

Supporting Information

An Epidermis-Like Hierarchical Smart Coating with a Hardness of Tooth Enamel

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Figure S17. A side view of the self-healing process on an e-LBL film with a thickness of h and a width of w (not shown in this schematic drawing). The white rectangular region denotes the crack with a width of z . c_1 , c_2 and c_3 are polymer concentrations in the film, crack area and solution, respectively. k_{in} and k_{out} are the mass transfer coefficients into and out of the crack area.

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The dotted surface fluorescence may be due to the high roughness of l-LBL film growing on glass substrates. The insets in a and b indicate the thickness of the film.

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Table S1. A summary of modulus and hardness for different multilayers consisting of $\text{PVA}_{145\text{k}}$ obtained from nanoindentation.

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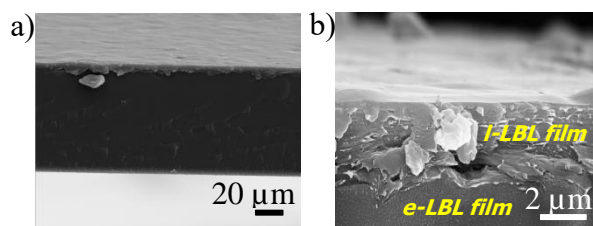


Figure S1. Cross-sectional SEM images of $(\text{PVA}_{145\text{k}}/\text{TA})_{50} @ (\text{PVA}_{145\text{k}}\text{-GO}/\text{TA})_{10}$.

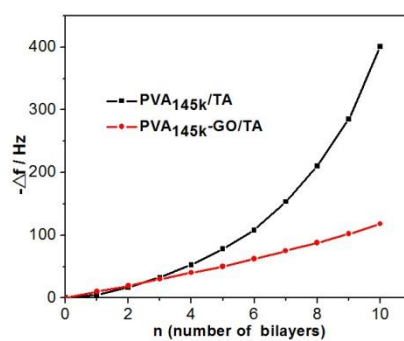


Figure S2. QCM results for the growth of first ten bilayers of $\text{PVA}_{145\text{k}}/\text{TA}$ and $\text{PVA}_{145\text{k}}\text{-GO}/\text{TA}$.

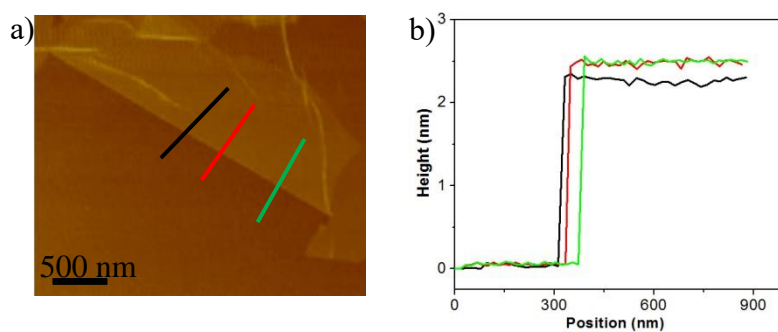


Figure S3. a) The AFM image and b) thickness profiles of GO.

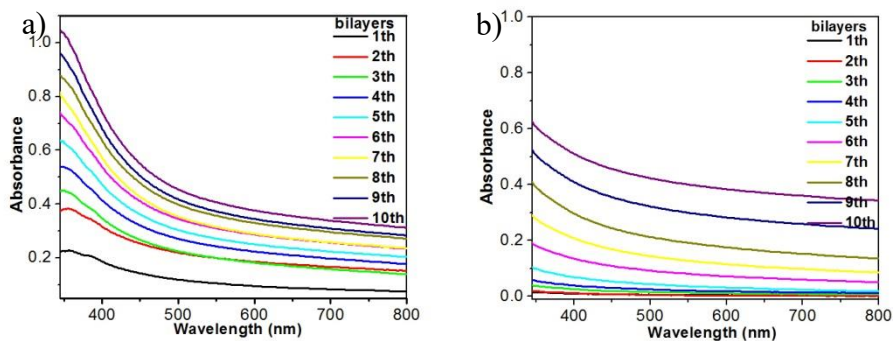


Figure S4. Compilation of UV-vis spectra of PVA_{47k}-GO/TA during the growth of first 10 bilayers on a) (PVA_{47k}/TA)₅₀ and b) glass substrates.

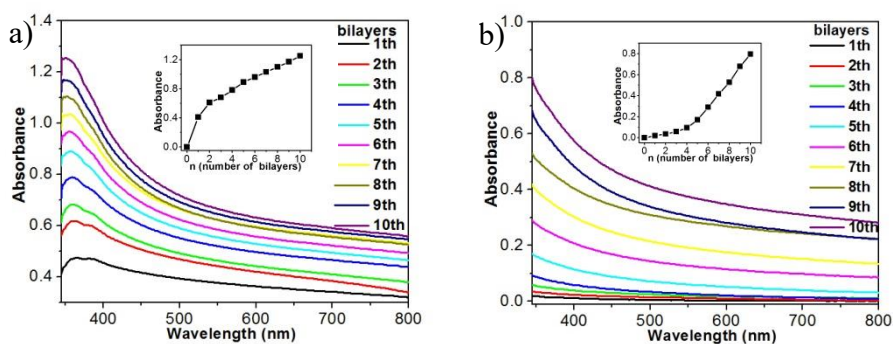


Figure S5. Compilation of UV-vis spectra of PVA_{145k}-GO/TA during the growth of the first 10 bilayers on a) (PVA_{145k}/TA)₅₀ and b) glass substrates. The insets are the plotting of absorbance at 345 nm against the number of bilayers.

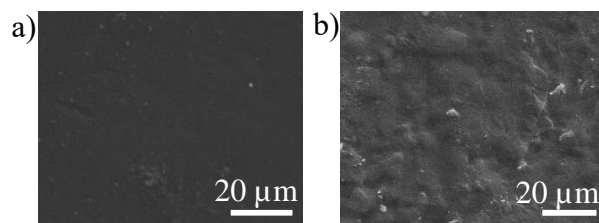


Figure S6. SEM images of (PVA_{145k}-GO/TA)₁₀ growing on a) (PVA_{145k}/TA)₅₀ and b) glass substrates.

