# Supporting Information

# Palladium-Catalyzed Regioselective C-H Alkenylation of Arylacetamides via Distal Weakly Coordinating Primary Amides as Directing Groups

Yogesh Jaiswal, Yogesh Kumar and Amit Kumar\*

Department of Chemistry, Indian Institute of Technology Patna, Bihta 801106, Bihar, India

## **Table of Contents**

1. Optimization of reaction conditions	S2-S4
S1. Optimization of solvents	
S2. Optimization of oxidants	
S3. Optimization of amount of oxidant	S3
S4. Optimization of time and temperature	
S5. Optimization of amount of ethyl acrylate	S4
S6. Optimization of amount of catalyst	S4
2. Intermolecular competition experiment	S4-S5
3. Synthesis of phenylacetamide- <i>d</i> <sub>5</sub>	S5-S6
4. Kinetic isotope effect experiment	S7
5. References	S7
6. NMR spectra	

#### 1. Optimization of reaction conditions.

## Table S1. Optimization by varying solvents<sup>a</sup>



<sup>*a*</sup>Reaction conditions: **1a** (0.2 mmol), **2a** (0.6 mmol), Pd(OAc)<sub>2</sub> (10 mol%), Ag<sub>2</sub>CO<sub>3</sub> (2.0 equiv), solvent (2.0 mL), at 100 °C for 36 h. <sup>*b*</sup>Isolated yield of **3a** through column chromatography.

## Table S2. Optimization by varying oxidants<sup>a</sup>



5.	CF <sub>3</sub> CO <sub>2</sub> Ag	53
6.	AgNO <sub>3</sub>	40
7.	$K_2S_2O_8$	8
8.	BQ/O <sub>2</sub>	54
9.	Cu(OAc) <sub>2</sub> /O <sub>2</sub>	51
10.		33
11.	BQ	42

<sup>*a*</sup>Reaction conditions: **1a** (0.2 mmol), **2a** (0.6 mmol), Pd(OAc)<sub>2</sub> (10 mol%), Oxidant (2.0 equiv), TFA (2.0 mL), at 100 °C for 36 h. <sup>*b*</sup>Isolated yield of **3a** through column chromatography.

#### Table S3. Optimization by varying amount of oxidant<sup>a</sup>



<sup>*a*</sup>Reaction conditions: **1a** (0.2 mmol), **2a** (0.6 mmol), Pd(OAc)<sub>2</sub> (10 mol%), BQ (X equiv.), TFA (2.0 mL), at 100 °C for 36 h. <sup>*b*</sup>Isolated yield of **3a** through column chromatography.

### Table S4.Optimization by varying time and temperature<sup>a</sup>

	NH <sub>2</sub>	+ U	Pd(OAc) <sub>2</sub> (10 mol%) BQ/O <sub>2</sub>	NH <sub>2</sub>
ļĻ į	Ö	CO <sub>2</sub> Et	TFA	
	1a	2a		3a CO <sub>2</sub> Et
	S.No.	Time (h)	Temperature (°C)	Yield <sup>b</sup> of 3a (%)
	1	15	100	54
	2	24	100	68
	3	36	100	72
	4	36	120	71
	5	36	80	66

<sup>*a*</sup>Reaction conditions: **1a** (0.2 mmol), **2a** (0.6 mmol),  $Pd(OAc)_2$  (10 mol%),  $BQ/O_2(1.0 \text{ equiv})$ , solvent (2.0 mL). <sup>*b*</sup>Isolated yield of **3a** through column chromatography.

#### Table S5. Optimization by varying amount of ethyl acrylate<sup>a</sup>



<sup>*a*</sup>Reaction conditions: **1a** (0.2 mmol), **2b** (X mmol),  $Pd(OAc)_2$  (10 mol%), BQ (1.0 equiv), solvent (2.0 mL). <sup>*b*</sup>Isolated yield of **3a** through column chromatography.

#### Table S6. Optimization of amount of catalyst<sup>a</sup>



S. No.	Pd(OAc) <sub>2</sub> (mol%)	Yield <sup>b</sup> of 3a (%)
1	2	45
2	5	68
3	10	72

<sup>*a*</sup>Reaction conditions: **1a** (0.2 mmol), **2a** (0.6 mmol), Pd(OAc)<sub>2</sub> (Xmol%), BQ(1.0 equiv), solvent (2.0 mL). <sup>*b*</sup>Isolated yield of **3a** through column chromatography.

#### 2. Intermolecular competition experiment between 1d and 1e:



#### Intermolecular competition experiment between 2a and 2g:



#### 3. Synthesis of Phenylacetamide-d<sub>5</sub>



**Procedure for synthesis of Benzyl bromide**- $d_7$ : By following the reported literature procedure,<sup>1</sup> to a 50 mL round bottom flask equipped with magnetic stir bar, were added toluene- $d_8(1.0 \text{ mL}, 10 \text{ mmol})$ , NBS (12 mmol, 2.2 g), CCl<sub>4</sub> (30 mL) followed by catalytic amount of tert. butylperoxybenzoate (30 mol%, 580 µL). Reaction was monitored by TLC. After 3h solution was cooled to room temperature. Reaction mixture was filtered to remove the precipitated succinimide. The filtrate was concentrated in vacuumand crude product was purified through column chromatography to give colourless oil.

