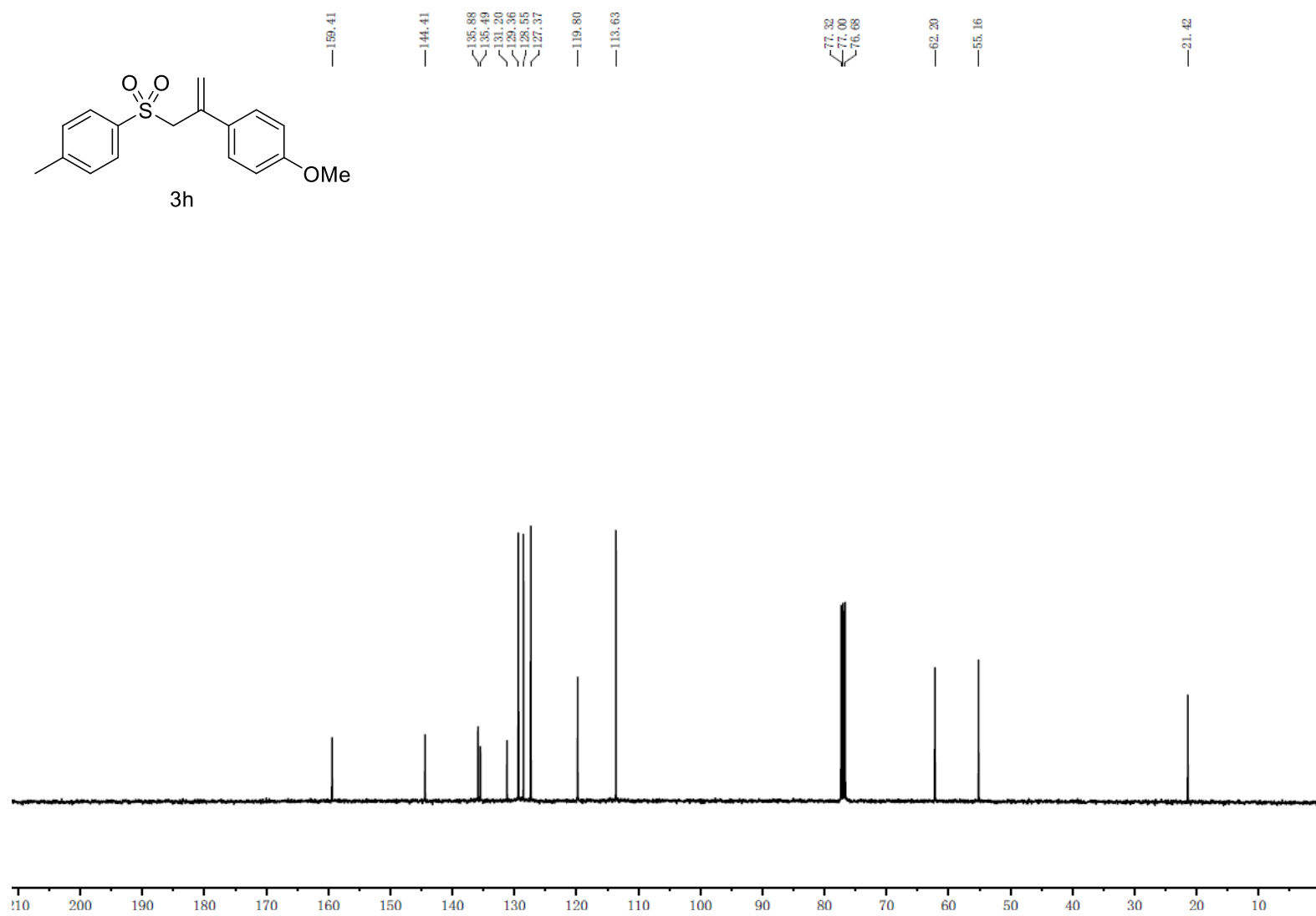


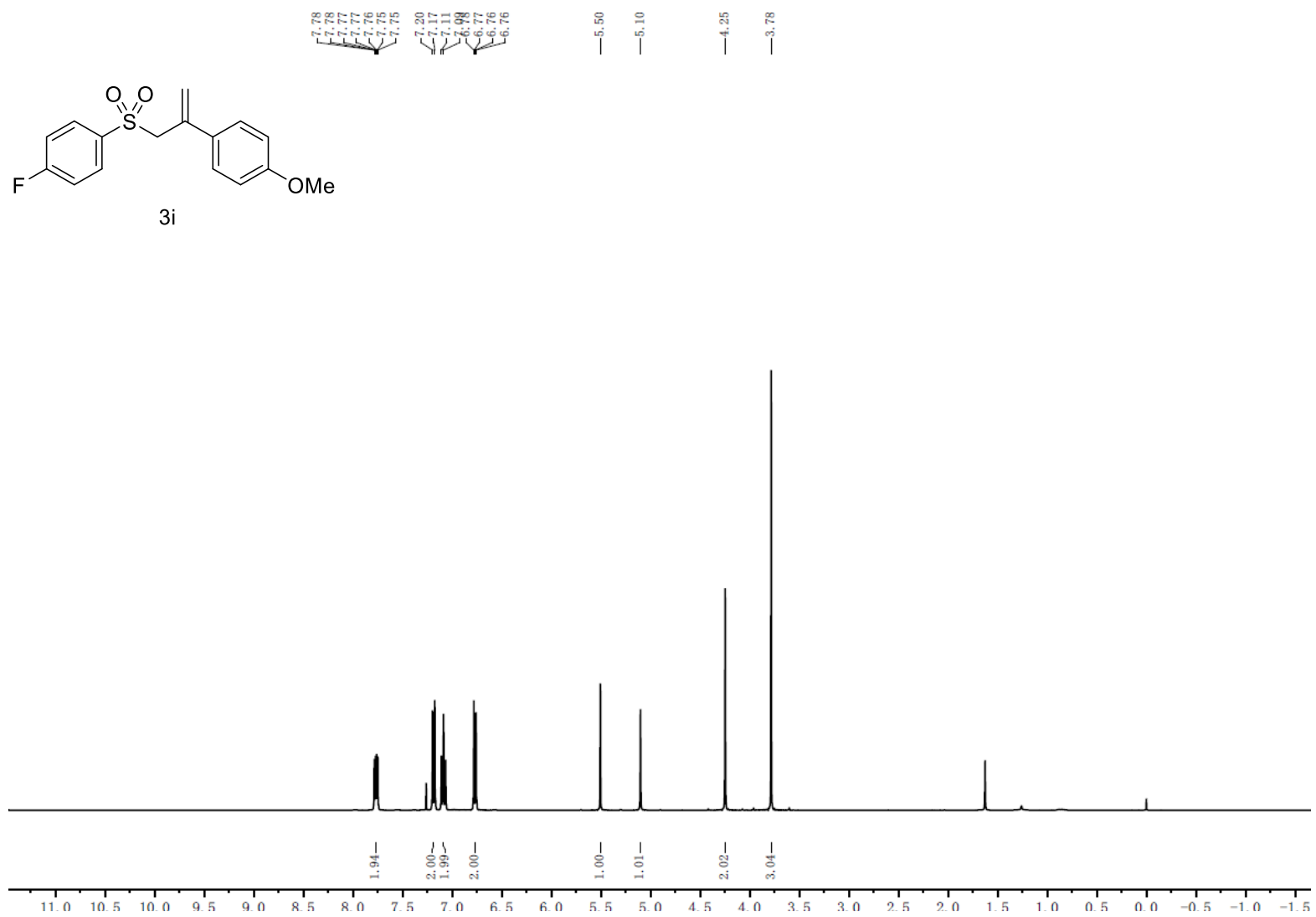


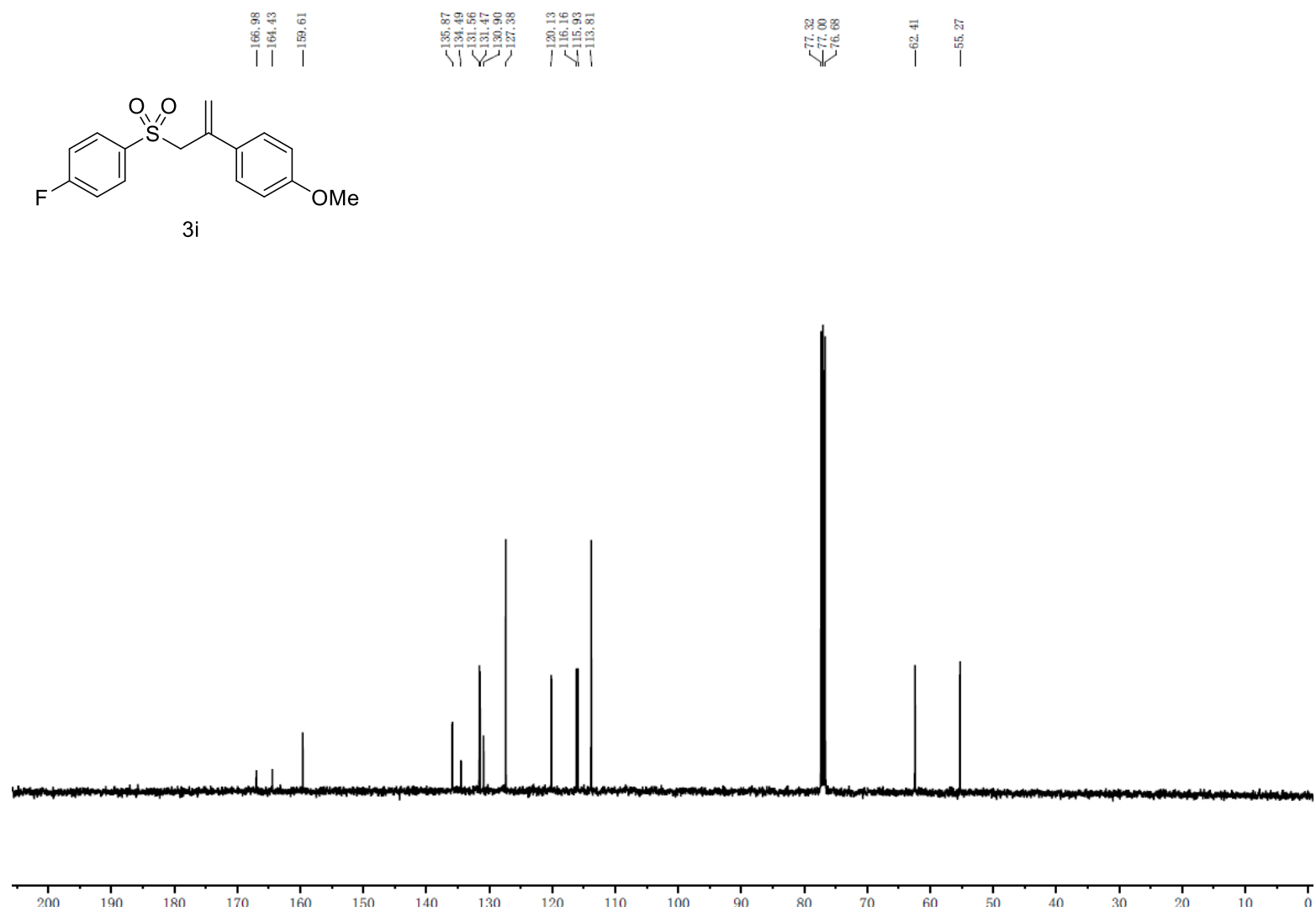


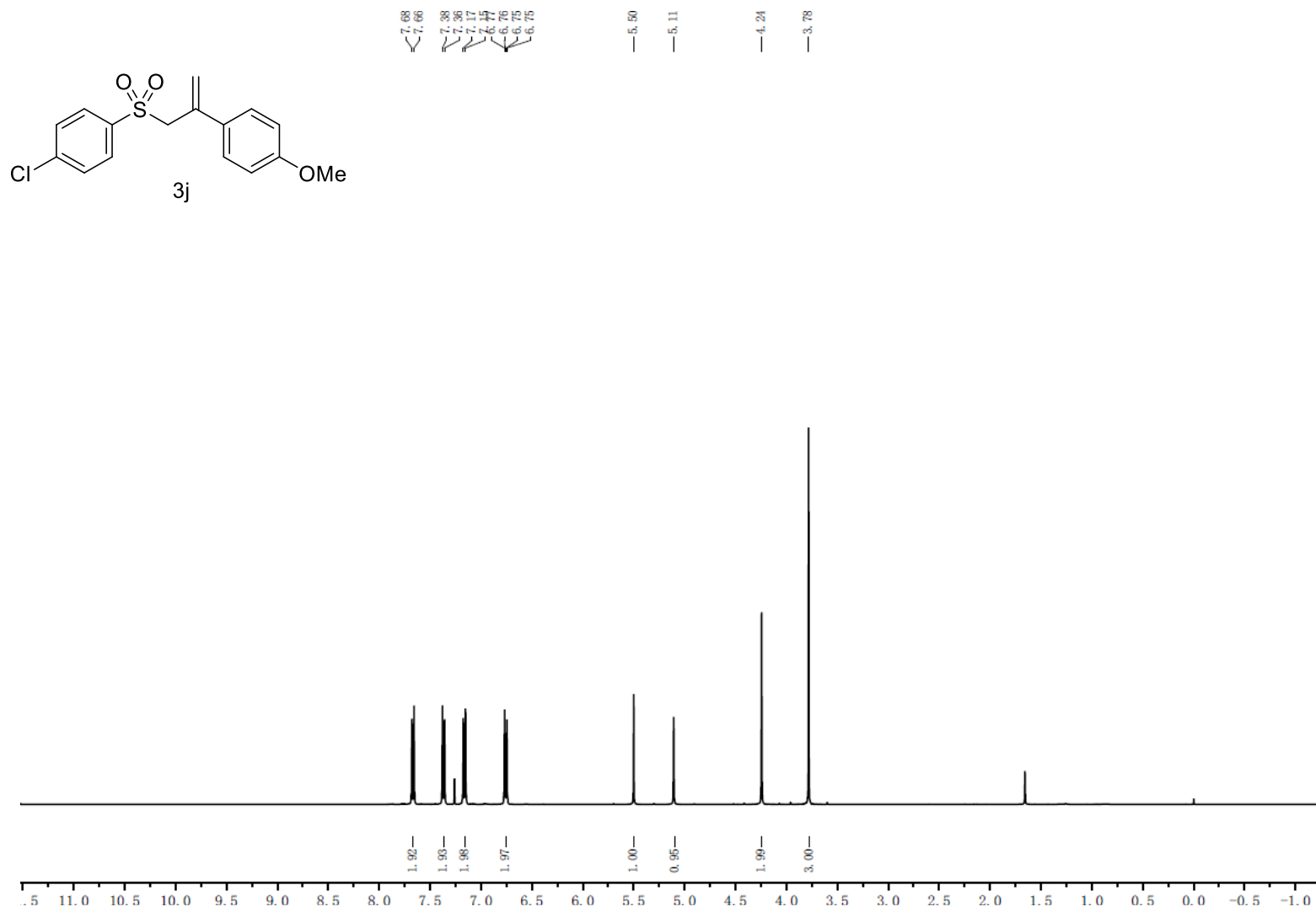


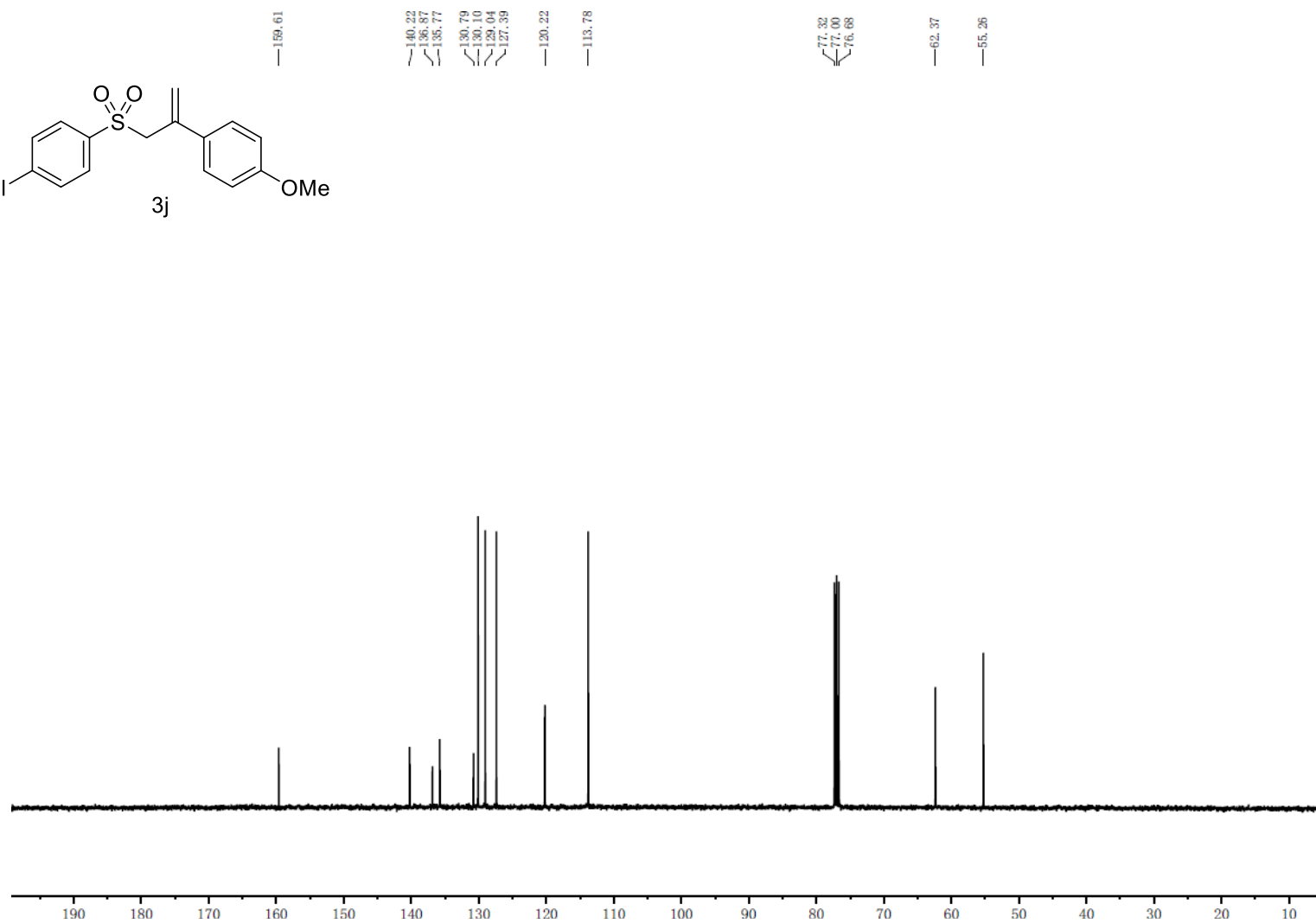
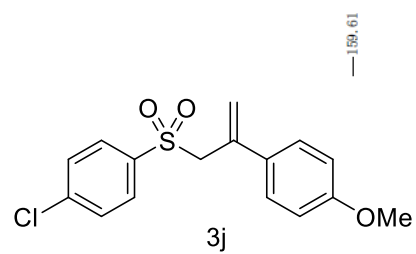