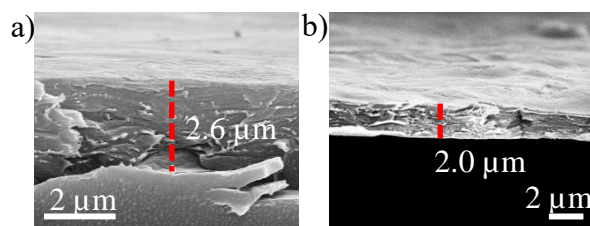


Figure S7. Cross-sectional SEM images of a) $(\text{PVA}_{145\text{k}}/\text{TA})_{50} @ (\text{PVA}_{145\text{k}}\text{-GO}/\text{TA})_{10}$ and b) $(\text{PVA}_{145\text{k}}\text{-GO}/\text{TA})_{10}$ on glass substrates.

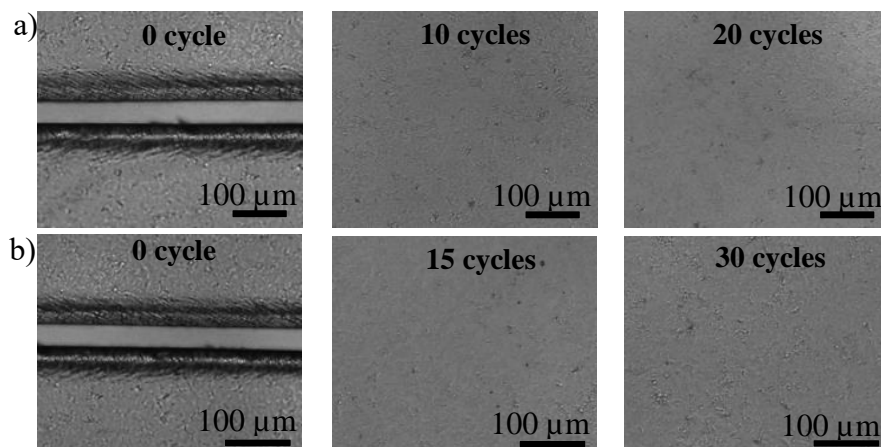


Figure S8. Optical images of a) $(\text{PVA}_{47\text{k}}/\text{TA})_{50} @ (\text{PVA}_{47\text{k}}\text{-GO}/\text{TA})_{10}$ and b) $(\text{PVA}_{47\text{k}}/\text{TA})_{80} @ (\text{PVA}_{47\text{k}}\text{-GO}/\text{TA})_{10}$ with different scratch/healing cycles.

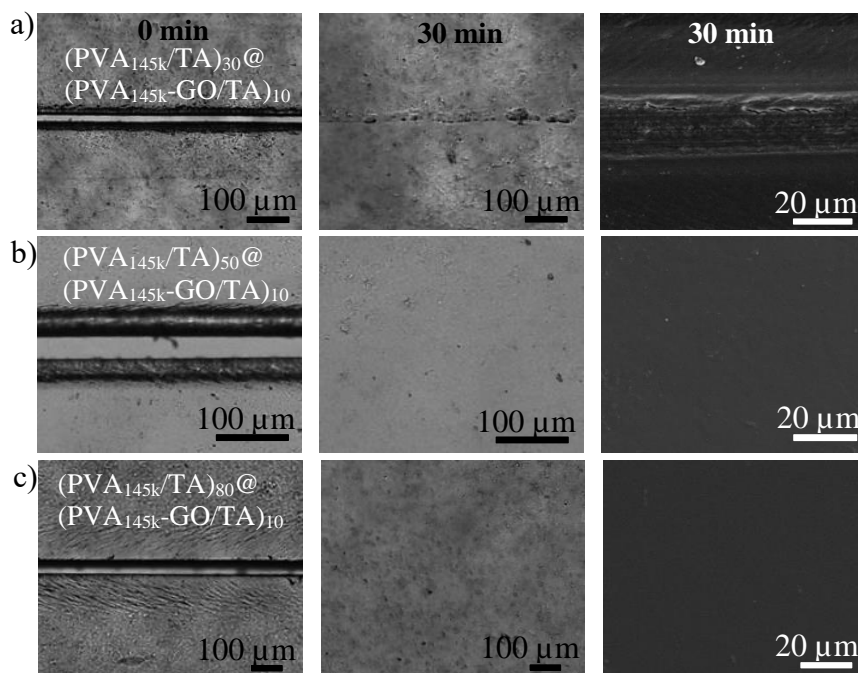


Figure S9. Optical and SEM images of a) $(\text{PVA}_{145\text{k}}/\text{TA})_{30} @ (\text{PVA}_{145\text{k}}\text{-GO}/\text{TA})_{10}$, b) $(\text{PVA}_{145\text{k}}/\text{TA})_{50} @ (\text{PVA}_{145\text{k}}\text{-GO}/\text{TA})_{10}$ and c) $(\text{PVA}_{145\text{k}}/\text{TA})_{80} @ (\text{PVA}_{145\text{k}}\text{-GO}/\text{TA})_{10}$ with a 50 μm wide cut throughout the film before and after immersion in water for 30 min.

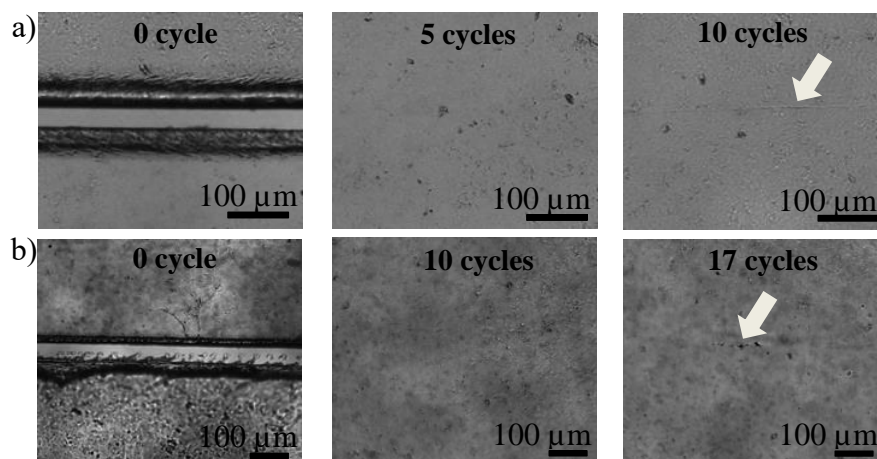


Figure S10. Optical images of a) $(\text{PVA}_{145\text{k}}/\text{TA})_{50} @ (\text{PVA}_{145\text{k}}\text{-GO}/\text{TA})_{10}$ and b) $(\text{PVA}_{145\text{k}}/\text{TA})_{80} @ (\text{PVA}_{145\text{k}}\text{-GO}/\text{TA})_{10}$ with different cutting-healing cycles. The insets indicate the emergence of unrecovered surface scratch.

