

Supporting Information

Spatial Isolation of Conformational Isomers of Hydroquinone and Its Water Cluster Using the Stark Deflector

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1. Dependence of deflection on voltages and carrier gas of 3-AP conformers.

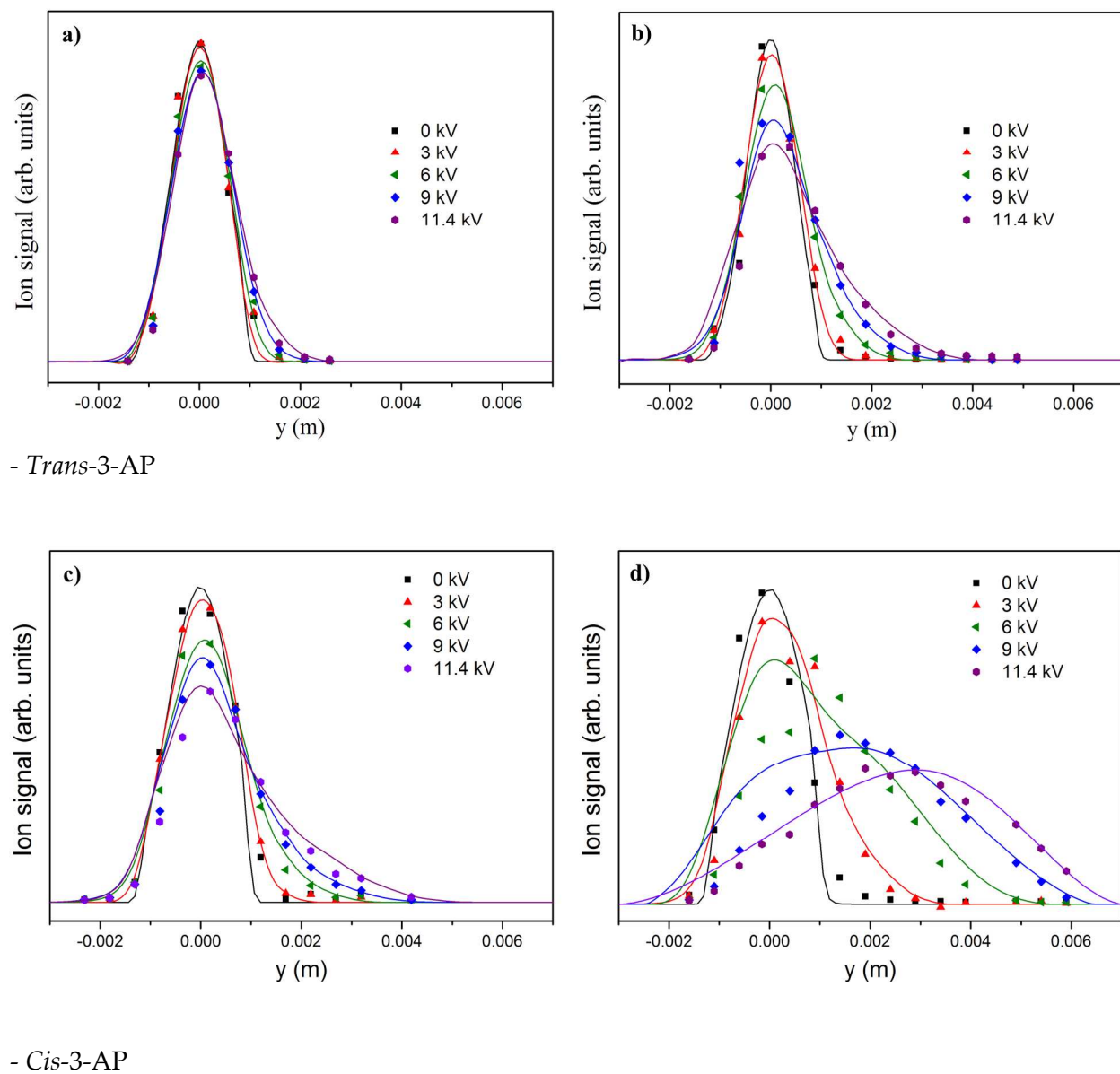
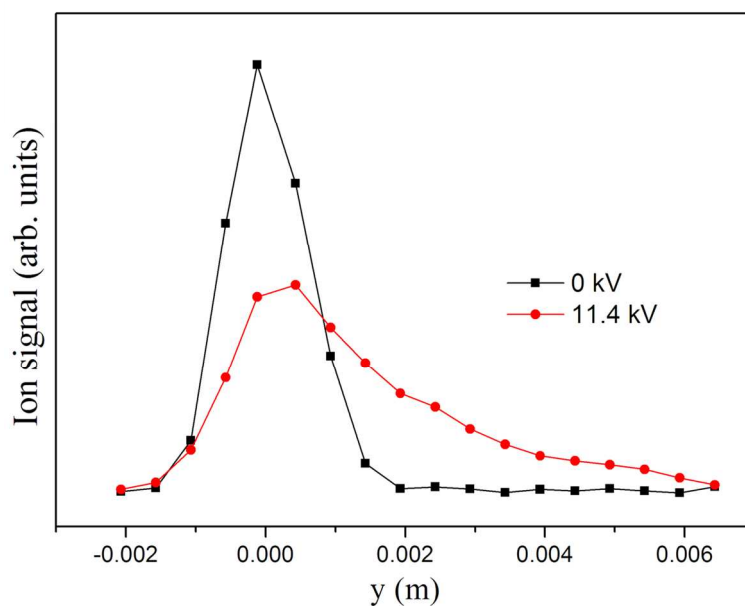


