

Supporting Information for:

Density-Functional Tight-Binding Molecular Dynamics Simulations of
Excess Proton Diffusion in Ice I_h, Ice I_c, Ice III, and Melted Ice VI
Phases

Aditya Wibawa Sakti,^a Yoshifumi Nishimura,^b Chien-Pin Chou,^b and Hiromi Nakai^{a,b,c,d,*}

^aDepartment of Chemistry and Biochemistry, School of Advanced Science and Engineering,
Waseda University, Tokyo 169-8555, Japan

^bResearch Institute for Science and Engineering, Waseda University, Tokyo 169-8555, Japan

^cCREST, Japan Science and Technology Agency, Tokyo 102-0075, Japan

^dESICB, Kyoto University, Kyoto 615-8520, Japan

*E-mail: nakai@waseda.jp. Tel: +81-3-5286-3452.

Details of Pre-Equilibrated Ice Structures

Table S1. Unit cell sizes and densities of ice I_h obtained with TIP4P/ice^{S1} force field^a

T [K]	Unit cell size [Å ³]	Density [g/cm ³]
50	22.530×23.334×21.875	0.936
70	22.586×23.393×21.930	0.929
120	22.627×23.435×21.970	0.924
230	22.739×23.551×22.078	0.910
250	22.752×23.565×22.091	0.909
270	22.778×23.592×22.117	0.906

^aNumber of water molecules is 360 and target pressure is 0.101325 MPa.

Table S2. Unit cell sizes and densities of ice I_c obtained with TIP4P/ice^{S1} force field^a

T [K]	Unit cell size [Å ³]	Density [g/cm ³]
50	20.107×18.526×18.550	0.935
70	20.117×18.535×18.559	0.933
120	20.138×18.554×18.578	0.930
230	20.222×18.632×18.656	0.919
250	20.241×18.649×18.673	0.916
270	20.249×18.667×18.681	0.914

^aNumber of water molecules is 216 and target pressure is 0.101325 MPa.

Table S3. Unit cell sizes and densities of ice III obtained with TIP4P/2005^{S2} force field^a

P [MPa]	T [K]	Unit cell size [Å³]	Density [g/cm³]
280	230	19.673×20.082×21.024	1.166
	250	19.685×20.095×21.037	1.164
	270	19.710×20.119×21.063	1.160
290	230	19.446×19.851×20.782	1.207
	250	19.489×19.895×20.828	1.200
	270	19.532×19.938×20.873	1.192
300	230	19.277×19.677×20.600	1.240
	250	19.323×19.725×20.650	1.231
	270	19.393×19.797×20.725	1.217

^aNumber of water molecules is 324.

Table S4. Unit cell sizes and densities of melted ice VI obtained with TIP4P/2005^{S2} force field^a

T [K]	Unit cell size [Å³]	Density [g/cm³]
230	18.141×19.216×16.878	1.372
250	18.155×19.231×16.891	1.368
270	18.170×19.247×16.904	1.365

^aNumber of water molecules is 270 and target pressure is 1100 MPa.