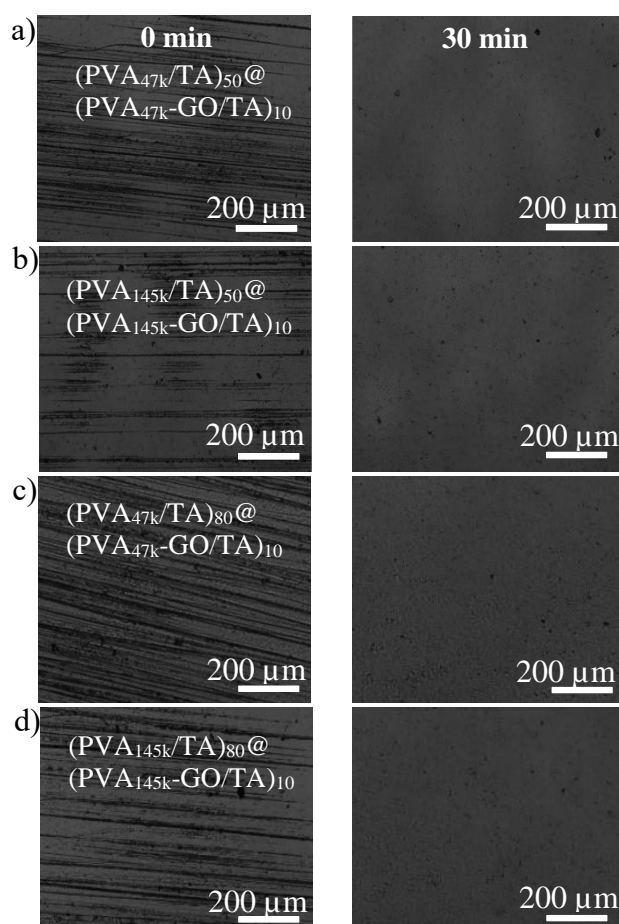


Figure S11. Optical images of a) $(\text{PVA}_{47\text{k}}/\text{TA})_{50} @ (\text{PVA}_{47\text{k}}\text{-GO}/\text{TA})_{10}$, b) $(\text{PVA}_{145\text{k}}/\text{TA})_{50} @ (\text{PVA}_{145\text{k}}\text{-GO}/\text{TA})_{10}$, c) $(\text{PVA}_{47\text{k}}/\text{TA})_{80} @ (\text{PVA}_{47\text{k}}\text{-GO}/\text{TA})_{10}$, and d) $(\text{PVA}_{145\text{k}}/\text{TA})_{80} @ (\text{PVA}_{145\text{k}}\text{-GO}/\text{TA})_{10}$ scratched by a sand paper before and after immersion in water for 30 min.

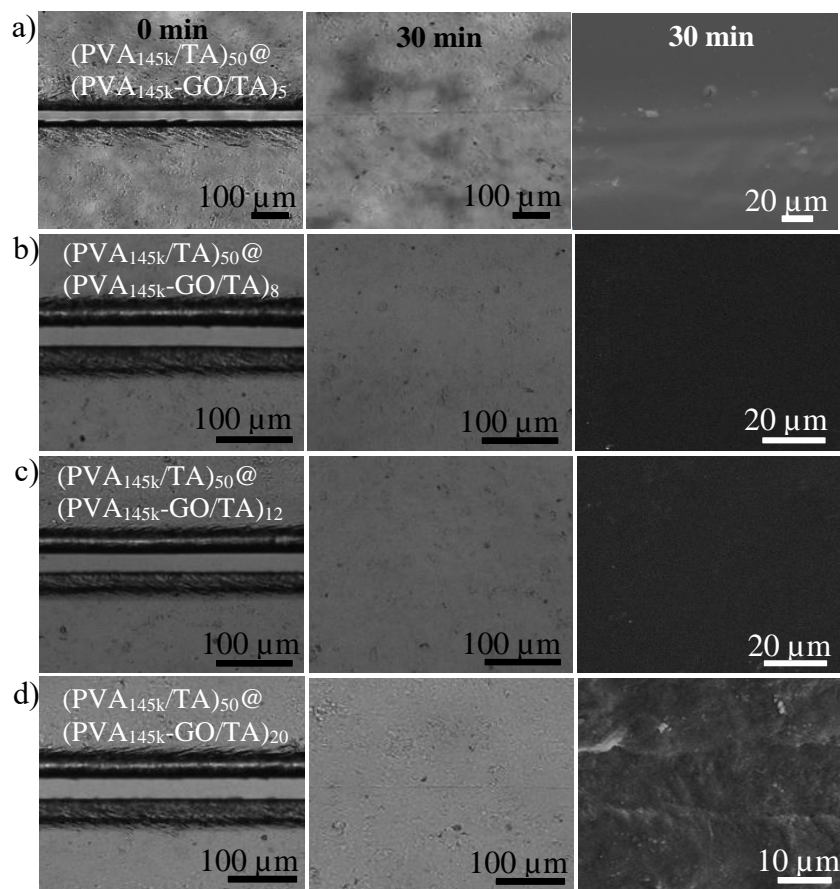


Figure S12. Optical and SEM images of a) $(PVA_{145k}/TA)_{50}@ (PVA_{145k}-GO/TA)_5$, b) $(PVA_{145k}/TA)_{50}@ (PVA_{145k}-GO/TA)_8$ c) $(PVA_{145k}/TA)_{50}@ (PVA_{145k}-GO/TA)_{12}$ and d) $(PVA_{145k}/TA)_{50}@ (PVA_{145k}-GO/TA)_{20}$ with a 50 μm wide cut throughout the film before and after immersion in water for 30 min.

