

Supporting Information:

Hierarchical porous carbon nanospheres for efficient removal of toxic organic water contaminants of phenol and methylene blue

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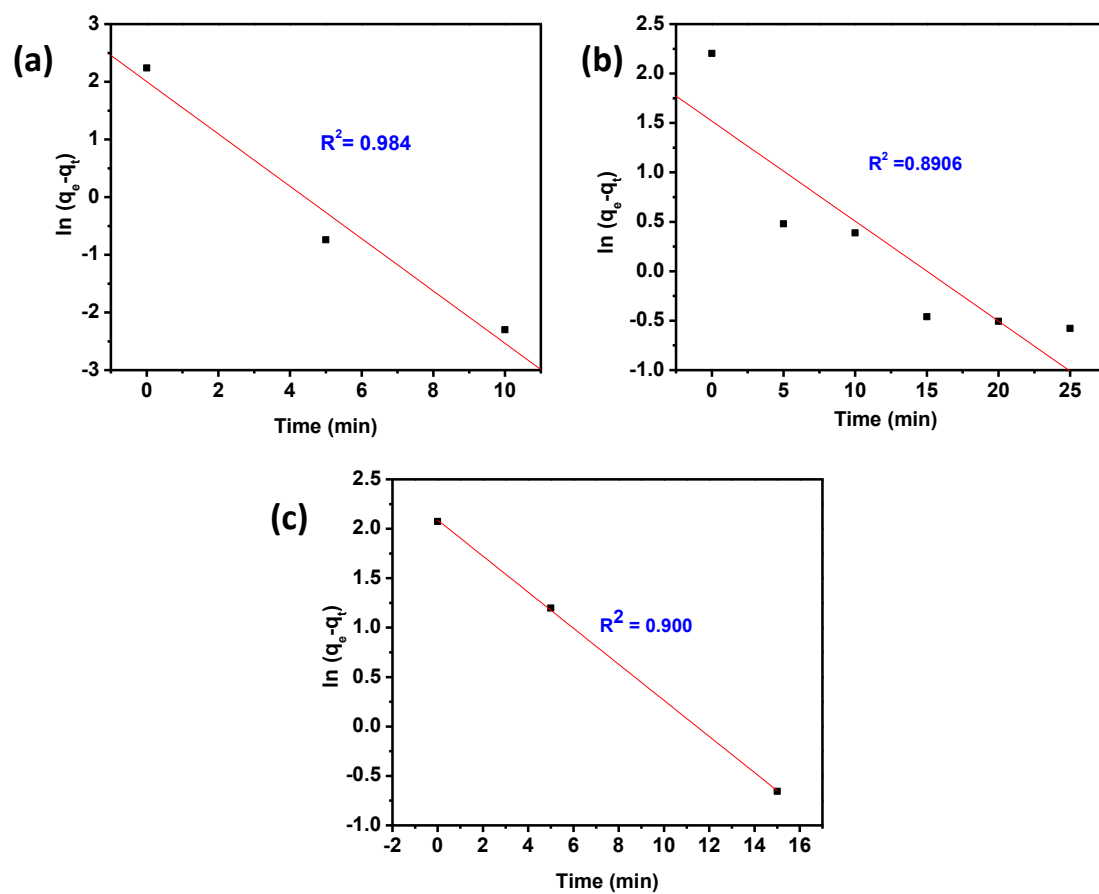


Fig. S1: Pseudo-first order kinetic plots for the adsorption of phenol by porous carbon nanospheres: (a) CF108, (b) CF127 and (c) CL64.

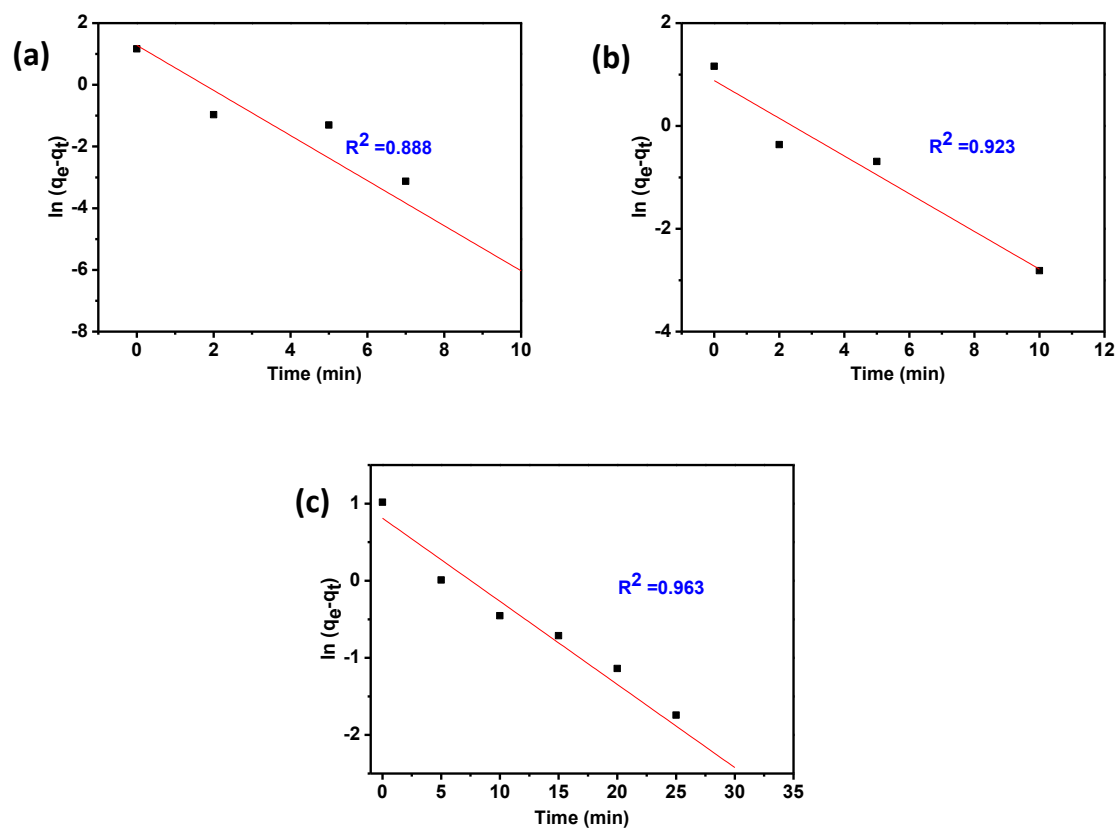


Fig. S2: Pseudo-first order kinetic plots for the adsorption of MB by porous carbon nanospheres: (a) CF108, (b) CF127 and (c) CL64.

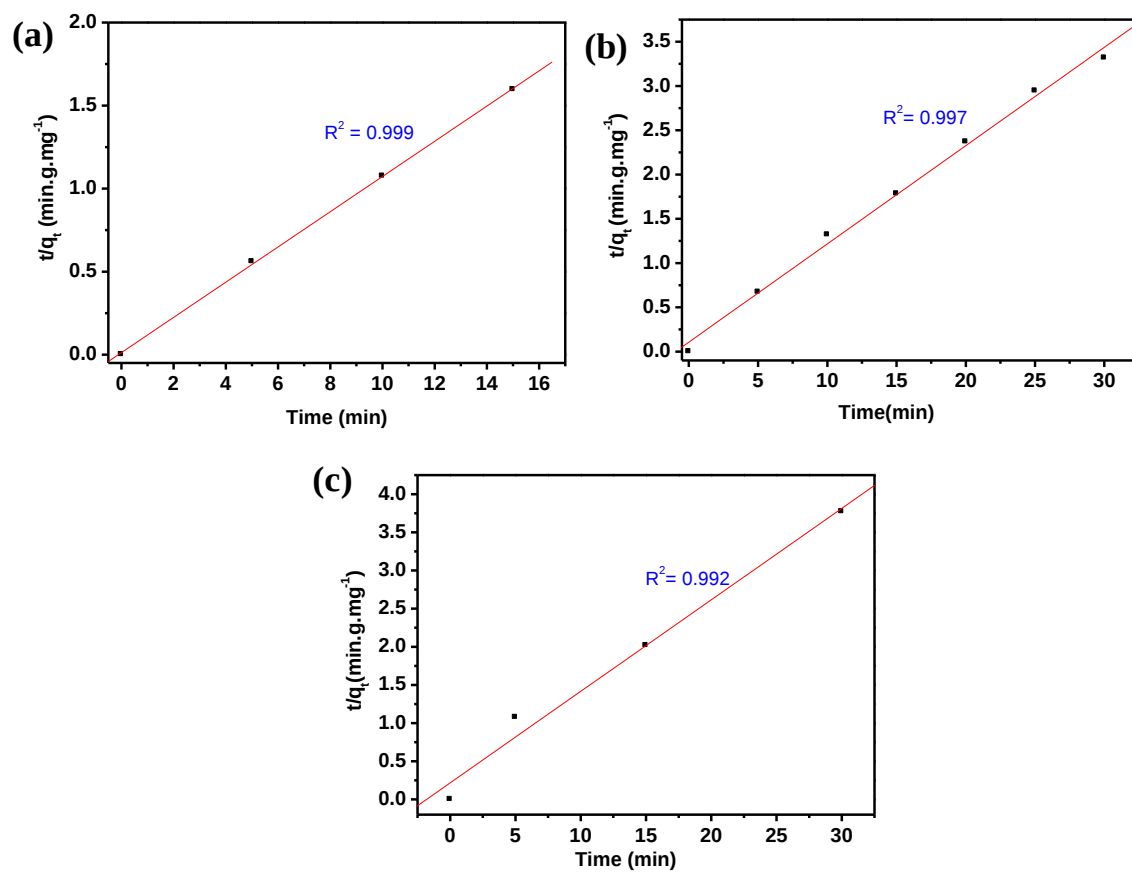


Fig. S3: Pseudo-second order kinetic plots for the adsorption of phenol by porous carbon nanospheres: (a) CF108, (b) CF127 and (c) CL64.