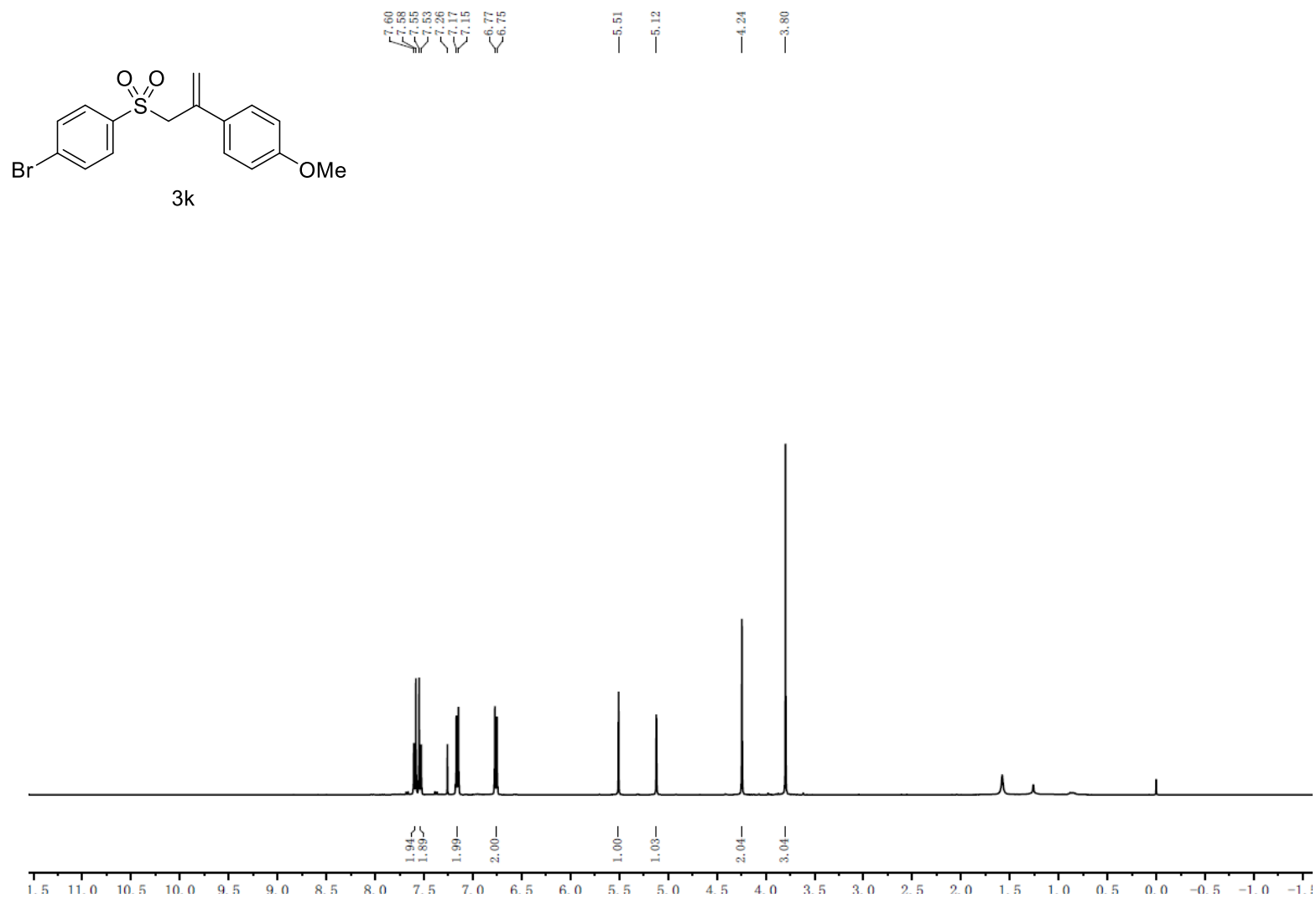


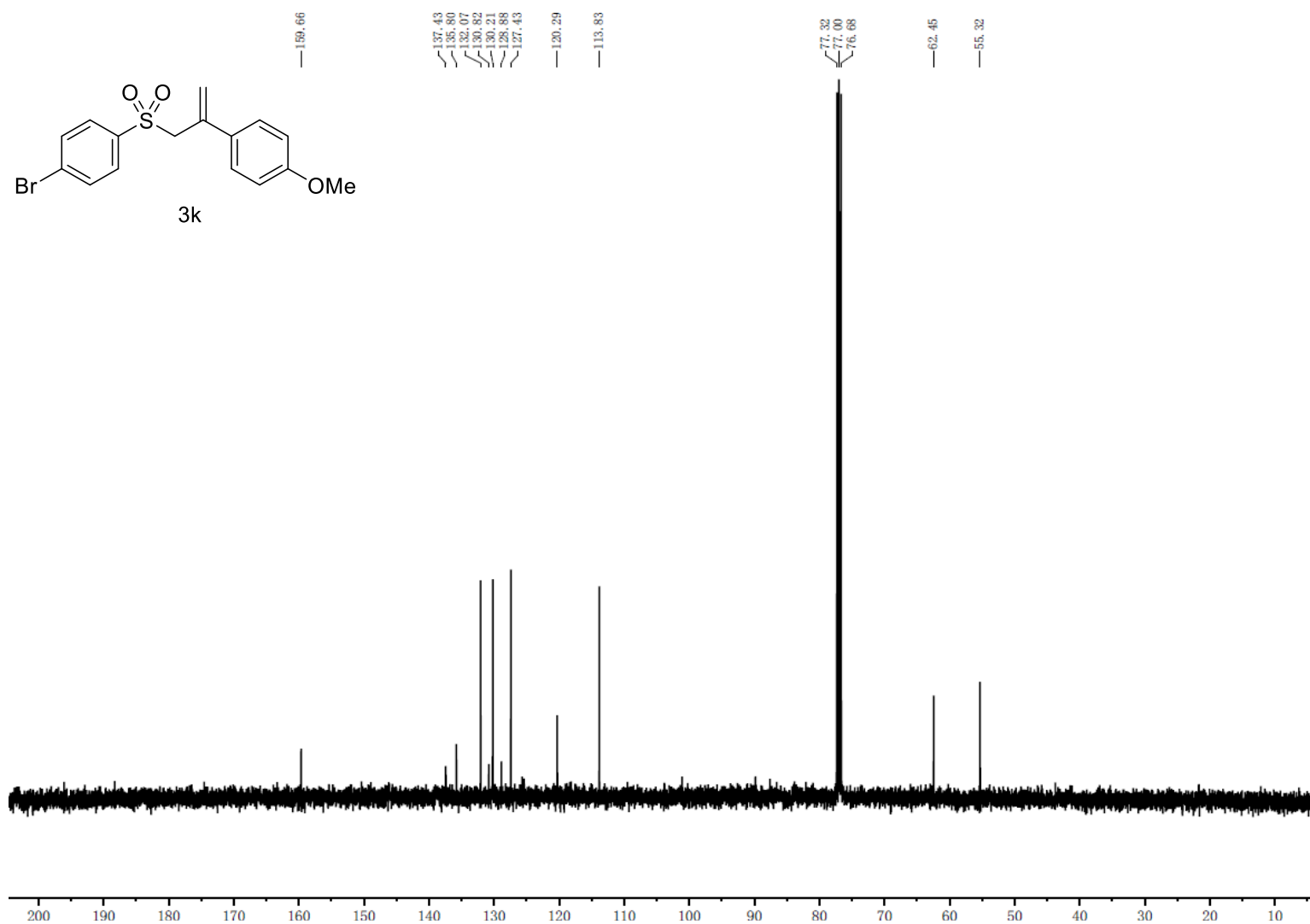


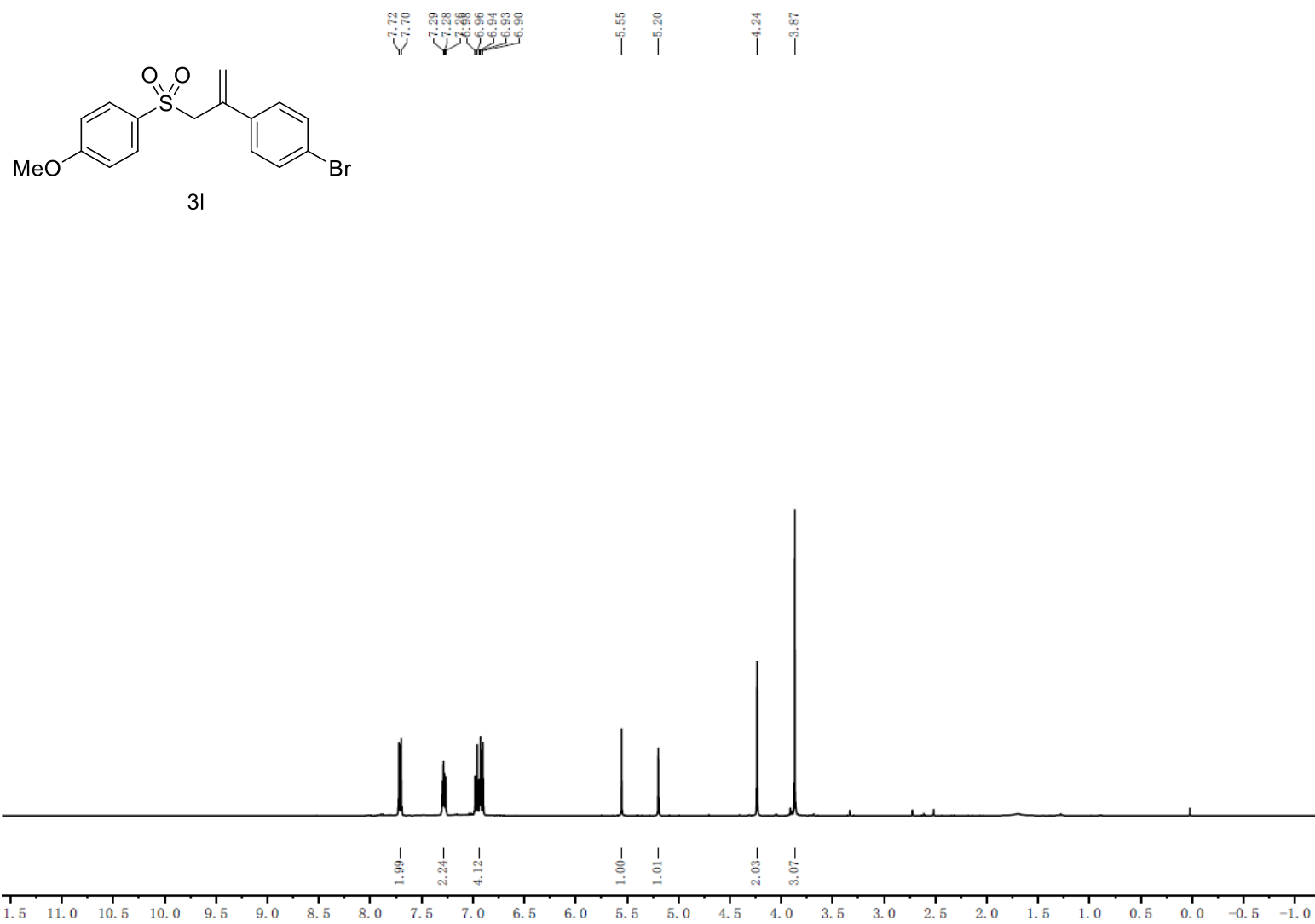


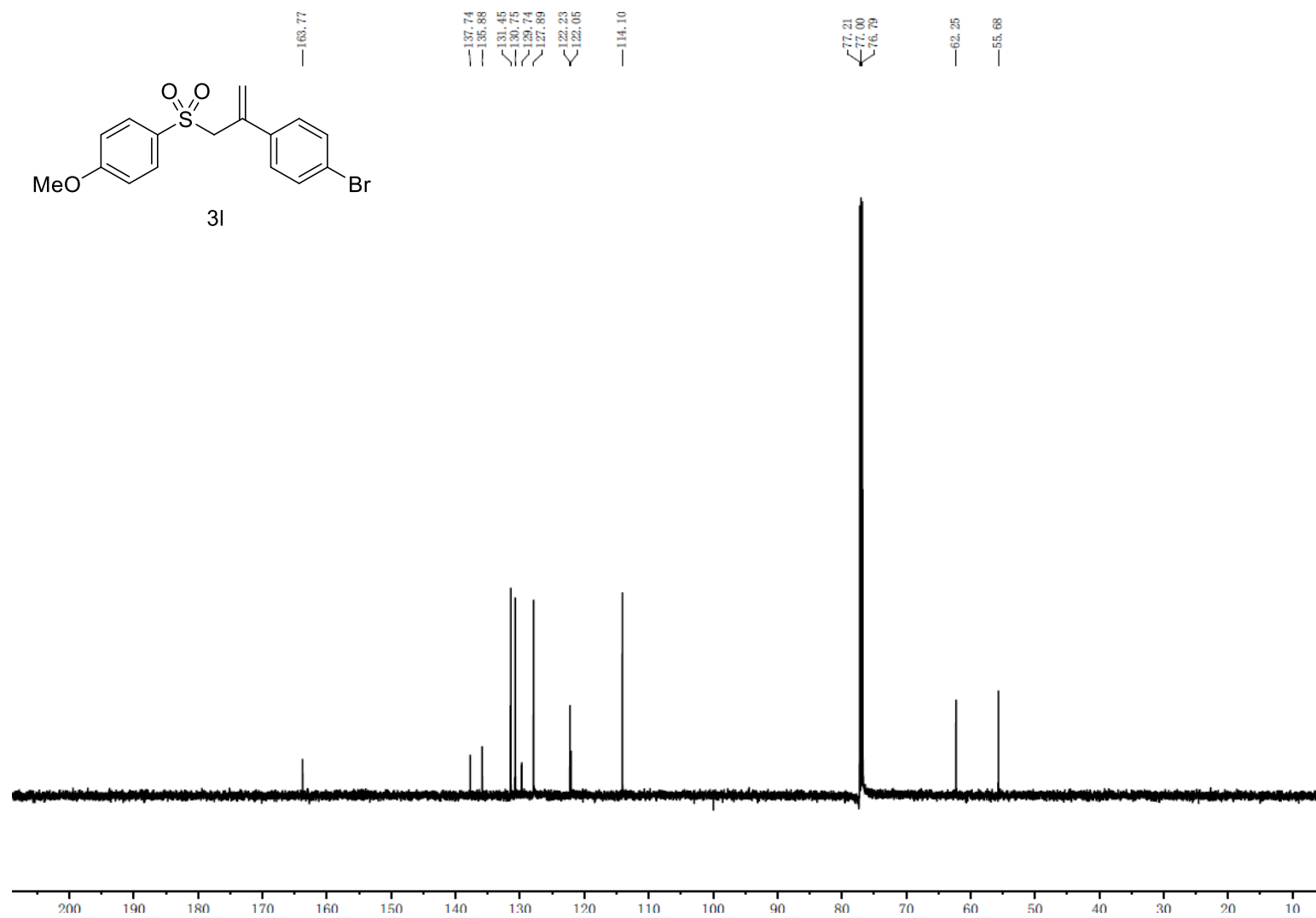


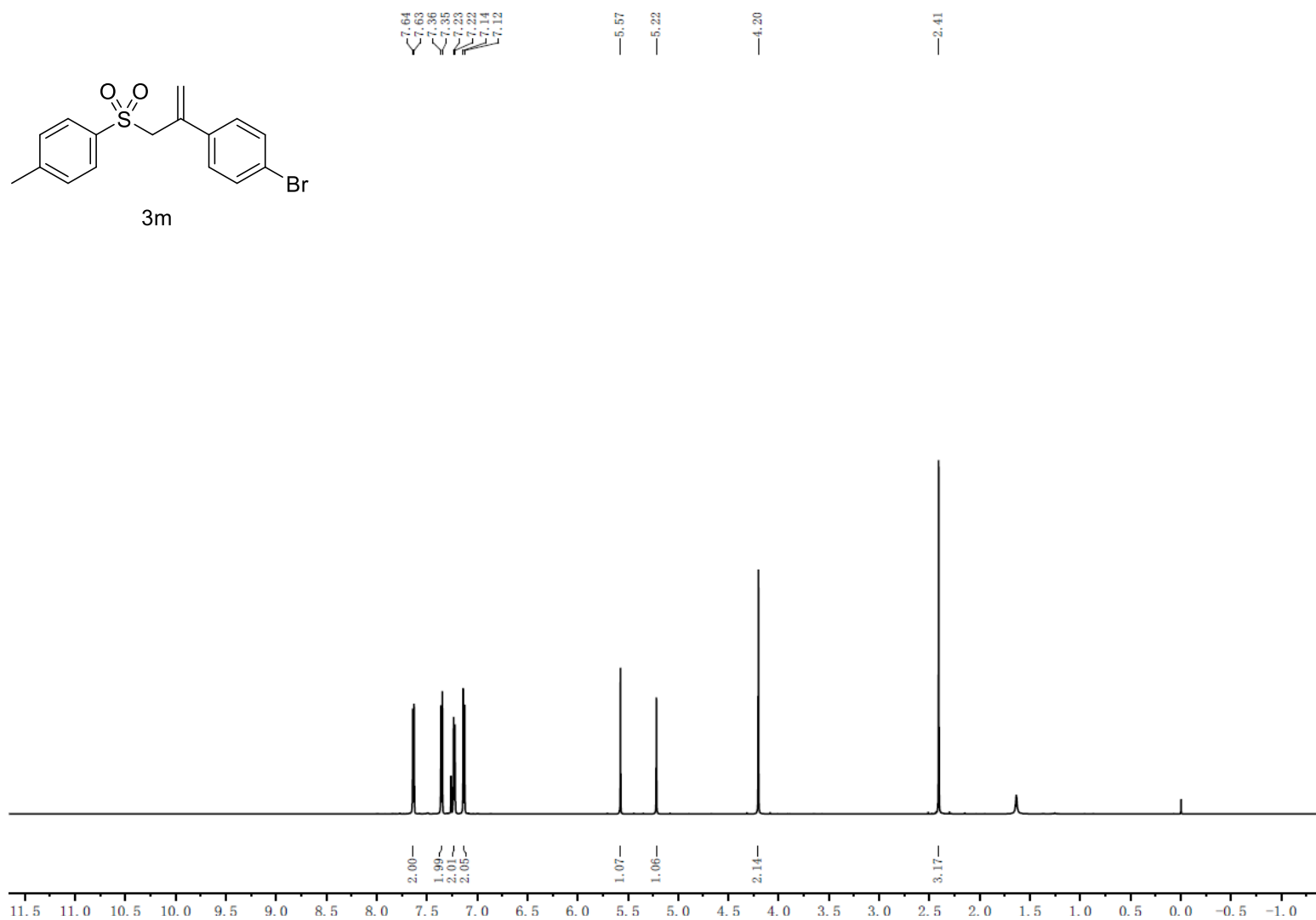


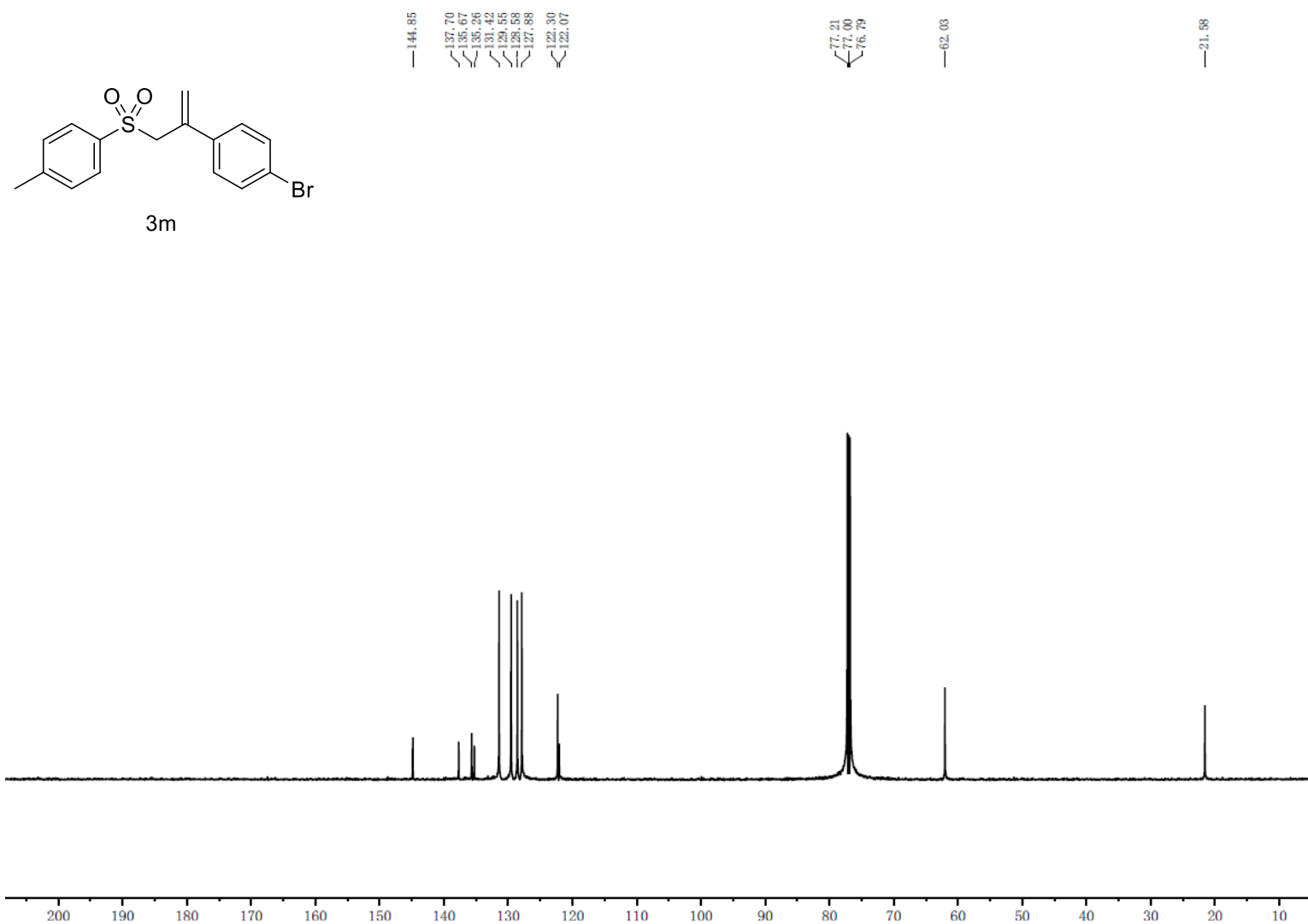


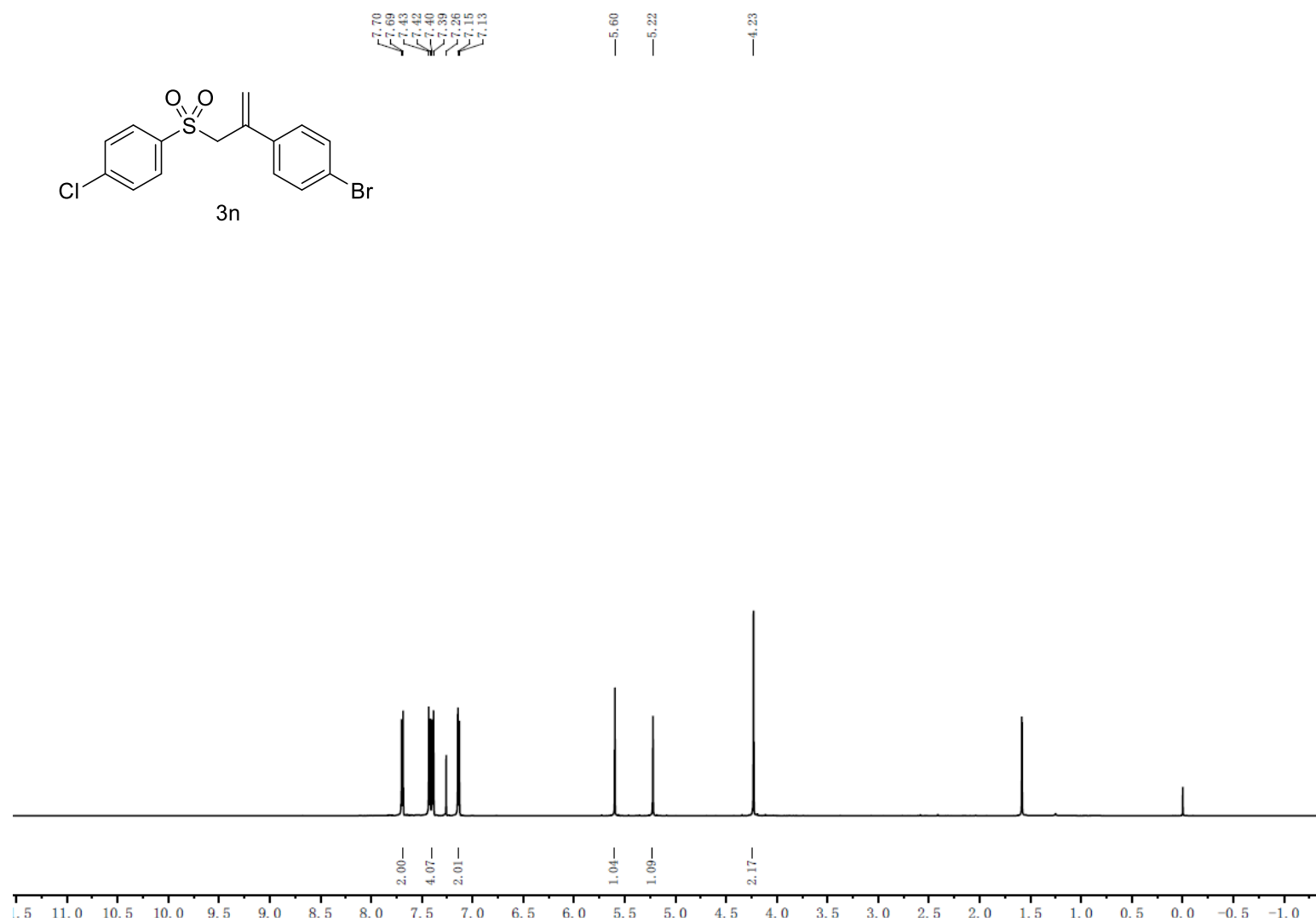


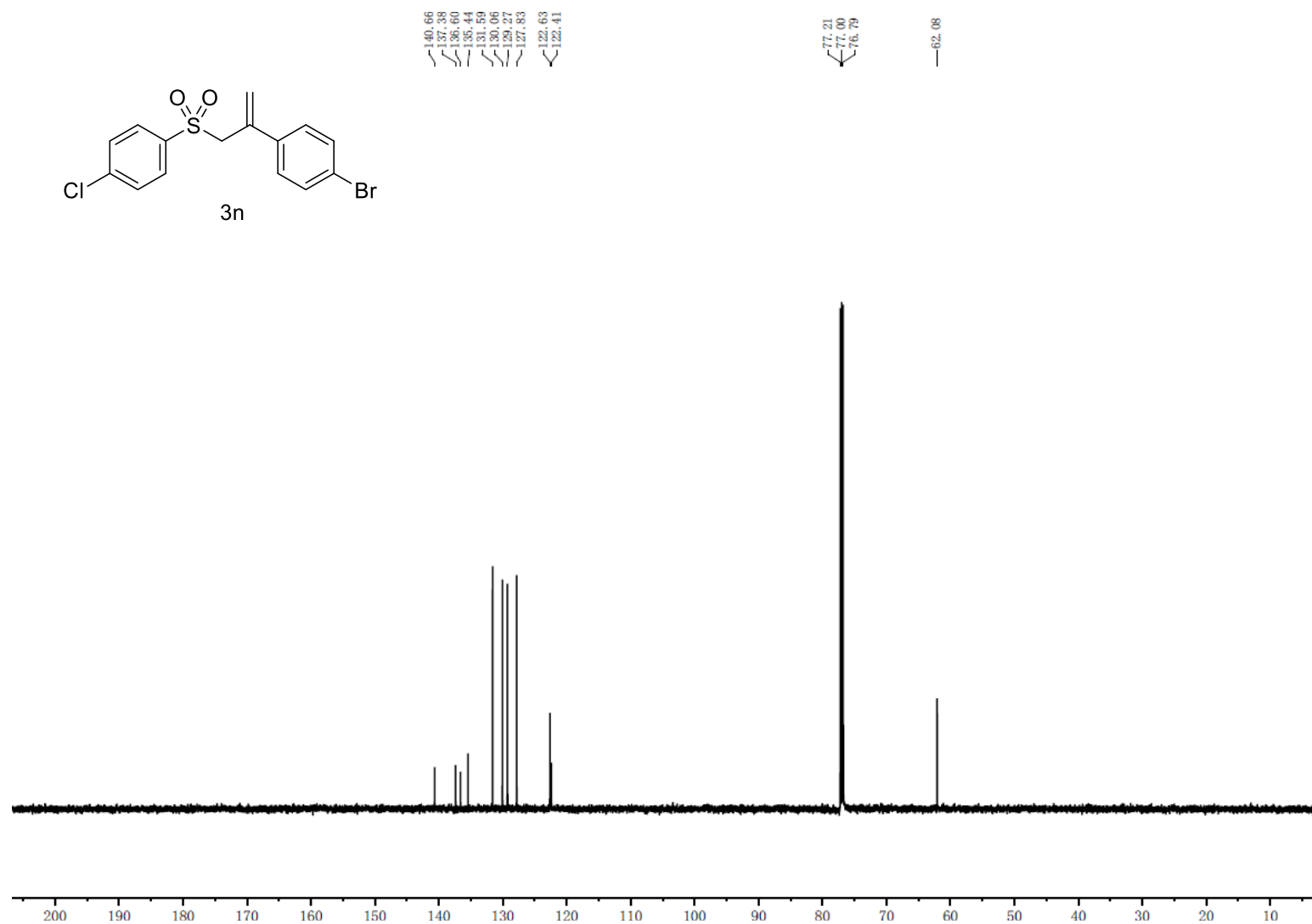
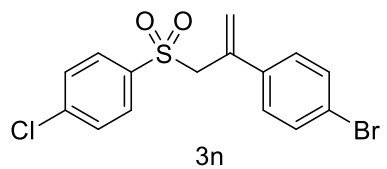


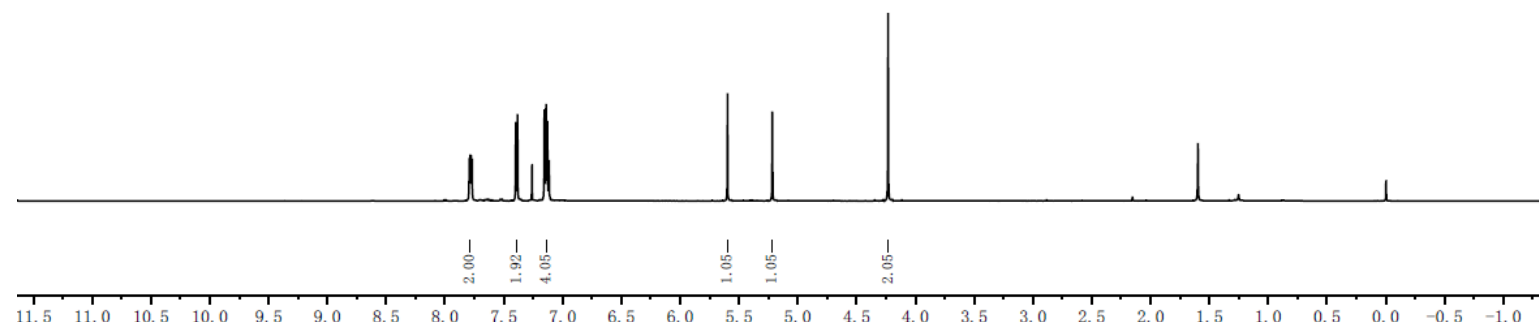
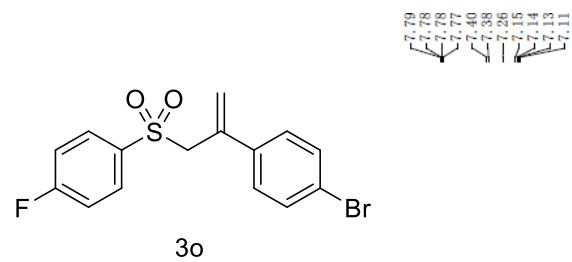