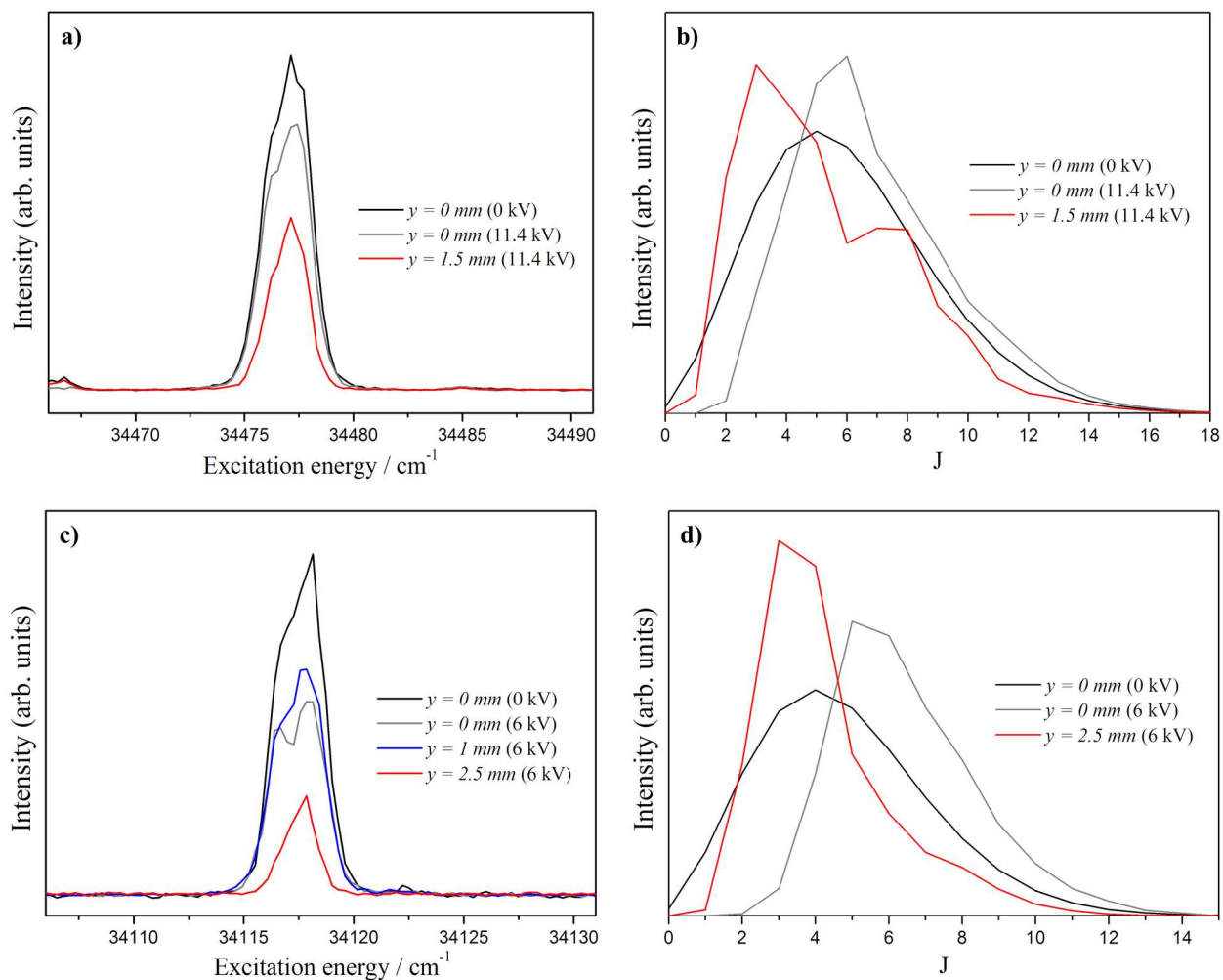
Figure S1. Deflection profiles of *trans*, *cis*-3-AP seeded in (a, c) Ne and (b, d) Ar at the voltages from 0 kV to 11.4 kV. Symbols represent experiment and lines are simulations.

2. Deflection
trans-3-AP
on.



profiles of
seeded in Xen-

Figure S2. Experimental deflection profiles of *trans*-3-AP seeded in Xe at the deflector voltage of 11.4 kV taken by monitoring the S_1 - S_0 origin signal intensity. The average deflection position is estimated to be ~ 1 mm.