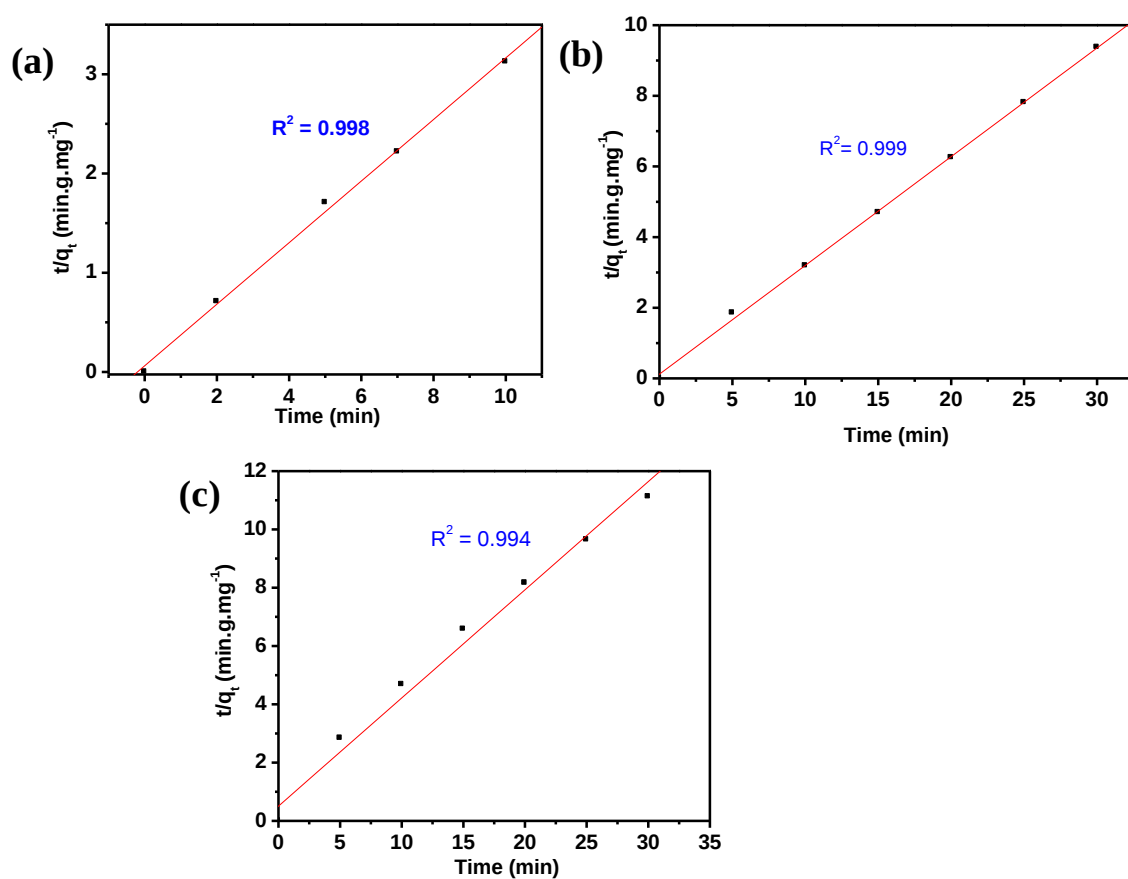


Fig. S4: Pseudo-second order kinetic plots for the adsorption of MB by the porous carbon nanospheres: (a) CF108, (b) CF127 and (c) CL64.

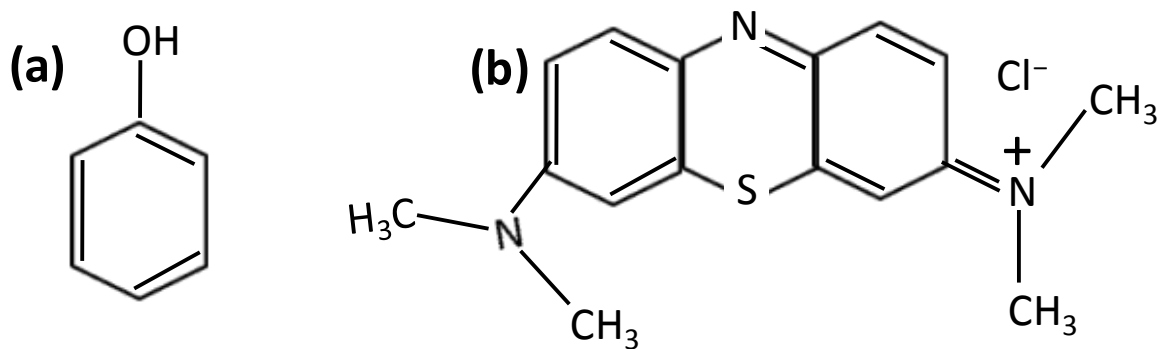


Fig. S5: Structure of (a) phenol and (b) methylene blue (MB).

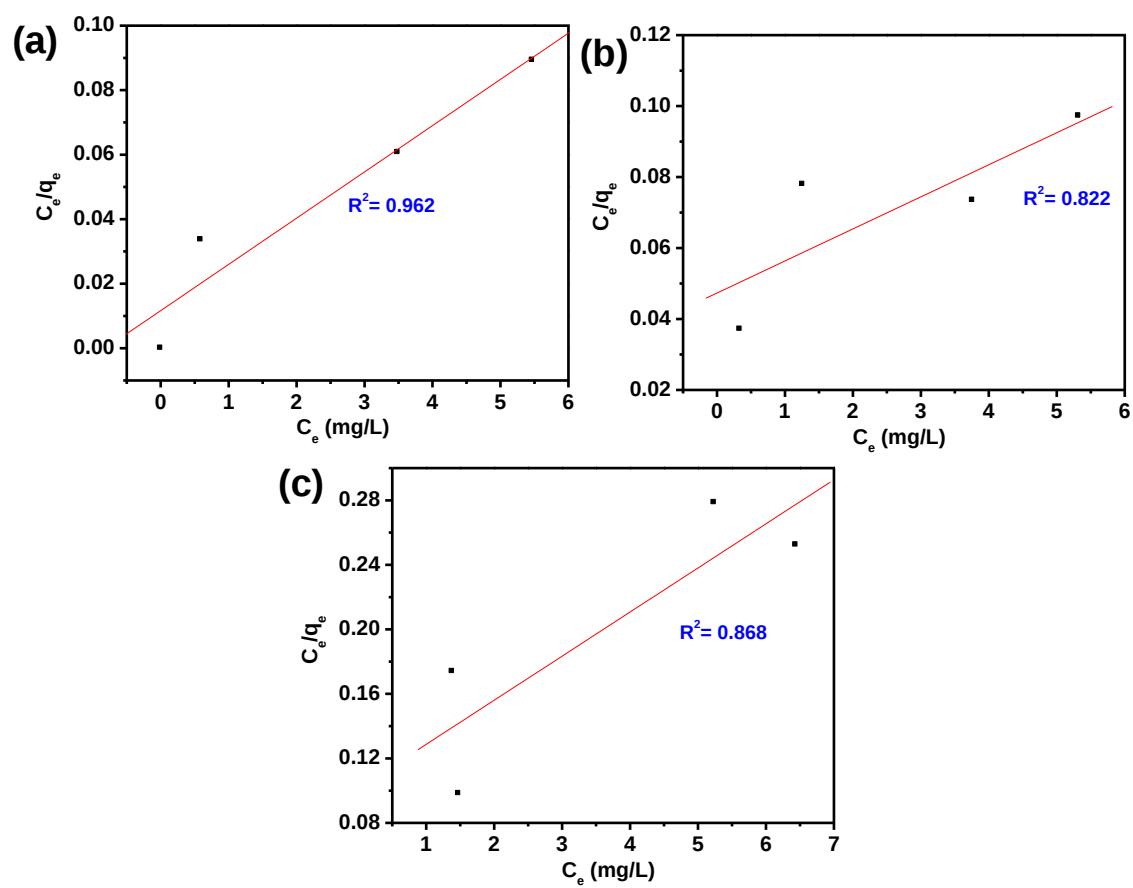


Fig. S6: Langmuir isotherms for phenol adsorption on the porous carbon nanospheres with different adsorbent doses: (a) CF108, (b) CF127 and (c) CL64.

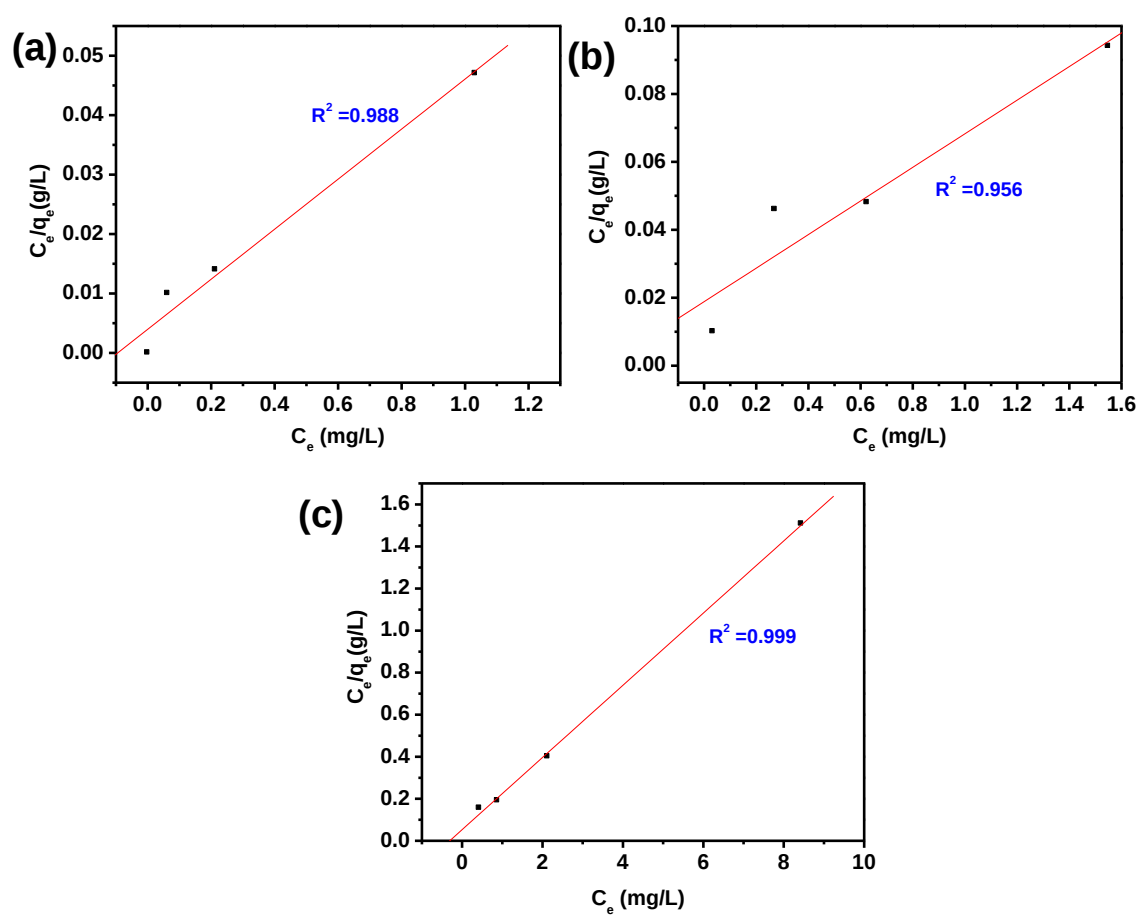


Fig. S7: Langmuir isotherms for MB adsorption on the porous carbon nanospheres with different adsorbent doses: (a) CF108, (b) CF127 and (c) CL64.