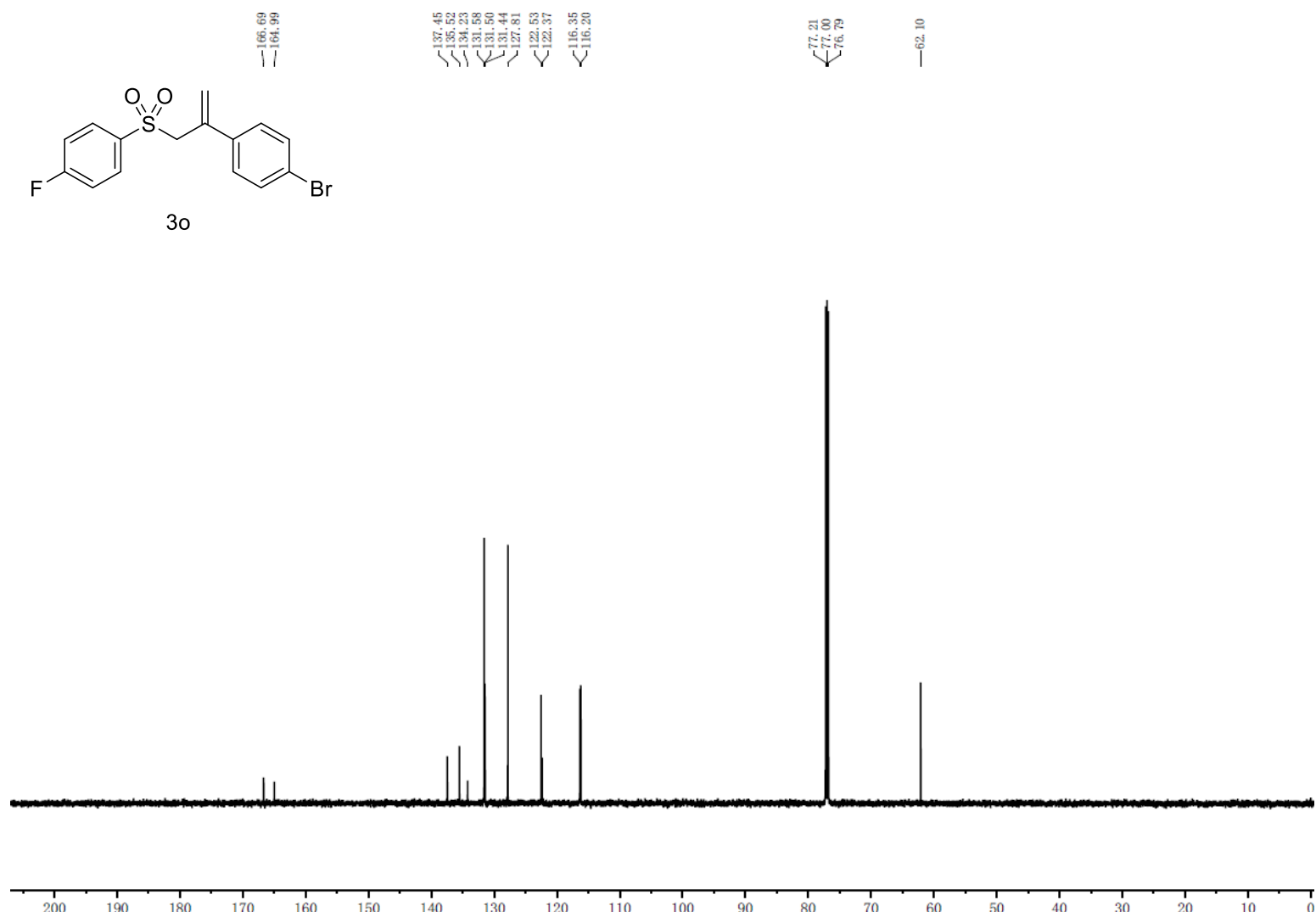


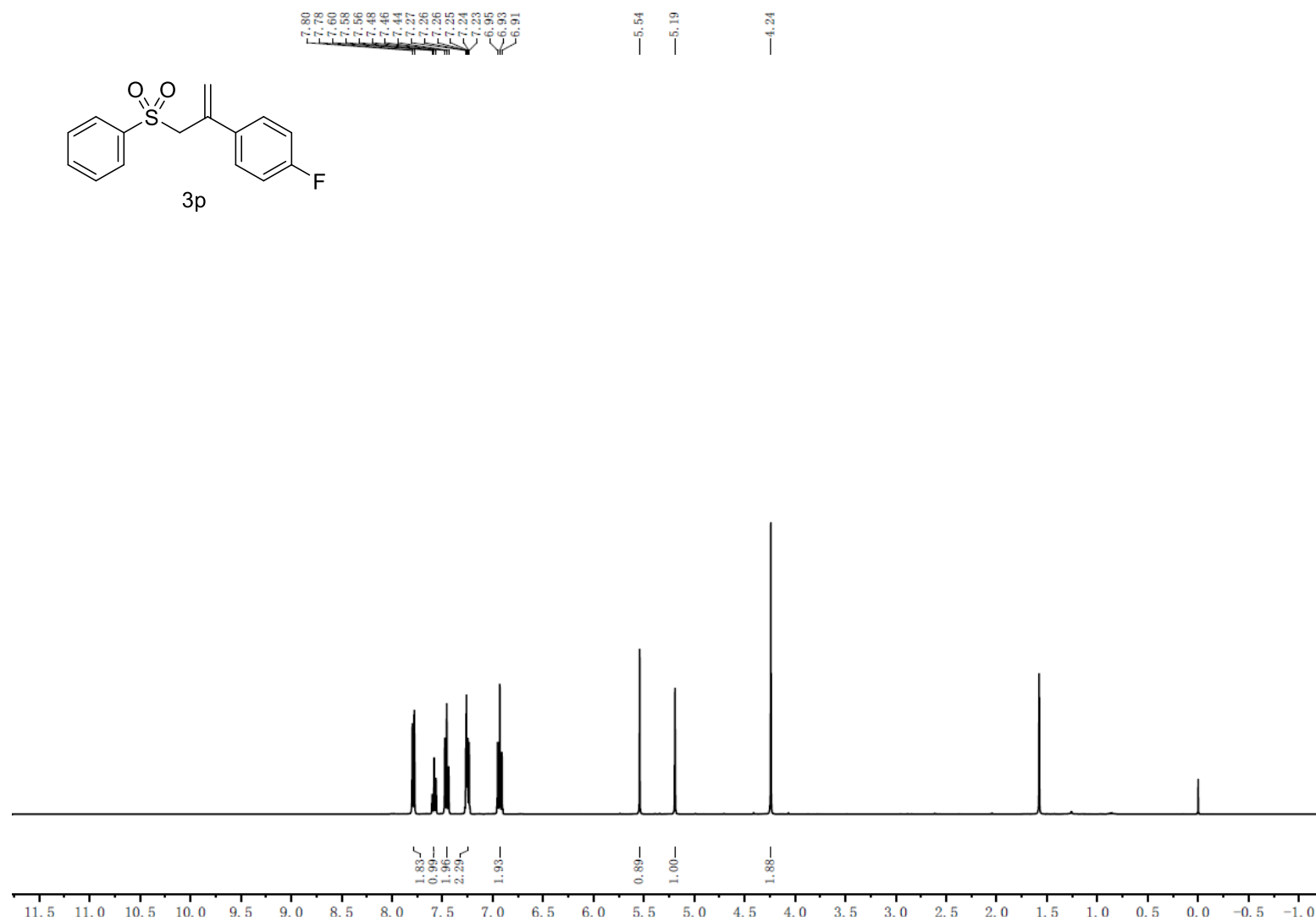


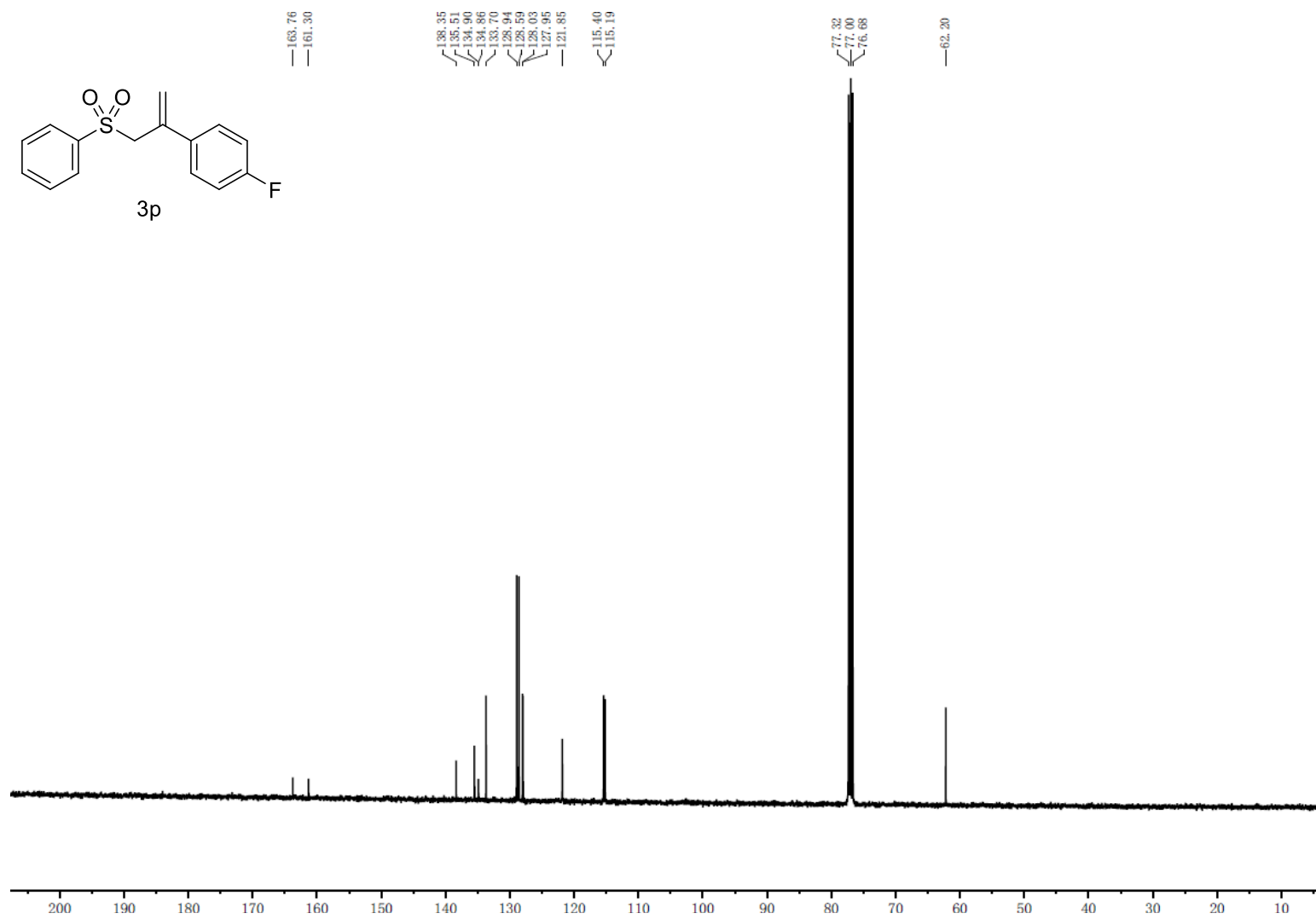


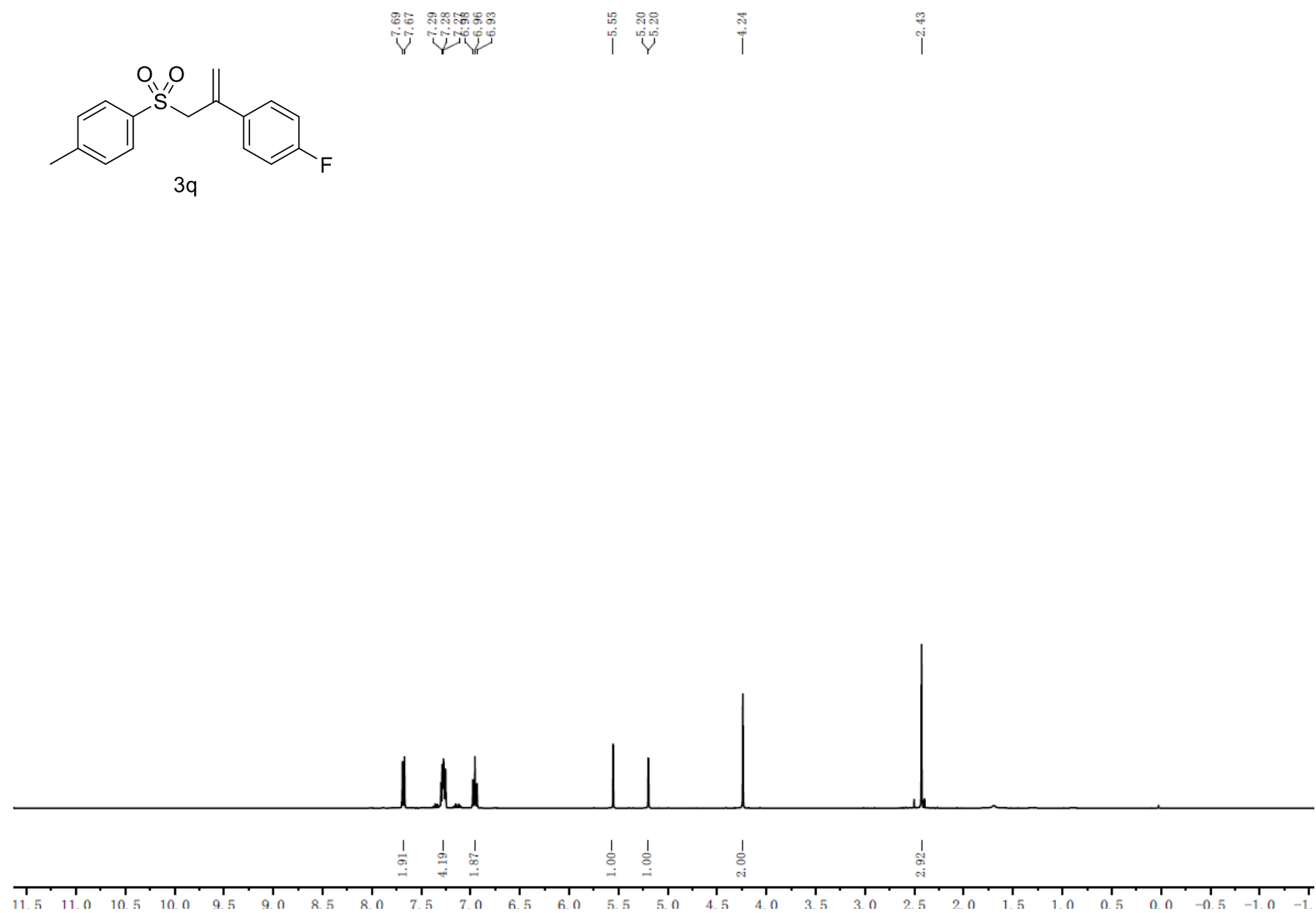


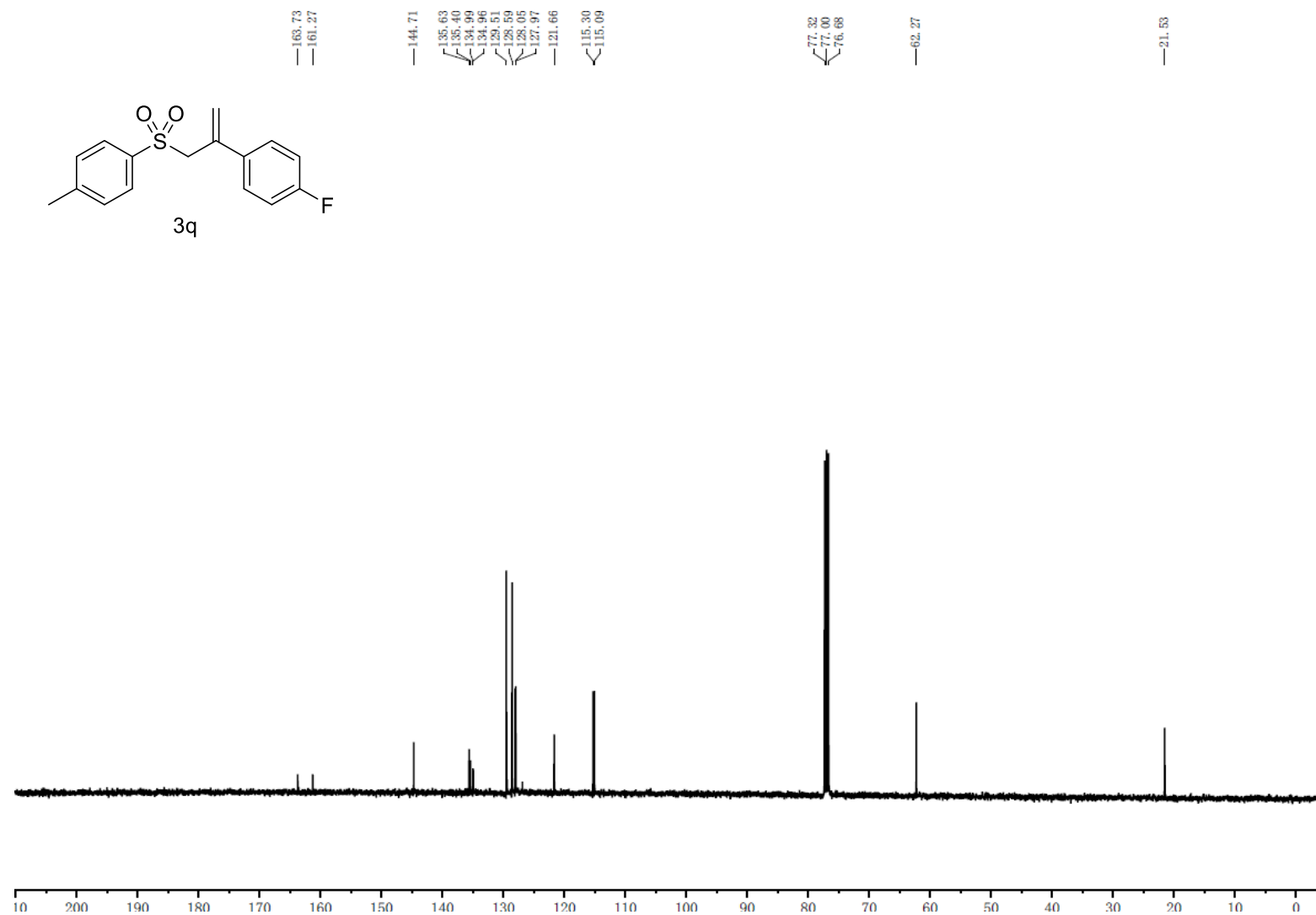


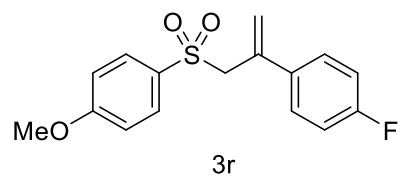


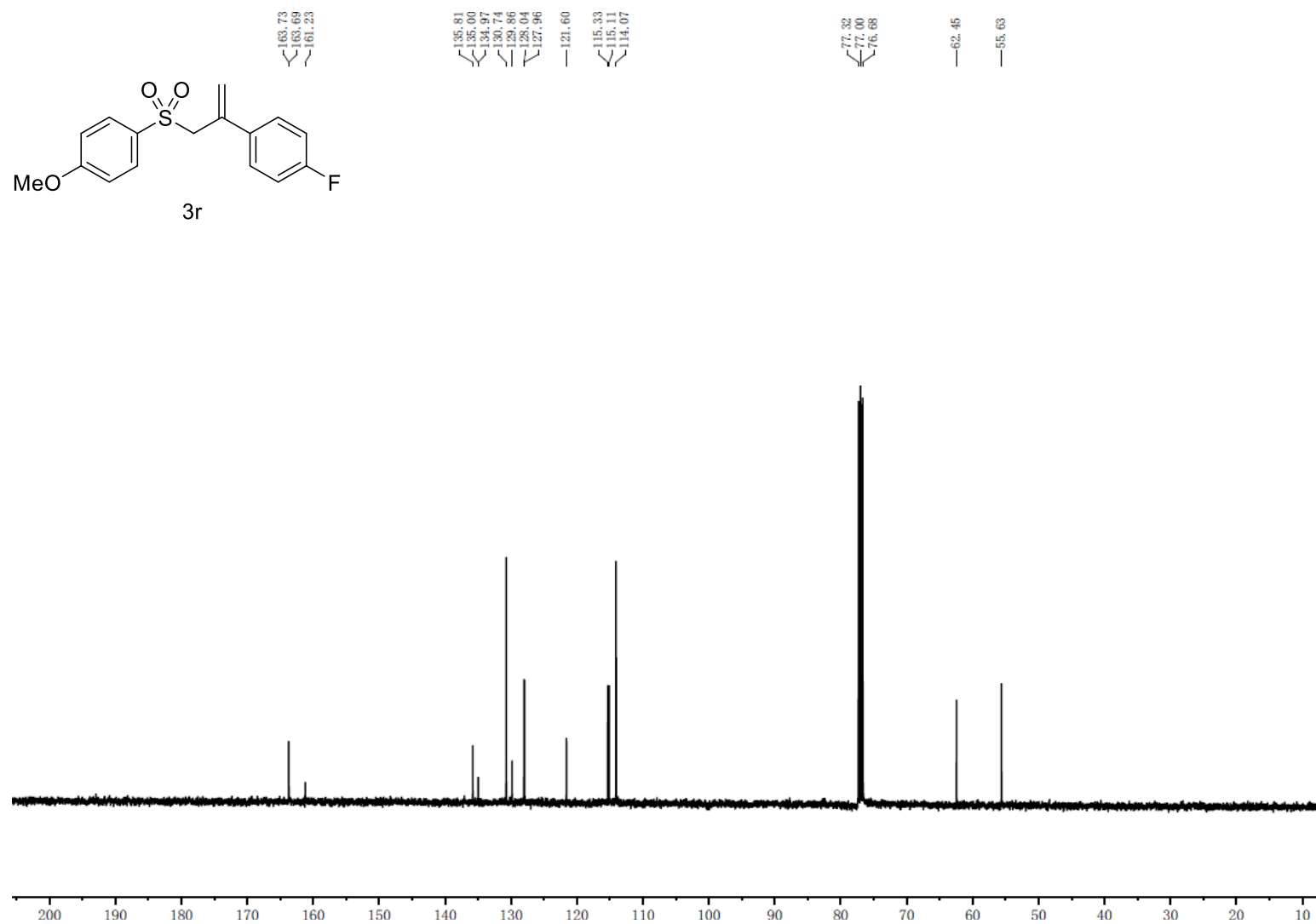


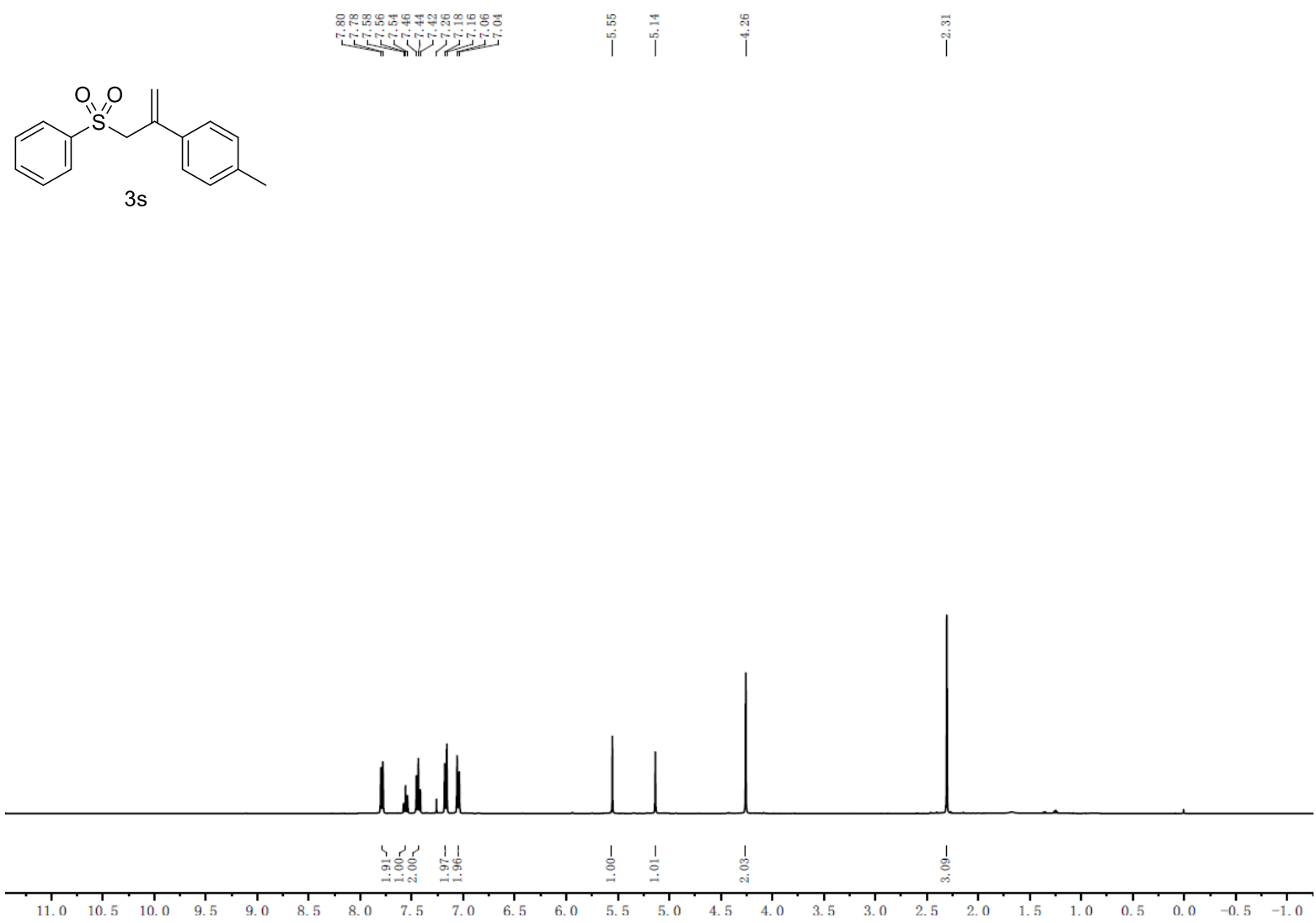


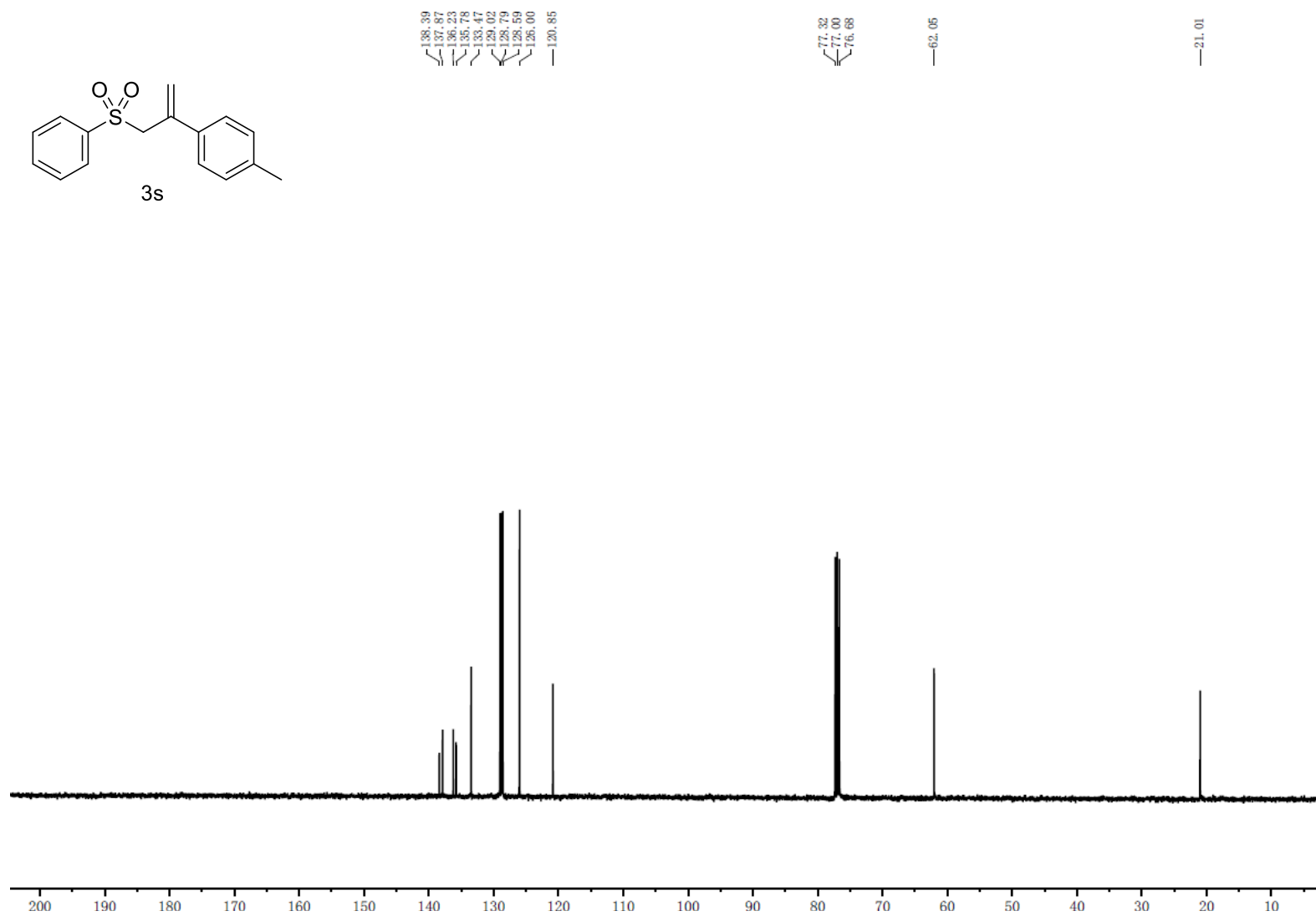


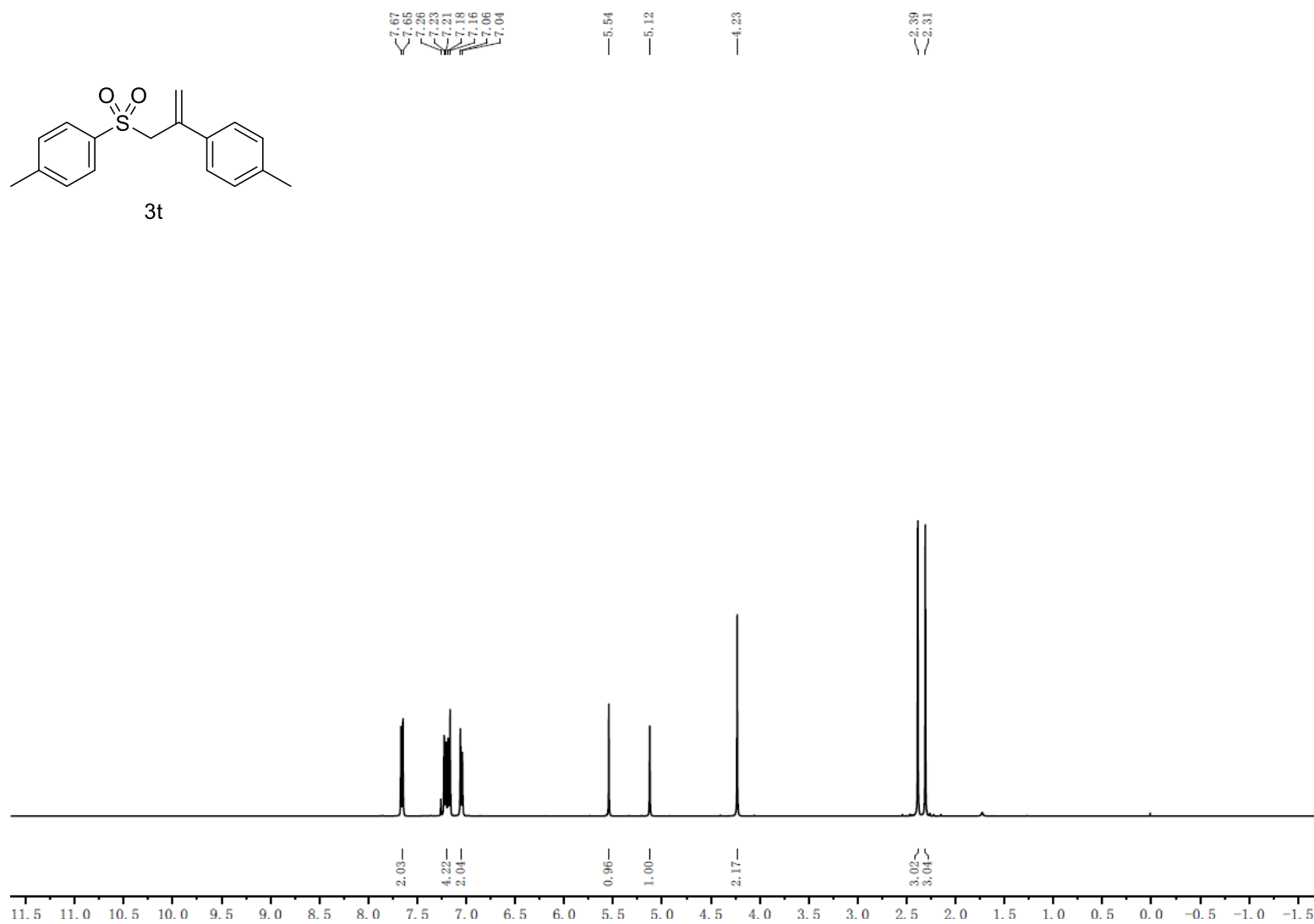


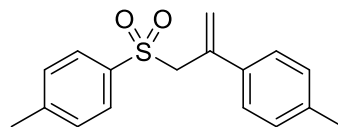




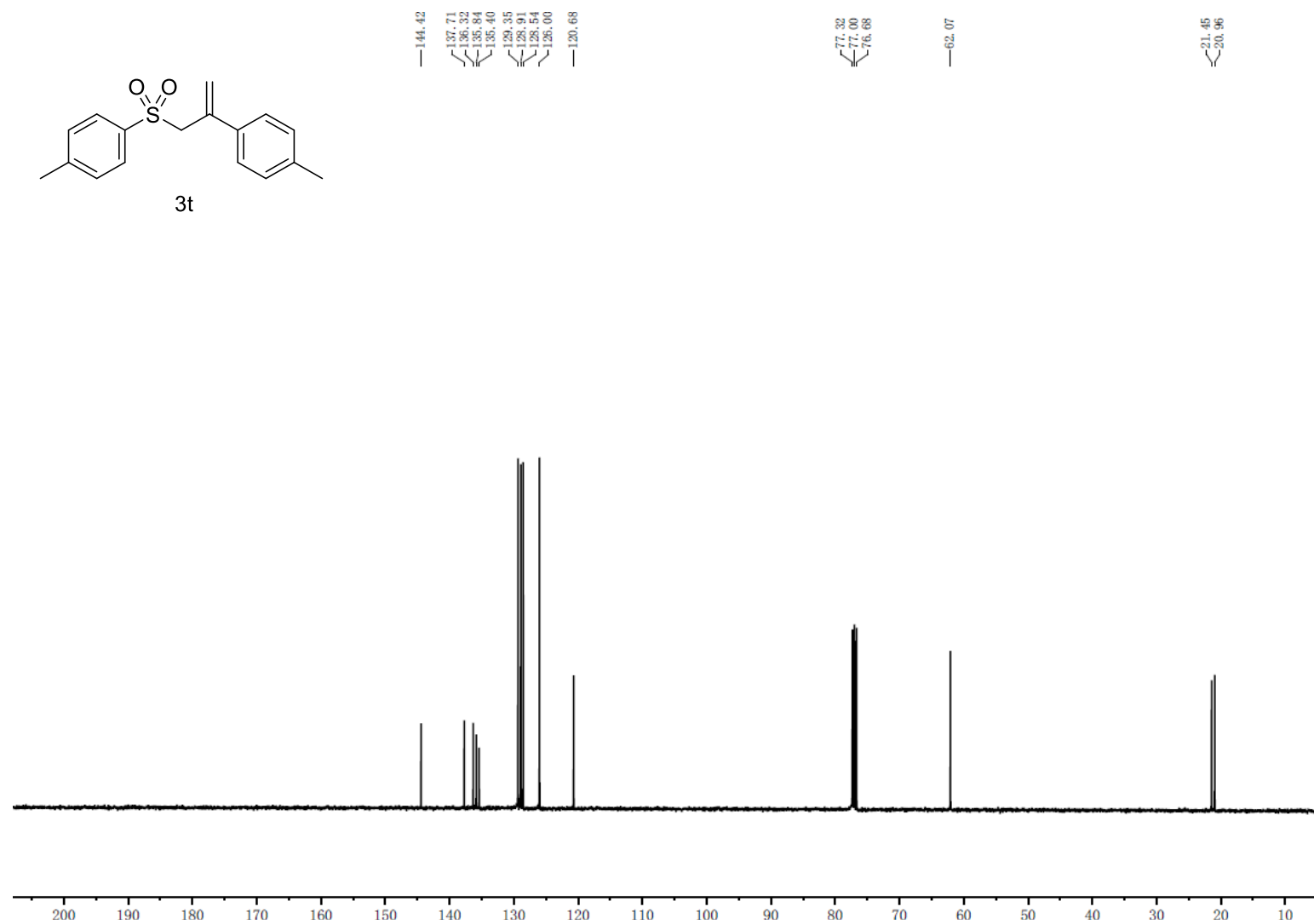