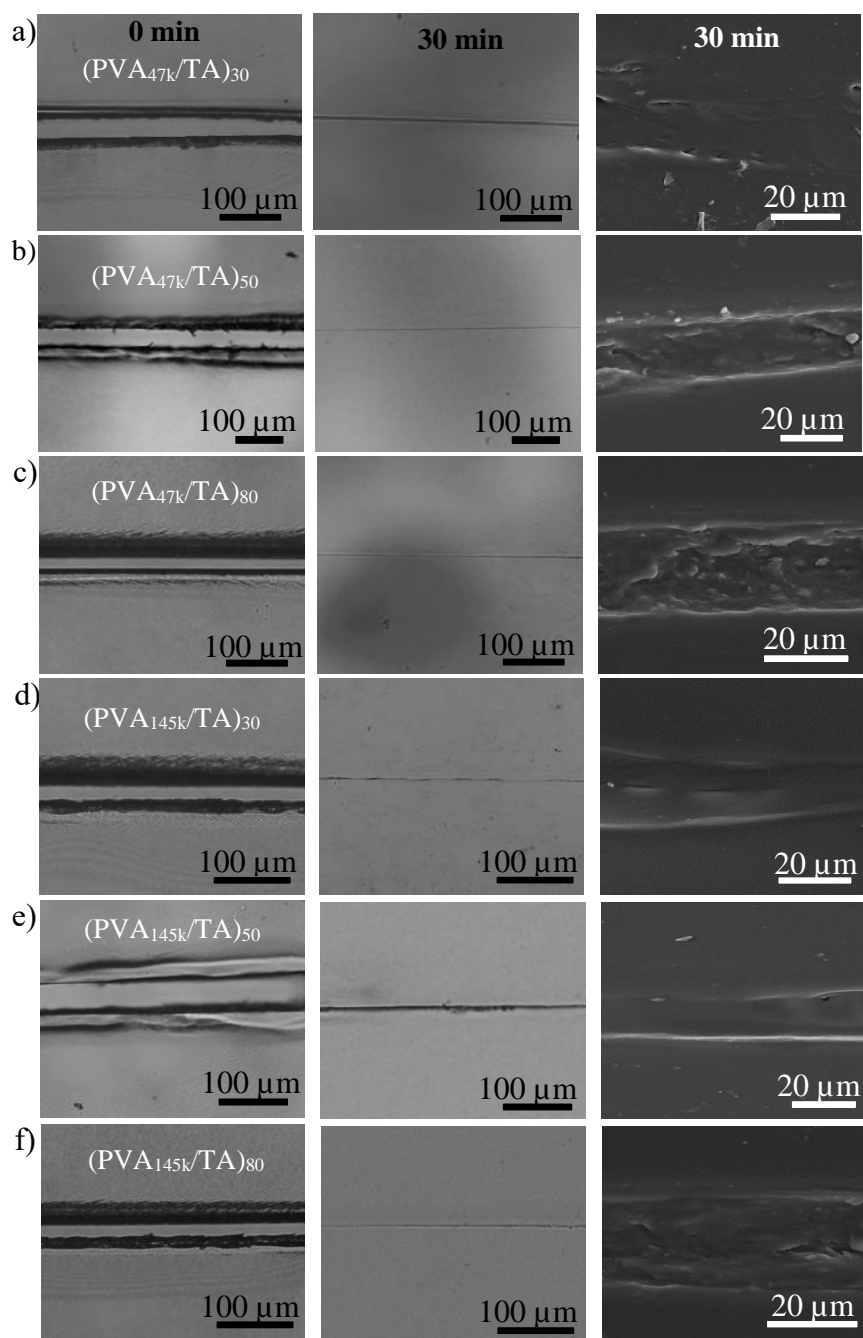


Figure S13. Optical and SEM images of a) (PVA_{47k}/TA)₃₀, b) (PVA_{47k}/TA)₅₀, c) (PVA_{47k}/TA)₈₀, d) (PVA_{145k}/TA)₃₀, e) (PVA_{145k}/TA)₅₀ and f) (PVA_{145k}/TA)₈₀ with a 50 μm wide cut throughout the film before and after immersion in water for 30 min.