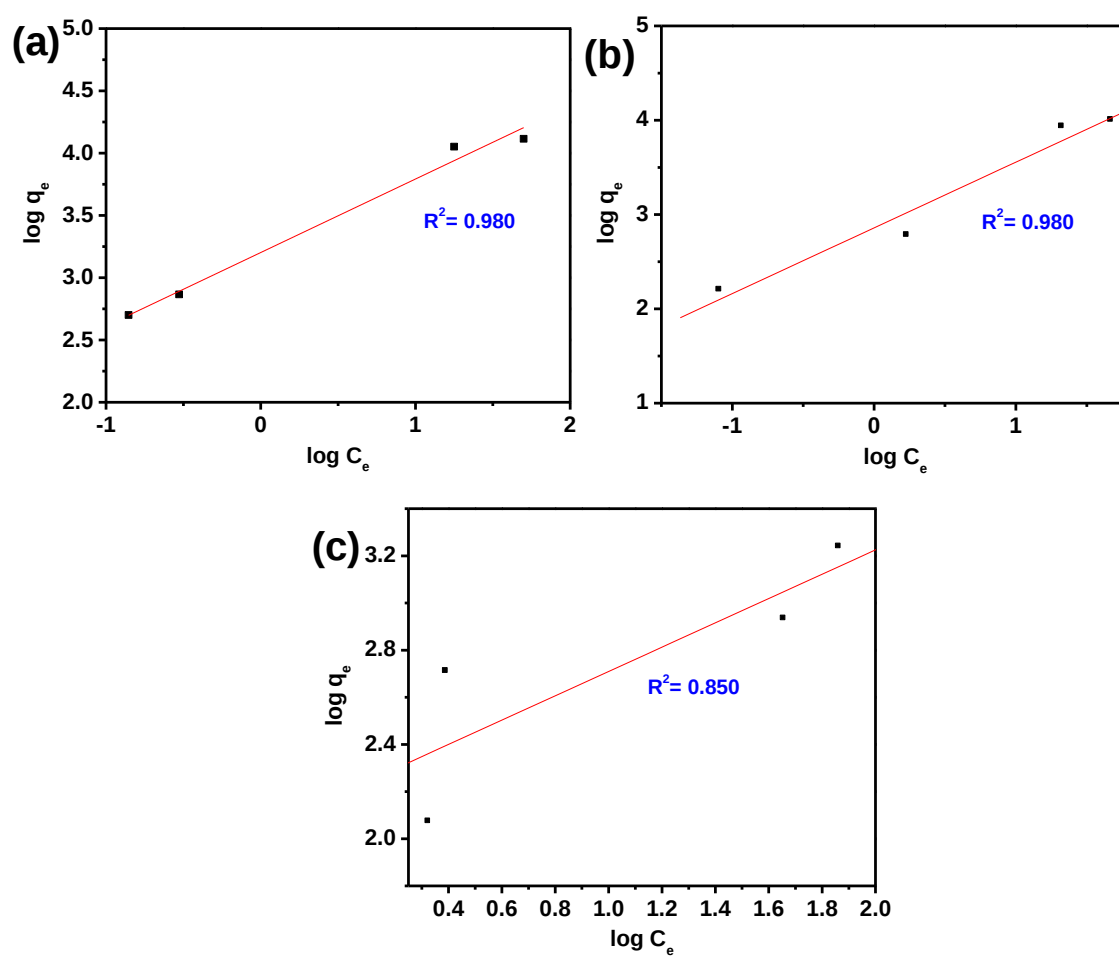


Fig. S8: Freundlich isotherms for phenol adsorption on the porous carbon nanospheres with different adsorbent doses: (a) CF108, (b) CF127 and (c) CL64.

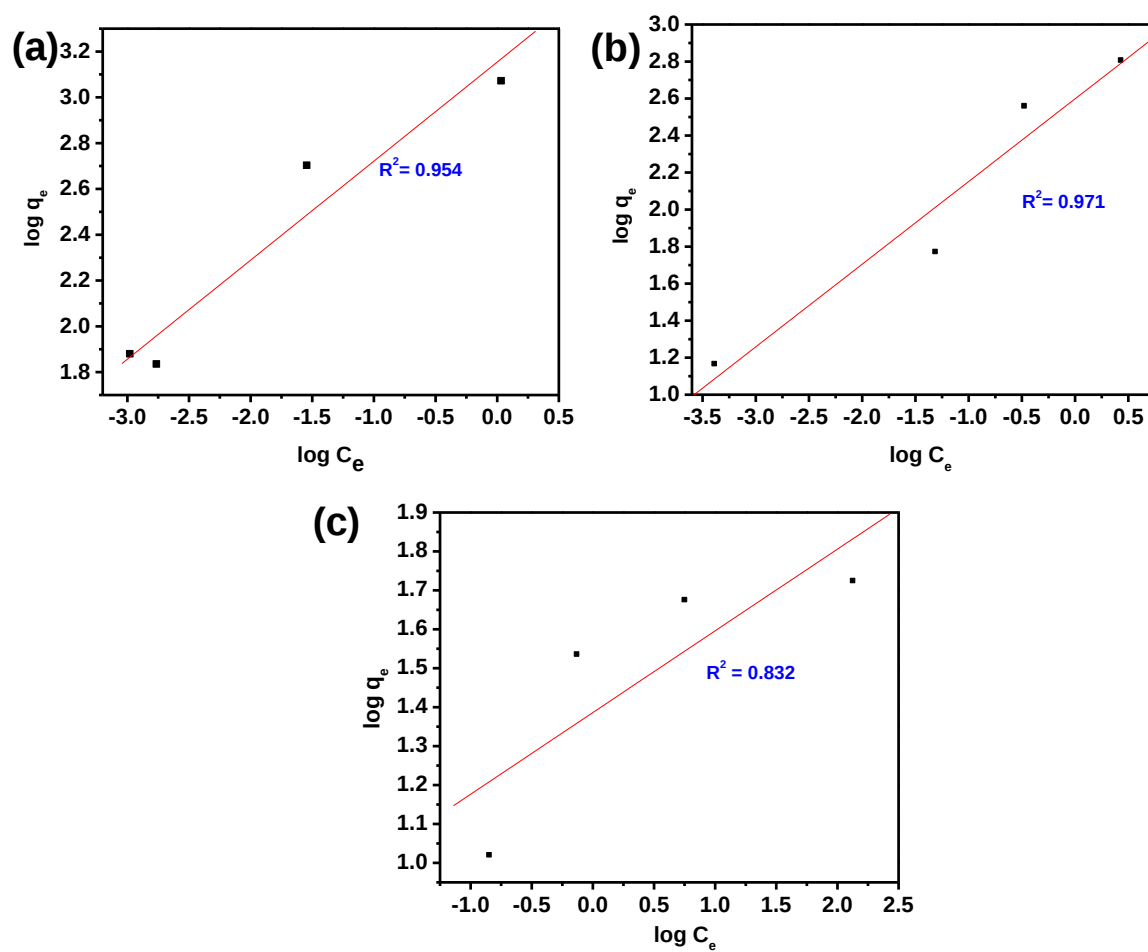


Fig. S9: Freundlich isotherms for MB adsorption on the porous carbon nanospheres with different adsorbent doses: (a) CF108, (b) CF127 and (c) CL64.

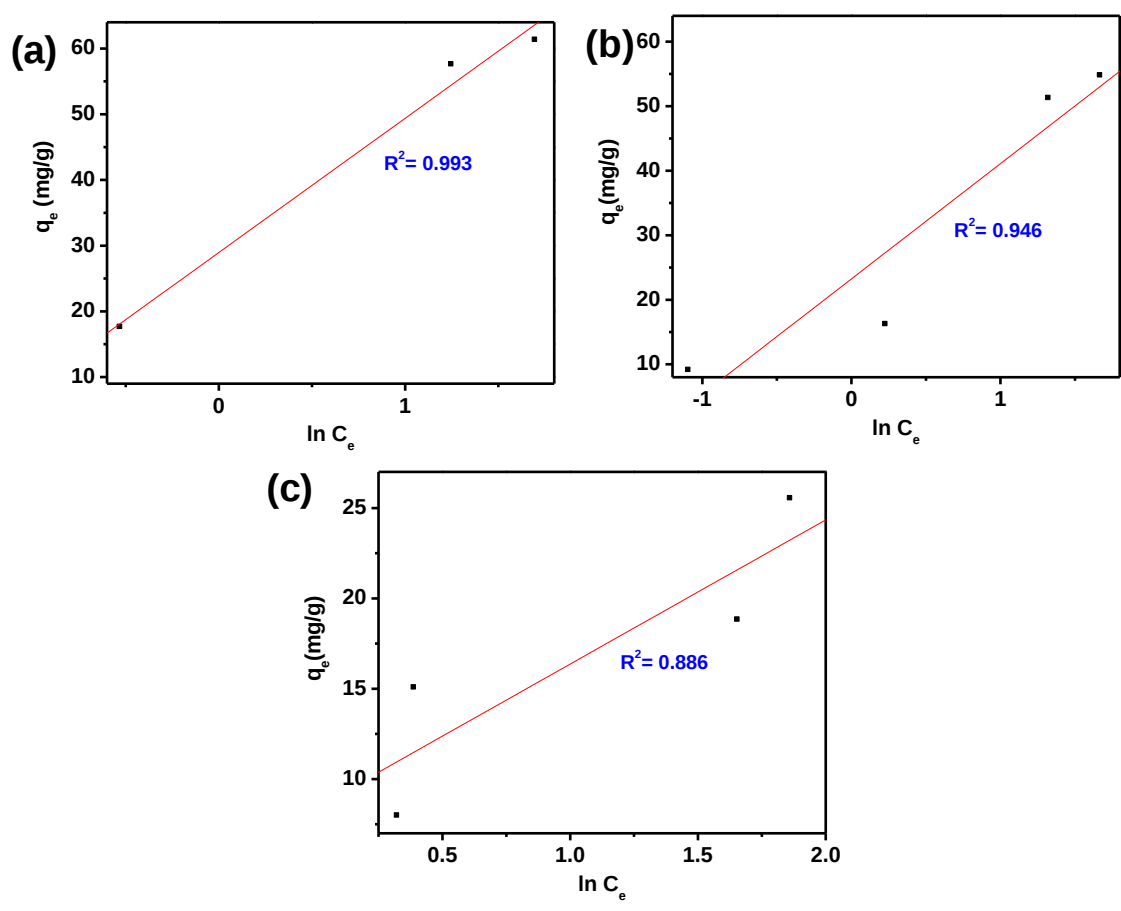


Fig. S10: Temkin isotherms for phenol adsorption on the porous carbon nanospheres with different adsorbent doses: (a) CF108, (b) CF127 and (c) CL64.