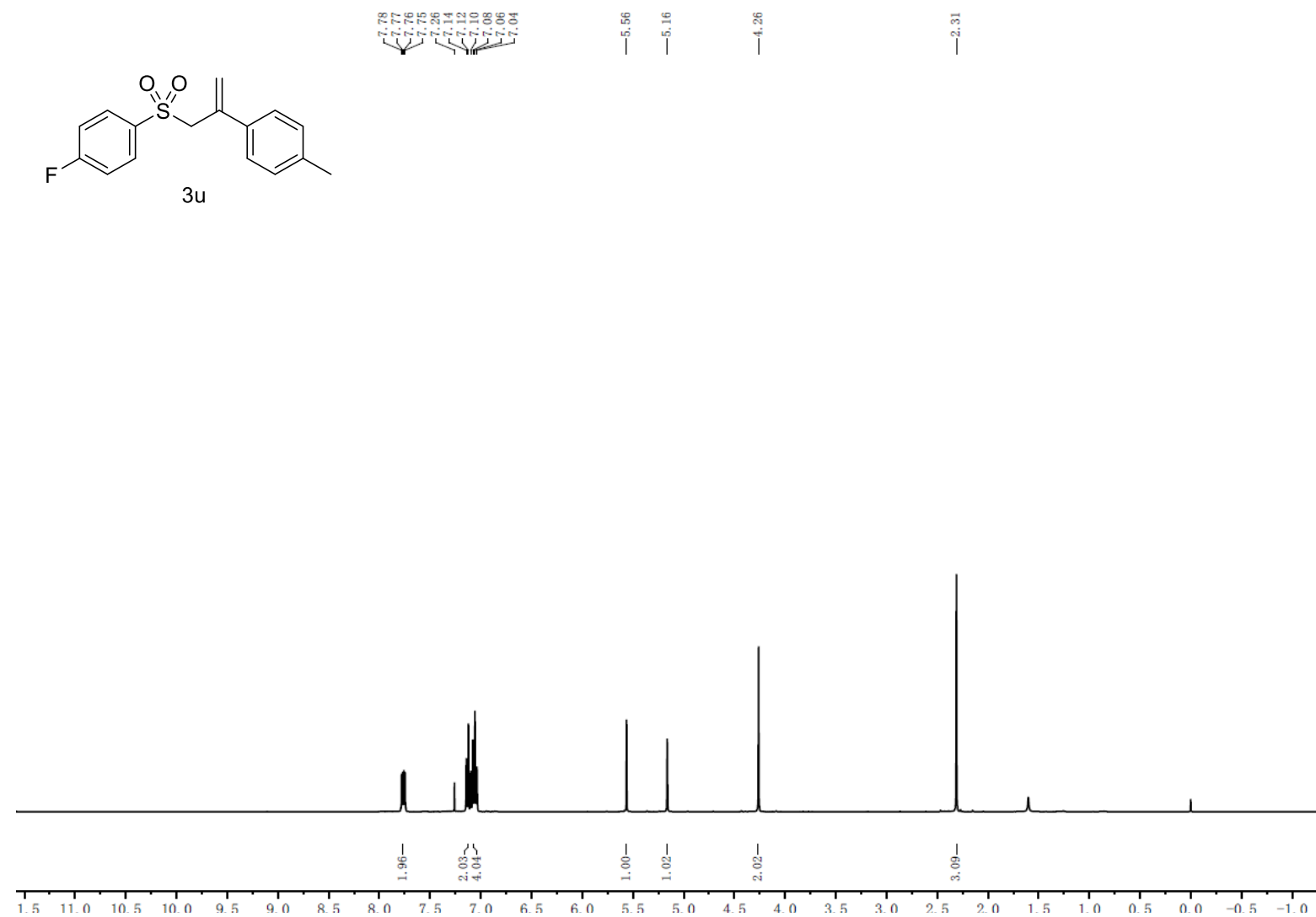


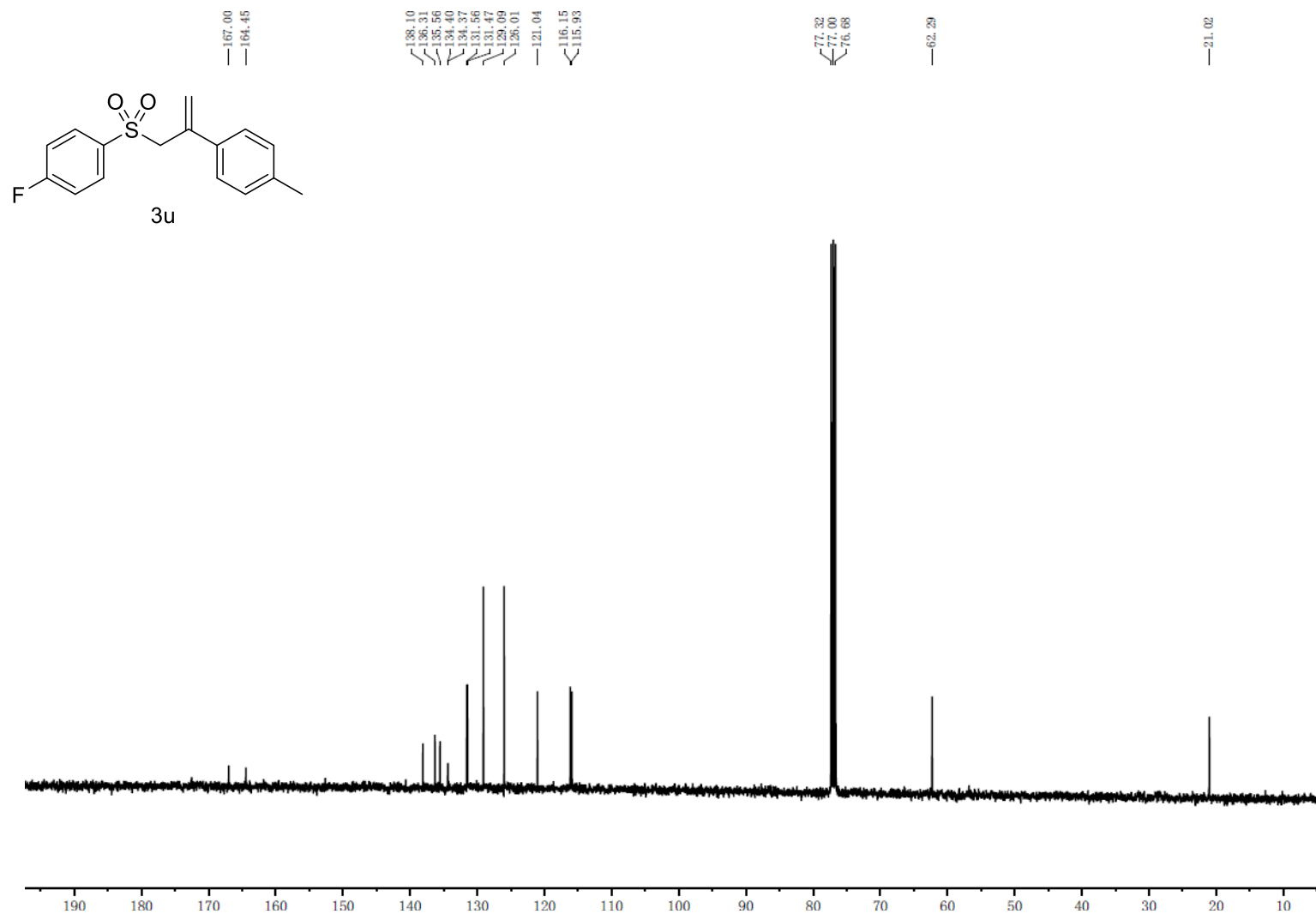


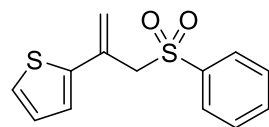


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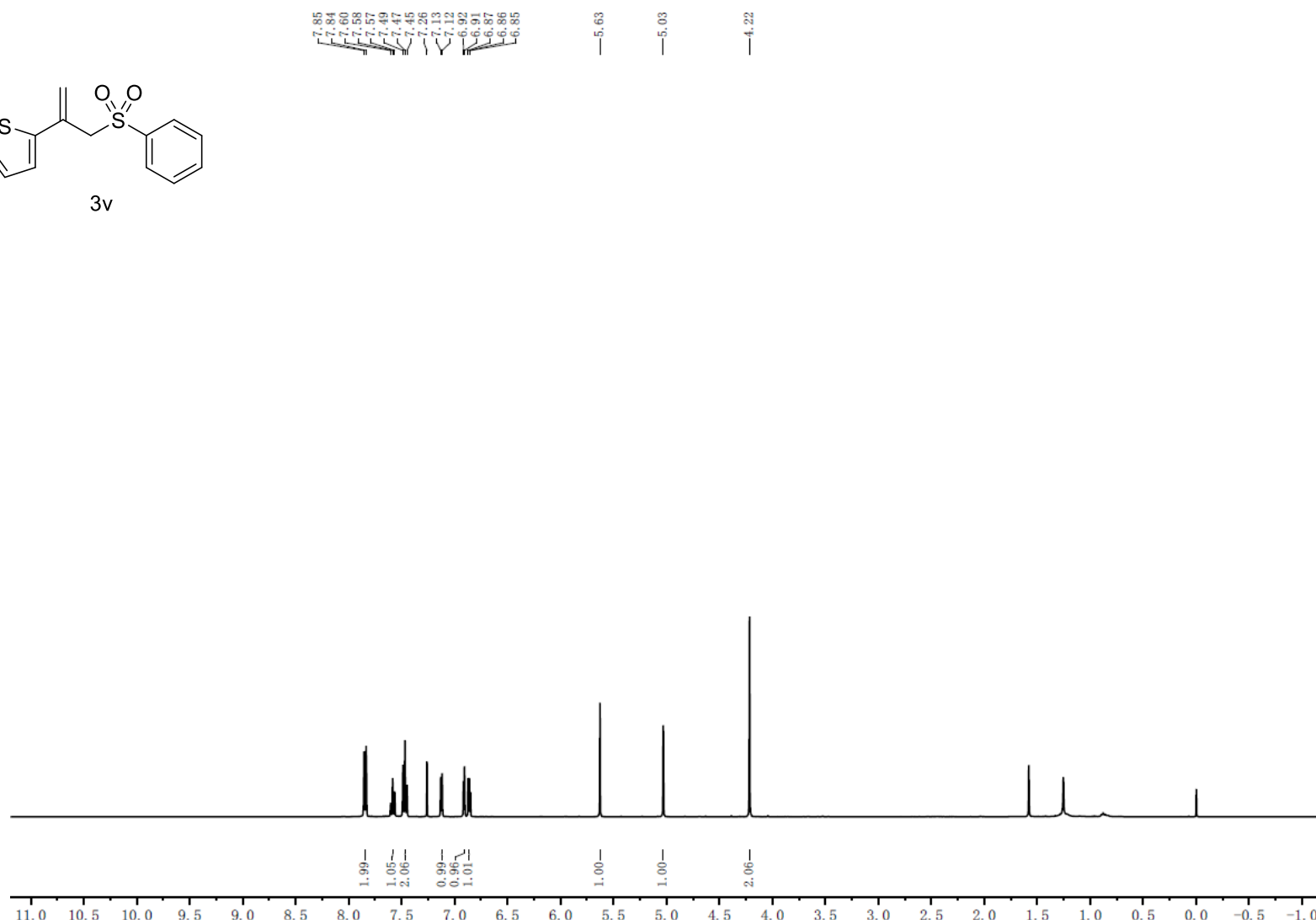


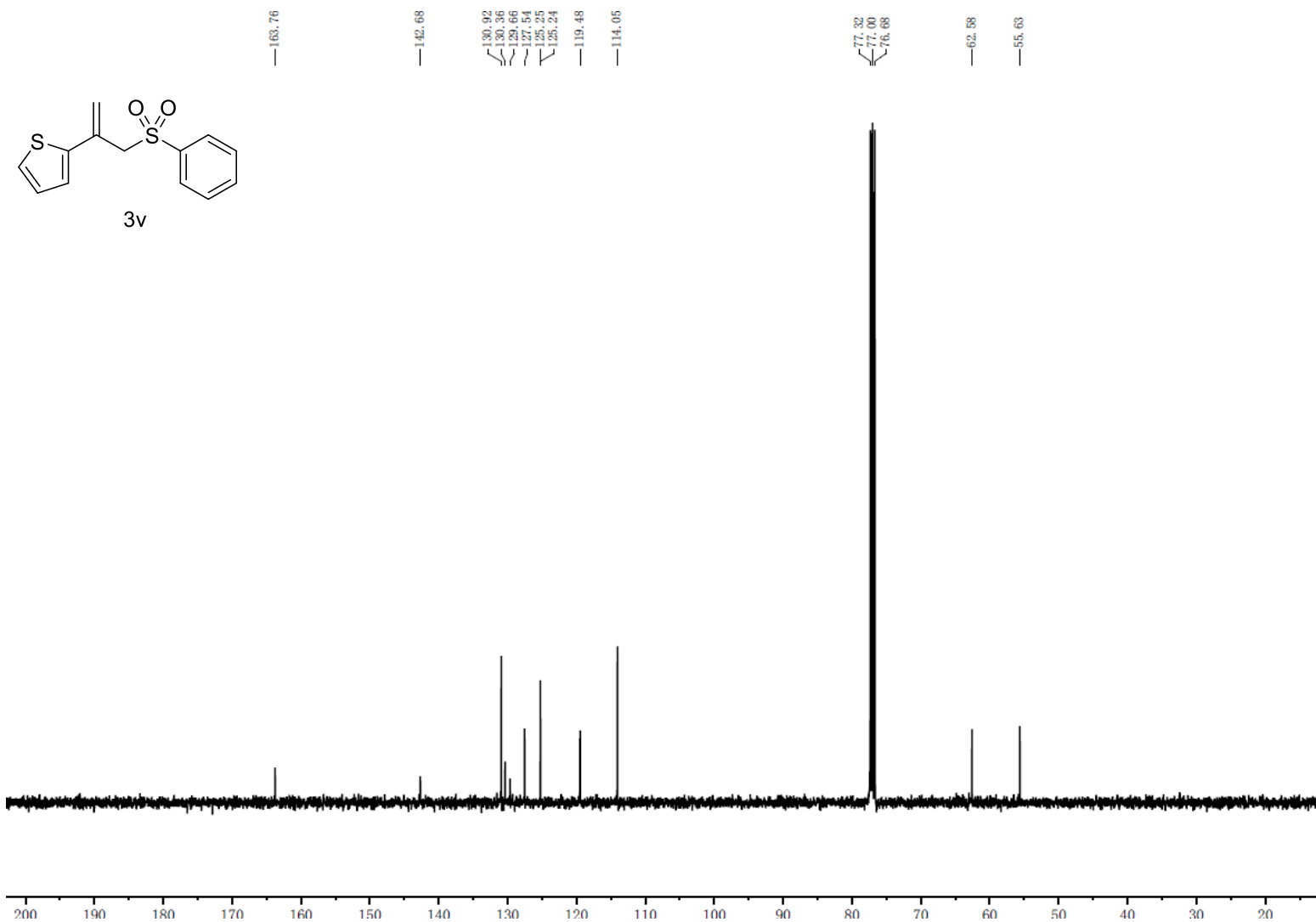


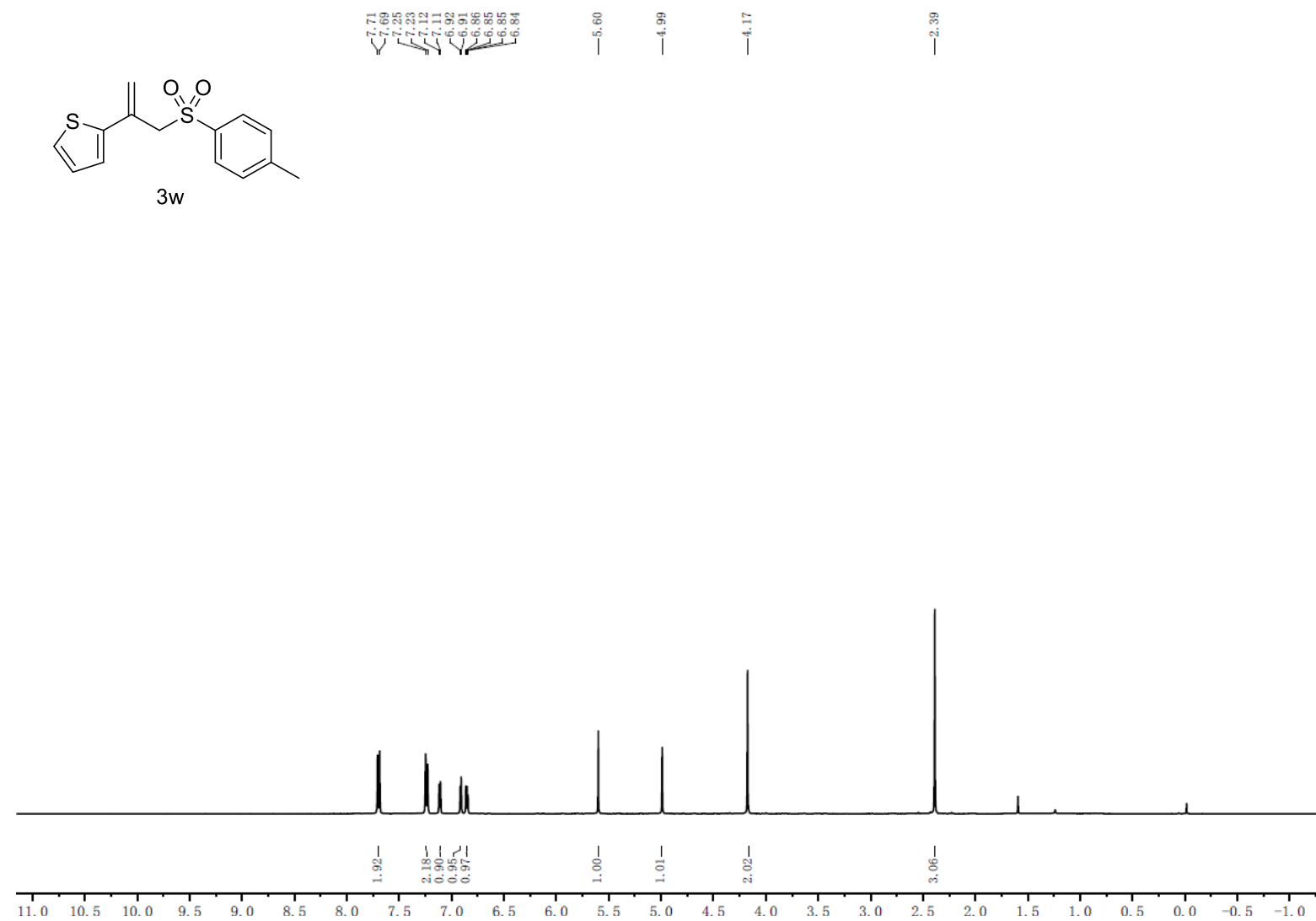


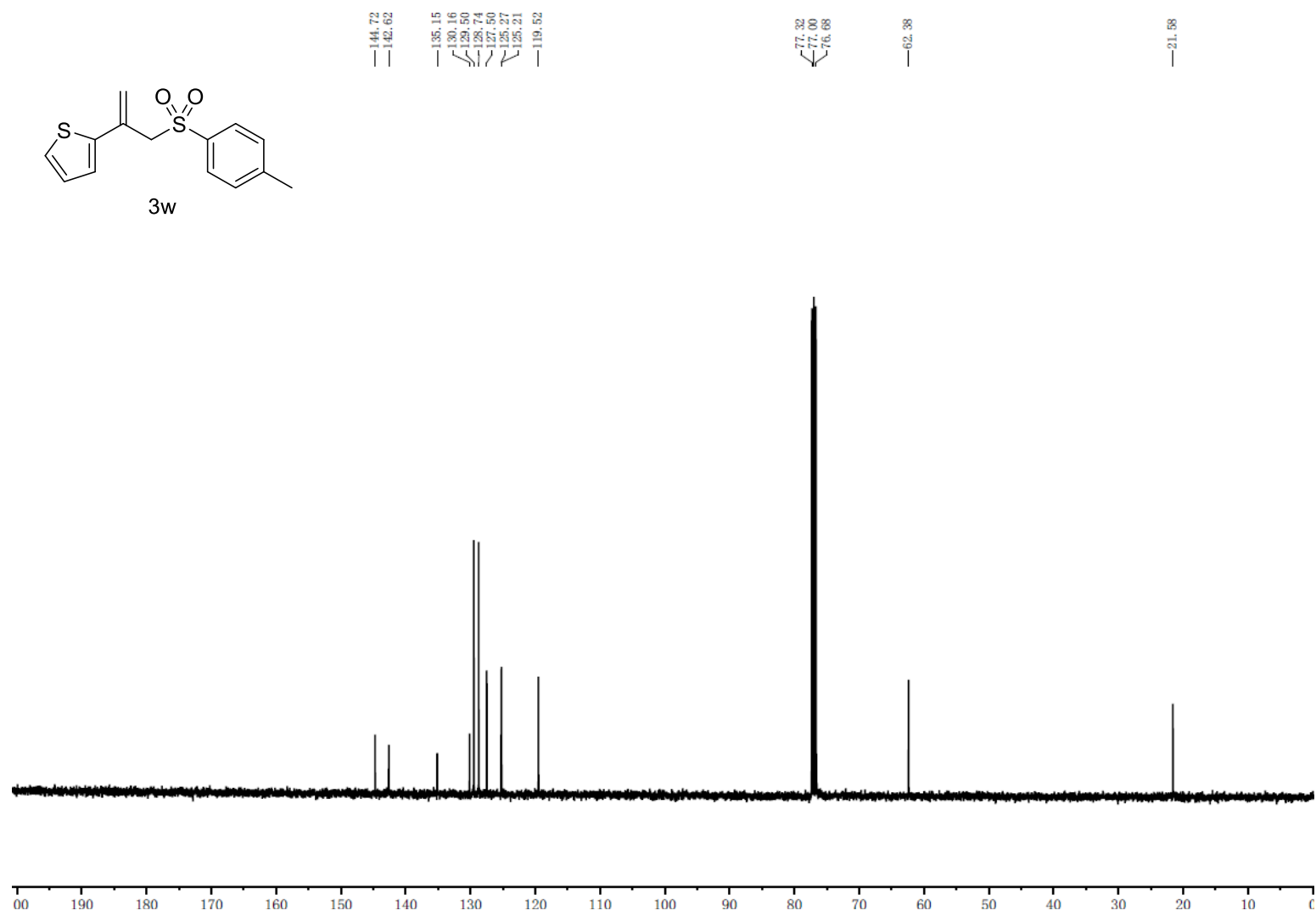


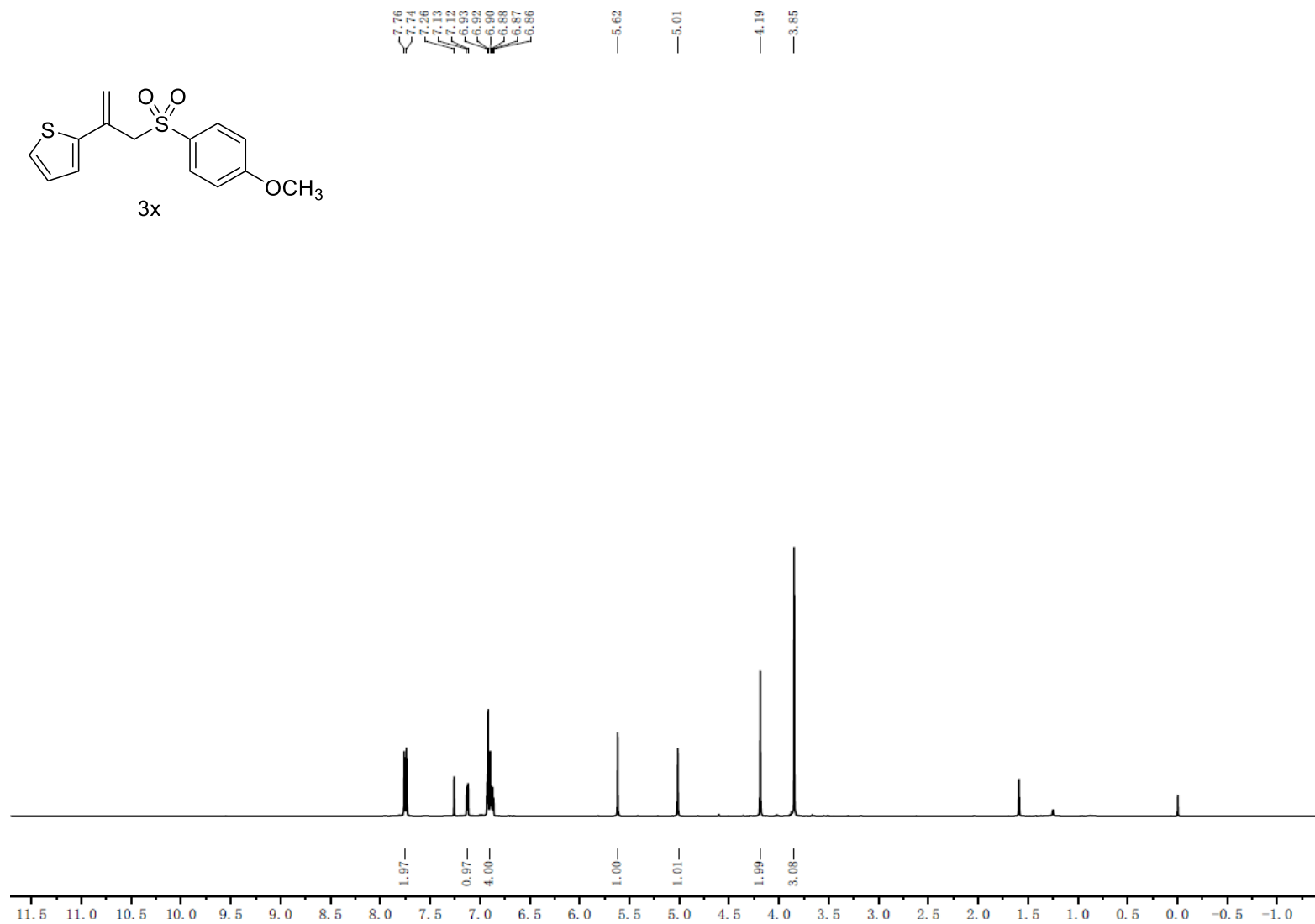
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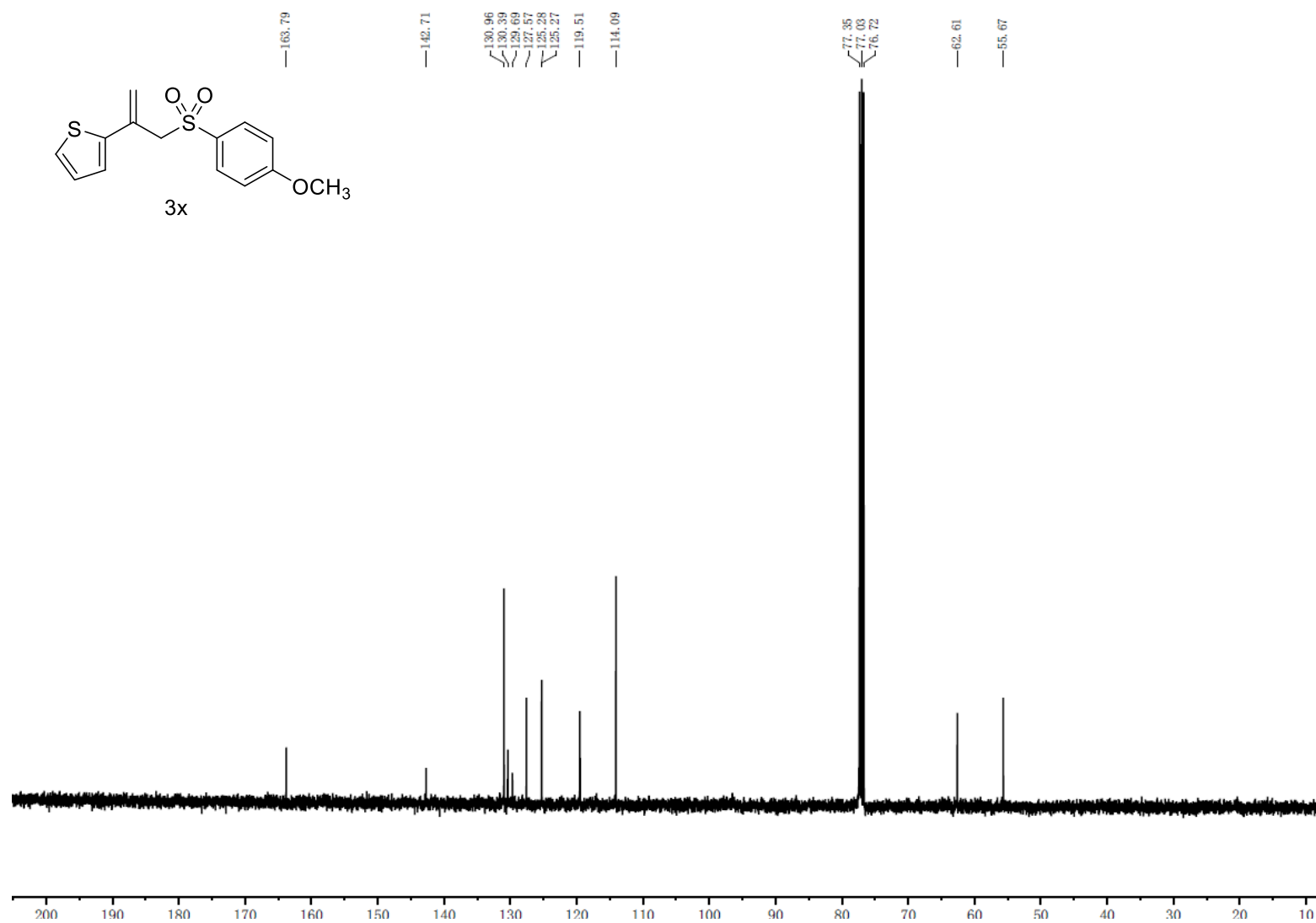


