

Supporting Information

Composites of Layered $M(\text{HPO}_4)_2$ ($M = \text{Zr}$, Sn , and Ti) with Reduced Graphene Oxide as Anode Materials for Lithium Ion Batteries

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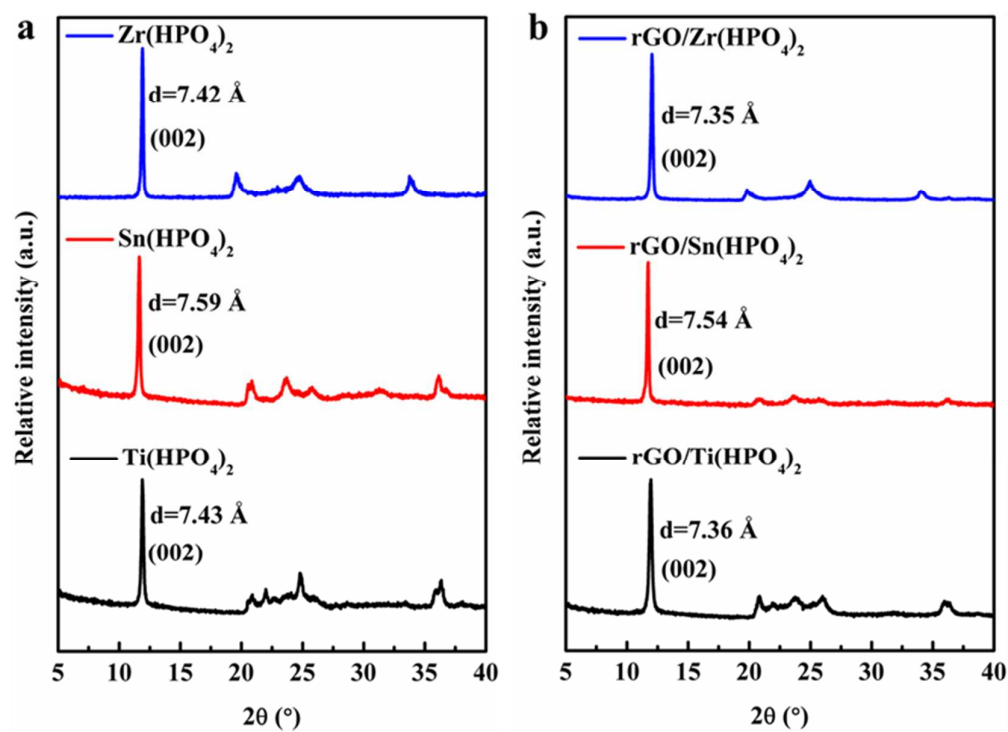


Figure S1. XRD patterns of (a) $M(\text{HPO}_4)_2$, and (b) $\text{rGO}/M(\text{HPO}_4)_2$ with rGO -to- $M(\text{HPO}_4)_2$ ratio of 5:100 (in weight). The interlayer d-spacings of the materials are labeled in the Figure.

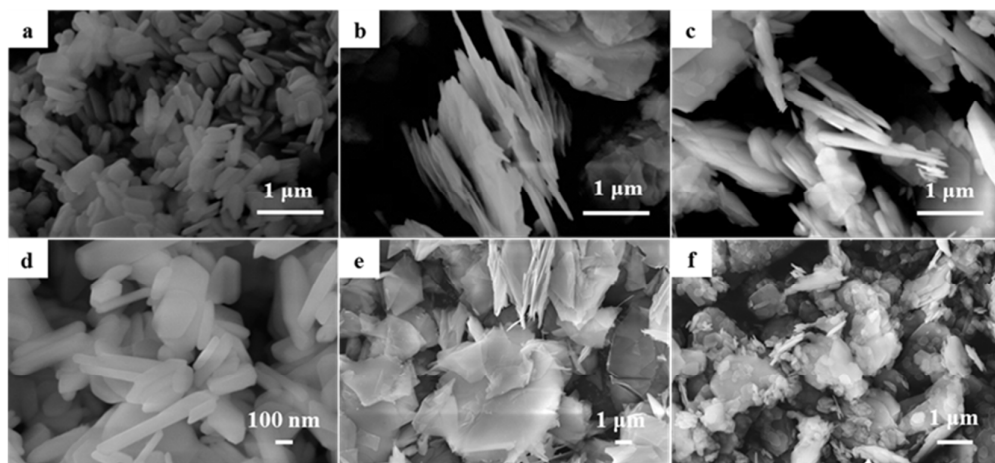


Figure S2. FE-SEM images of metal phosphates hydrates before and after dehydration, (a) $\text{Zr}(\text{HPO}_4)_2 \cdot \text{H}_2\text{O}$, (b) $\text{Sn}(\text{HPO}_4)_2 \cdot \text{H}_2\text{O}$, (c) $\text{Ti}(\text{HPO}_4)_2 \cdot \text{H}_2\text{O}$, (d) $\text{Zr}(\text{HPO}_4)_2$, (e) $\text{Sn}(\text{HPO}_4)_2$, and (f) $\text{Ti}(\text{HPO}_4)_2$, revealing that all three metal phosphates have similar thin flake morphology with the thickness ranged in nanometers and lateral dimensions over a few micrometers.