3. Stark-R2PI spectra and rotational state distributions at deflected regions of 3-AP.

Figure S3. Stark-R2PI spectra of (a) *trans*-3-AP and (c) *cis*-3-AP seeded in Ar at the origin band and corresponding normalized rotational state distributions (b, d) populated at specific deflected regions in the deflection profiles at simulated T_{rot} (Table 1). Since rotational states with lowest J 's deflect most, relative population of low J 's stands out in the deflected region. Accordingly,

full width at half maximum (FWHM) of the origin peak of *cis(trans)*-3-AP becomes shortened from 2.5 (2.4) cm^{-1} to 1.7 (2.0) cm^{-1} , respectively.

4. Stark energy
3-AP, HyQ, and
ers.

curves for (*trans*, *cis*)-
HyQ-H₂O conform-

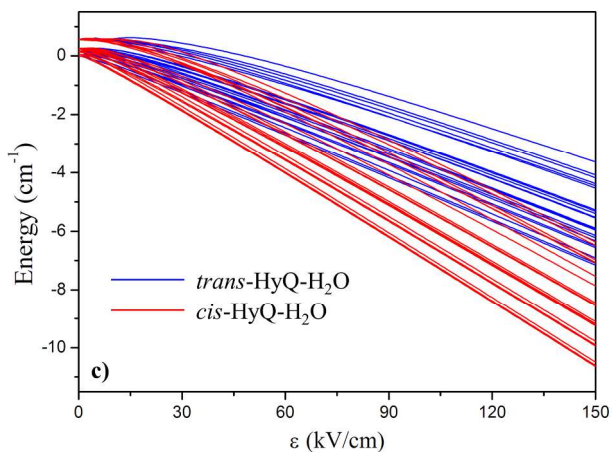
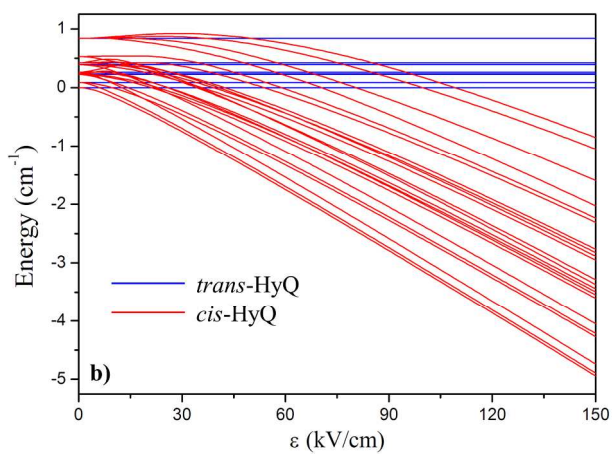
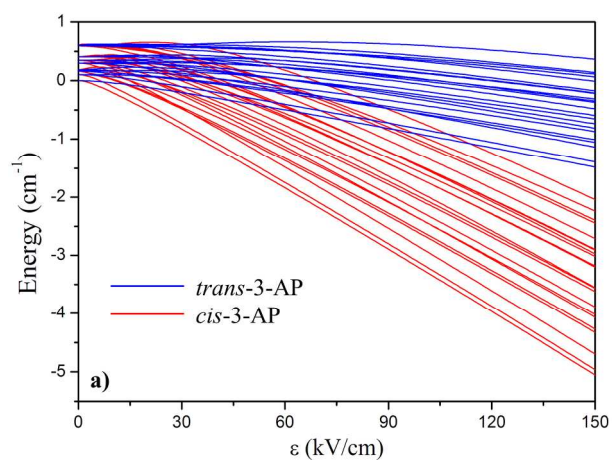


Figure S4. Stark energy shifts of 3-AP, HyQ and HyQ-H₂O conformational isomers for the lowest rotational quantum states ($J \leq 2$) as a function of applied electric field strengths.^{1,2}

5. Molecular parameters for simulation of deflection profile

Table S1. Molecular mass, rotational constants, and electric dipole moments along the principal axis.

	<i>cis(trans)</i> -3-AP ^a	<i>cis</i> ^b (<i>trans</i> ^c)-HyQ	<i>cis(trans)</i> -HyQ-H ₂ O ^d
Molecular mass			
Mass (g/mol)	109.13	110.11	128.13
Rotational Constants			
A (MHz)	3734.93 (3730.17)	5614.14 (5615.1)	3844.31 (3820.5)
B (MHz)	1823.21 (1828.26)	1481.59 (1481.7)	757.35 (760.58)
C (MHz)	1226.49 (1228.19)	1172.67 (1172.8)	634.59 (636.20)
Electric dipole moments			
μ_A (D)	1.772 (0.5563)	0 (0)	-2.8280 (-2.6244)
μ_B (D)	1.517 (0.5375)	2.38 (0)	3.6821 (1.7094)
μ_C (D)	0 (0)	0 (0)	0 (0)
μ_{Total} (D)	2.33 (0.77)	2.38 (0)	4.64 (3.13)

^a The values from ref. *Phys. Chem. Chem. Phys.*, 10, 666-673 (2008).³

^b The values from ref. *J. Chem. Phys.*, 100, 8569 (1994).⁴

^c The values from ref. *J. Chem. Phys.*, 99, 5078 (1993).⁵

^d Calculated values from B3LYP/6-311++G(3df,3pd).

Supporting References

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