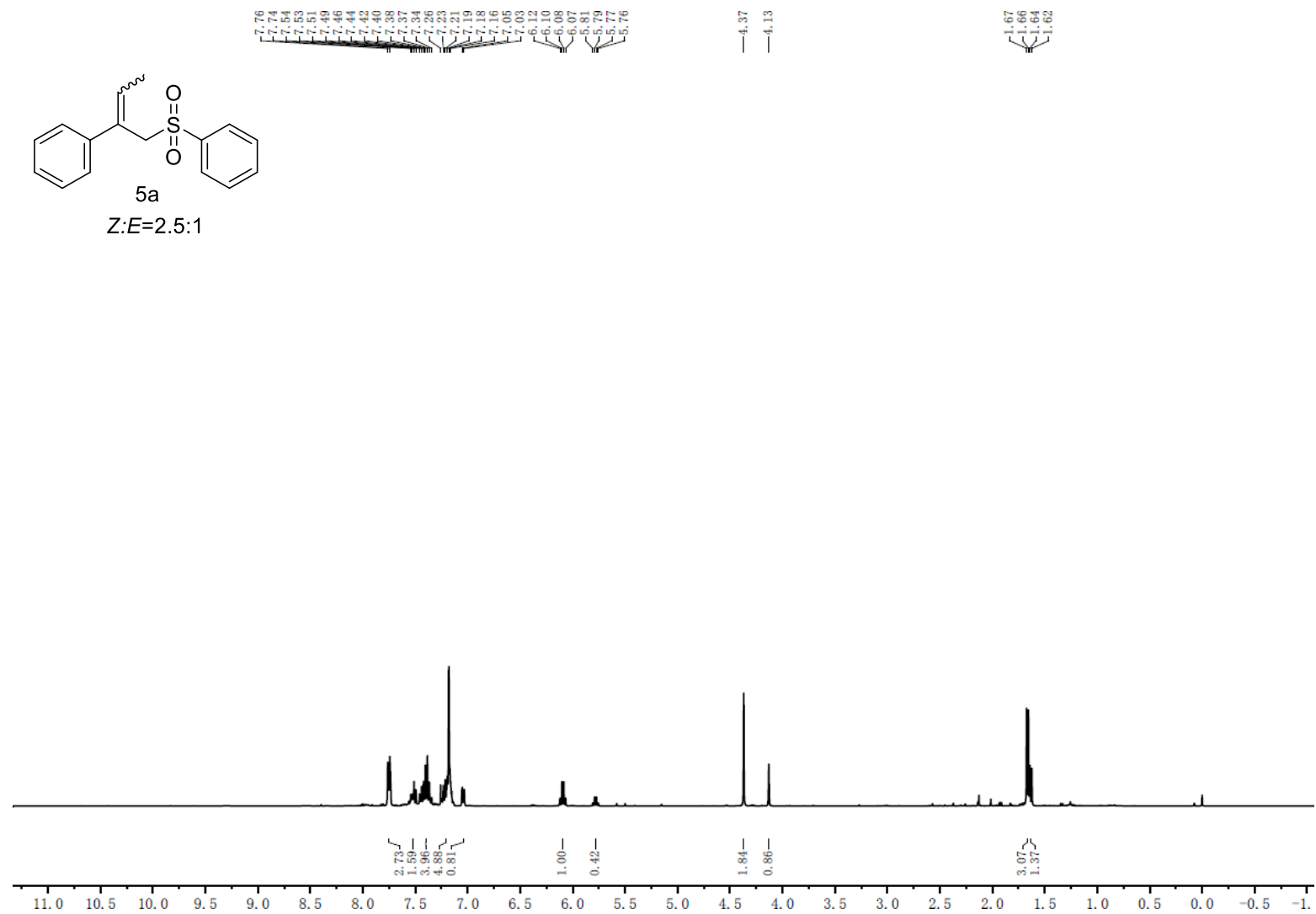


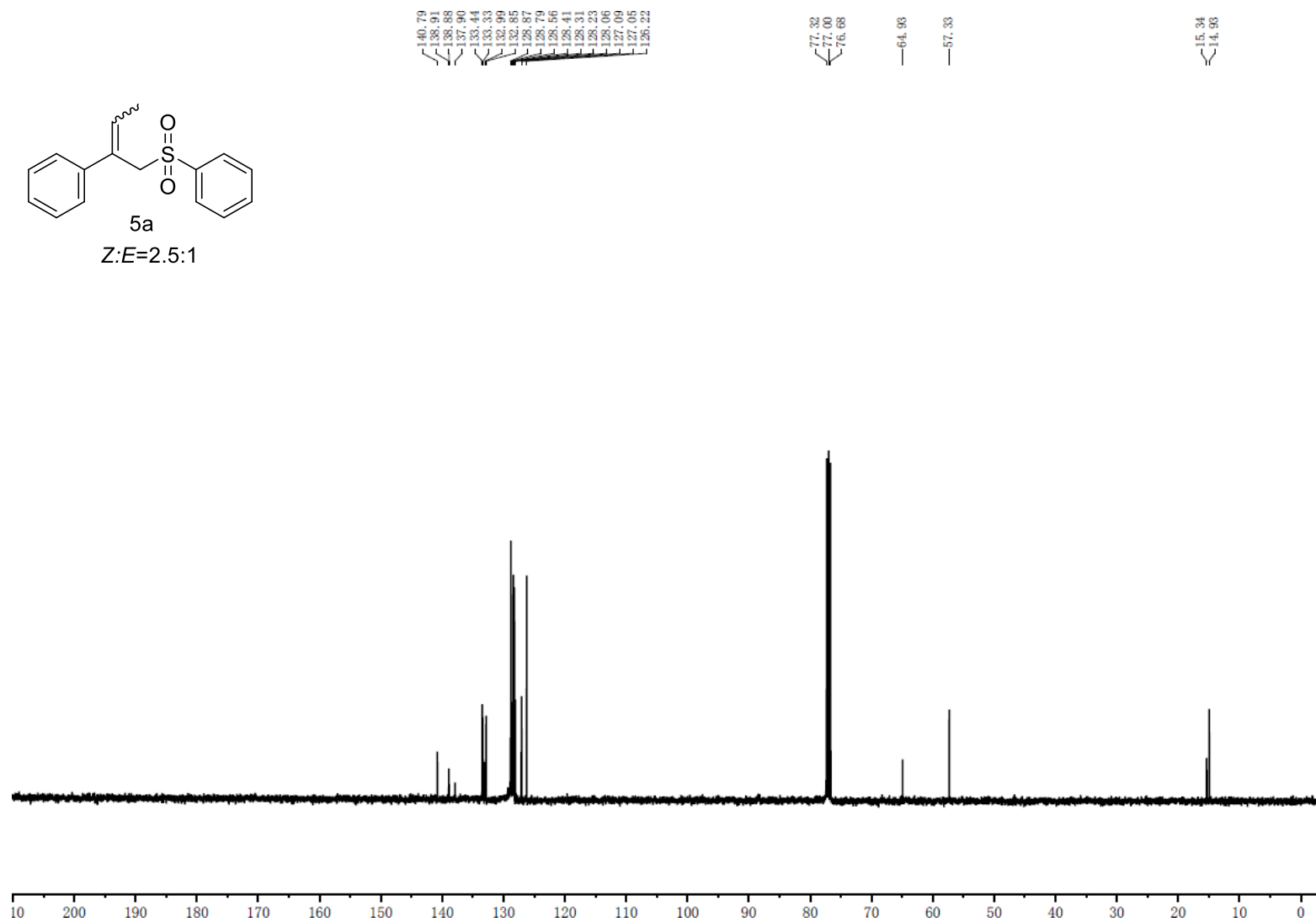
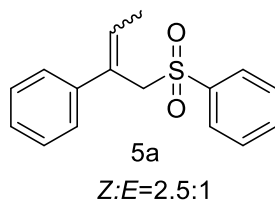


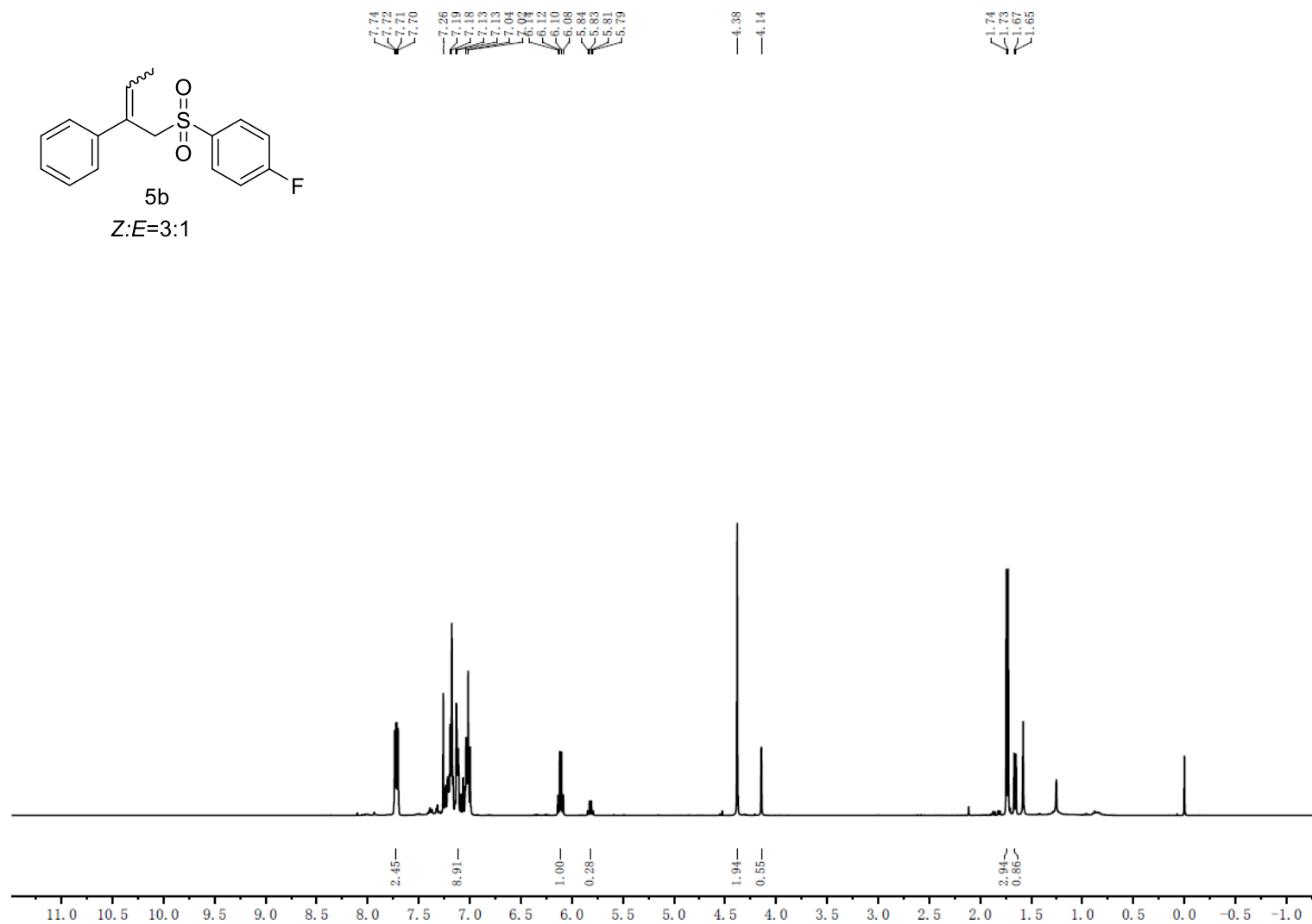


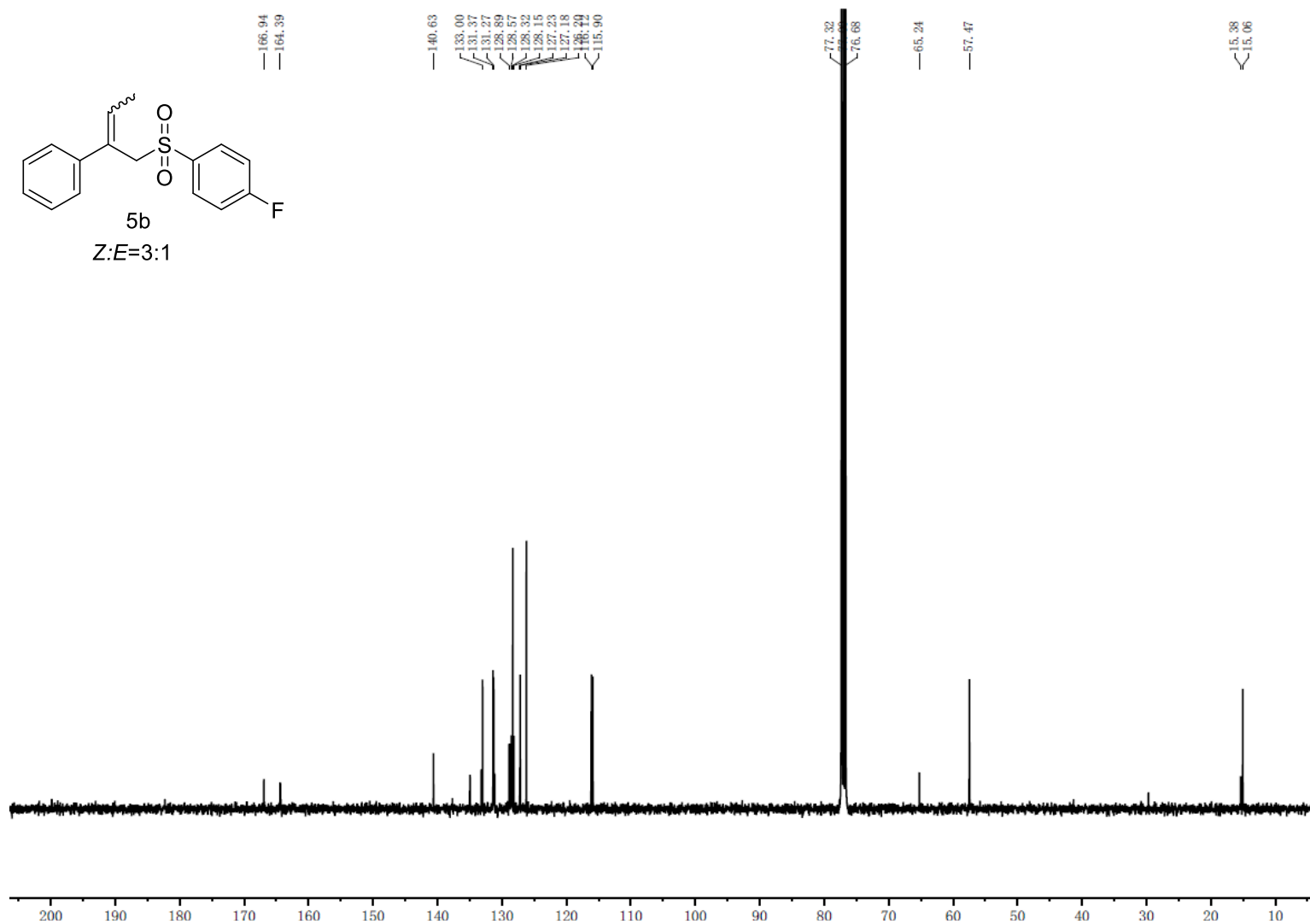


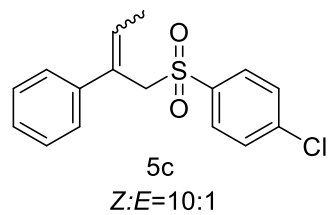








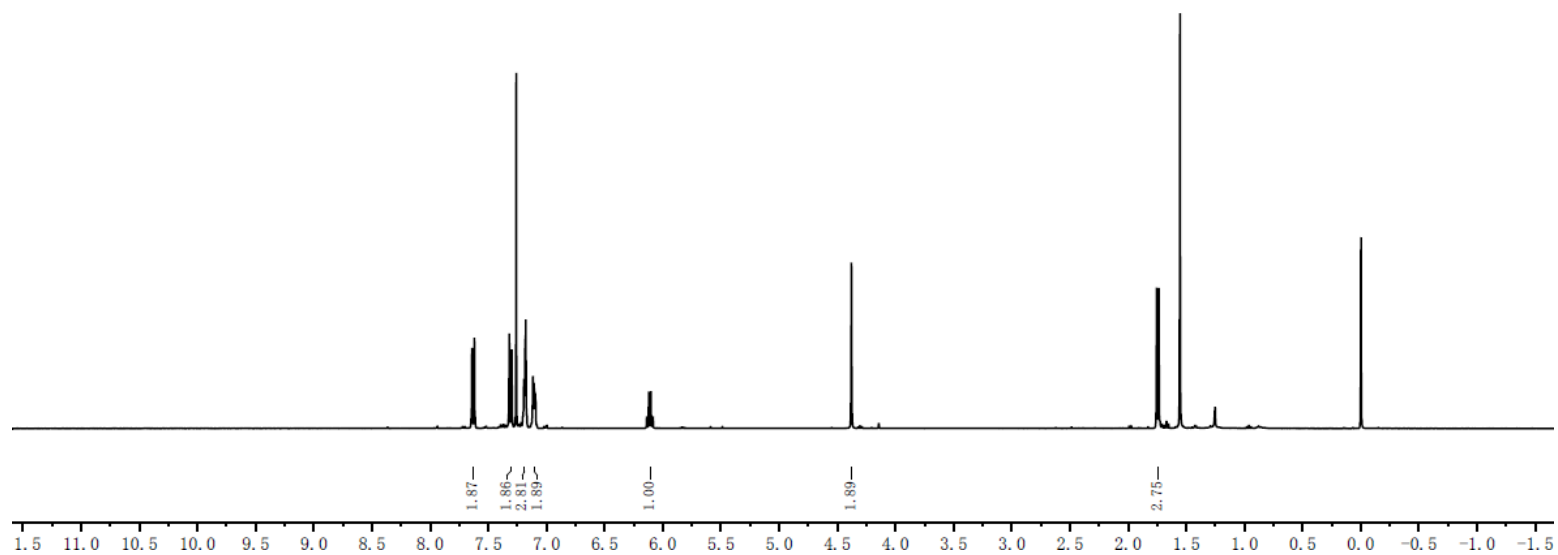
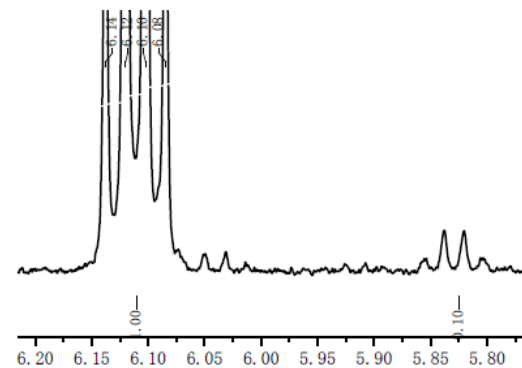


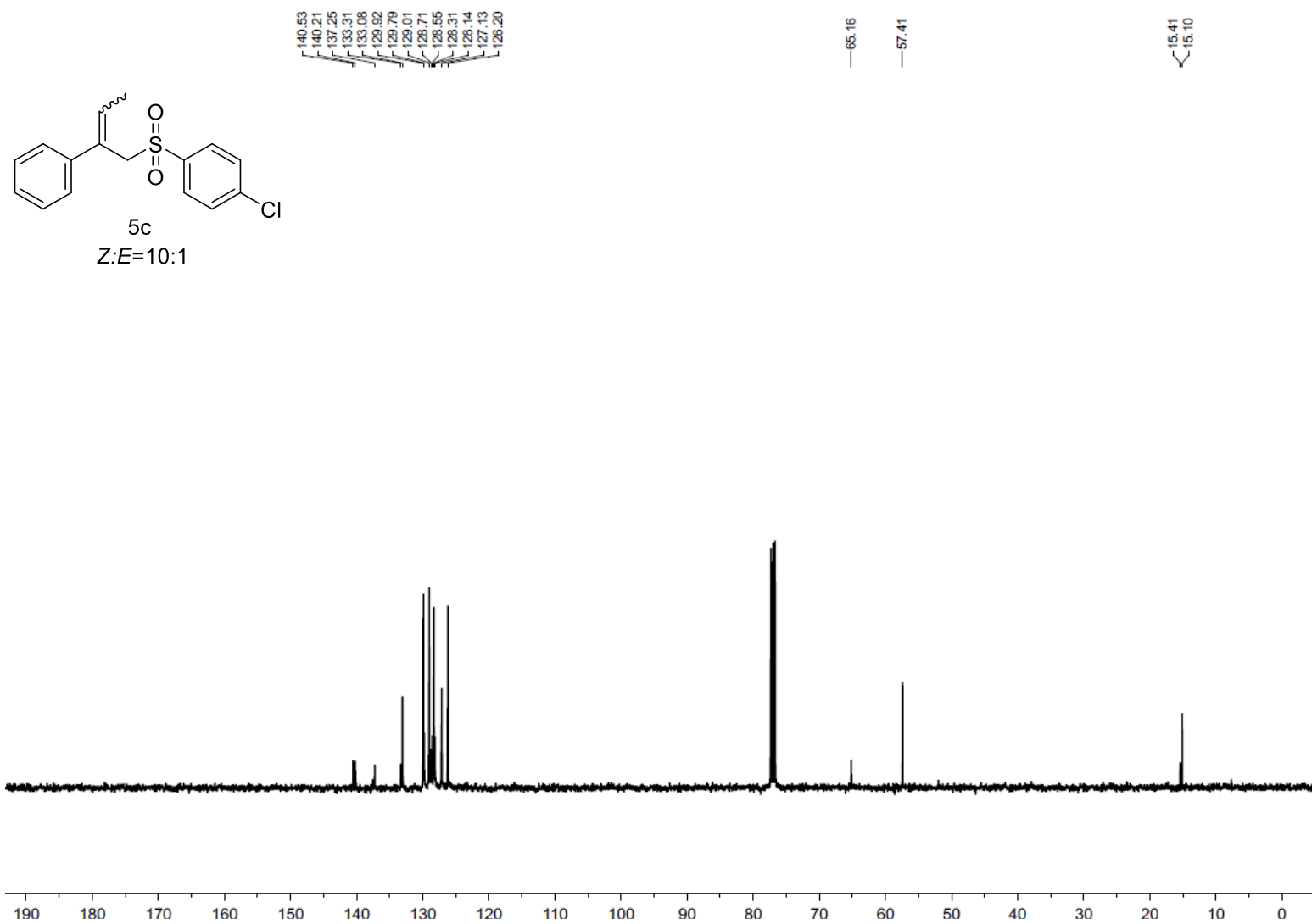


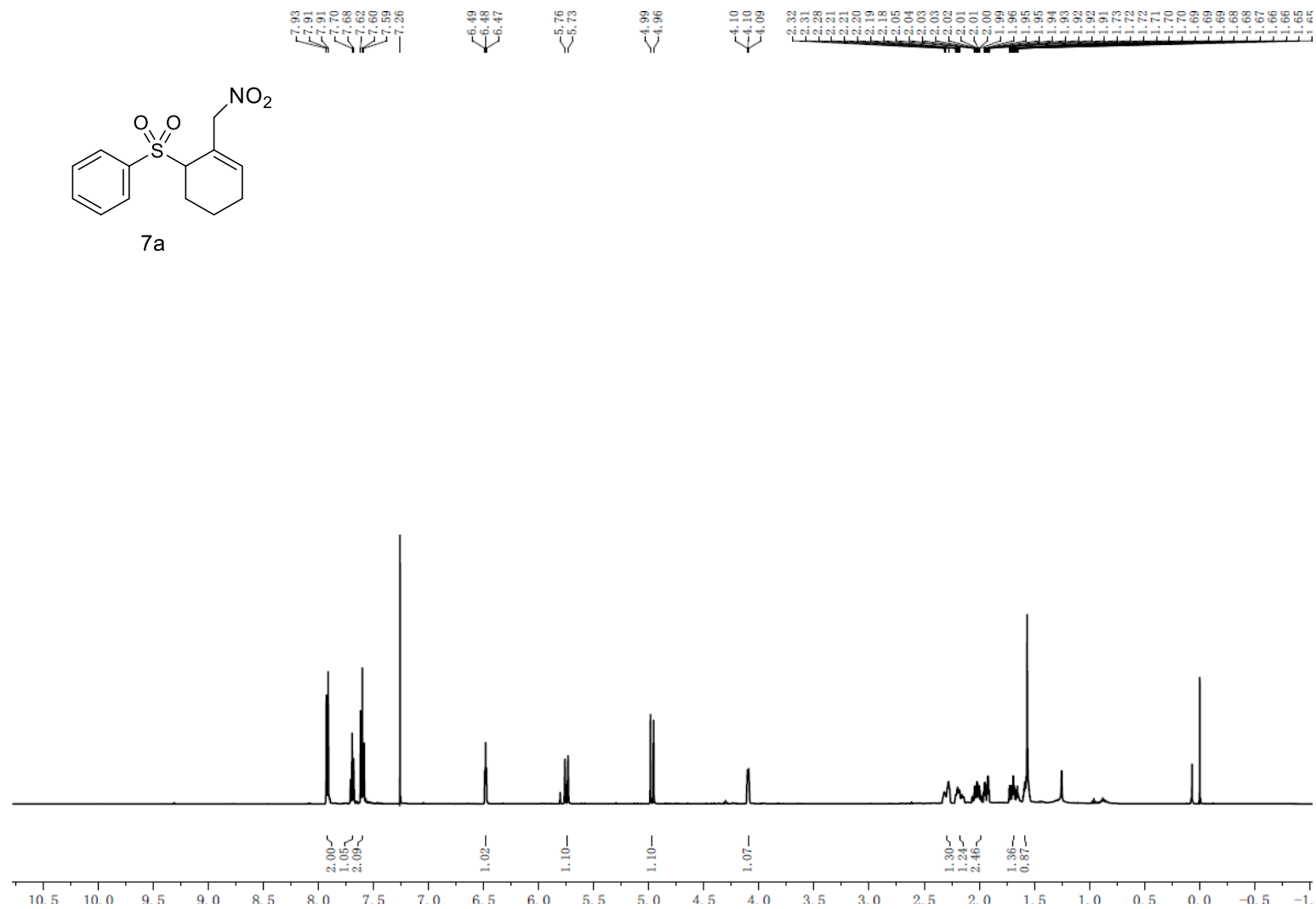
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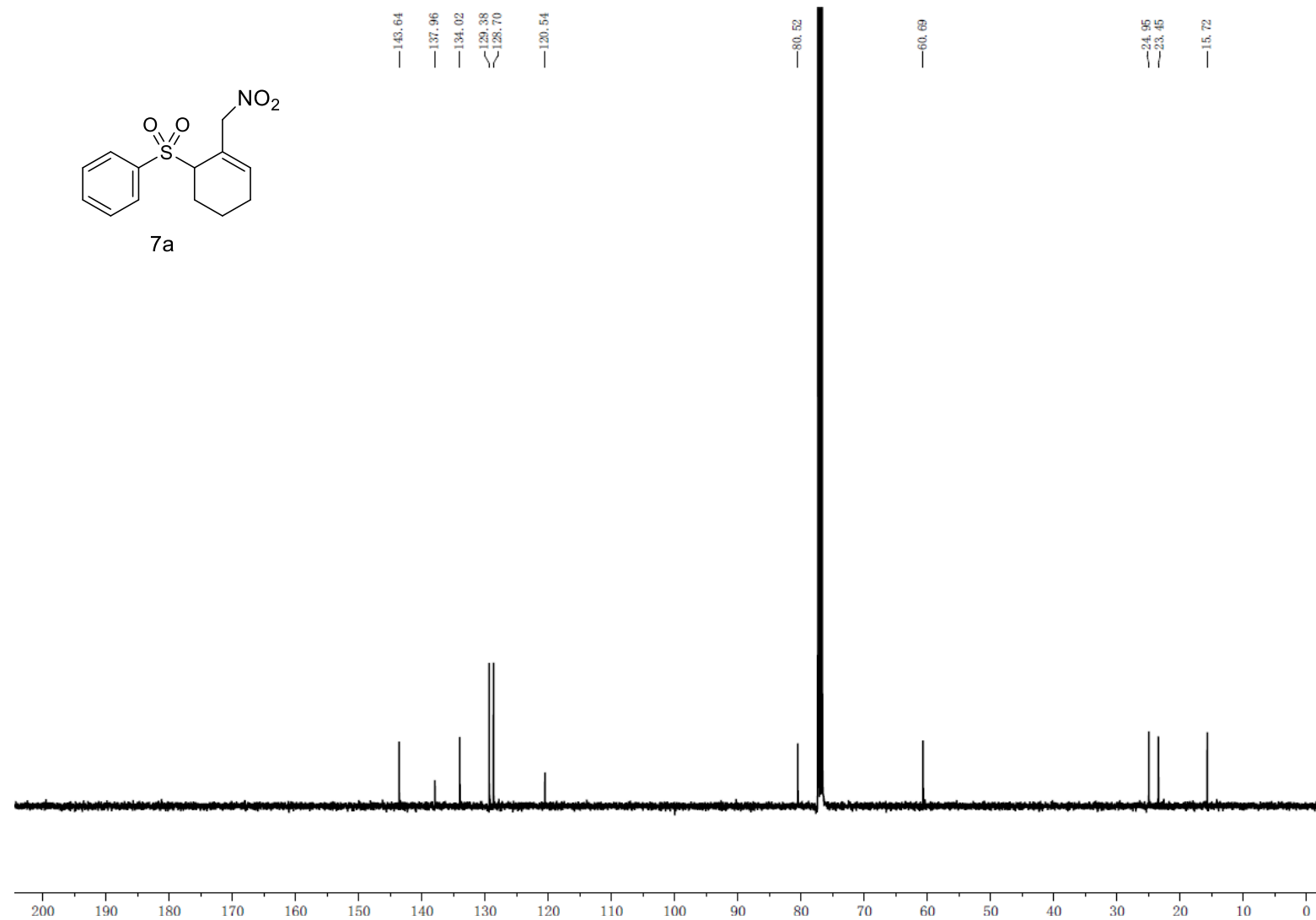
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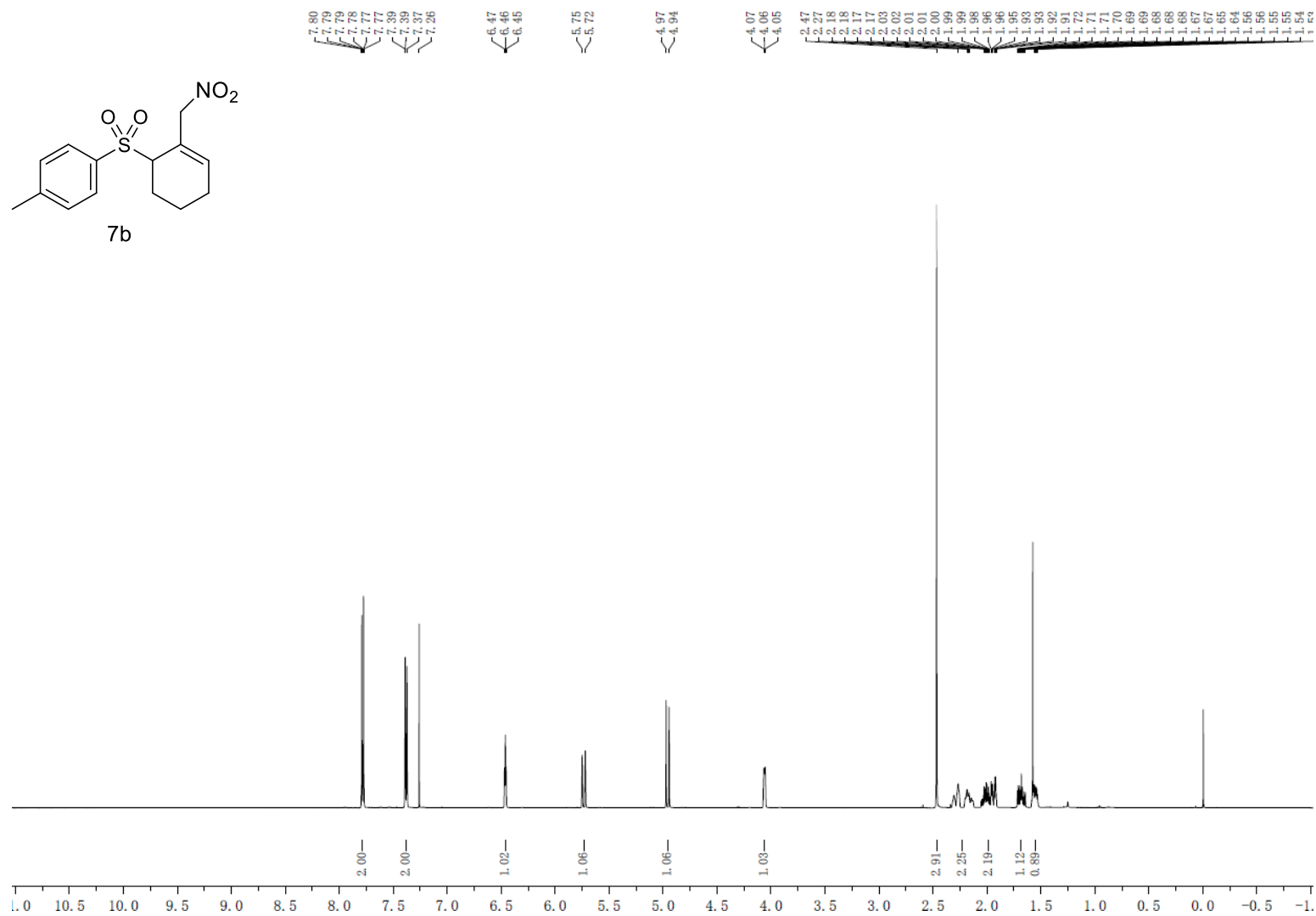
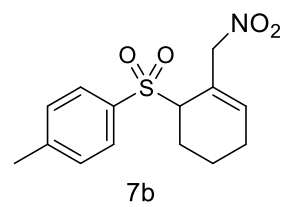
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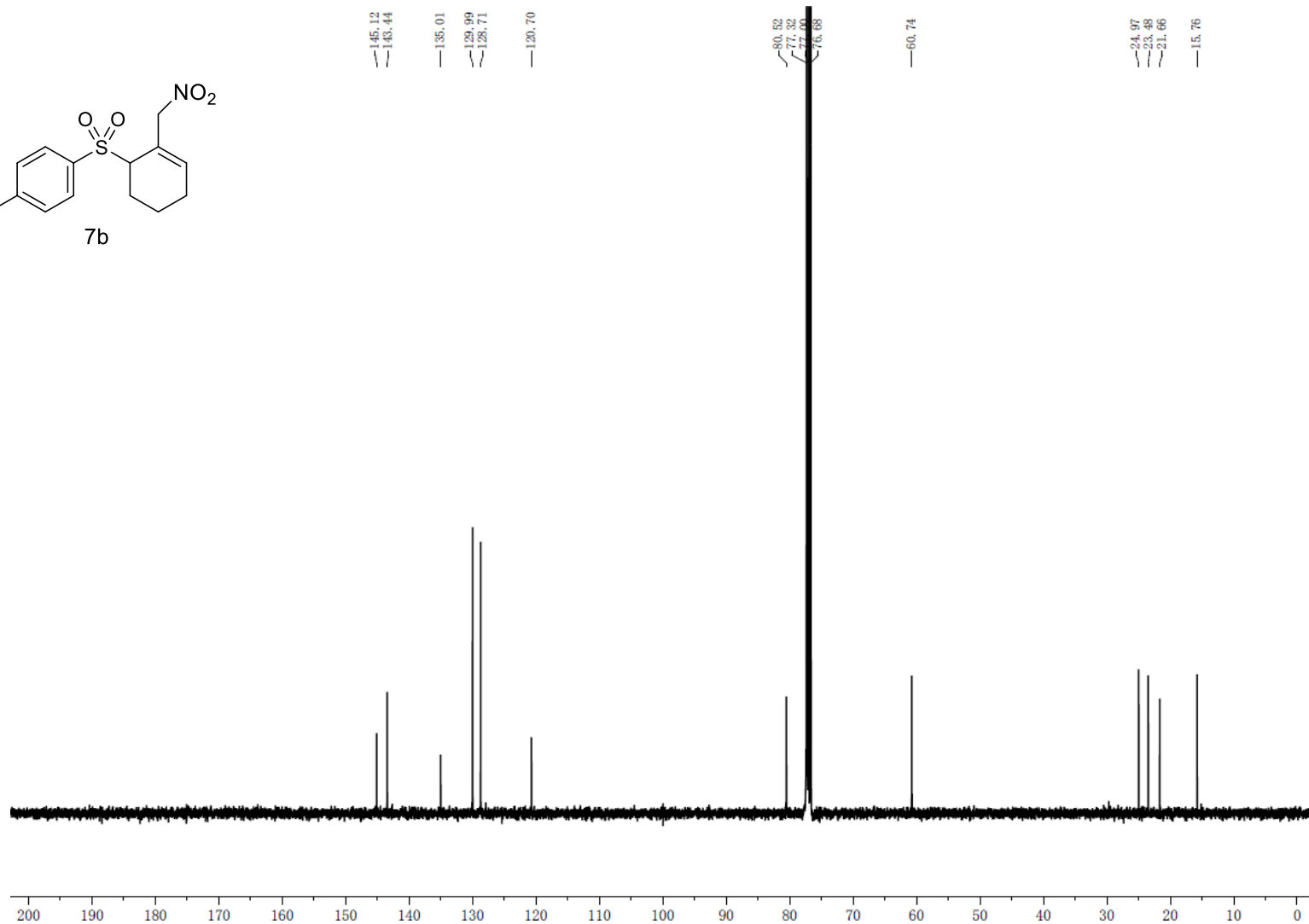
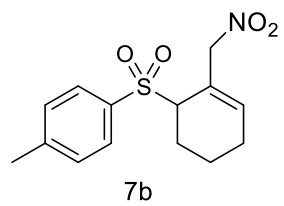