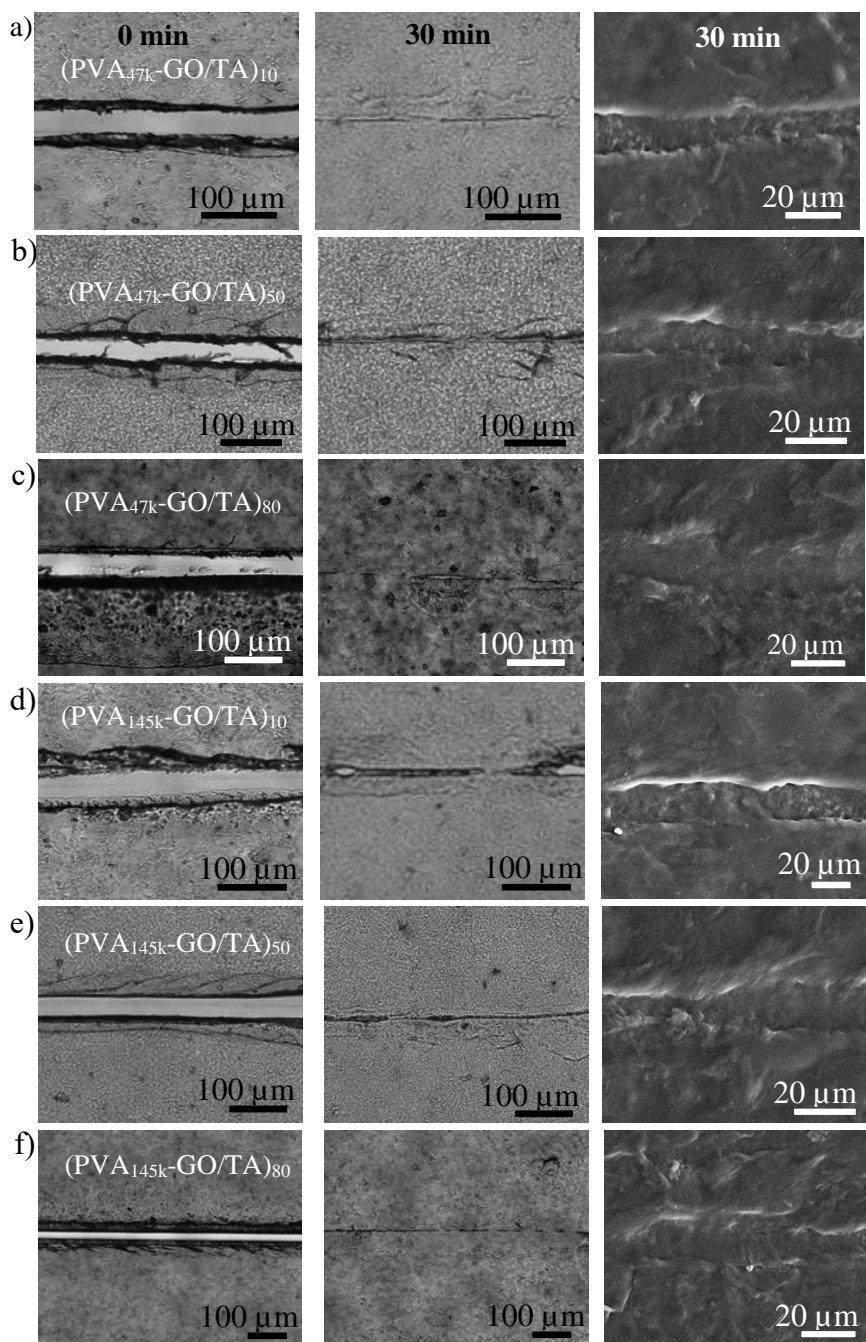


Figure S14 Optical and SEM images of a) (PVA_{47k}-GO/TA)₁₀, b) (PVA_{47k}-GO/TA)₅₀, c) (PVA_{47k}-GO/TA)₈₀, d) (PVA_{145k}-GO/TA)₁₀, e) (PVA_{145k}-GO/TA)₅₀ and f) (PVA_{145k}-GO/TA)₈₀ with a 50 μm wide cut throughout the film before and after immersion in water for 30 min.

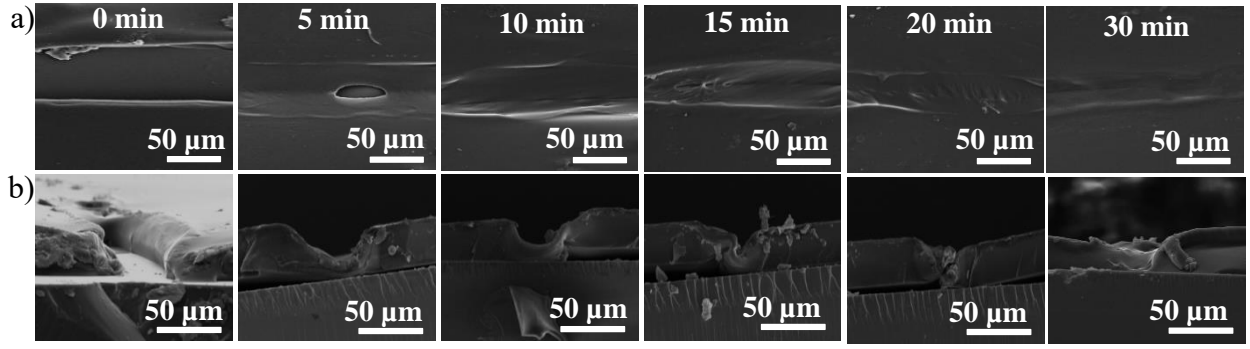


Figure S15. a) Surface and b) cross-sectional SEM images of (PVA_{47k}/TA)₅₀ with a 50 μm wide cut throughout the film before and after immersion in water for different lengths of time.

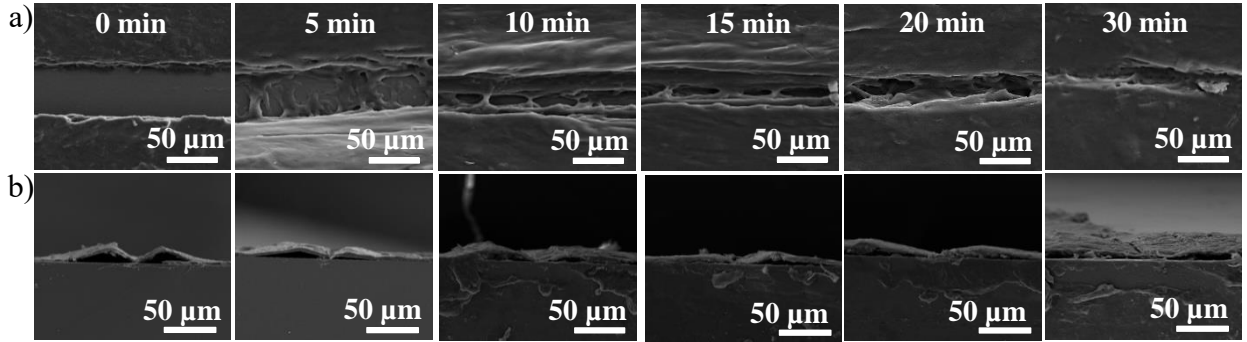


Figure S16. a) Surface and b) cross-sectional SEM images of (PVA_{47k}-GO/TA)₁₀ with a 50 μm wide cut throughout the film before and after immersion in water for different lengths of time.

Details about the development of theoretical model: The flux (J) into and out of the crack area (Figure S17) can be expressed as

$$J_{in} = Ak_{in}(c_1 - c_2) = hwk_{in}(c_1 - c_2) \quad (1)$$

$$J_{out} = A'k_{out}(c_2 - c_3) = zwk_{out}(c_2 - c_3) \quad (2)$$

where k_{in} and k_{out} are the mass transfer coefficients into and out of the crack with the cross-sectional areas of A and A' , respectively; h is the thickness of the film, z is the length of the crack and w is the width of the film; c_1 , c_2 and c_3 are polymer concentrations in the film, crack area and solution, respectively.

We also have

$$V \frac{dc_2}{dt} = Ak_{in}(c_1 - c_2) - A'k_{out}(c_2 - c_3) \quad (3)$$

where V is the volume of the crack part.