Radial Distribution Functions of Ice III

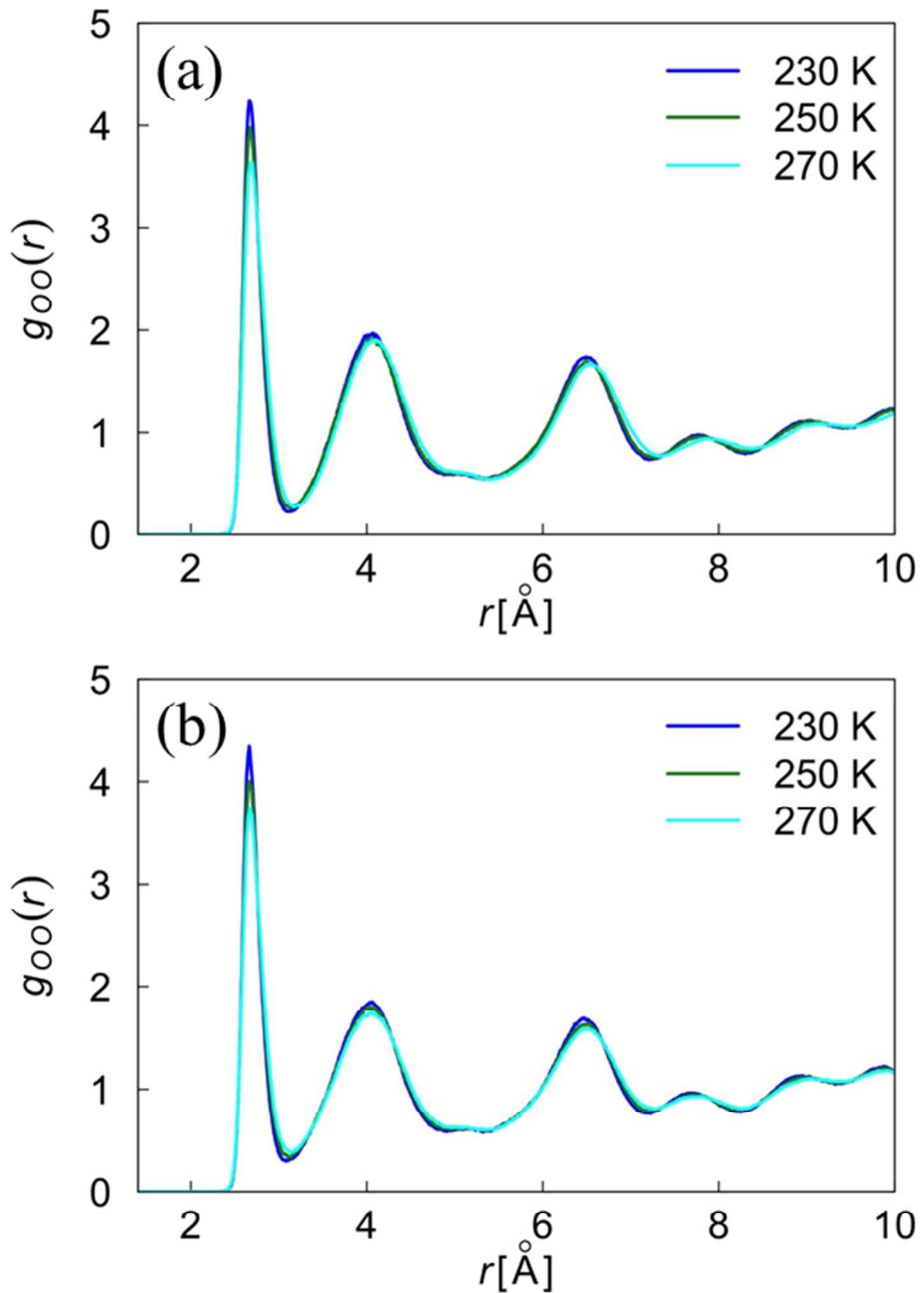


Figure S1. Temperature dependence of the O–O pair RDF for ice III pre-equilibrated at (a) 290 and (b) 300 MPa.

Statistics of Bond Order Parameter Calculations for Ice

Table S5. Mean of averaged form of cubic ($\langle q_4(i) \rangle$) and hexagonal ($\langle q_6(i) \rangle$) bond order parameters and its standard deviation for ice I_h

T [K]	Mean		Standard deviation	
	$\langle q_4(i) \rangle$	$\langle q_6(i) \rangle$	$\langle q_4(i) \rangle$	$\langle q_6(i) \rangle$
50	0.3650	0.5428	0.0143	0.0150
70	0.3641	0.5379	0.0148	0.0164
120	0.3621	0.5252	0.0152	0.0166
230	0.3528	0.4822	0.0219	0.0282
250	0.3526	0.4821	0.0219	0.0282
270	0.3479	0.4497	0.0348	0.0400

Table S6. Mean of averaged form of cubic ($\langle q_4(i) \rangle$) and hexagonal ($\langle q_6(i) \rangle$) bond order parameters and its standard deviation for ice I_c

T [K]	Mean		Standard deviation	
	$\langle q_4(i) \rangle$	$\langle q_6(i) \rangle$	$\langle q_4(i) \rangle$	$\langle q_6(i) \rangle$
50	0.4914	0.5921	0.0325	0.0346
70	0.4942	0.5868	0.0072	0.0153
120	0.4875	0.5718	0.0234	0.0361
230	0.4500	0.4876	0.0402	0.0533
250	0.4404	0.4765	0.0402	0.0601
270	0.4021	0.3962	0.0587	0.0679

Table S7. Mean of averaged form of cubic ($\langle q_4(i) \rangle$) and hexagonal ($\langle q_6(i) \rangle$) bond order parameters and its standard deviation for ice III

P [MPa]	T [K]	Mean		Standard deviation	
		$\langle q_4(i) \rangle$	$\langle q_6(i) \rangle$	$\langle q_4(i) \rangle$	$\langle q_6(i) \rangle$
280	230	0.3298	0.1926	0.0339	0.0383
	250	0.3217	0.1917	0.0397	0.0369
	270	0.3207	0.1945	0.0385	0.0354
290	230	0.3271	0.2012	0.0353	0.0361
	250	0.3249	0.1933	0.0324	0.0350
	270	0.3311	0.1941	0.0347	0.0365
300	230	0.3070	0.1898	0.0414	0.0375
	250	0.3049	0.1937	0.0412	0.0326
	270	0.2968	0.2024	0.0374	0.0392

Table S8. Mean of averaged form of cubic ($\langle q_4(i) \rangle$) and hexagonal ($\langle q_6(i) \rangle$) bond order parameters and its standard deviation for melted ice VI

T [K]	Mean		Standard deviation	
	$\langle q_4(i) \rangle$	$\langle q_6(i) \rangle$	$\langle q_4(i) \rangle$	$\langle q_6(i) \rangle$
230	0.1386	0.1739	0.0408	0.0337
250	0.1186	0.1717	0.0332	0.0342
270	0.1222	0.1752	0.0356	0.0372

References

- (S1) Abascal, J. L. F.; Sanz, E.; García Fernández, R.; Vega, C. A Potential Model for the Study of Ices and Amorphous Water: TIP4P/Ice. *J. Chem. Phys.* **2005**, *122*, 234511.
- (S2) Abascal, J. L. F.; Vega, C. A General Purpose Model for the Condensed Phases of Water: TIP4P/2005. *J. Chem. Phys.* **2005**, *123*, 234505.