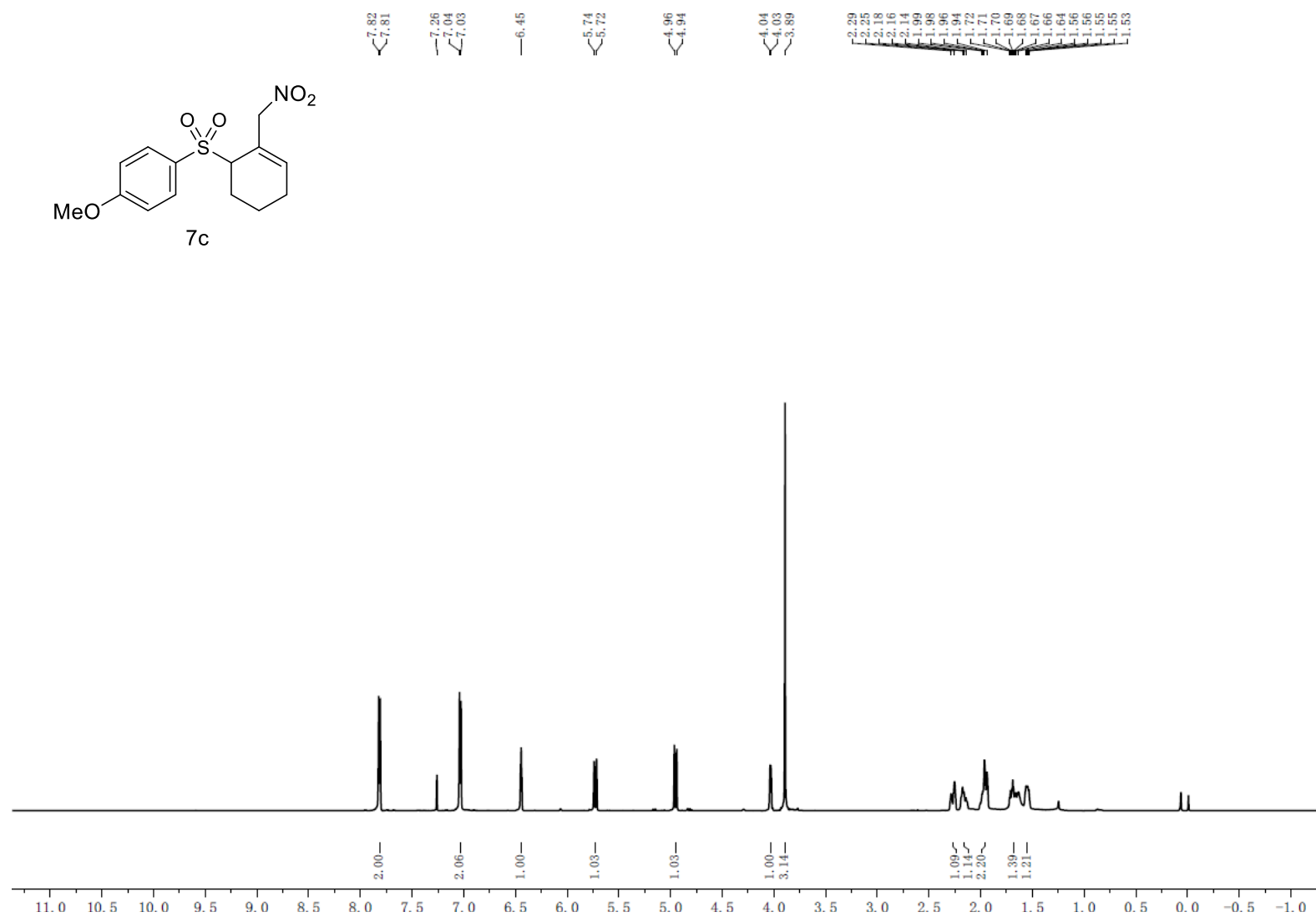


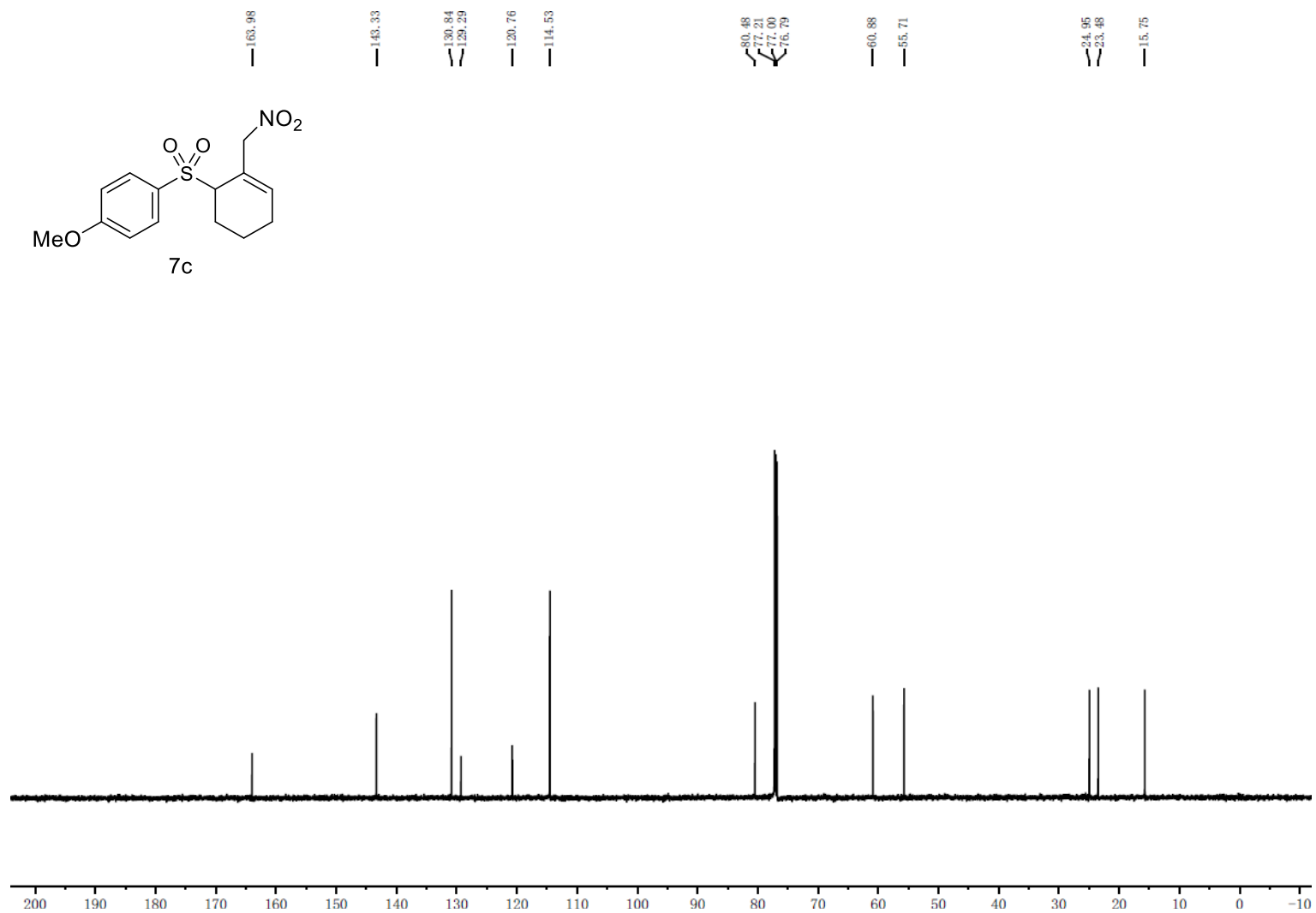


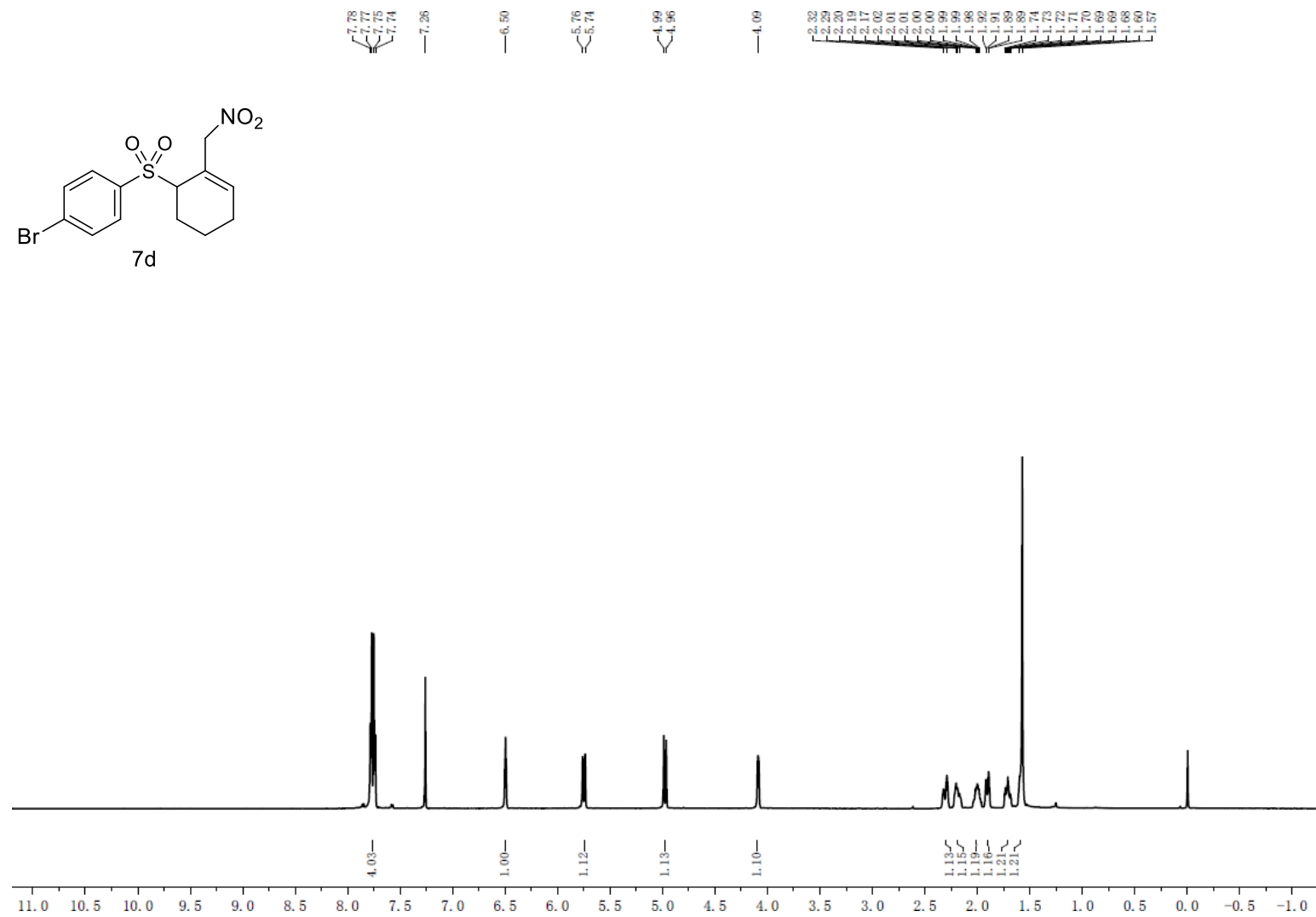
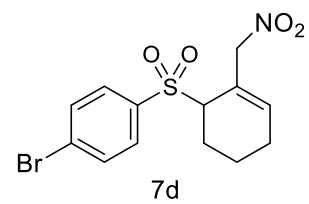


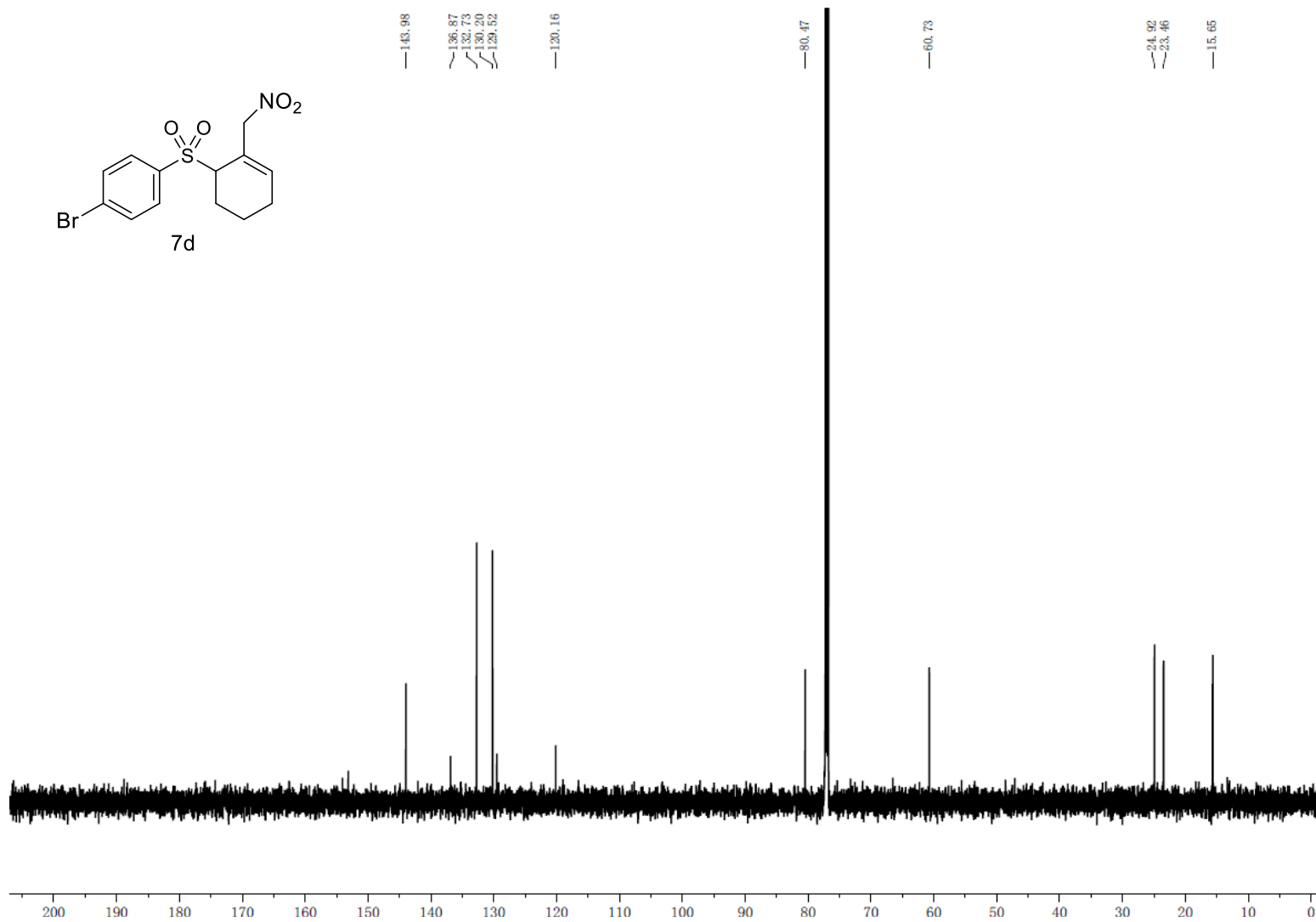


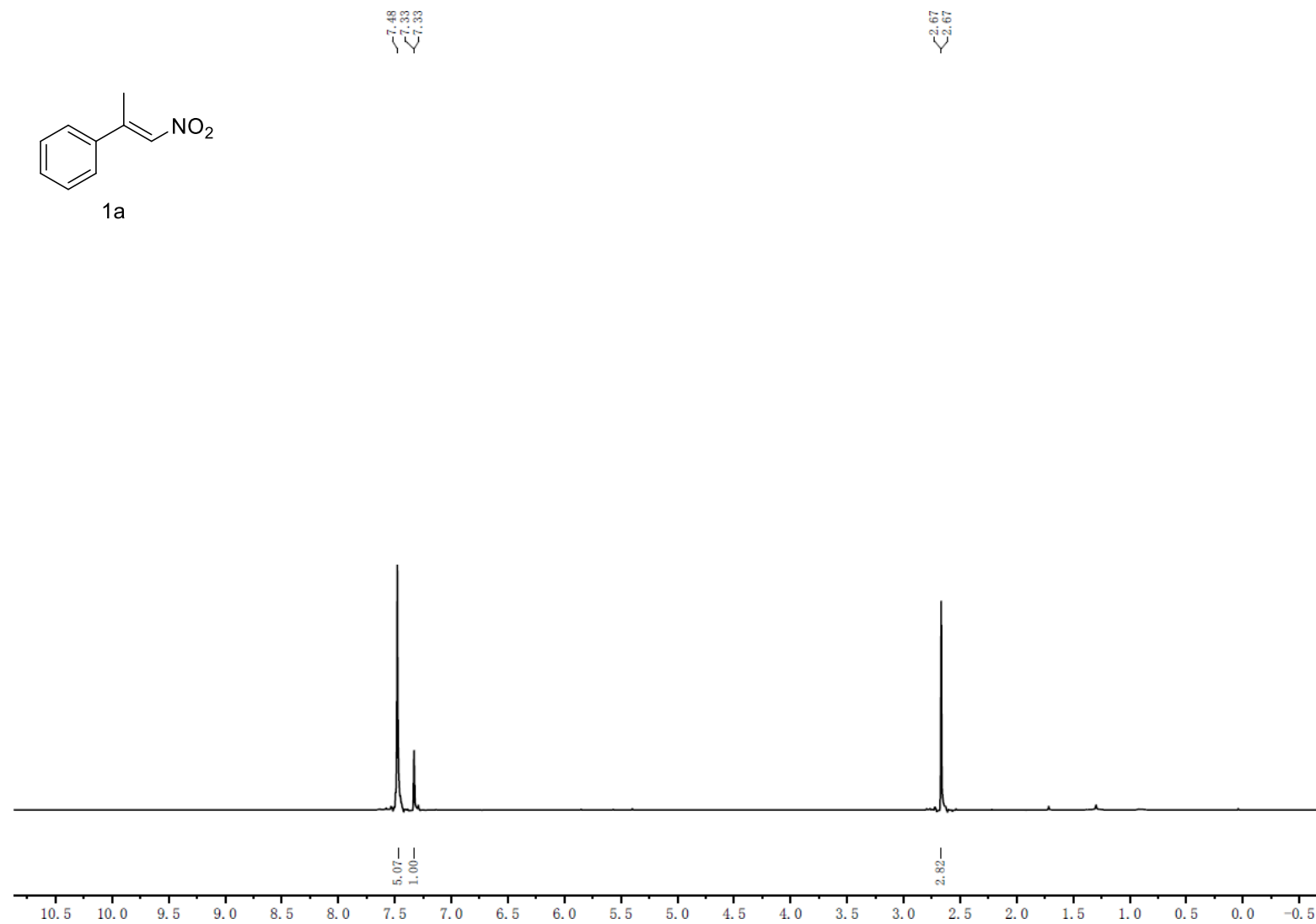
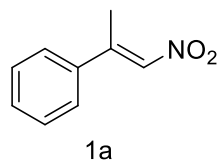


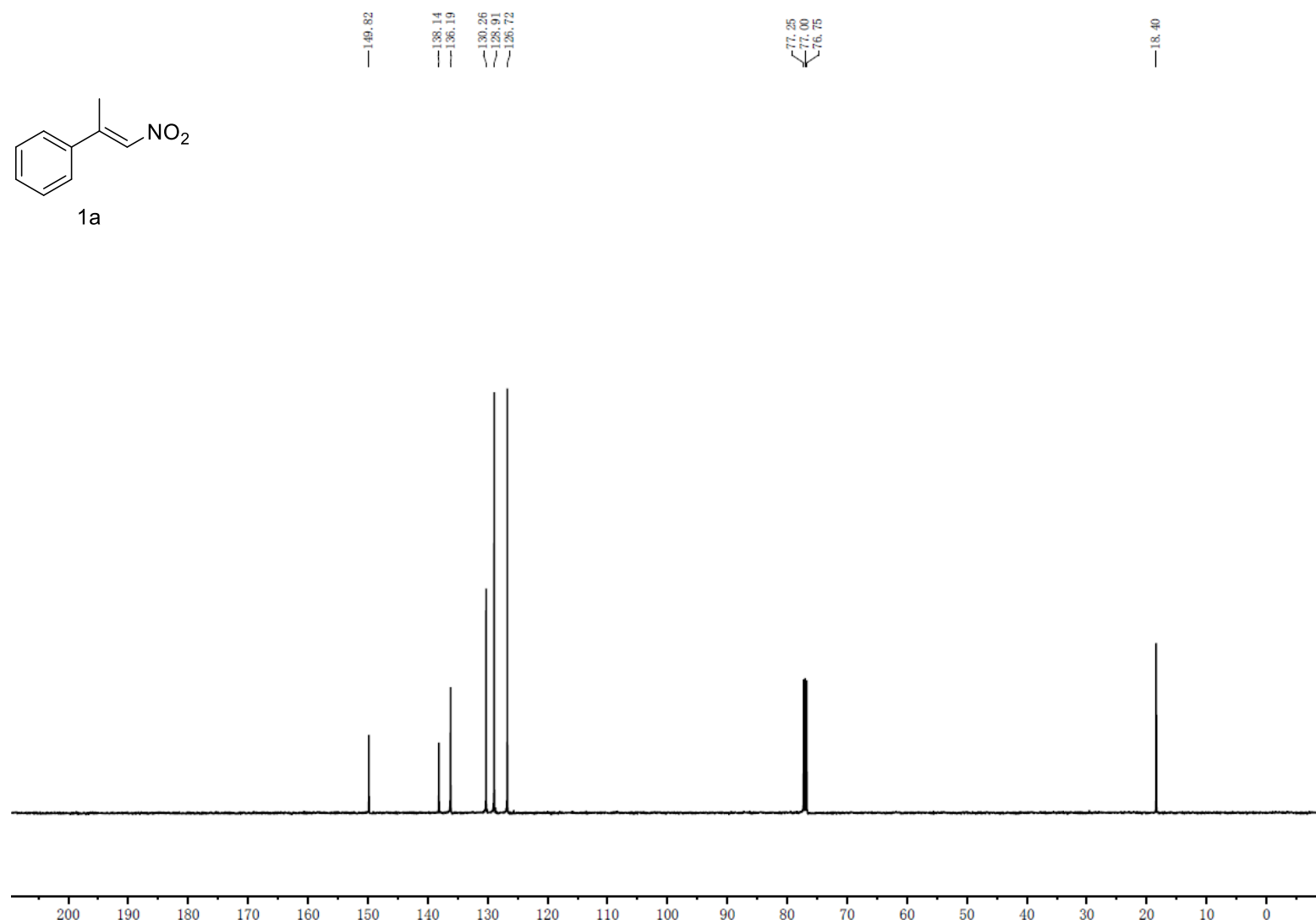
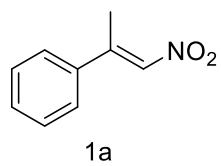