We use the initial condition ($t = 0$; $c_2 = 0$) to integrate the mass balance which gives

$$c_2 = \frac{Ak_{in}c_1 + A'k_{out}c_3}{Ak_{in} + A'k_{out}} \left(1 - e^{-\left(\frac{Ak_{in} + A'k_{out}}{V}\right)t} \right) \quad (4)$$

Here c_1 can be regarded as a constant and c_3 can be essentially zero due to the large volume of the solution. Also consider that k_{out} is much smaller than k_{in} due to the interactions between PVA and TA, we have

$$c_2 = c_1 \left(1 - e^{-\left(\frac{Ak_{in}}{V}\right)t} \right) \quad (5)$$

After a time of t_e , an equilibrium is established,

$$J_{in} = J_{out} \quad (6)$$

So

$$h k_{in}(c_1 - c_2) = z k_{out}(c_2 - c_3) \quad (7)$$

Combining (5) and (7) gives

$$h k_{in} e^{-\left(\frac{k_{in}A}{V}\right)t_e} c_1 = z k_{out} \left(1 - e^{-\left(\frac{k_{in}A}{V}\right)t_e} \right) c_1 \quad (8)$$

Then we have

$$e^{-\left(\frac{k_{in}A}{V}\right)t_e} = \frac{z k_{out}}{h k_{in} + z k_{out}} \quad (9)$$

According to (9), t_e is in reverse proportion to k_{out} . On the other hand, suppose that after a time period (t_r), a complete repair can be achieved. According to (5), t_r is mainly determined by and reversely proportional to k_{in} .

As at the equilibrium, no further accumulation in the crack area can be expected, to make a full recovery, it is necessary to have

$$t_r < t_e$$

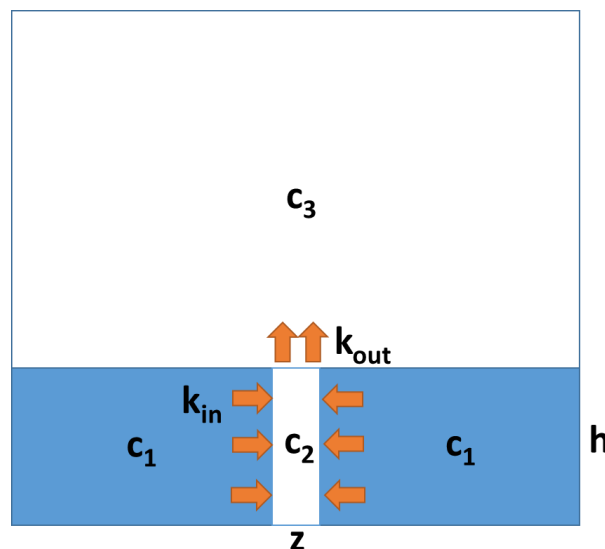


Figure S17. A side view of the self-healing process on an e-LBL film with a thickness of h and a width of w (not shown in this schematic drawing). The white rectangular region denotes the crack with a width of z . c_1 , c_2 and c_3 are polymer concentrations in the film, crack area and solution, respectively. k_{in} and k_{out} are the mass transfer coefficients into and out of the crack area.

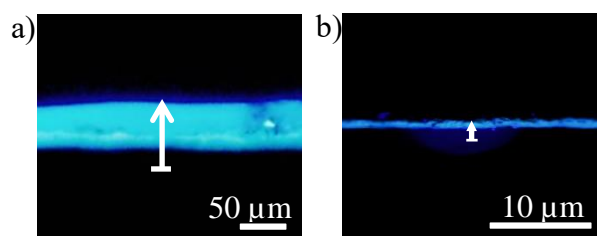


Figure S18. Cross-sectional fluorescent optical images of a) (PVA_{145k}/TA)₅₀ and (PVA_{145k}-GO/TA)₁₀ after their immersion into PVA_{145k}-c solutions for 5 min and 1 min, respectively. The immersion time is chosen to avoid complete diffusion across the film so that the diffusion rate can be calculated.

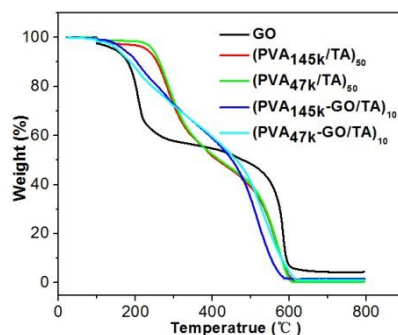


Figure S19. TGA results for GO, (PVA_{145k}/TA)₅₀, (PVA_{47k}/TA)₅₀, (PVA_{145k}-GO/TA)₁₀ and (PVA_{47k}-GO/TA)₁₀.

The determination of GO contents in the film: Thermogravimetric analysis (TGA) was used to calculate the contents of GO in l-LBL film (Figure S19). (PVA/TA)₅₀ can decompose completely when reaching 620 °C under oxygen atmosphere, however, GO does not fully decompose at a temperature as high as 800 °C. The above fact provides a simple method to calculate the percentage of GO in (PVA-GO/TA)₁₀. In Eq. 1, a and b are the remaining contents of GO and (PVA-GO/TA)₁₀ at 800 °C, respectively; x is therefore the contents of GO in (PVA-GO/TA)₁₀.

$$ax = b \quad (1)$$

According to Eq. 1, there are approximately 30 % and 35 % GO in (PVA_{47k}-GO/TA)₁₀ and (PVA_{145k}-GO/TA)₁₀, respectively.

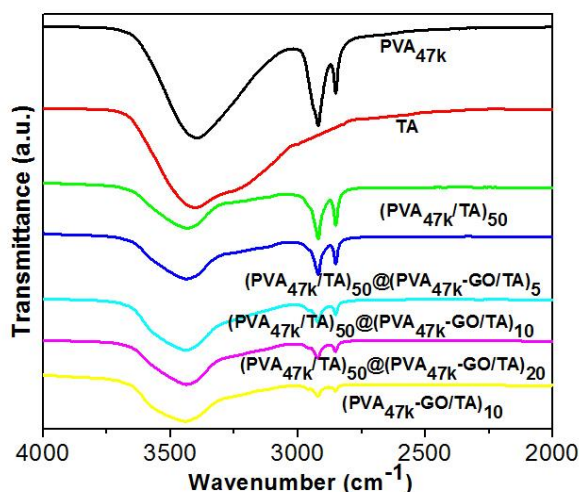


Figure S20. IR spectra of PVA_{47k}, TA and the solutions after immersion of (PVA_{47k}/TA)₅₀, (PVA_{47k}/TA)₅₀@(PVA_{47k}-GO/TA)₅, (PVA_{47k}/TA)₅₀@(PVA_{47k}-GO/TA)₁₀, (PVA_{47k}/TA)₅₀@(PVA_{47k}-GO/TA)₂₀, and (PVA_{47k}-GO/TA)₁₀ into 50 mL DI water for 30 min.