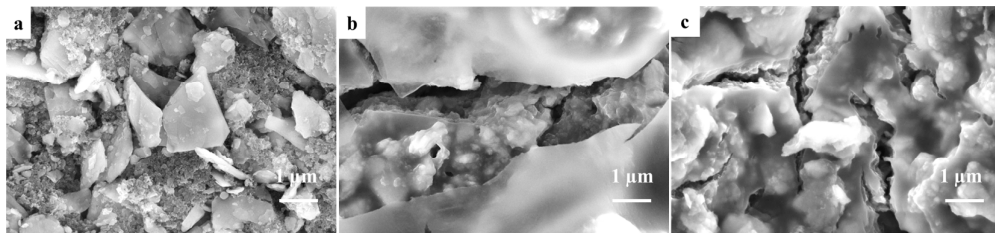


Figure S3. FE-SEM images of the electrodes using rGO/Sn(HPO₄)₂ composites with rGO-to-Sn(HPO₄)₂·H₂O ratio of 5:100 (in weight) at different charge-discharge states. (a) without charge-discharge process, (b) after 1000 charge-discharge cycles at 1 A g⁻¹ and (c) after 1000 charge-discharge at 5 A g⁻¹.

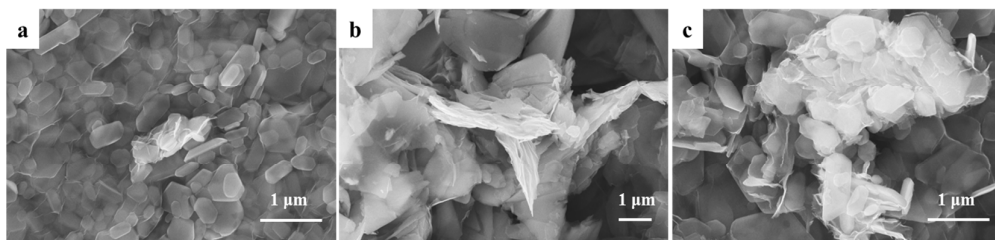


Figure S4. FE-SEM images of (a) rGO/Zr(HPO₄)₂ composite with rGO-to-Zr(HPO₄)₂ ratio of 20:100 (in weight), (b) rGO/Sn(HPO₄)₂ composite with rGO-to-Sn(HPO₄)₂ ratio of 10:100 (in weight), and (c) rGO/Ti(HPO₄)₂ composite with rGO-to-Ti(HPO₄)₂ ratio of 20:100 (in weight).

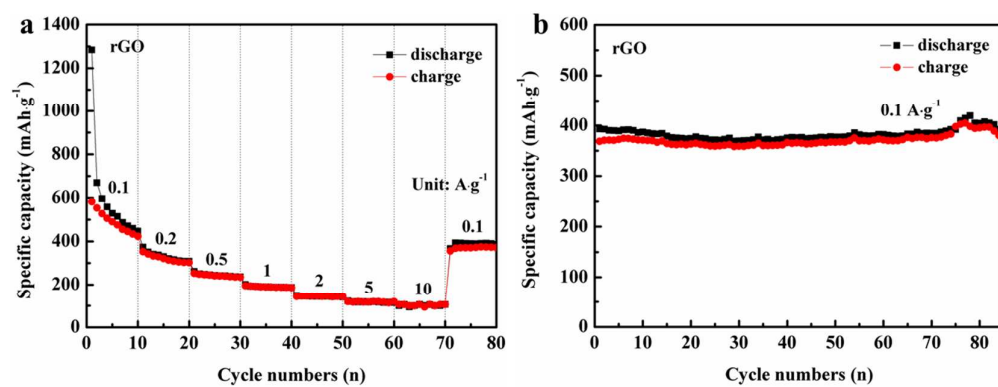


Figure S5. (a) Galvanostatic charge-discharge curves of bare rGO as anode material for LIBs, measured at current densities from 0.1 to 10 A g⁻¹; (b) Cycling performance of rGO anode at 0.1 A g⁻¹.