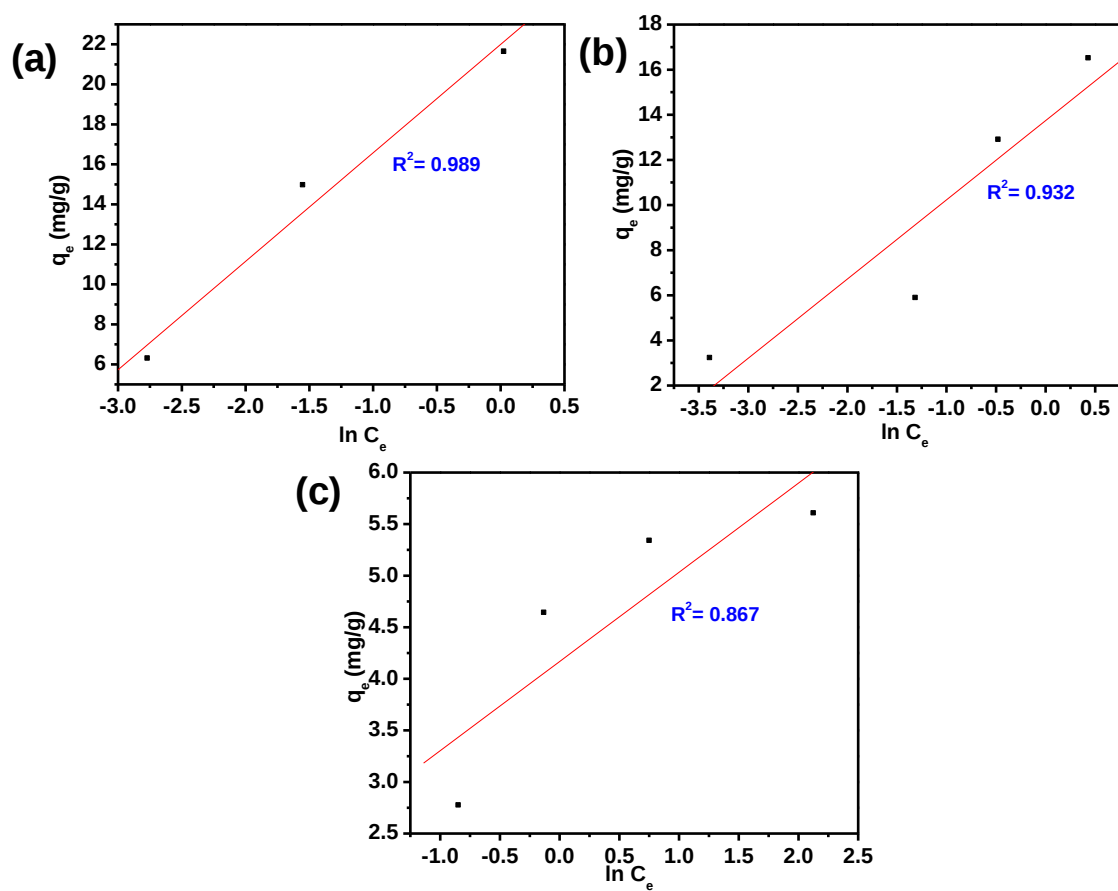


Fig. S11: Temkin isotherms for MB adsorption on the porous carbon nanospheres with different adsorbent doses: (a) CF108, (b) CF127 and (c) CL64.

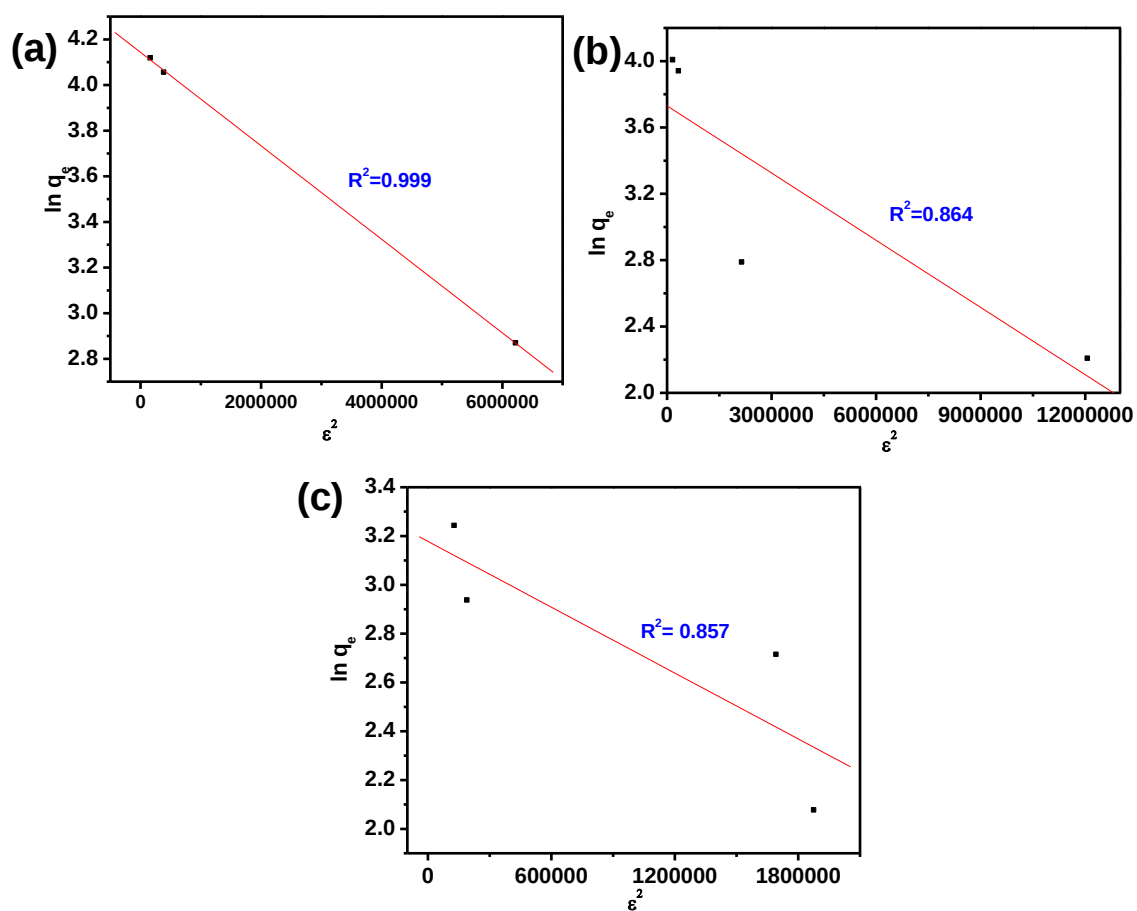


Fig. S12: Dubinin–Radushkevich isotherms for phenol adsorption on the porous carbon nanospheres with different adsorbent doses: (a) CF108, (b) CF127 and (c) CL64.

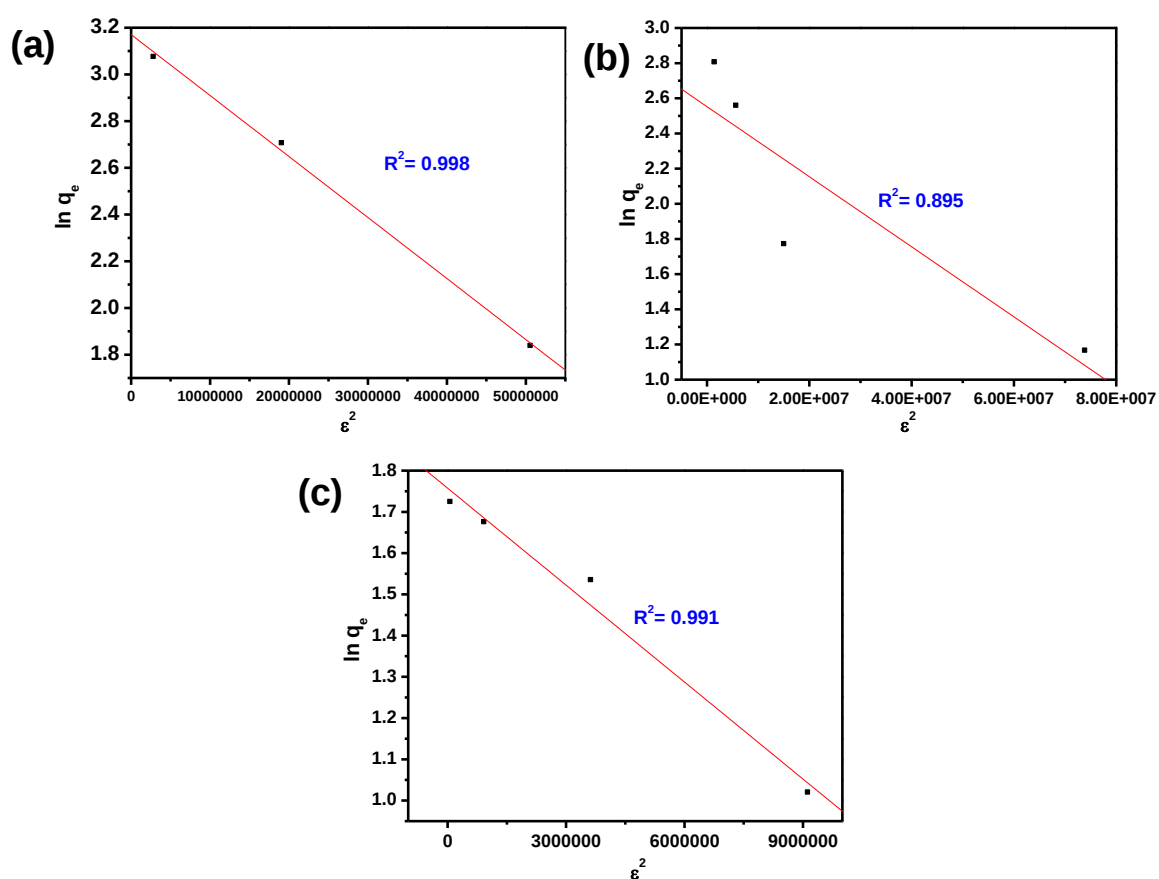


Fig. S13: Dubinin–Radushkevich isotherms for MB adsorption on the porous carbon nanospheres with different adsorbent doses: (a) CF108, (b) CF127 and (c) CL64.

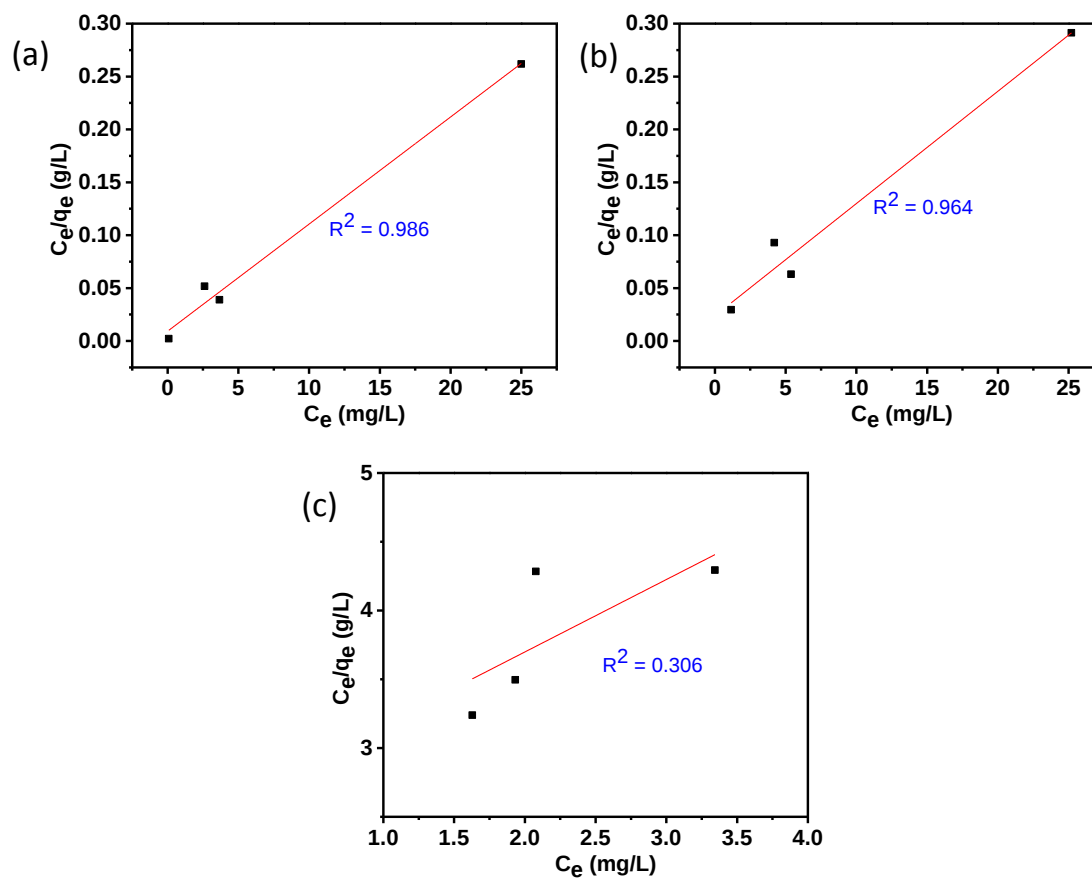


Fig. S14: Langmuir isotherms for phenol adsorption on the porous carbon nanospheres with different adsorbate concentration : (a) CF108, (b) CF127 and (c) CL64.