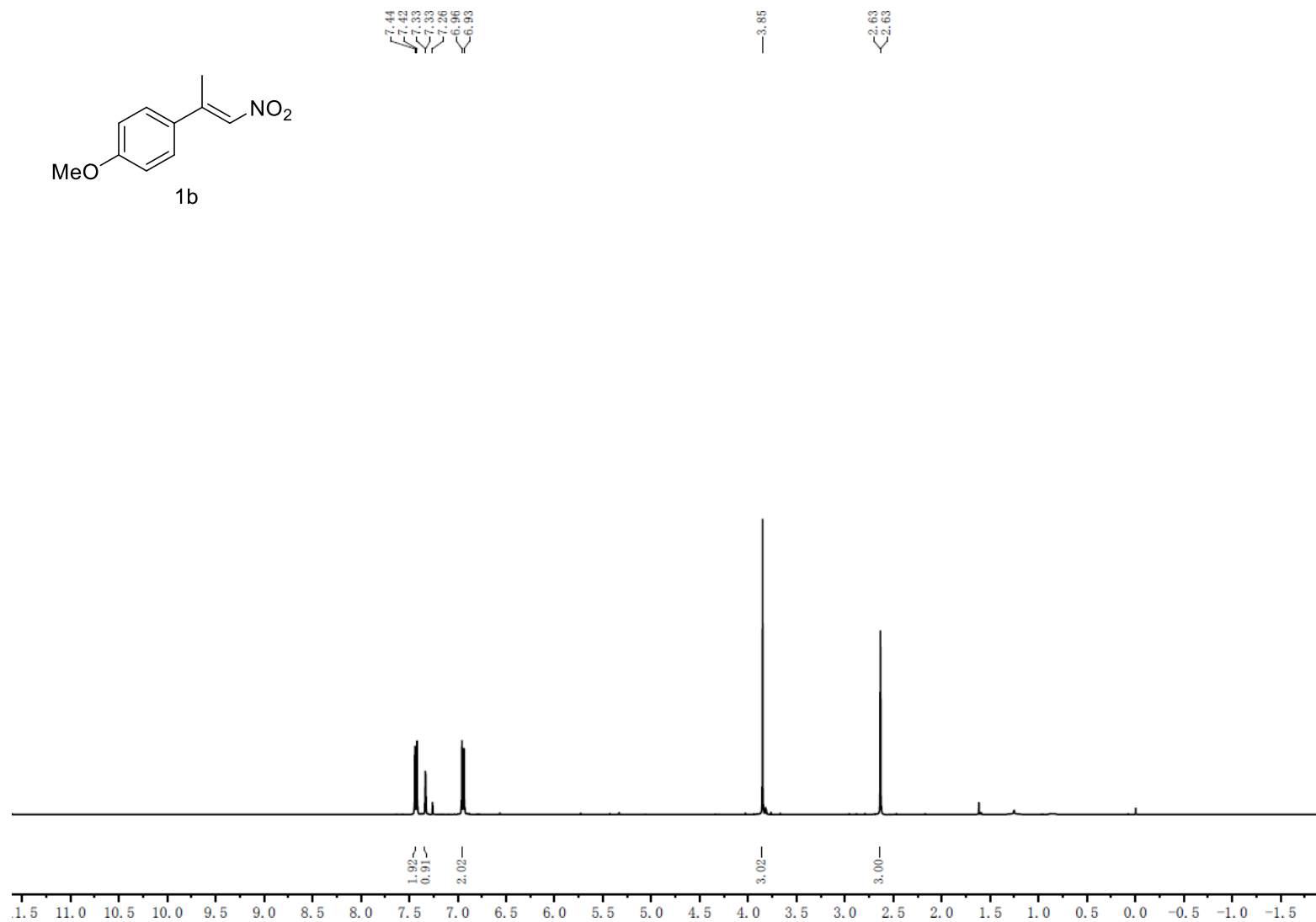
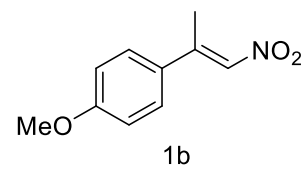


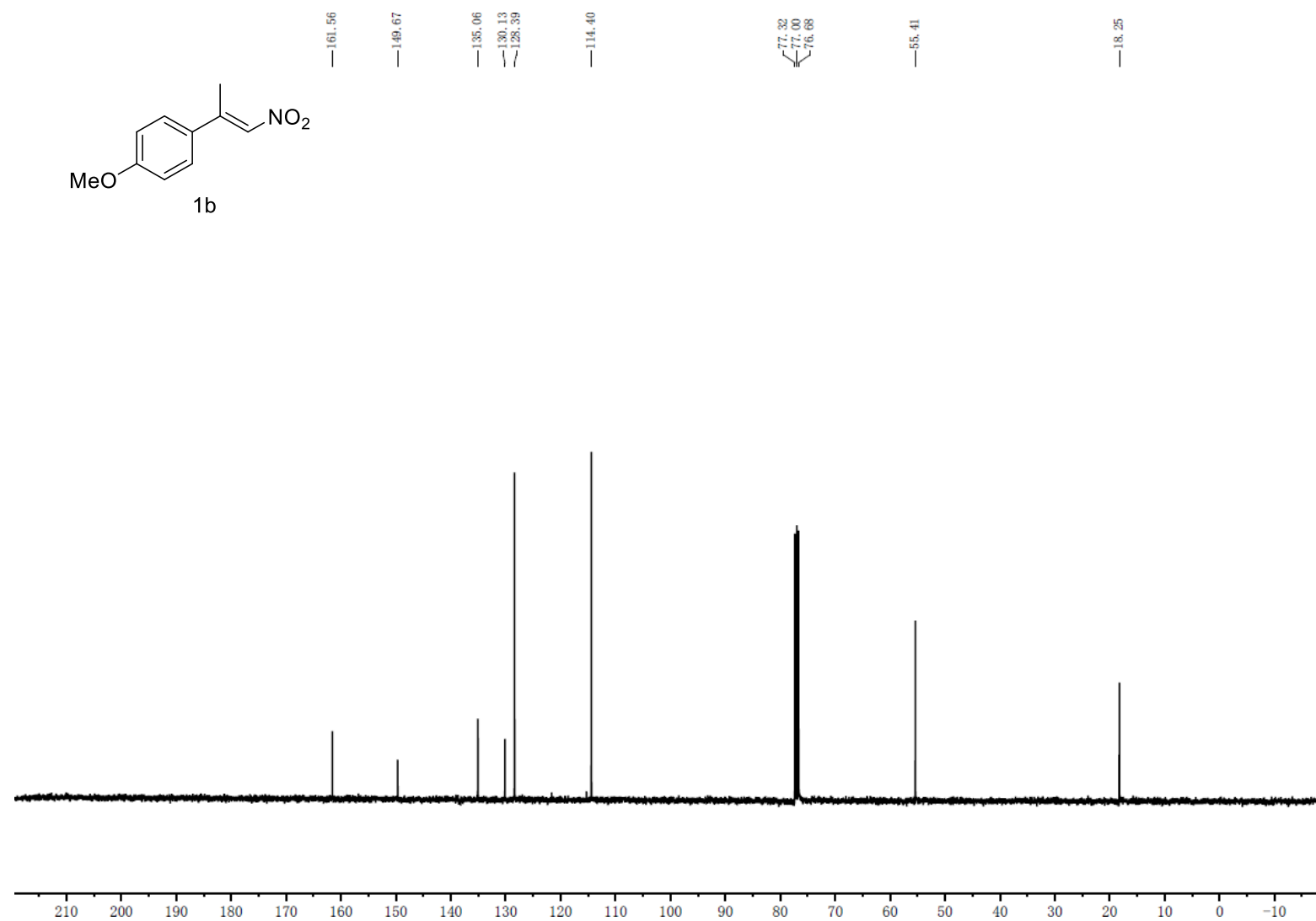
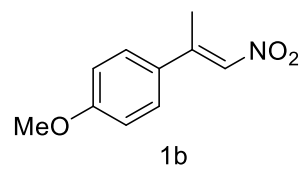


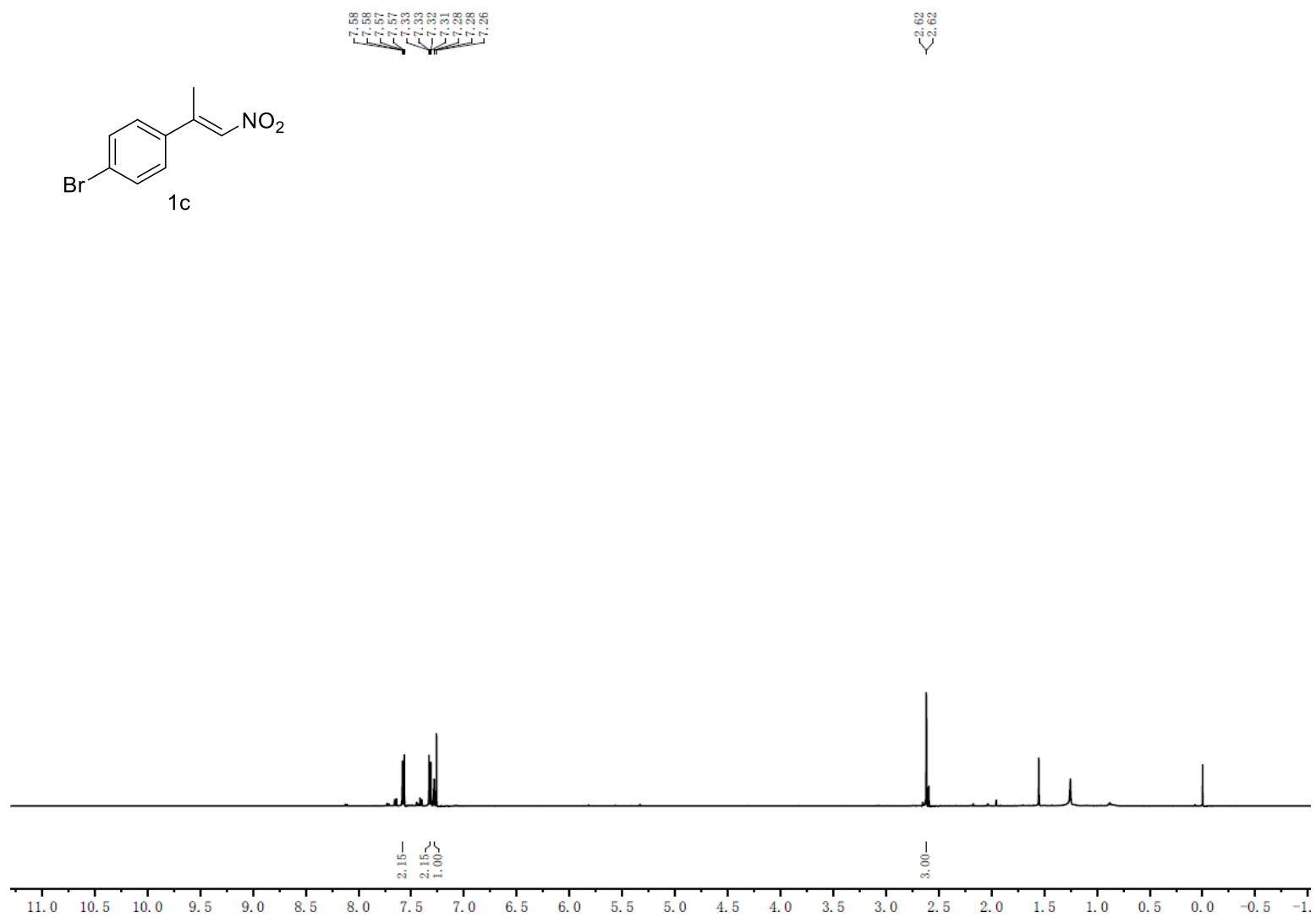


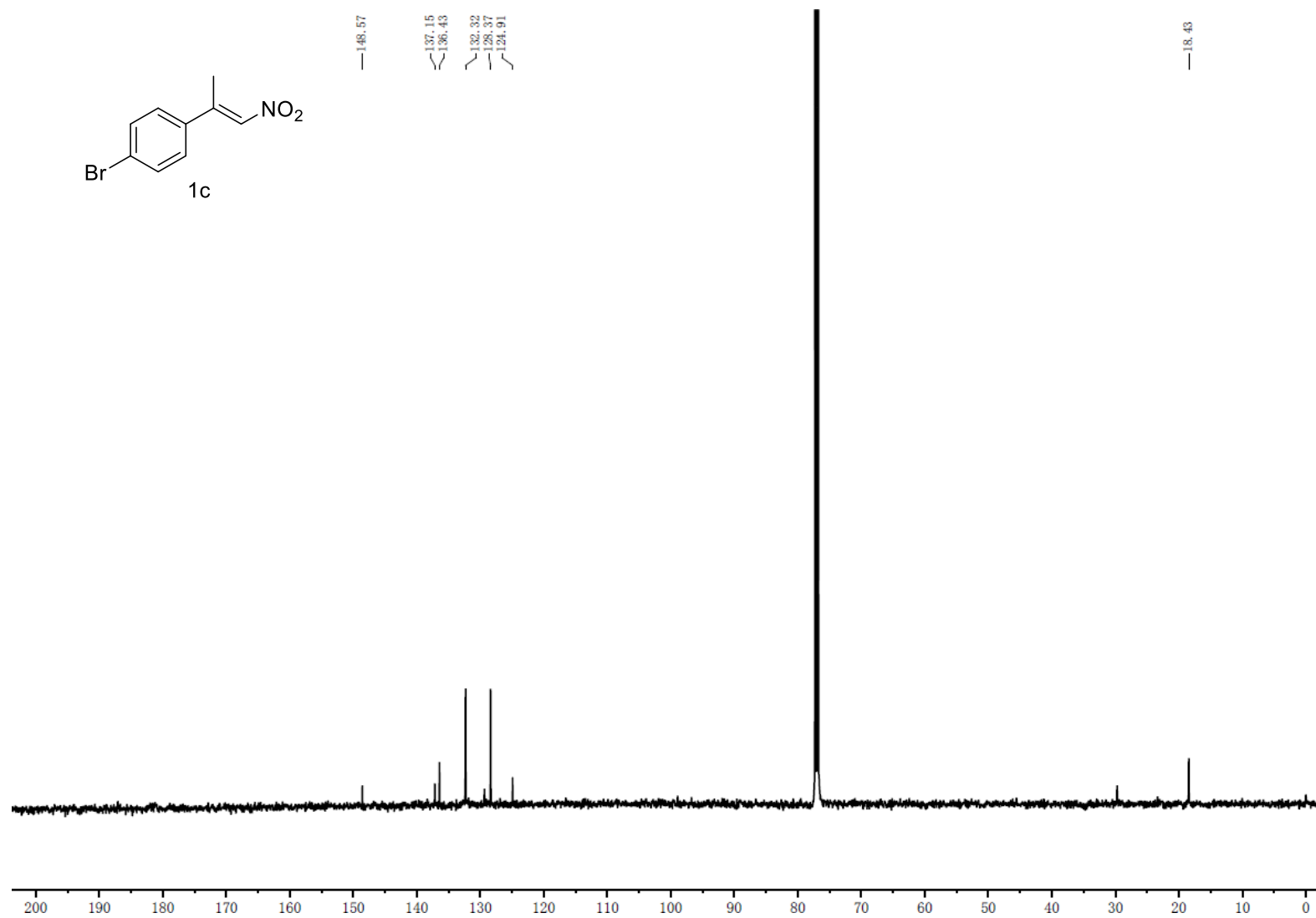


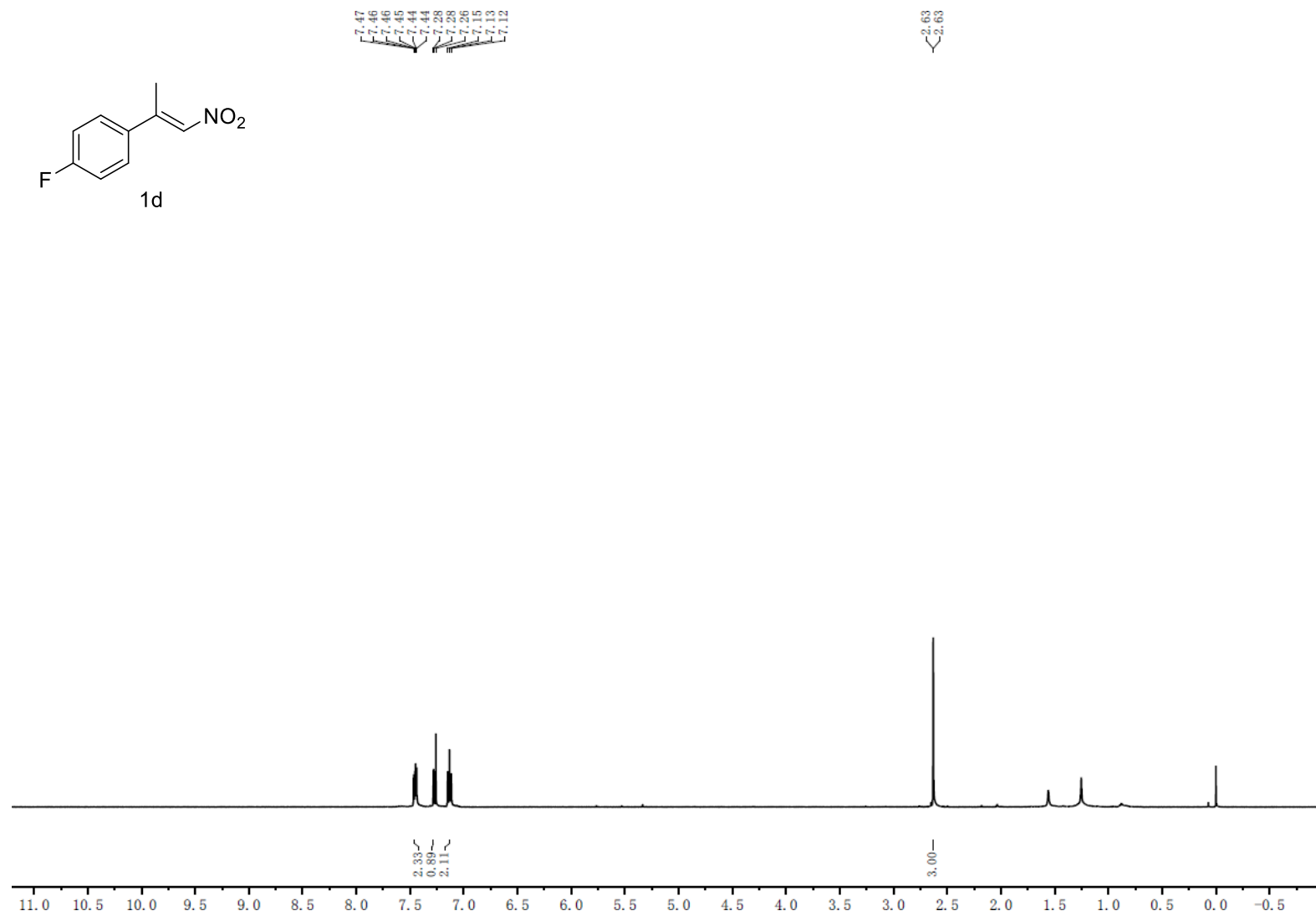


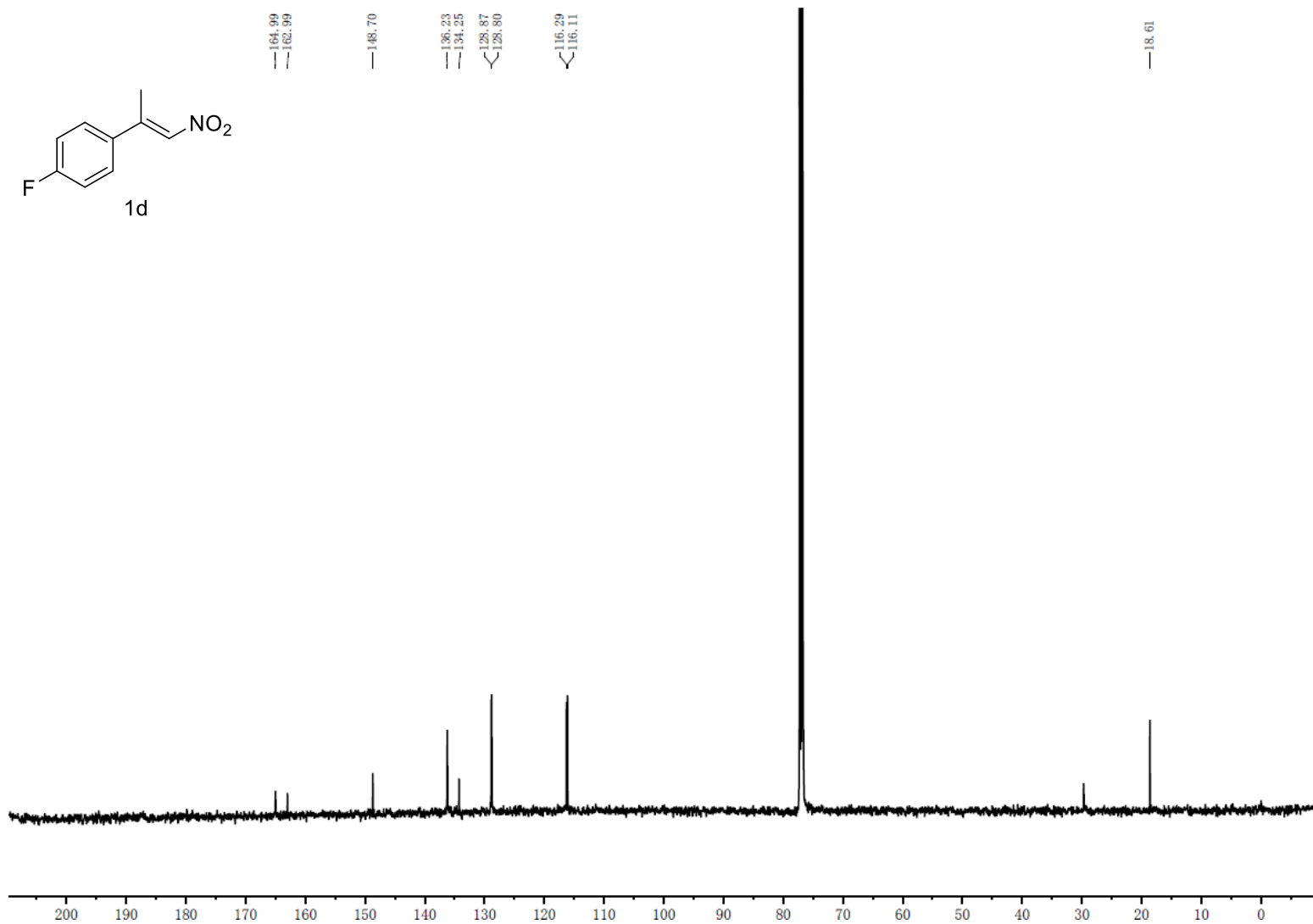


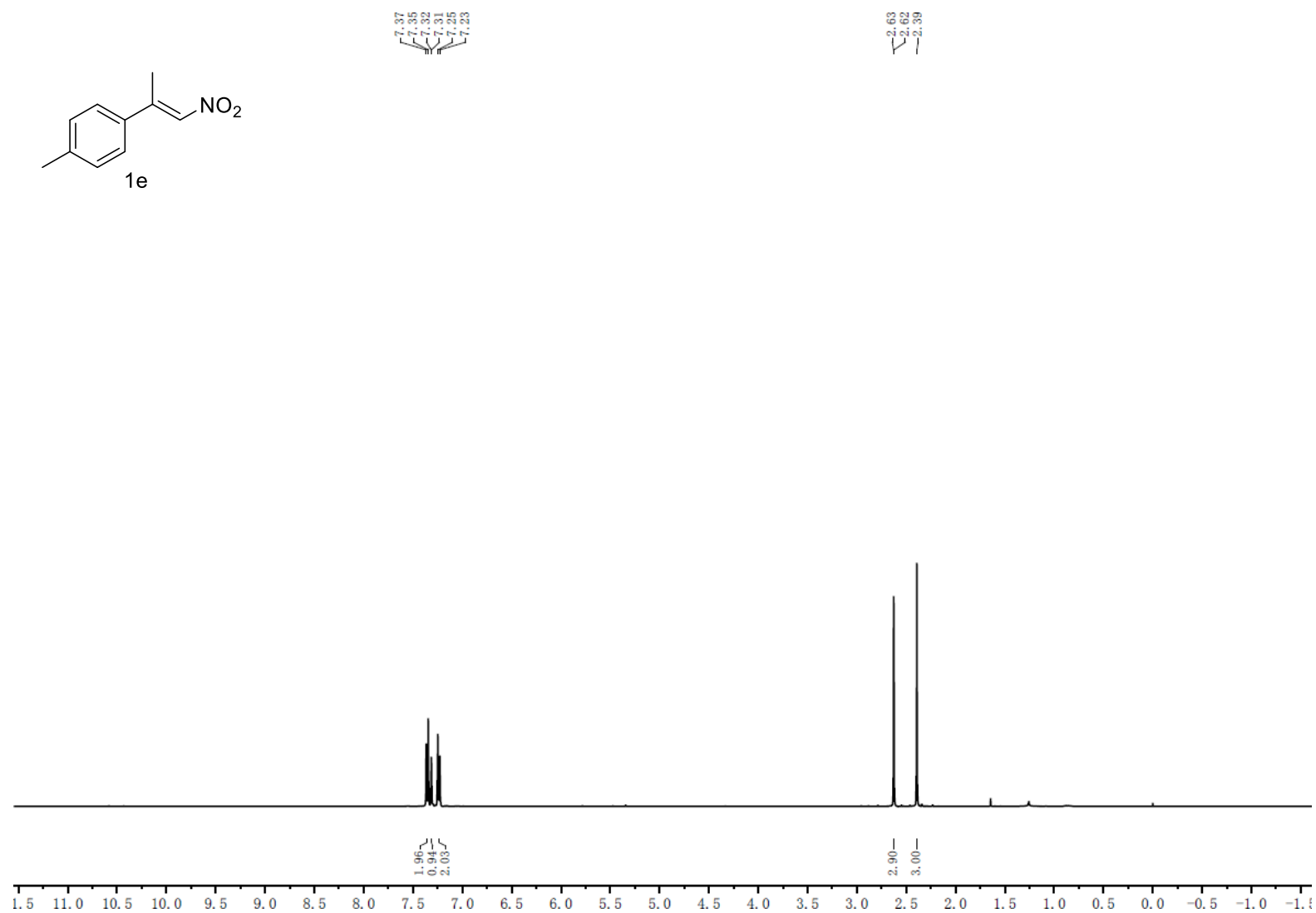


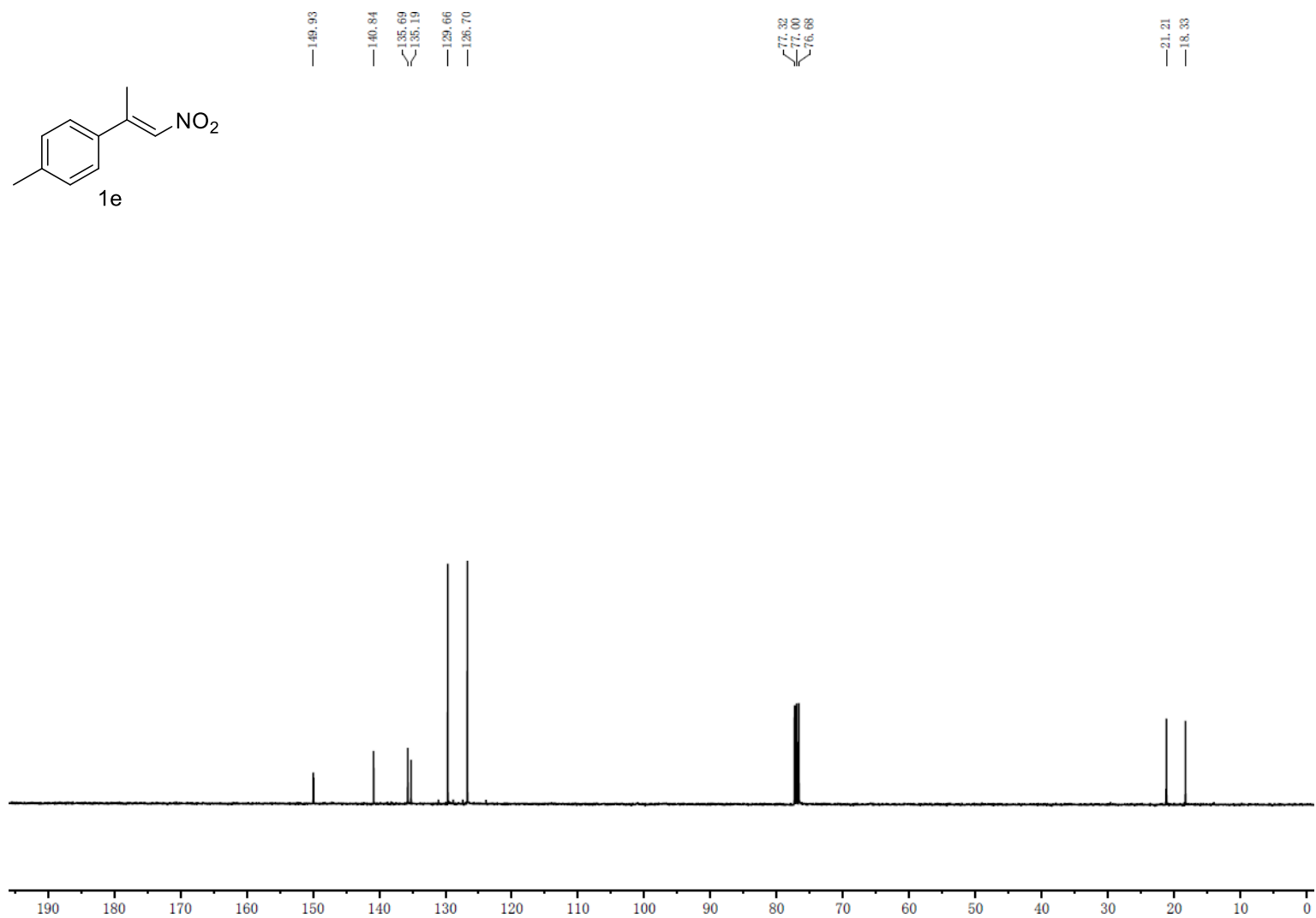


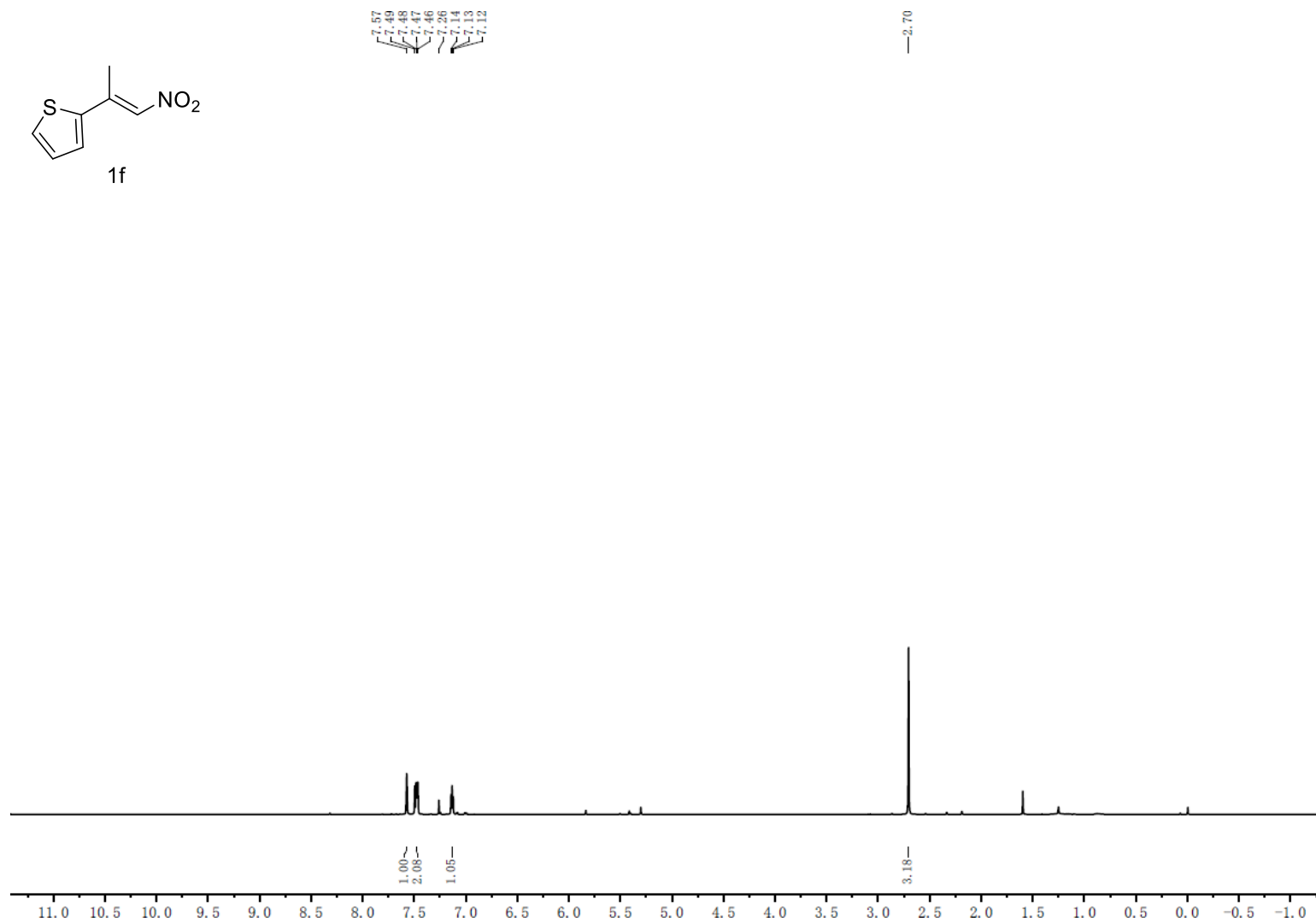
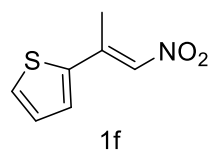