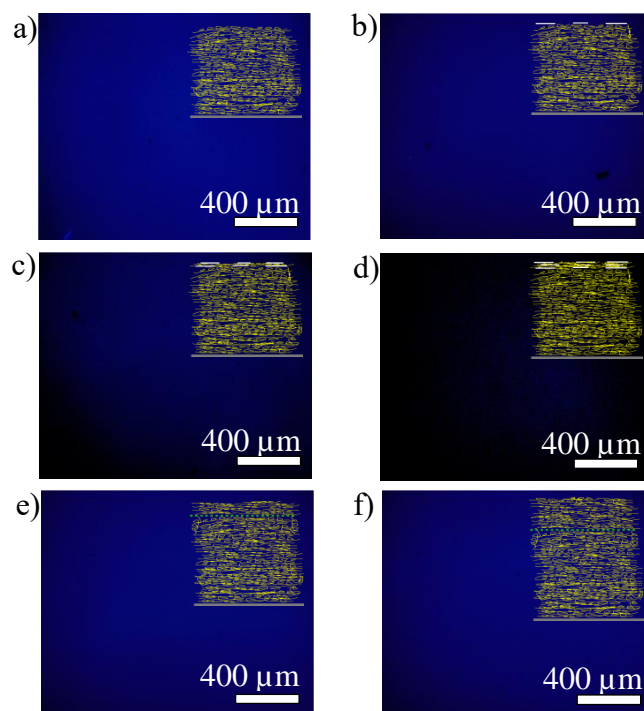


Figure S21. Surface fluorescent optical images of $(\text{PVA}_{145\text{k}}/\text{TA})_{50}(\text{PVA}_{145\text{k}}\text{-c}/\text{TA})$ after the additional depositions of a) 0, b) 5, c) 10 and d) 20 bilayers of $\text{PVA}_{145\text{k}}\text{-GO}/\text{TA}$ and e) 5 and f) 10 bilayers of $\text{PVA}_{145\text{k}}/\text{TA}$.

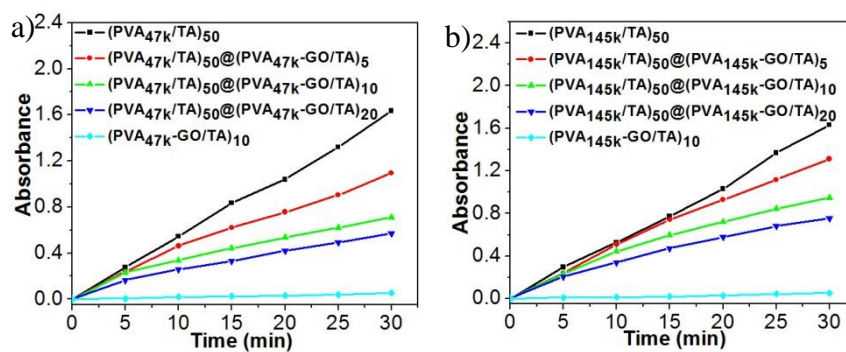


Figure S22. The plotting of TA absorbance against different lengths of immersion time in 50 mL water for a) $(\text{PVA}_{47\text{k}}\text{-GO}/\text{TA})_{10}$ and $(\text{PVA}_{47\text{k}}/\text{TA})_{50}$ with different bilayers of $(\text{PVA}_{47\text{k}}\text{-GO}/\text{TA})$ and b) $(\text{PVA}_{145\text{k}}\text{-GO}/\text{TA})_{10}$ and $(\text{PVA}_{145\text{k}}/\text{TA})_{50}$ with different bilayers of $(\text{PVA}_{145\text{k}}\text{-GO}/\text{TA})$ on the top.

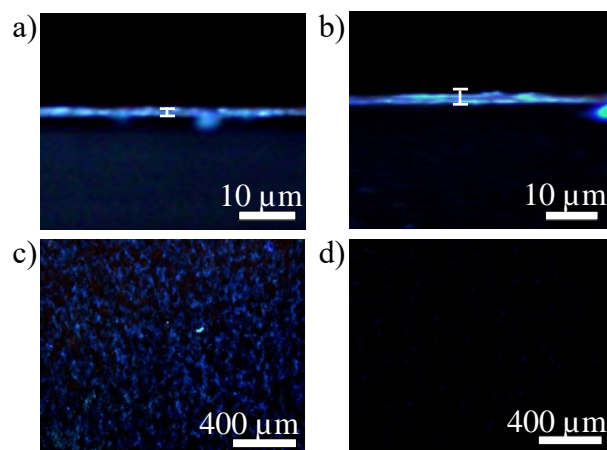


Figure S23 a,b) Cross-sectional and c,d) surface fluorescent optical images of a,c) (PVA_{47k}-c-GO/TA)(PVA_{47k}-GO/TA)₉ and b,d) (PVA_{47k}-c-GO/TA)(PVA_{47k}-GO/TA)₁₉. The dotted surface fluorescence may be due to the high roughness of l-LBL film growing on glass substrates. The insets in a and b indicate the thickness of the film.

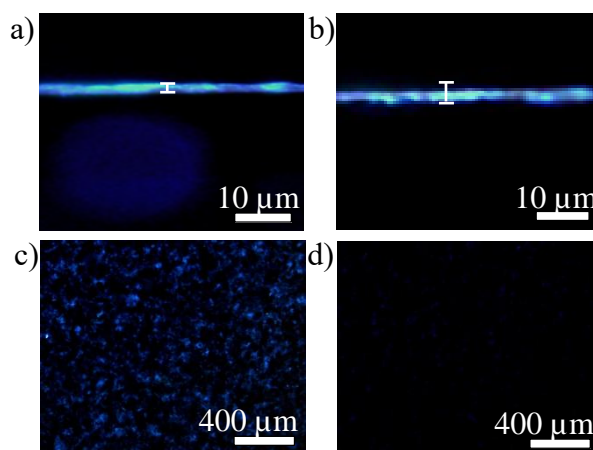


Figure S24. a,b) Cross-sectional and c,d) surface fluorescent optical images of a,c) (PVA_{145k}-c-GO/TA)(PVA_{145k}-GO/TA)₉ and b,d) (PVA_{145k}-c-GO/TA)(PVA_{145k}-GO/TA)₁₉. The dotted surface fluorescence may be due to the high roughness of l-LBL film growing on glass substrates. The insets in a and b indicate the thickness of the film.