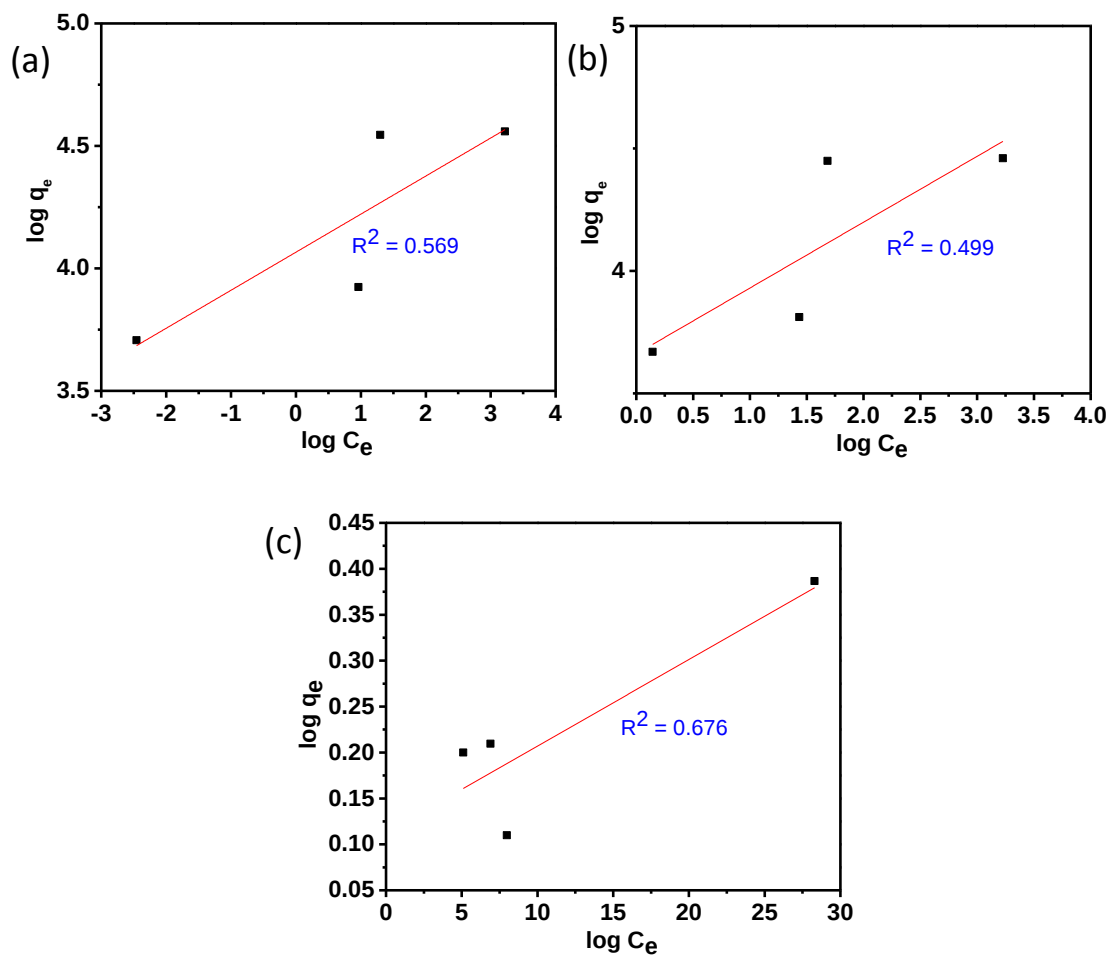


Fig. S15: Freundlich isotherms for phenol adsorption on the porous carbon nanospheres with different adsorbate concentration :(a) CF108, (b) CF127 and (c) CL64.

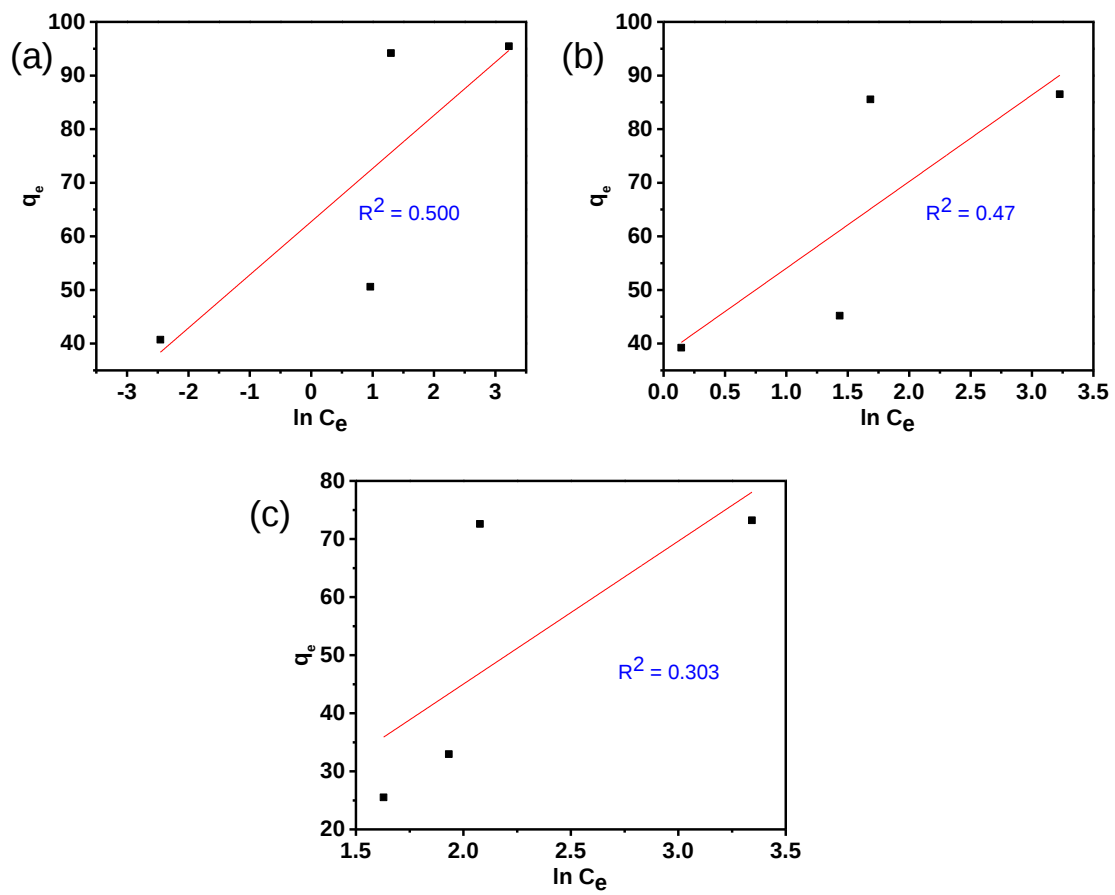


Fig. S16: Temkin isotherms for phenol adsorption on the porous carbon nanospheres with different adsorbate concentration :(a) CF108, (b) CF127 and (c) CL64.

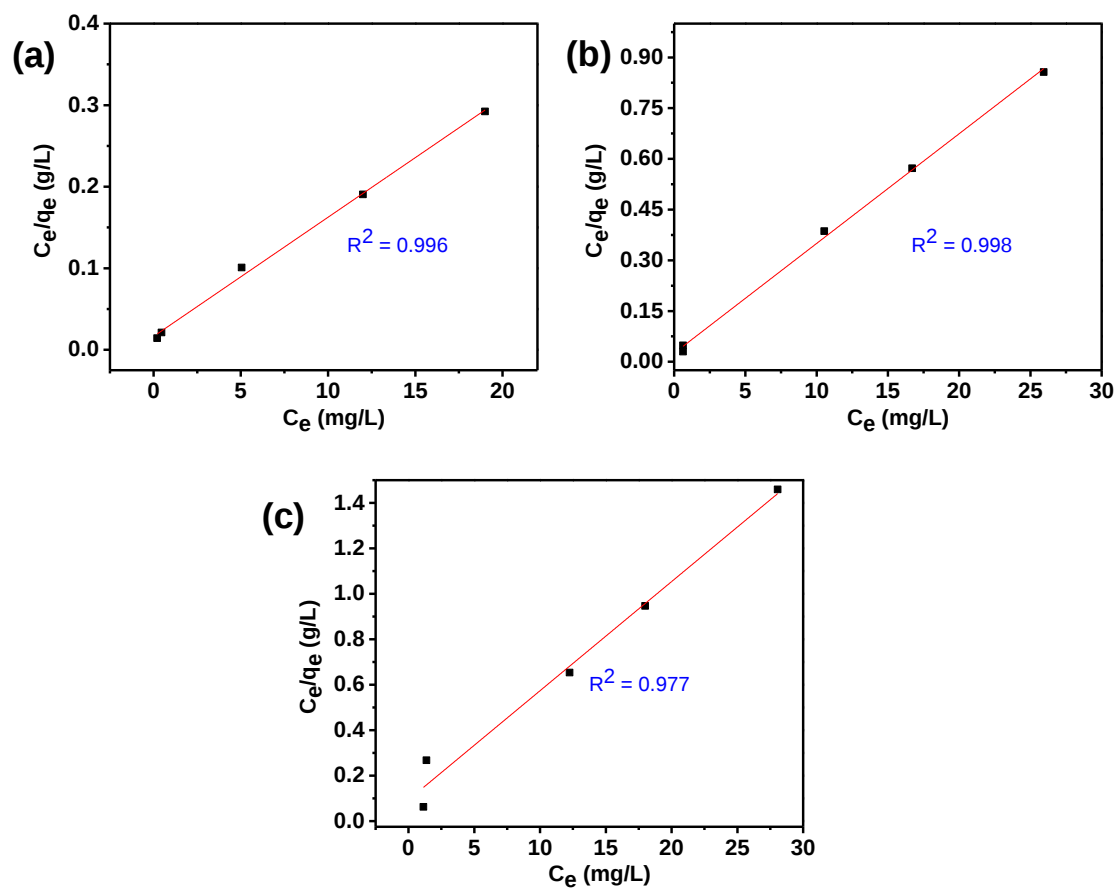


Fig. S17: Langmuir isotherms for MB adsorption on the porous carbon nanospheres with different adsorbate concentration :(a) CF108, (b) CF127 and (c) CL64.