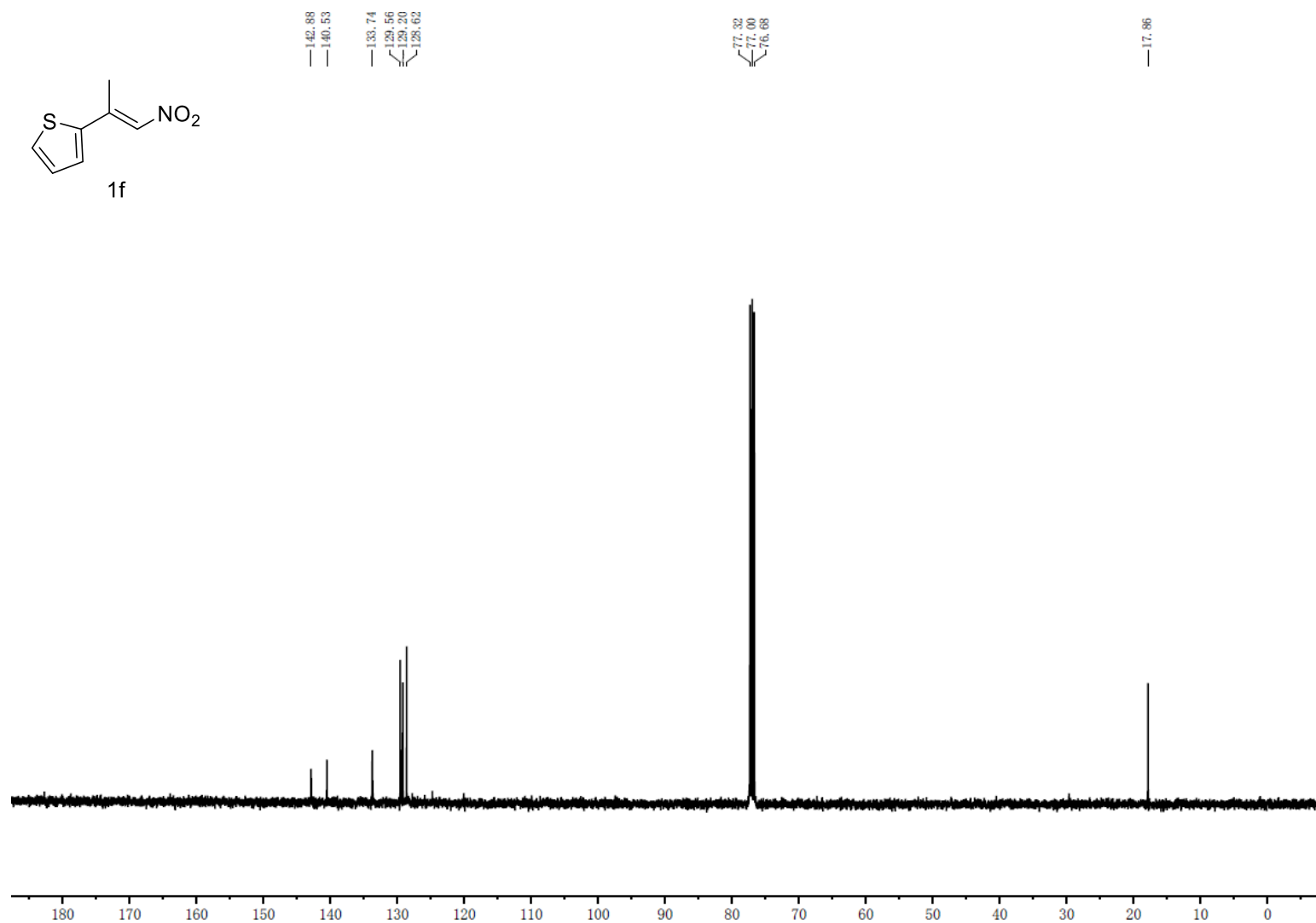
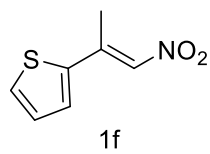


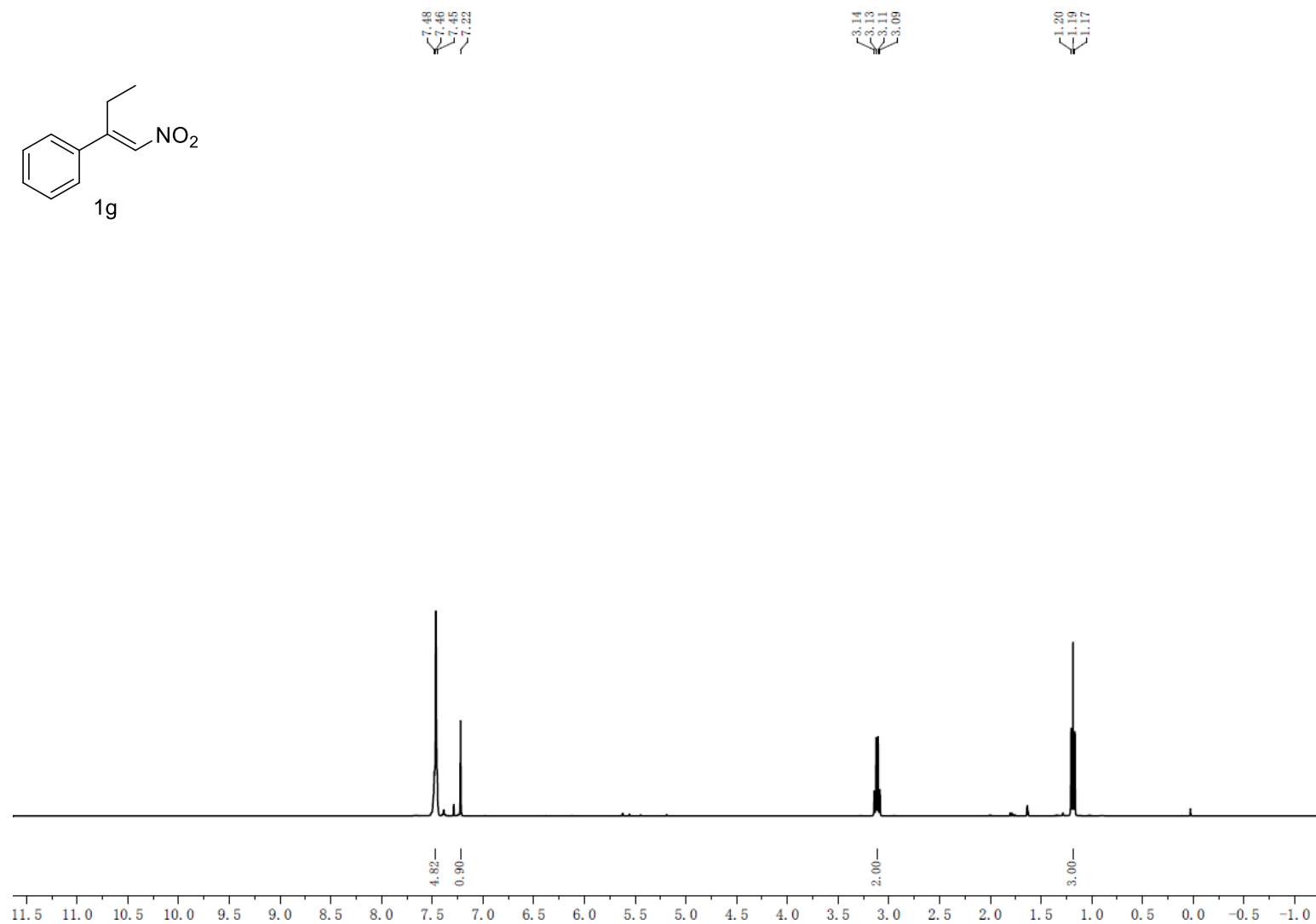
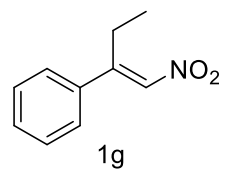


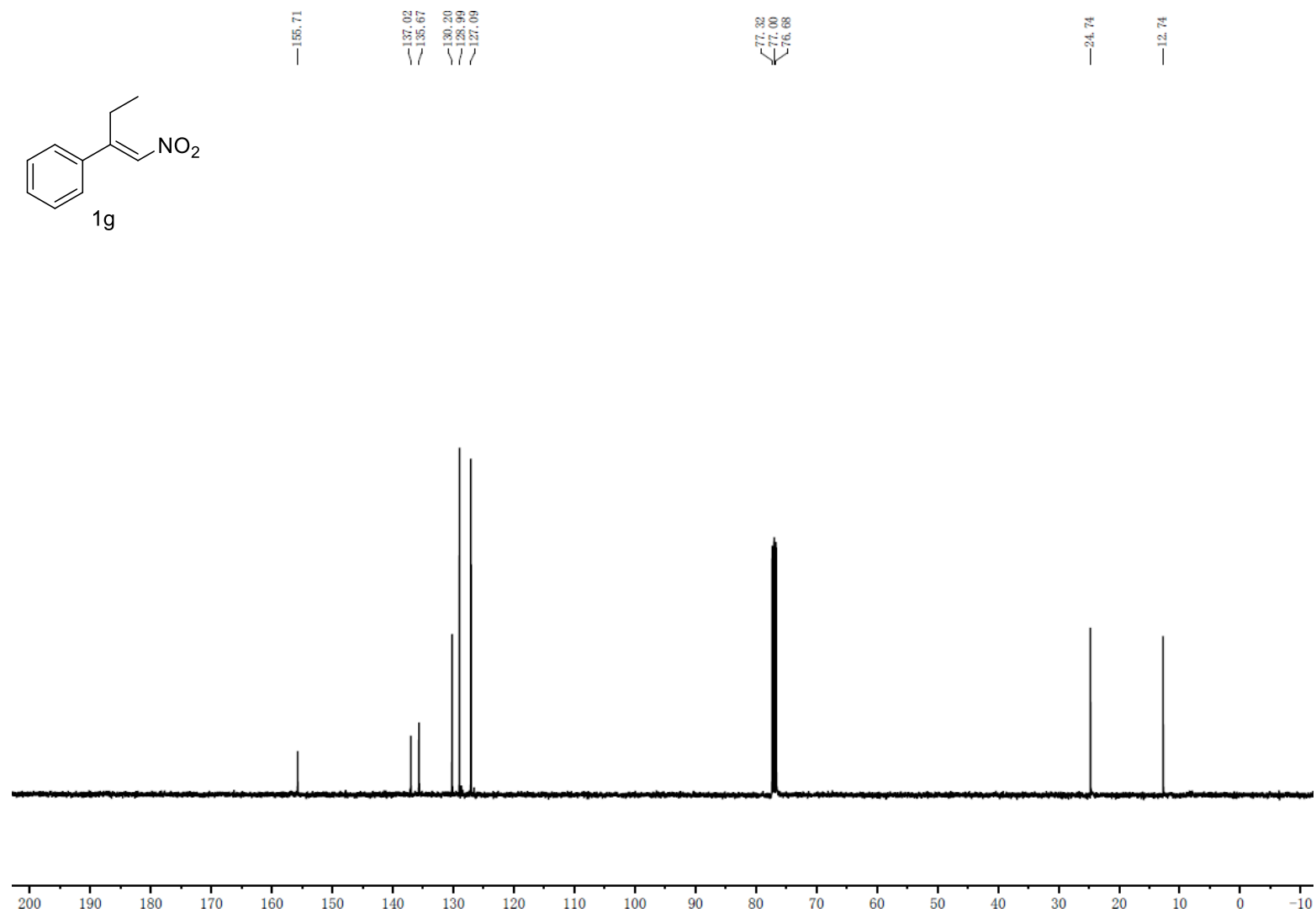
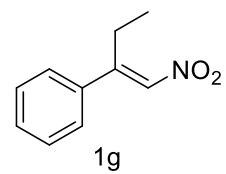


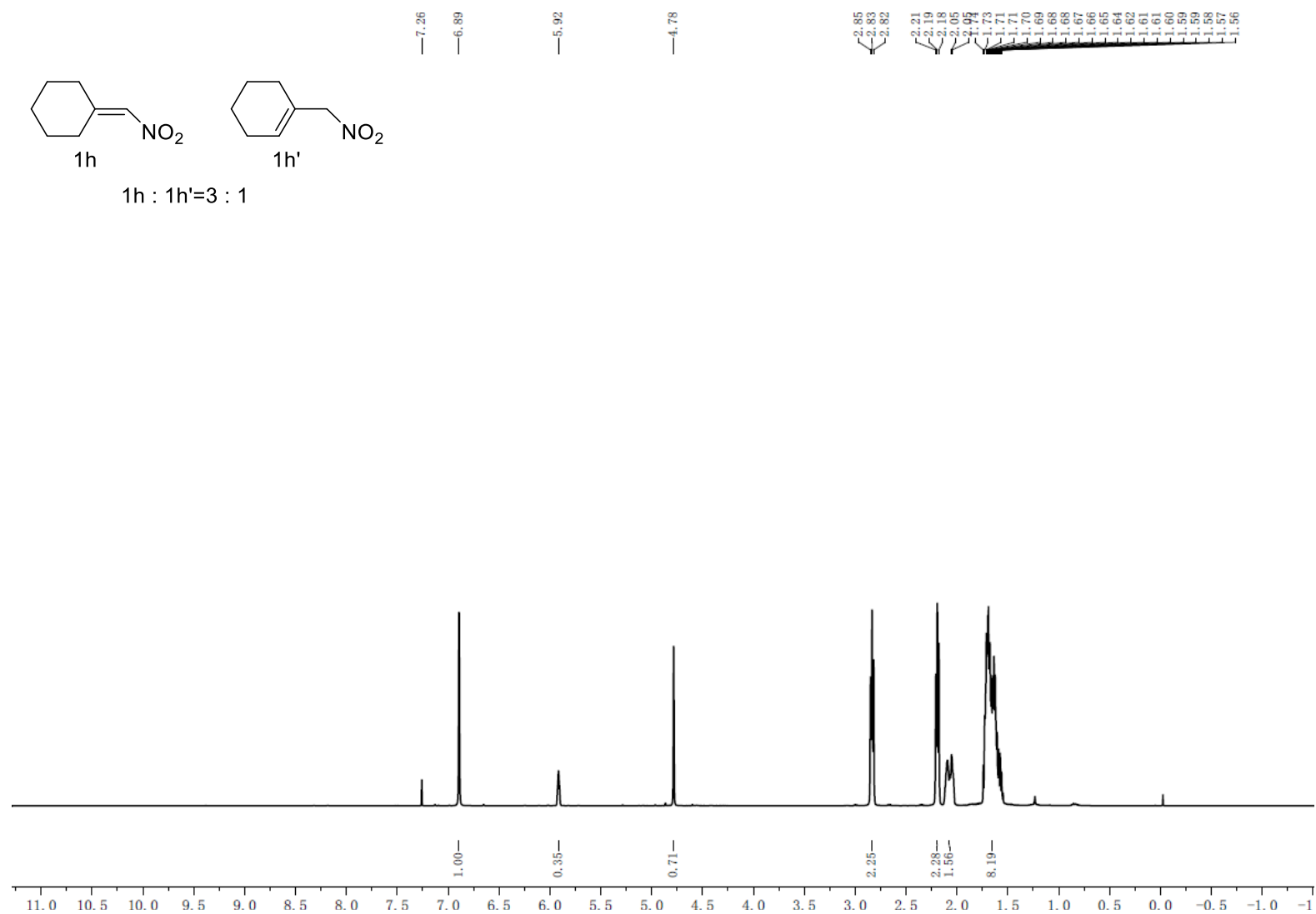


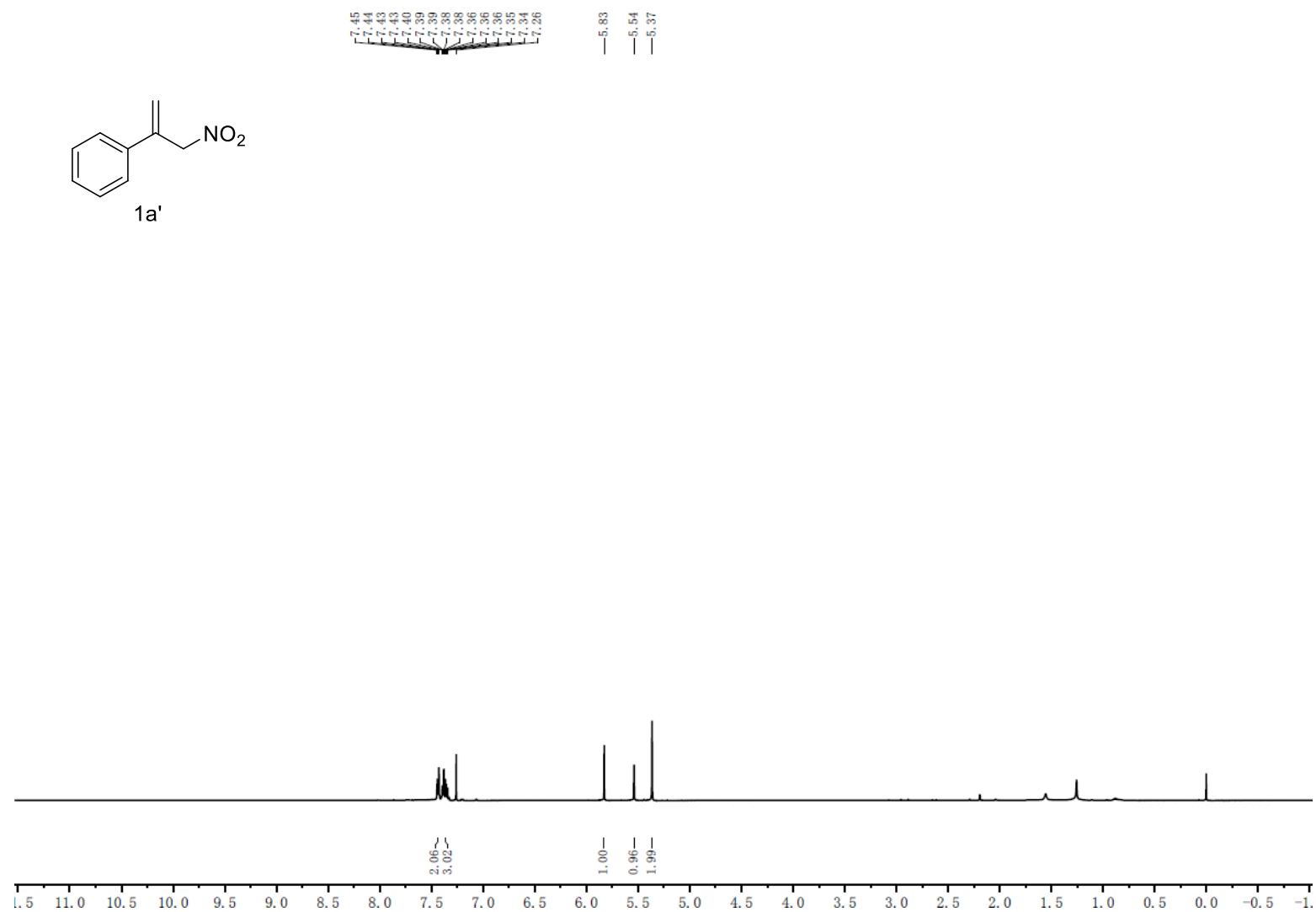


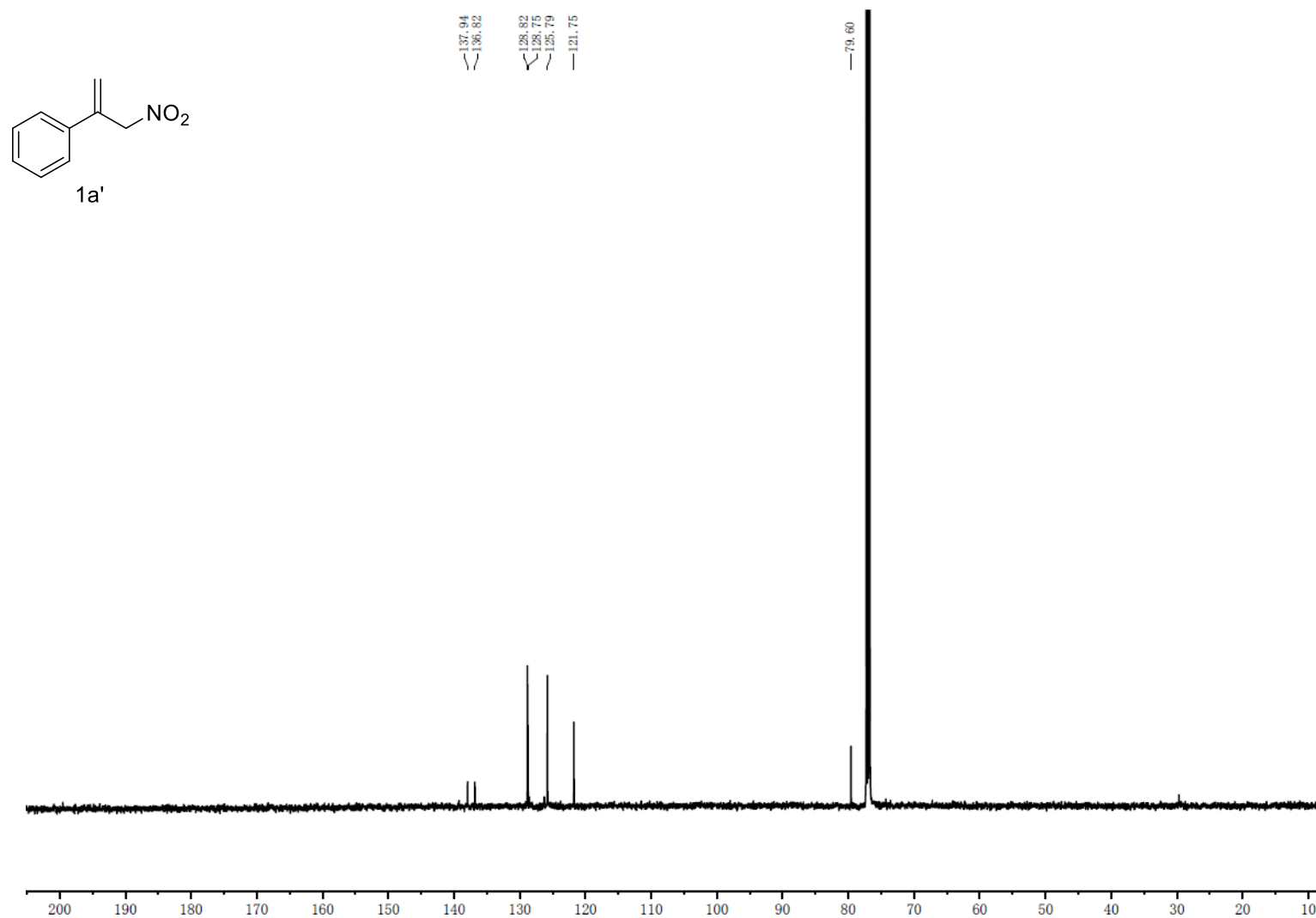








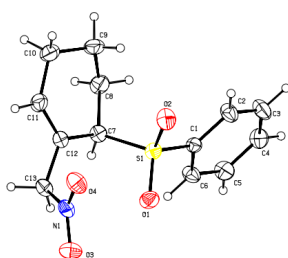




V. X-ray diffraction analysis of compounds **7a** and **8**

The crystallographic data for the single crystal of the complexes **7a** and **8** were collected on a CrysAlis PRO 1.171.39.7a (Rigaku OD, 2015) employing graphite-monochromated MoK α radiation ($\lambda = 0.71072$ Å). Empirical absorption corrections were applied using the CrysAlisPro program at 300 K and 293 K respectively. The structure was solved by direct methods and refined by least-squares. Program used to solve structure is SHELXS (Sheldrick, 2008); program used to refine structure is SHELXH (Sheldrick, 2015). Hydrogen atoms of –CH, –CH₂, –CH₃, groups were placed in geometrically calculated positions and were included in the refinement process using riding model with isotropic thermal parameters.

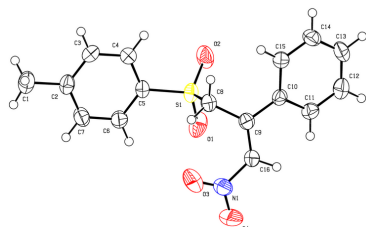
Crystallography data and structure refinement for **7a**. (CCDC 1565191) Thermal ellipsoids are shown at 30% probability level.



Empirical formula	C ₁₃ H ₁₅ NO ₄ S
Formula weight	281.32
Temperature/K	296(2)
Wavelength/Å	0.71073
Crystal system	monoclinic
a/Å	8.2267(15)
b/Å	12.296(2)
c/Å	13.654(3)
α /°	90
β /°	105.690(3)
γ /°	90
Volume/Å ³	1329.7(4)
Z	4
$\rho_{\text{calc}}/\text{cm}^{-3}$	1.405
μ/mm^{-1}	0.253
F(000)	592
Crystal size/mm ³	0.23 × 0.1 × 0.1
2 θ range for data collection/°	3.06 to 25.92
Reflections collected	11535
Independent reflections	2140 [$R_{\text{int}} = 0.0497$, $R_{\text{sigma}} = 0.0514$]
Data/restraints/parameters	2140/0/172

Goodness-of-fit on F^2	1.165
Final R indexes [$I \geq 2\sigma(I)$]	$R_1 = 0.0456$, $wR_2 = 0.1060$
Final R indexes [all data]	$R_1 = 0.0768$, $wR_2 = 0.1195$

Crystallography data and structure refinement for **8**. (CCDC 1565190) Thermal ellipsoids are shown at 30% probability level.



Empirical formula	$C_{16}H_{15}NO_4S$
Formula weight	317.35
Temperature/K	296(2)
Wavelength/Å	0.71073
Crystal system	orthorhombic
$a/\text{Å}$	5.6914(9)
$b/\text{Å}$	7.7326(12)
$c/\text{Å}$	33.904(5)
$\alpha/^\circ$	90
$\beta/^\circ$	90
$\gamma/^\circ$	90
Volume/Å ³	1492.1(4)
Z	4
$\rho_{\text{calc}}/\text{cm}^3$	1.413
μ/mm^{-1}	0.235
$F(000)$	664
Crystal size/mm ³	$0.15 \times 0.12 \times 0.1$
2θ range for data collection/ $^\circ$	2.70 to 30.41
Reflections collected	10718
Independent reflections	2708 [$R_{\text{int}} = 0.0237$, $R_{\text{sigma}} = 0.0216$]
Data/restraints/parameters	2708/0/200
Goodness-of-fit on F^2	1.135
Final R indexes [$I \geq 2\sigma(I)$]	$R_1 = 0.0373$, $wR_2 = 0.0986$
Final R indexes [all data]	$R_1 = 0.0381$, $wR_2 = 0.0993$