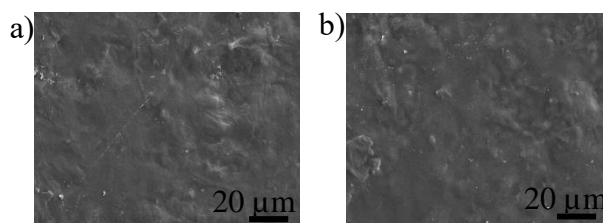


Figure S25. SEM images of a) (PVA_{47k}/TA)₅₀@(PVA_{47k}-GO/TA)₂₀ and b) (PVA_{145k}/TA)₅₀@(PVA_{145k}-GO/TA)₂₀.

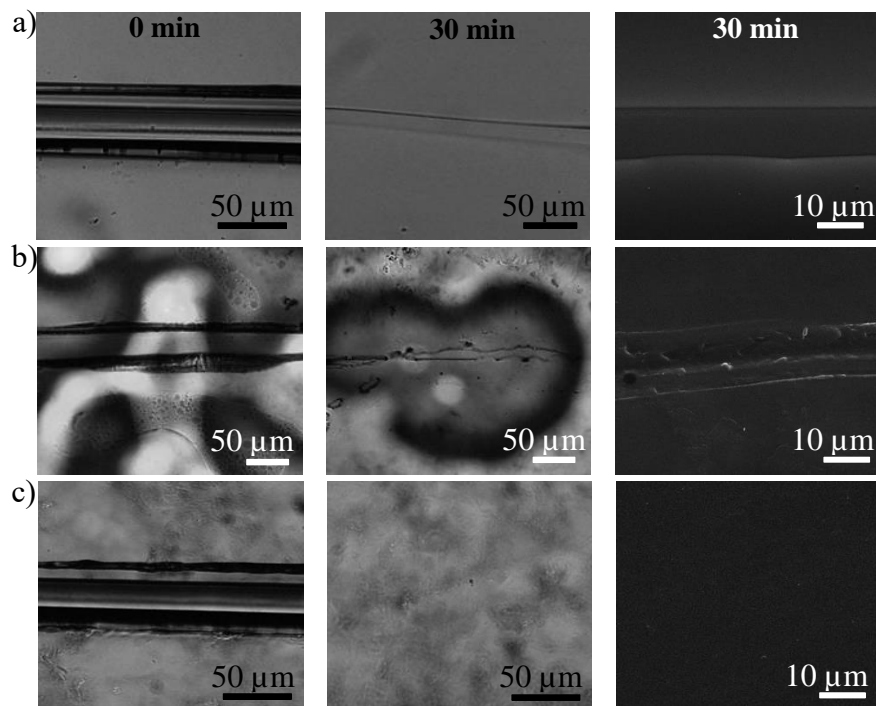


Figure S26. Optical and SEM images of a) (PEG/TA)₅₀, b) (PEG-GO/TA)₁₀ and c) (PEG/TA)₅₀@(PEG-GO/TA)₁₀ with a 50 μm wide cut throughout the film before and after immersion in water for 30 min.

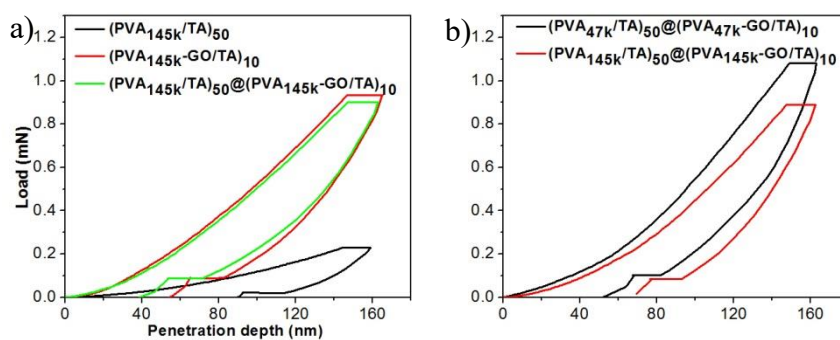


Figure S27. Load-displacement curves for a) (PVA_{145k}/TA)₅₀, (PVA_{145k}-GO/TA)₁₀, (PVA_{145k}/TA)₅₀@(PVA_{145k}-GO/TA)₁₀ and b) the crack area after self-healing for (PVA_{47k}/TA)₅₀@(PVA_{47k}-GO/TA)₁₀ and (PVA_{145k}/TA)₅₀@(PVA_{145k}-GO/TA)₁₀.

Samples	Modulus E (GPa)	Hardness H (GPa)
(PVA _{145k} /TA) ₅₀	8.7±0.1	0.44±0.02
(PVA _{145k} -GO/TA) ₁₀	26.7±0.4	2.18±0.11
(PVA _{145k} /TA) ₅₀ @(PVA _{145k} -GO/TA) ₁₀	25.1±2.2	2.11±0.22

Table S1. A summary of modulus and hardness for different multilayers consisting of PVA_{145k} obtained from nanoindentation.

Modes	Samples	Modulus E (GPa)	Hardness H (GPa)
Before self-healing	(PVA _{47k} /TA) ₅₀ @(PVA _{47k} -GO/TA) ₁₀	31.4±1.8	2.27±0.09
	(PVA _{145k} /TA) ₅₀ @(PVA _{145k} -GO/TA) ₁₀	25.1±2.2	2.11±0.22
After self-healing	(PVA _{47k} /TA) ₅₀ @(PVA _{47k} -GO/TA) ₁₀	31.3±1.9	2.21±0.05
	(PVA _{145k} /TA) ₅₀ @(PVA _{145k} -GO/TA) ₁₀	24.3±0.8	2.09±0.26

Table S2. A summary of modulus and hardness for hybrid films before and after self-healing obtained from nanoindentation.