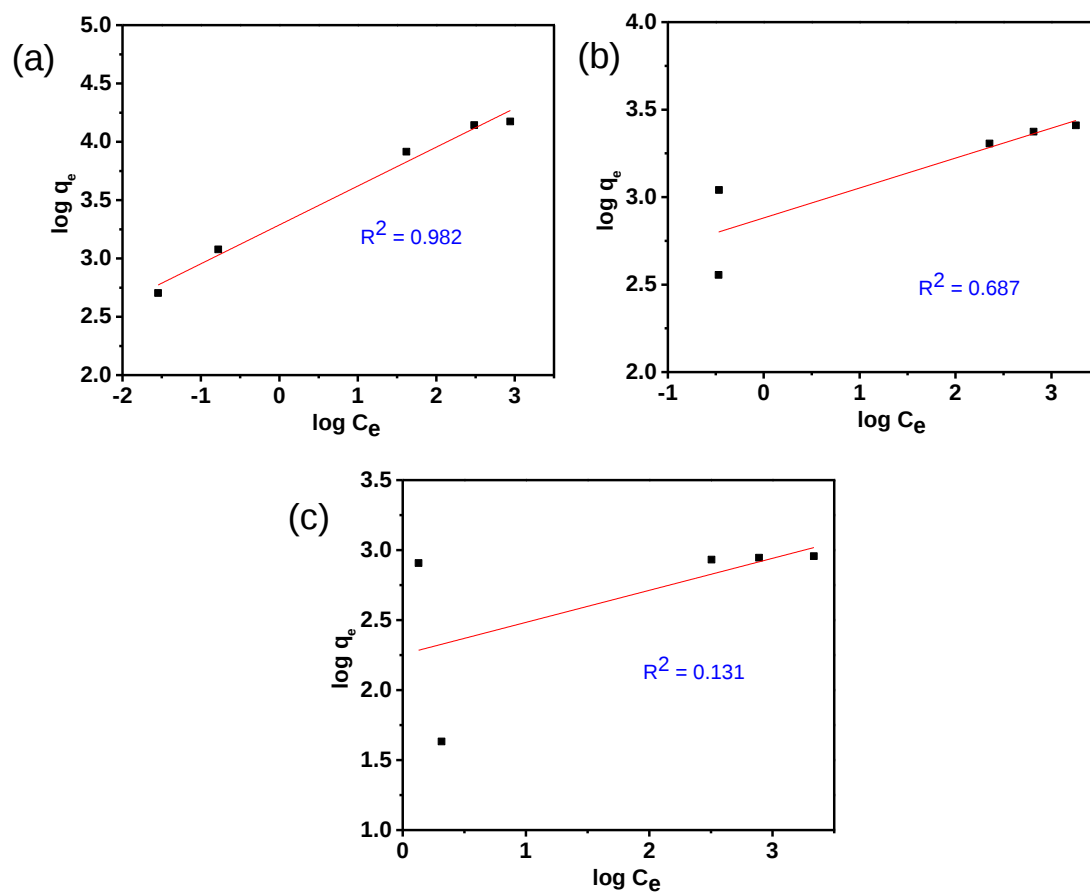


Fig. S18: Freundlich isotherms for MB adsorption on the porous carbon nanospheres with different adsorbate concentration : (a) CF108, (b) CF127 and (c) CL64.

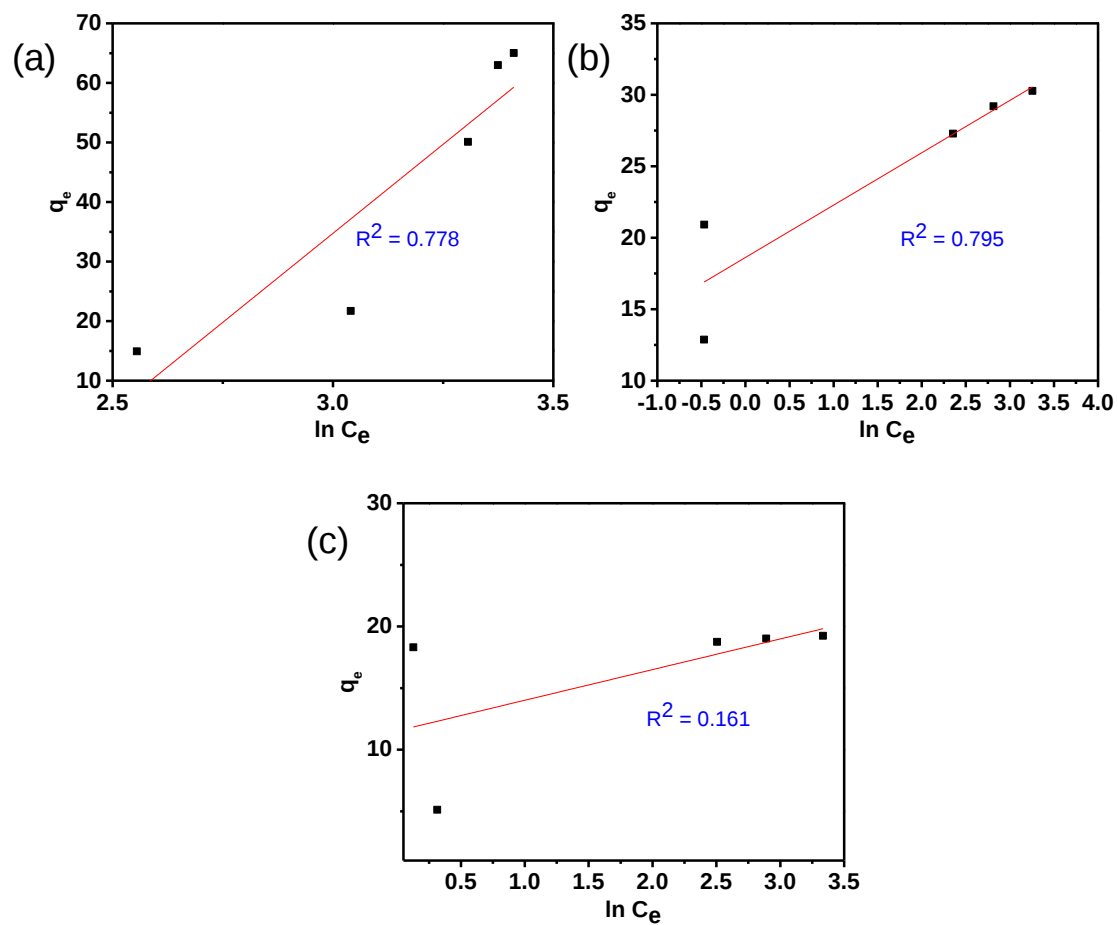


Fig. S19: Temkin isotherms for MB adsorption on the porous carbon nanospheres with different adsorbate concentration : (a) CF108, (b) CF127 and (c) CL64.

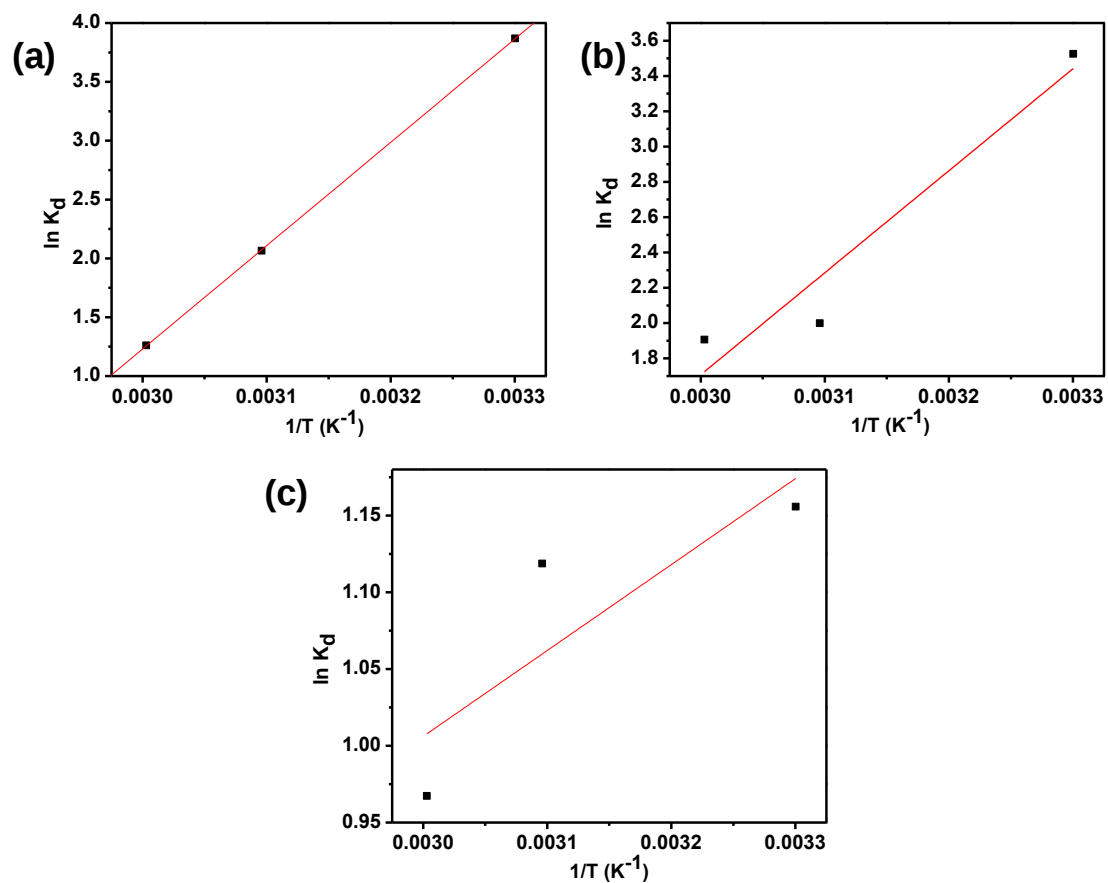


Fig. S20: Plot of $\ln K_d$ versus $1/T$ for phenol adsorption on the porous carbon nanospheres: (a) CF108, (b) CF127 and (c) CL64.