**Procedure for synthesis of phenyl acetonitrile**- $d_3$ : By following the reported literature procedure,<sup>2</sup> to a 50 mL round bottom flask equipped with magnetic stir bar, were added benzyl bromide- $d_7$  (6 mmol, 1.06g), TMSCN (7.2 mmol, 0.98 mL), K<sub>2</sub>CO<sub>3</sub> (7.2 mmol, 995 mg) and acetonitrile (20 mL). The solution was stir under reflux for 12 h. After cooling to room temperature, solvent was removed and reaction mixture was extracted with ethyl acetate (30 mL x 3). The organic layer was washed with water and brine. The solution was concentrated

undervacuum and crude product was purified by column chromatography to give colorless liquid.

**Procedure for synthesis of phenyl acetamide**- $d_5$ : By following the reported literature procedure,<sup>3</sup> to a 25 mL round bottom flask equipped with magnetic stir bar, were added phenyl acetonitrile- $d_5$  (1.6 mmol, 195 mg), tetra butyl ammonium hydroxide (4.0 ml, 25% solution), ethanol (10 mL). The solution was stir under reflux for 12 h. After cooling to room temperature, solvent was evaporated in vacuum. The reaction mixture was extracted with ethyl acetate (30 mL x 3). The organic layer was washed with brine and dried over sodium sulphate. After concentrated in vacuum crude product was purified by column chromatography to give white solid (110 mg, 57%).





## 4. Intermolecular competition experiment between 1a and 1a-d<sub>5</sub> to find KIE:

### 5. References

- (1) Z, Hong. J. Label. Compd. Radiopharm. 2008, 51, 293.
- (2) Yabe, O.; Mizufune, H.; Ikemoto, T.Synlett. 2009, 8, 1291.
- (3) Veisi, H.; Maleki, B.; Hamelian, M.; Ashrafi, S. S. RSC Adv. 2015, 5, 6365.

6. <sup>1</sup>H and <sup>13</sup>C{<sup>1</sup>H} Spectra of 3a



 $^1H$  and  $^{13}C\{^1H\}$  Spectra of  $\bf 3b$ 

190

180 170

160 150

140

130 120





90

80 70

100 f1 (ppm)

110

50

40 30

20

10

60

-0 --500

Ó

 $^1H$  and  $^{13}C\{^1H\}$  Spectra of 3c



 $^1H$  and  $^{13}C\{^1H\}$  Spectra of 3d



 $^1H$  and  $^{13}C\{^1H\}$  Spectra of 3e



<sup>1</sup>H and <sup>13</sup>C $\{^{1}H\}$  Spectra of **3f** 



 $^1H$  and  $^{13}C\{^1H\}$  Spectra of 3g



S14

 $^1H$  and  $^{13}C\{^1H\}$  Spectra of 3h



 $^1H$  and  $^{13}C\{^1H\}$  Spectra of 3i



 $^1H$  and  $^{13}C\{^1H\}$  Spectra of 4a



 $^1H$  and  $^{13}C\{^1H\}$  Spectra of 4b



 $^1H$  and  $^{13}C\{^1H\}$  Spectra of 4c



 $^1H$  and  $^{13}C\{^1H\}$  Spectra of 4d



 $^1H$  and  $^{13}C\{^1H\}$  Spectra of 4e



 $^1H$  and  $^{13}C\{^1H\}$  Spectra of 4f



 $^1H$  and  $^{13}C\{^1H\}$  Spectra of 4g



<sup>1</sup>H and <sup>13</sup>C $\{^{1}H\}$  Spectra of **4h** 



 $^1H$  and  $^{13}C\{^1H\}$  Spectra of 4i



 $^1H$  and  $^{13}C\{^1H\}$  Spectra of 4j



 $^1H$  and  $^{13}C\{^1H\}$  Spectra of 4k



 $^1H$  and  $^{13}C\{^1H\}$  Spectra of 41



 $^1H$  and  $^{13}C\{^1H\}$  Spectra of  $\bf 6a$ 



<sup>1</sup>H and <sup>13</sup>C $\{^{1}H\}$  Spectra of **6b** 



<sup>1</sup>H and <sup>13</sup>C $\{^{1}H\}$  Spectra of **6c** 



 $^1H$  and  $^{13}C\{^1H\}$  Spectra of 6d



 $^1H$  and  $^{13}C\{^1H\}$  Spectra of 7a



 $^1H$  and  $^{13}C\{^1H\}$  Spectra of 7b



 $^1H$  and  $^{13}C\{^1H\}$  Spectra of 7c



 $^1H$  and  $^{13}C\{^1H\}$  Spectra of  $\boldsymbol{9}$ 