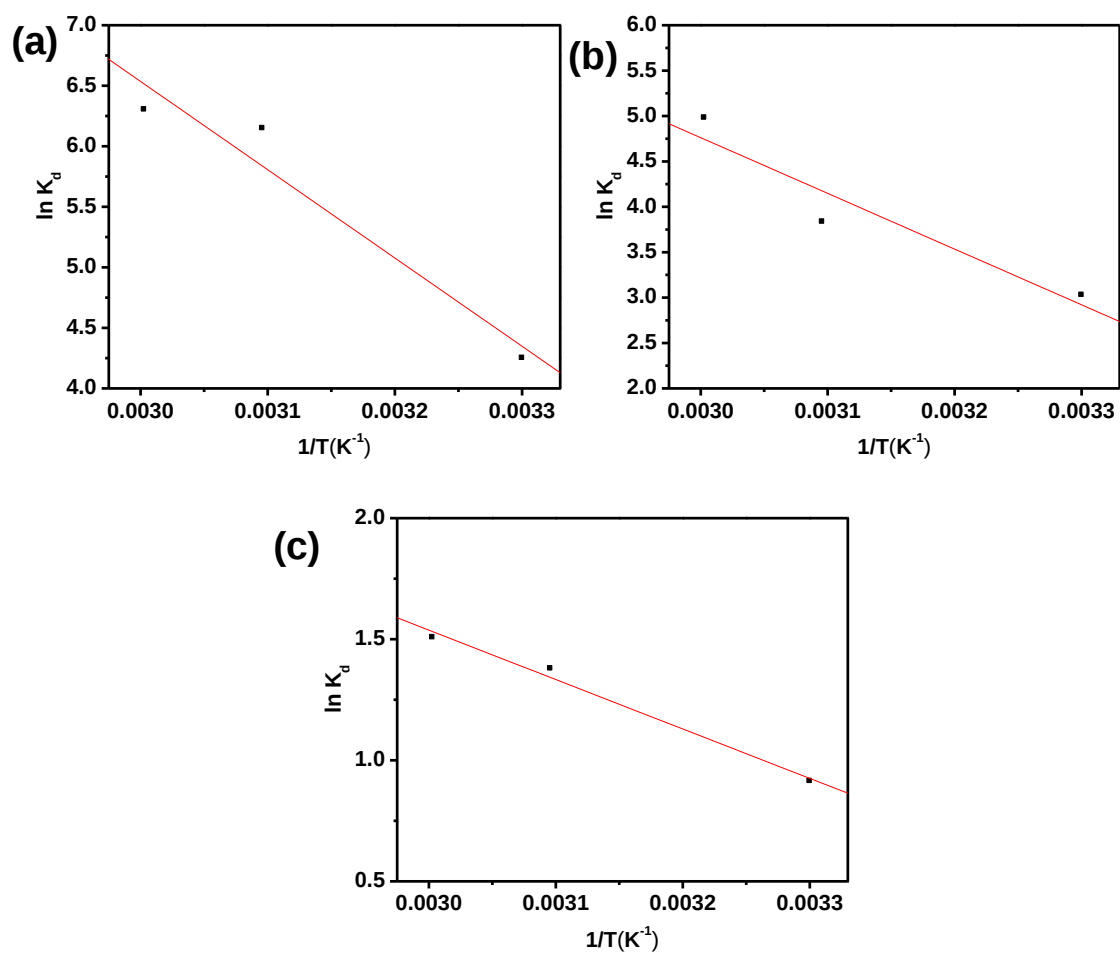


Fig. S21: Plot of $\ln K_d$ versus $1/T$ for MB adsorption on the porous carbon nanospheres: (a) CF108, (b) CF127 and (c) CL64.

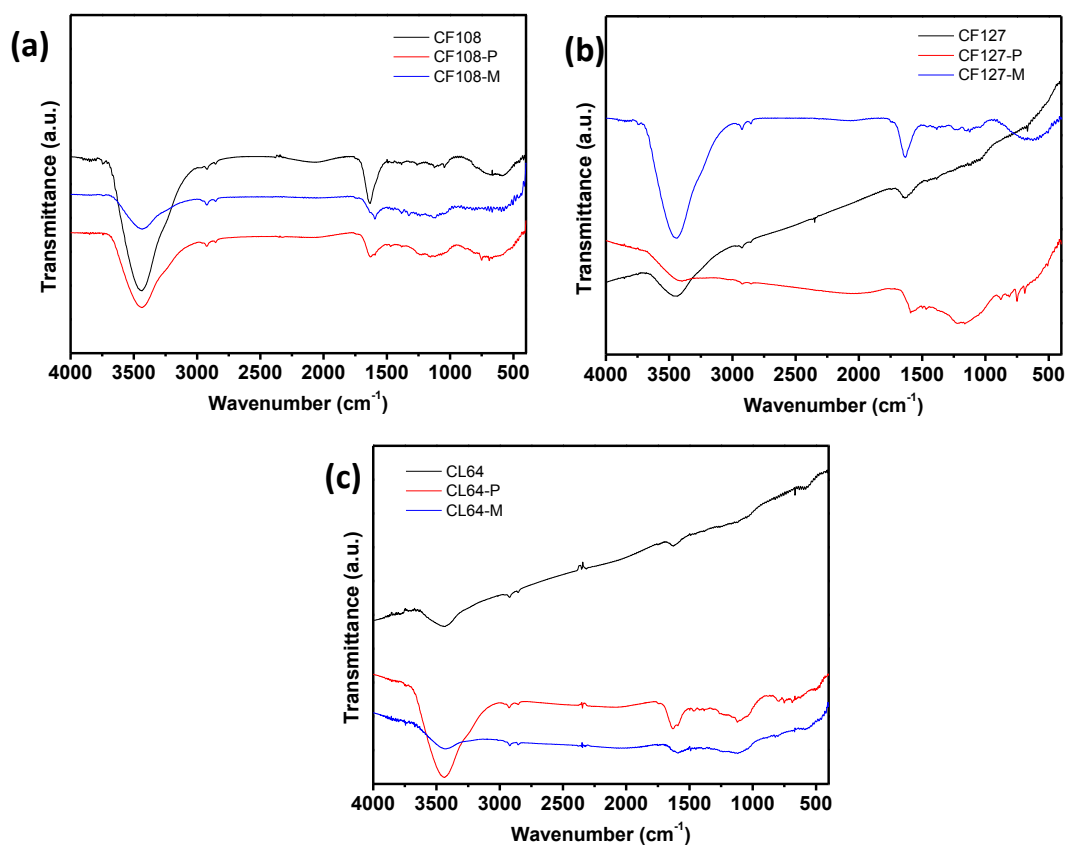


Fig. S22: FTIR spectra of porous carbon nanospheres before adsorption ((a) CF108, (b) CF127 and (c) CL64), after phenol adsorption ((a) CF108-P, (b) CF127-P and (c) CL64-P) and MB adsorption ((a) CF108-M, (b) CF127-M and (c) CL64-M).

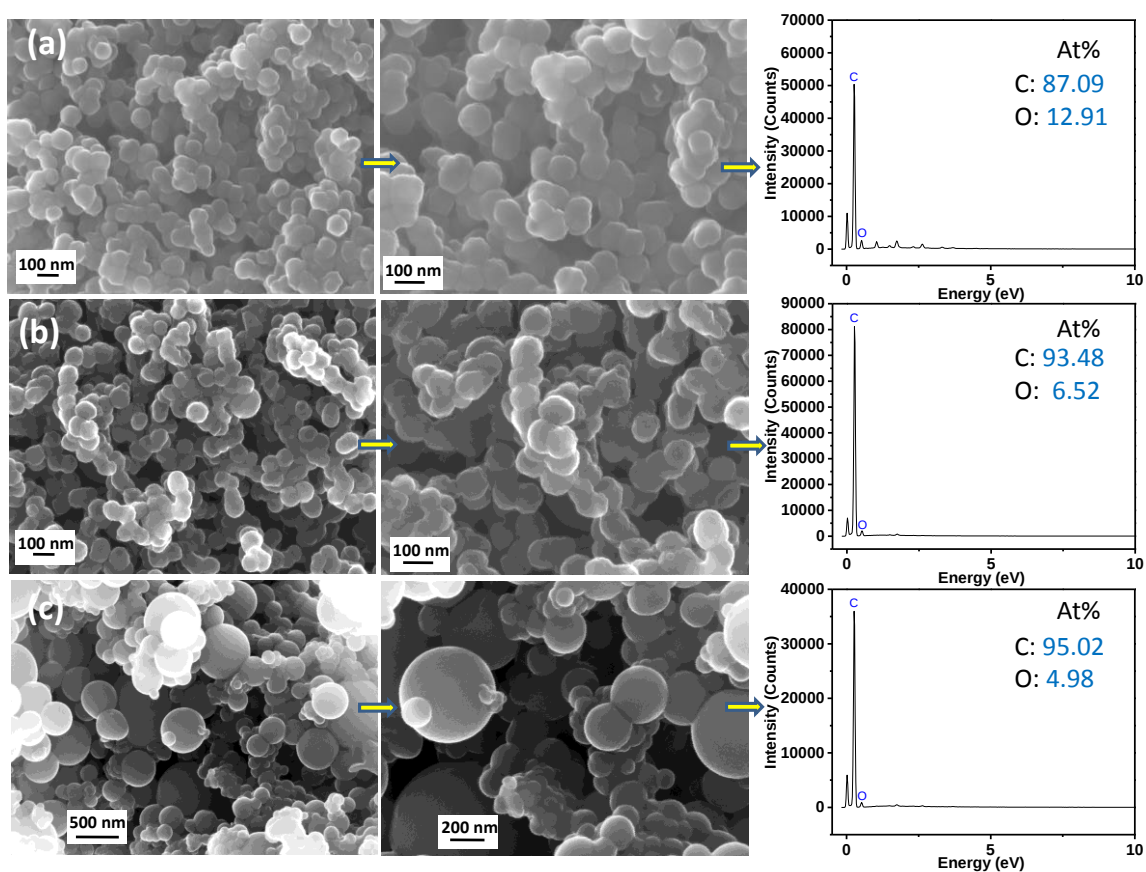


Fig. S23: FESEM images (low and high magnification) and EDX analysis of the samples (a) CF108-P, (b) CF127-P and (c) CL64-P after adsorption of phenol.

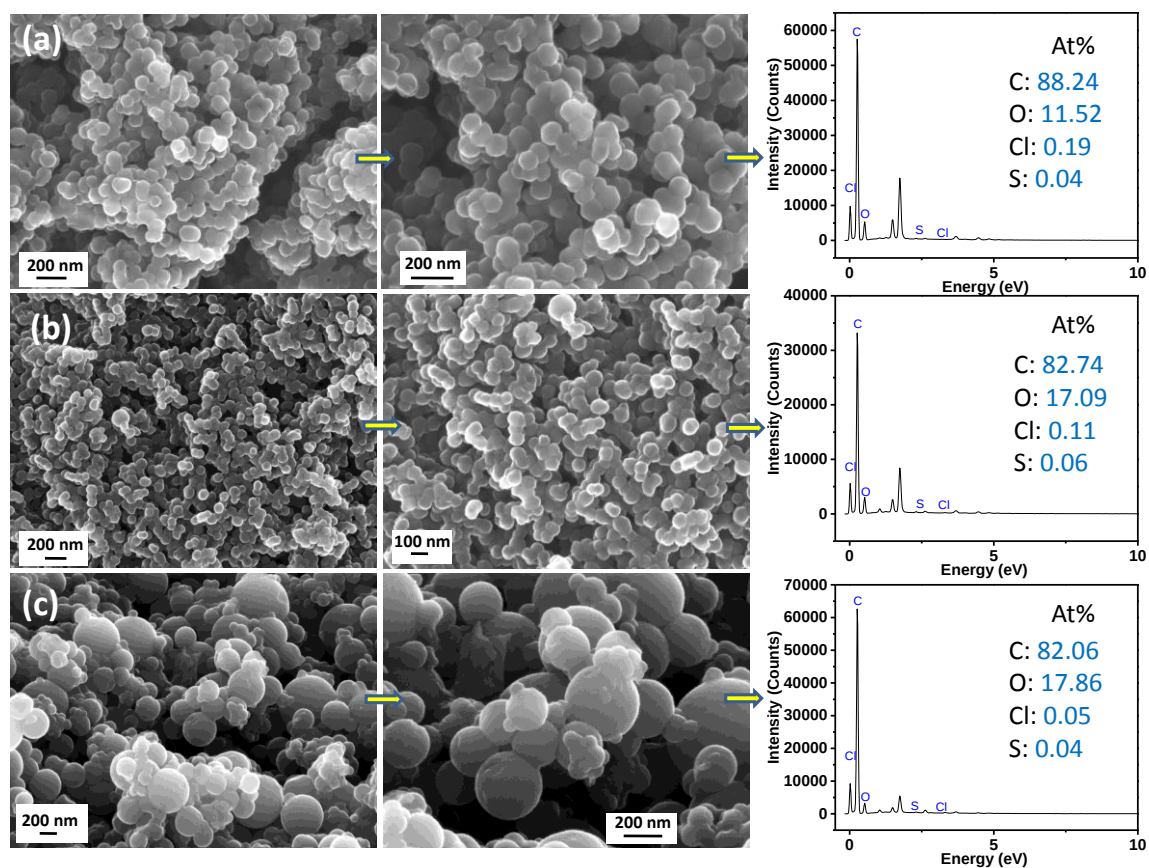


Fig. S24: FESEM images (low and high magnification) and EDX analysis of the samples (a) CF108-M, (b) CF127-M and (c) CL64-M after adsorption of MB.

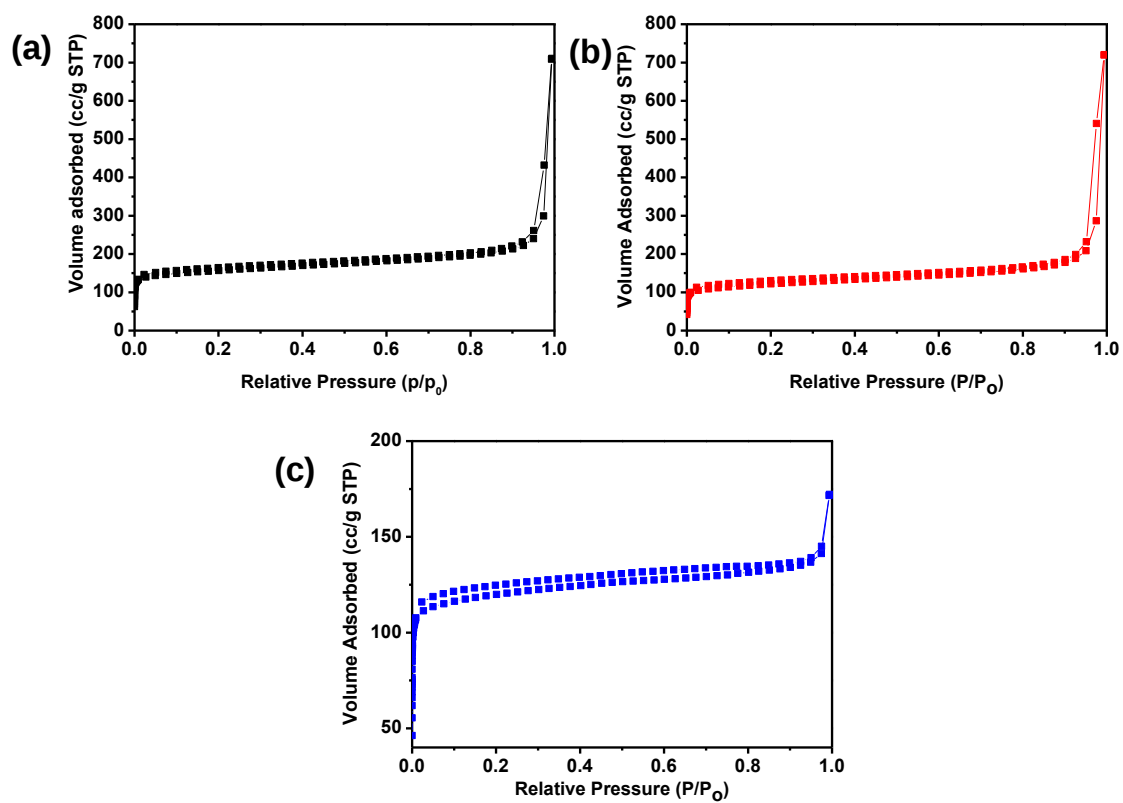


Fig. S25: N₂ adsorption-desorption isotherms of porous carbon nanospheres after phenol adsorption: (a) CF108-P, (b) CF127-P and (c) CL64-P.

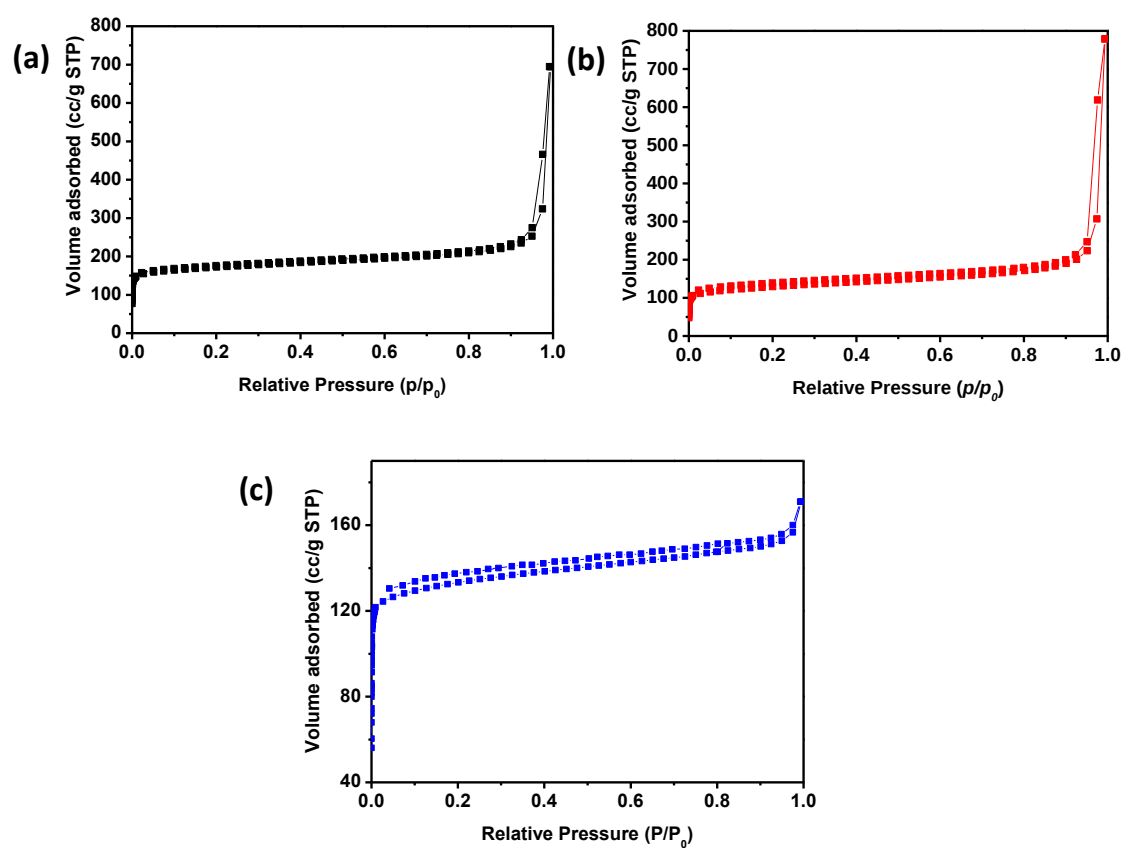


Fig. S26: N₂ adsorption-desorption isotherms of porous carbon nanospheres after MB adsorption: (a) CF108-M, (b) CF127-M and (c) CL64-M.

Table S1: Parameters of kinetic models for the adsorption of phenol and MB

Sample ID	Kinetic Model	Adsorption of phenol		
		K_1 (min ⁻¹)	R ²	
CF108	Pseudo-first order	0.454	0.984	
CF127		0.101	0.890	
CL64		0.180	0.900	
		Adsorption of MB		
CF108		0.731	0.888	
CF127		0.360	0.923	
CL64		0.100	0.963	
		Pseudo-second order	Adsorption of phenol	
	q _e (mg/g)		K_2 (gmg ⁻¹ min ⁻¹)	R ²
CF108	9.4		0.905	0.999
CF127	9.0		0.114	0.997
CL64	7.9		0.073	0.992
	Adsorption of MB			
CF108	3.2		19.53	0.998
CF127	3.196		8.158	0.999
CL64	2.768		1.517	0.994

Table S2: Parameters obtained from Langmuir, Freundlich and Temkin using different adsorbate concentration

Sample ID	Adsorption of phenol				Adsorption of MB			
	Langmuir				Langmuir			
	q _m (mg/g)	K _L (L/mg)	R ²		q _m (mg/g)	K _L (L/mg)	R ²	
CF108	99	1.09	0.986		68.49	0.88	0.996	
CF127	94.25	0.43	0.964		30.86	1.27	0.998	
CL64	75	0.12	0.676		20.83	0.51	0.977	
	Freundlich				Freundlich			
	n _F	K _F	R ²		n _F	K _F (mg/g)	R ²	
CF108	6.45	58.32	0.569		3.00	26.77	0.982	
CF127	3.72	38.86	0.499		5.84	17.83	0.687	
CL64	1.92	13.46	0.306		4.38	9.53	0.131	
	Temkin				Temkin			
	A _T (L/mg)	b _T	B	R ²	A _T (L/mg)	b _T	B	R ²
CF108	558.84	253.9	9.92	0.500	0.089	42.1	59.79	0.778
CF127	10.42	155.8	16.16	0.470	161.92	688.1	3.66	0.795
CL64	0.843	102.3	24.61	0.303	102.53	1012.5	2.48	0.161

Table S3: Surface area of porous carbon nanospheres after phenol and MB adsorption

Sample	Surface area (m^2g^{-1})		
	S_{BET}	$S_{\text{micropore}}$	S_{mesopore}
CF108-P	568.6	308.9	259.7
CF127-P	439.0	159.9	279.1
CL64-P	438	375.5	62.45
CF108-M	602.4	347.5	254.9
CF127-M	467.2	162.6	304.6
CL64-M	486.8	423.4	63.4